

Cyclopropanation and Carbonyl Olefination Utilizing 2-(Alk-1-yn-1-yl)-2-(trialkylsilyl)-1,3-dithianes via Regioselective Generation of Titanium Alkynylcarbene Complexes

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 $Cp_2Ti[P(OEt)_3]_2$ -promoted reactions of 2-(alk-1-yn-1-yl)-2-(trialkylsilyl)-1,3-dithianes $(RS)_2C(Si)C \equiv CR$ with terminal olefins and carbonyl compounds produced (trialkylsilylethynyl)cyclopropanes and 1-(trialkylsilyl)alk-3-en-1-ynes, respectively. These compounds were suggest to be produced via the formation of intermediary titanium α -(trialkylsilylethynyl)carbene complexes $Cp_2Ti = C(R)C \equiv CSi$ in preference to their regioisomers, α -(trialkylsilyl)alkynylcarbene complexes $Cp_2Ti = C(Si)C \equiv CR$.

Introduction

The unique reactivities of transition metal carbene complexes and their application to organic synthesis including olefin metathesis, cyclopropanation, and carbonyl olefination have attracted considerable attention.¹ We have developed a variety of transformations of organic compounds utilizing a thioacetal-titanocene(II) system, in which titanium carbene complexes are assumed to be formed as active intermediates. Thus alkyl-, alkenyl-, and alkynylcarbene complexes were readily generated by desulfurizative titanation of thioacetals and related organosulfur compounds with the titanocene(II) species Cp₂Ti[P(OEt)₃]₂ (1).² This methodology, however, suffers a serious drawback in that the formation of α-substituted alkylidenetitanocenes is seriously affected by steric hindrance;³ only thioacetals derived from less hindered ketones such as cyclobutanone can be employed for the generation of the alkylidene complexes.⁴

Previously, we have found that cyclopropanes were produced by reaction of terminal olefins with the titanium alkynylcarbene complexes generated from β , γ -acetylenic thioacetals.⁵ Although various types of heteroatom-substituted alkynylcarbene complexes have been pre-

Here we report highly regioselective cyclopropanation and carbonyl olefination utilizing α -silylalkynyl thioacetals **2** which involve unprecedented allylic rearrangement directed by the trialkylsilyl group.⁷

Results and Discussion

Preparation of 2-(Alk-1-yn-1-yl)-2-(trialkylsilyl)- 1,3-dithianes 2. The alkynyl thioacetals employed in this

pared and applied to organic synthesis,¹ only limited work has been done on the chemistry of nonheteroatom-substituted alkynylcarbene complexes.⁶ Therefore we were intrigued with the generation of highly substituted alkynylcarbene complexes from the acetylenic thioacetals bearing an α-substituent. We expected that the less bulky alkynyl group would alleviate the steric crowding of the thioacetals, and hence these organotitanium species could be obtained. Our major interest in the chemistry of highly substituted alkynylcarbene complexes is their regioselective formation, which is of crucial importance for their synthetic application.

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SCHEME 1. Preparation of Alkynyl Thioacetals

OEt
$$^{+}$$
 HS SH $^{-}$ AlCl₃ / THF S $^{-}$ Alcl₃ / The S $^{-}$ Alcl₃ Alcl₃ / The S $^{-}$ Alcl₃ Alcl₃ Alcl₃ Alcl₃ Alc

SCHEME 2. Cyclopropanation with 2-Methyl-2-(5-phenylpent-1-yn-1-yl)-1,3-dithiane (7)

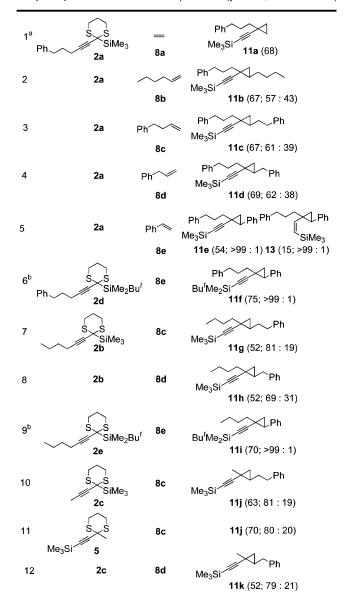
study were prepared as summarized in Scheme 1. Transthioacetalization of β , γ -acetylenic acetals **3** with 1,3-propanedithiol in the presence of AlCl₃ gave the 2-alkynyl-1,3-dithianes **4**. The α -(trialkylsilyl)- β , γ -acetylenic thioacetals **2** were obtained by successive treatment of **4** with buthyllithium and trialkylchlorosilanes. 2-Methyl-2-(trimethylsilylethynyl)-1,3-dithiane (**5**) was prepared by the treatment of 2-(trimethylsilylethynyl)-1,3-dithiane (**6**)⁸ with butyllithium followed by the alkylation with methyl iodide. The alkylation of the dithiane **4a** with methyl iodide also gave 2-methyl-2-(5-phenylpent-1-yn-1-yl)-1,3-dithiane (**7**).

Reaction of Alkynyl Thioacetals with Terminal Olefins in the Presence of the Titanocene(II) Reagent. Initially we examined the titanocene(II)-promoted cyclopropanation of the α -methyl- β , γ -acetylenic thioacetal 7 under ethylene (8a). Treatment of 7 with the titanocene(II) reagent 1 (3 equiv) under ethylene atmosphere afforded a mixture of the alkynylcyclopropane 9 and its regioisomer 10 in 61% yield with a ratio of 79:21 in favor of the less steric crowding cyclopropane (Scheme 2).

When more sterically hindered 2-(5-phenylpent-1-yn-1-yl)-2-(trimethylsilyl)-1,3-dithiane (**2a**) was treated with **1** (2 equiv) under **8a** at 25 °C for 1 h, the trimethylsilylethynyl-substituted cyclopropane **11a** was exclusively produced in 68% yield (Table 1, entry 1). Similar titanocene(II)-promoted reactions of the α -trialkylsilyl- β , γ -acetylenic thioacetals **2** with various terminal olefins **8**

TABLE 1. Regioselective Formation of the (Trialkylsilylethynyl)cyclopropanes 11

entry acetylenic thioacetal 2 olefin 8 product 11 (yield / %; ratio of isomers)



 a Carried out under ethylene. b 3 equiv of 1 and 4 equiv of 8e were used.

SCHEME 3. Silyl Group Directed Regioselective Cyclopropanation with the α -(Trialkylsilyl)- β , γ -acetylenic Thioacetals 2

also gave the (trialkylsilylethynyl)cyclopropanes 11 without formation of their possible regioisomers 12 (Scheme 3). When styrene (8e) was used as an olefin component, partial reduction of the triple bond of the alkynylcyclopropane proceeded to form the alkenylsilane 13 (entry 5). The coupling constant of its vinyl protons (14.5 Hz) suggests that its configuration is Z. This reduction was

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SCHEME 4. Reaction Pathway for the Regioselective Formation of (Trialkylsilylethynyl)cyclopropanes 11

SCHEME 5. Formation of the Dideuterio Alkynylsilanes 16

2a
$$\frac{1 \text{ (3 eq)}}{\text{Me}_3 \text{Si}}$$
 $\frac{\text{D}}{\text{D}}$ $\frac{\text{D}}{\text{Ph}}$ $\frac{\text{D}}{\text{D}}$ $\frac{\text{D}}{\text{SiMe}_3}$ $\frac{\text{D}}{\text{SiMe}_3}$ $\frac{\text{D}}{\text{SiMe}_3}$ $\frac{\text{D}}{\text{SiMe}_3}$ $\frac{\text{D}}{\text{SiMe}_3}$ $\frac{\text{D}}{\text{SiMe}_3}$ $\frac{\text{TiCp}_2}{\text{SiMe}_3}$ $\frac{\text{TiCp}_2}{\text{SiMe}_3}$ $\frac{\text{TiCp}_2}{\text{SiMe}_3}$ $\frac{\text{TiCp}_2}{\text{SiMe}_3}$ $\frac{\text{TiCp}_2}{\text{SiMe}_3}$

suppressed by use of more bulky tert-butyldimethylsilylsubstituted acetylenic thioacetals (entries 6 and 9). Interestingly, the position γ to the thioacetal group of 2 was predominantly cyclopropanated in all the reactions examined. This regioselectivity is in sharp contrast to that observed in the reaction of α-unsubstituted thioacetals, in which cyclopropanation took place at the α-position.⁵ Furthermore, it is noteworthy that both the cyclopropanation of 4-phenylbut-1-ene (8c) with the α -(trimethylsilyl)- β , γ -acetylenic thioacetal **2c** and its regioisomer, γ -(trimethylsilyl)- β , γ -acetylenic thioacetal **5**, regioselectively produced the (trimethysilylethynyl)cyclopropane 11j with the same ratio of stereoisomers (entries 10 and 11). These results indicate that the regioselectivity of the cyclopropanation depends on the relative steric bulkiness of α (R¹)- and γ (R²)-substituents of acetylenic thioacetals $R^2C \equiv CC(SR)_2R^1$.

The regioselectivity of the formation of (trialkylsilylethynyl)cyclopropanes 11 strongly suggests exclusive formation of intermediary α -(trialkylsilylethynyl)carbene complexes 14 (Cp₂Ti=C(R)C=CSi). Accordingly, we assume that the cyclopropanation proceeds through the formation of the titanacyclobutane intermediate 15 formed from 14 and a terminal olefin 8. Subsequent reductive elimination affords the cyclopropane 11 (Scheme 4).

Indeed, the formation of alkynylcarbene complex 14 was supported by the fact that the dideuterated 1- and 2-alkynylsilanes 16a and 16b (>95% D) were produced by successive treatment of the thioacetal 2a with the titanocene(II) reagent 1 and deuterium oxide (Scheme

Reaction of the Alkynylcarbene Complexes with Dimethylphenylsilanes

5). It is rational to assume that **16b** is produced by deuteration of the alkynylcarbene complex 14a with allylic rearrangement. However, the result does not exclude the formation of the α-(trimethylsilyl)alkynylcarbene complex 17.

Reaction of the Alkynylcarbene Complexes with **Dimethylphenylsilane.** A more persuasive evidence for the intermediary trialkylsilylethynly-substituted carbene complex 14 was obtained by using a carbenoid insertiontype reaction of titanium carbene complexes with group 14 metal hydrides¹⁰ as a probe. Desulfurizative titanation of 2a with 1 in the presence of dimethylphenylsilane produced propargylsilane 18a as a major product along with the allenylsilane 19 (Scheme 6). It is apparent that these compounds are formed by the [2+1] and [2+3] carbenoid insertion-type reactions of the (trialkylsilylethynyl)carbene complex 14a (Scheme 7). No formation of the propargylsilane 20 or allenylsilane 21, which would be formed via the α -trialkylsilyl carbene complex 17, was observed, thus indicating exclusive formation of 14a by the desulfurization of **2a** with **1**. The selective generation of the silvlethynyl carbene complex is further supported by the fact that a similar reaction of 2c with dimethylphenylsilane gave only the propargylsilane 18b, which was also produced by the reaction of the isomeric silylethynyl thioacetal **5**. The mechanism for the regioselective formation of **14a** is obscure at present. Although it is reasonable to assume that the desulfurizative titanation of 2 with 1 proceeds with allylic rearrangement to give 14, the isomerization of the initially formed carbene complexes such as 17 to 14 cannot be excluded because the equilibrium between regioisomers of rhenium, 6b,d manganese, 6e and rhodium6h alkynylcarbene complexes, $LnM=C(R)C\equiv CR'$ and $LnM=C(R')C\equiv CR$, was observed.

Titanocene(II)-Promoted Reaction of the α-Substituted Alkynyl Thioacetals 2 with Carbonyl Compounds. On the basis of the regioselective formation of 14 suggested by the above investigation, we next examined the regioselective preparation of enynes 22 by the olefination of carbonyl compounds 23 with the alkynylcarbene complexes 14 (Scheme 8). Contrary to our expectations, the successive treatment of acetylenic thioacetal 2a with the titanocene(II) reagent 1 (3 equiv) and

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SCHEME 7. Reaction Pathway for the Formation of Propargyl- and Allenylsilanes

SCHEME 8. Carbonyl Olefination Utilizing the Acetylenic Thioacetals ${\bf 2}$

TABLE 2. Reaction of the Acetylenic Thioacetals 2 with Dialkyl Ketones 23

ntry ti	hioacetal 2	ketone 23	product 22	yield / % (22 : 24
1	2a	O 23a	Ph , 2	4a 75 (80 : 20)
2	2d	23a	Ph Bu ^l Me ₂ Si 22b	4b 72 (94 : 6)
3	2d Pl	O Ph 23b	Ph Ph Ph , 2	4c 60 (88 : 12)
4	2d	O Ph 23c	Ph , 2 Bu'Me ₂ Si 22d	4d 67 (88 : 12)
5	2e	23b	Bu ^f Me ₂ Si Ph , 2	4e 69 (87 : 13)
6	2f	23b	Bu ^f Me ₂ Si Ph	71
7	2f	23c	Bu ⁴ Me ₂ Si	68 (54 : 46)

3-pentanone (23a) (1.2 equiv) produced a mixture of the two regioisomers of the carbonyl olefination products, the conjugated dienes 22a and 24a (80:20) in 75% yield (Table 2, entry 1). The regioselectivity increased when the more sterically hindered *tert*-butyldimethylsilylsubstituted counterpart 2d was employed (entry 2). Although the reactions of dialkyl ketones and acetophenone with acetylenic thioacetals generally gave mixtures of regioisomers, complete regioselectivity was observed

TABLE 3. Reaction of the Acetylenic Thioacetals 2 with Sterically Less Hindered Ketones 23

entry	thioacetal 2	ketone 23	product 22	yield / %
1	2d	O 23d	Ph Bu'Me ₂ Si 22h	68
2	2d	0 23e	Ph Bu'Me ₂ Si 22i	71 (83 : 17)
3	2d	23f	Ph Bu'Me ₂ Si 22j	72
4	2d Bı		Ph Bu'Me ₂ Si 22k	60 Bu ^t
5	2d P	h—(O 23h	Ph Bu'Me ₂ Si 22I	72 Ph
6	2d	23i	Ph Bu'Me ₂ Si 22m	70 (85 : 15
7	2e	23g	Bu'Me ₂ Si 22n	69 Bu ^t
8	2e	23h	Bu'Me ₂ Si 22o	58 Ph
9	2f	23h	Bu ^t Me ₂ Si	66 Ph

when the less bulky methyl-substituted thioacetal **2f** was employed (entries 6 and 7).

Unlike the reactions of dialkyl ketones, less bulky ketones such as acetone, butan-2-one, and 4-substituted cyclohexanones were exclusively transformed into the corresponding silylacetylenes 22, regardless of the γ -sub-

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TABLE 4. Reaction of the Acetylenic Thioacetals 2 with **Esters and Lactones 23**

entry	thioacetal 2	ester or lactone 23	product 22 ^a	yield / %
1	2d	O Ph OEt	Ph OEt	63
2	2d	23 j	Ph O	70
3	2d	23kO=0 231	Ph Bu ^f Me ₂ Si	71
4	2f	O Ph OMe	Bu ^t Me ₂ Si Ph	63
5	2f	Ph OEt	OEt Bu ^t Me ₂ Si Ph	68

^a Single stereoisomers were obtained.

stituents of the β, γ -acetylenic thioacetals **2** (Table 3). The dienyne 22m was also obtained in good yield by the reaction of 2d with cyclohex-2-en-1-one (23i) (entry 6).

Similarly to the conventional carbonyl olefinations with titanium carbene complexes,¹¹ the present reaction is applicable to the transformation of esters and lactones into enynes 22 (Table 4). In contrast to the reactions of dialkyl ketones, the reaction proceeded regioselectively to produce silylethynyl group-substituted enol ethers as sole products. It is also of special interest that every reaction examined gave a single stereoisomer. Olefination of carboxylic acid derivatives utilizing a thioacetaltitanocene(II) system generally produces Z stereoisomers, in which there is less steric interaction between the alkyl group originating from the acyl group and the alkyl substituent at the β carbon. 12 Therefore, it is reasonable to assume that the configuration of these enol ethers is E. A similar stereoselectivity was also observed in other olefination reactions of esters with use of titanium carbene complexes.¹³

Protodesilylation of the (Trialkylsilylethynyl)cyclopropanes 11 and 1-(Trialkylsilyl)alk-3-en-1ynes 22. The major products of the reactions can be transformed into synthetically useful terminal alkynes (Scheme 9). Treatment of the (trialkylsilyethynyl)cyclopropanes 11 with tetrabutylammonium fluoride (TBAF) gave the ethynylcyclopropanes 25. Similarly the trialkyl-

SCHEME 9. Protodesilylation of the (Trialkylsilylethynyl)cyclopropanes 11 and 1-(Trialkylsilyl)alk-3-en-1-ynes 22

11 TBAF
$$H_{2}O - THF$$

$$R^{1}$$

$$R^{2}$$

$$R^{2}$$

$$R^{2}$$

$$R^{3}$$

$$R^{4}$$

$$R^{3}$$

$$R^{4}$$

$$R^{3}$$

$$R^{4}$$

$$R^{3}$$

$$R^{4}$$

$$R^{3}$$

$$R^{4}$$

$$R^{4}$$

$$R^{4}$$

$$R^{5}$$

$$R^{4}$$

$$R^{5}$$

$$R^{4}$$

$$R^{5}$$

$$R^{5}$$

$$R^{6}$$

$$R^{6}$$

$$R^{6}$$

TABLE 5. Protodesilylaton of the (Trimethylsilylethynyl)cyclopropanes 11 and 1-(tert-Butyldimethylsilyl)alk-3-en-1-ynes 22

entry	silylacetylene 11 or 22	terminal alkyne 25 or 26	yield/% (ratio of stereoisomers)
1	11b	25a	92 (56:44)
2	11c	25b	78 (62:38)
3	11g	25c	93 (83:17)
4	$22\mathbf{b}$	26a	74
5	22d	26b	75 (70:30)
6	22f	26c	76
7	22h	26d	76
8	22i	26e	77 (82:18)
9	22j	26f	80
10	$\mathbf{22k}$	26g	82
11	22o	26h	93

SCHEME 10. Pathways for the Reactions of the Alkynylcarbene Complexes 14

$$\begin{bmatrix} c_{p_2}T_1 & Y & X_{-Y} & T_1Cp_2 & X_{-Y} & R_{3S1} & 14 \end{bmatrix} \xrightarrow{X_{-Y}} \begin{bmatrix} X_{-Y} & X_{-Y} &$$

silyl group-substituted enynes 22 were transformed to the terminal alkynes 26 (Table 5).

Reaction Pathways. The results of the carbenoid insertion-type reactions indicate that only the α -(trialkylsilylethynyl)carbene complex 14 is generated by the desulfurization of the alkynyl thioacetal 2 with the lowvalent titanocene 1. Therefore all the reactions of 2 are explained by the formation of two types of organotitanium intermediates 27 and 28 (Scheme 10). The reaction sequence involving the formation of the propargyltitanium intermediate 27 via the more stable 4-membered transition state **29** and its metathesis-type degradation or reductive elimination affords the major products of cyclopropanation 11, carbenoid insertion-type reaction 18, or carbonyl olefination 22. The minor products of the latter two reactions 19 and 24 are produced by similar degradations of the allenyltitanium intermediate 28,

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which is formed via the less stable 6-membered transition state $\bf 30$. In the case of the reactions of sterically hindered dialkyl ketones or dimethylphenylsilane, the four-membered transition state $\bf 29$ is destabilized by steric repulsion between the α -substituent of the carbene complex and the substrate, hence the reaction partially follows the pathway involving the six-membered transition state $\bf 30$ to produce the minor products. On the contrary, a bulky tert-butyldimethylsilyl group destabilizes $\bf 30$, which would be the reason the carbonyl olefination with tert-butyldimethylsilyl group-substituted acetylenic thioacetals is more regioselective than the reaction of the trimethylsilyl group-substituted counterparts.

Conclusion

We have demonstrated that the highly regioselective reactions of alkynyl thioacetals with terminal olefins and carbonyl compounds are achieved by the introduction of the bulky trialkylsilyl group. The products of these reactions together with the results of the reaction of the thioacetals with dimethylphenylsilane suggested that the reaction preceded via the regioselective formation of silylethynyl group-substituted titanium carbene complexes. It should be noted that the reactions described here are convenient synthetic methods for the preparation of functionalized terminal alkynes.

Experimental Section

Preparation of 2-(5-Phenylpent-1-yn-1-yl)-1,3-dithiane (4a). A THF (50 mL) solution of AlCl₃ (5.87 g, 44 mmol) was cooled to 0 °C. To the solution was added 1,3-propanedithiol (2.0 mL, 20 mmol) under argon and then a THF (10 mL) solution of 1,1-diethoxy-6-phenylhex-2-yne (3a) (4.93 g, 20 mmol). After being stirred for 2 days, the reaction was quenched with H₂O and organic materials were extracted with ether. The organic layer was washed with 1 M NaOH, H₂O, and brine, and dried over Na2SO4. The solvent was removed under reduced pressure and the residue was chromatographed on silica gel (hexane/AcOEt = 98/2) to give 4a (3.74 g, 71%). **4a**: IR (neat) 3060, 3024, 2901, 2231, 1602, 1496, 1454, 1275, 1243, 1223, 1030, 910, 744, 700 cm $^{-1}$; H NMR δ 1.87 (tt, J =7.2, 7.2 Hz, 2H), 2.01-2.08 (m, 2H), 2.30 (dt, J = 2.2, 7.1 Hz, 2H), 2.71-2.85 (m, 4H), 3.17 (ddd, J = 5.9, 5.9, 13.8 Hz, 2H), 4.65 (s, 1H), 7.17–7.32 (m, 5H); 13 C NMR δ 18.3, 25.7, 28.4, 30.2, 33.6, 34.8, 76.7, 86.3, 125.9, 128.3, 128.5, 141.4. Anal. Calcd for C₁₅H₁₈S₂: C, 68.65; H, 6.91. Found: C, 69.01; H, 7.03.

In a similar manner, the 2-(alk-1-yn-1-yl)-1,3-dithianes ${\bf 4b}$ and ${\bf 4c}$ were prepared.

Preparation of 2-(*tert*-Butyldimethylsilyl)-2-(5-phen**ylpent-1-yn-1-yl)-1,3-dithiane (2d)**. To a THF (3.0 mL) solution of the dithiane 4a (262 mg, 1.0 mmol) was added butyllithium (1.37 M in hexane, 0.73 mL, 1.0 mmol) at -78 °C under argon. After the solution was stirred for 2 h, a THF (2 mL) solution of tert-butylchlorodimethylsilane (151 mg, 1.0 mmol) was added and the mixture was gradually warmed to room temperature and stirred overnight. The reaction was quenched by addition of a saturated aqueous solution of NH₄-Cl, and organic materials were extracted with ether. The extract was washed with H₂O and brine, dried over Na₂SO₄, and concentrated. The residue was chromatographed on silica gel (hexane/AcOEt = 99/1) to give **2d** (263 mg, 70%). **2d**: IR (neat) 3026, 2930, 2210, 1496, 1249, 917, 824, 698 cm⁻¹; ¹H NMR δ 0.29 (s, 6H), 1.79 (s, 9H), 1.79–2.18 (m, 4H), 2.39 (t, J = 7.1 Hz, 2H, 2.60 (ddd, J = 3.5, 3.5, 14.1 Hz, 2H), 2.76 (t, J= 7.6 Hz, 2H, 3.52 (ddd, J = 2.6, 13.3, 13.3 Hz, 2H), 7.15-7.33 (m, 5H); 13 C NMR δ -7.3, 18.8, 19.6, 25.9, 26.4, 28.0, 30.8, 34.3, 35.1, 80.2, 86.3, 125.9, 128.4, 128.5, 141.7. Anal. Calcd for $C_{21}H_{32}S_2Si$: C, 66.96; H, 8.56. Found: C, 66.75; H, 8.52.

The α -(trialkylsilyl)- β , γ -acetylenic thioacetals $2\mathbf{a}-\mathbf{c},\mathbf{e},\mathbf{f}$ were also obtained by a similar procedure described above.

Preparation of 2-Methyl-2-(trimethylsilylethynyl)-1,3-dithiane (5). Similarly to the preparation of α-(trialkylsilyl)- β , γ -acetylenic thioacetals **2**, the thioacetal **5** (193 mg, 70%) was prepared by the alkylation of 2-(trimethylsilylethynyl)-1,3-dithiane (**6**) (260 mg, 1.2 mmol) with MeI (0.09 mL, 1.4 mmol). **5**: IR (neat) 2957, 2903, 2157, 1419, 1252, 1159, 1067, 926, 859, 833, 764, 652 cm⁻¹; ¹H NMR δ 0.22 (s, 9H), 1.72–1.85 (m, 1H), 1.77 (s, 3H), 2.15 (dtt, J = 14.5, 2.8, 4.2 Hz, 1H), 2.79 (ddd, J = 3.6, 3.6, 13.8 Hz, 2H), 3.32 (ddd, J = 2.3, 12.7, 14.8 Hz, 2H); ¹³C NMR δ 0.13, 25.0, 28.8, 29.1, 41.3, 89.7, 105.1. Anal. Calcd for C₁₀H₁₈S₂Si: C, 52.11; H, 7.87. Found: C, 51.73; H, 7.94.

Similarly, 2-methyl-2-(5-phenylpent-1-yn-1-yl)-1,3-dithiane (7) (251 mg, 91%) was obtained by the alkylation of $\bf 4a$ (262 mg, 1 mmol) with MeI (0.08 mL, 1.2 mmol). 7: IR (neat) 2903, 2240, 1496, 1453, 1421, 1071, 740, 700 cm $^{-1}$; $^{1}{\rm H}$ NMR δ 1.80 (s, 3H), 1.77 – 1.93 (m, 3H), 2.11 – 2.14 (m, 1H), 2.34 (t, J=7.2 Hz, 2H), 2.75 (t, J=7.9 Hz, 2H), 2.80 (ddd, J=3.7, 3.7, 13.8 Hz, 2H), 3.35 (ddd, J=2.5, 13.6, 13.6 Hz, 2H), 7.17 – 7.32 (m, 5H); $^{13}{\rm C}$ NMR δ 18.4, 25.1, 29.1, 29.6, 30.6, 35.0, 41.4, 81.6, 85.1, 125.9, 128.4, 128.5, 141.5. Anal. Calcd for $\rm C_{16}H_{20}S_2$: C, 69.51; H, 7.29. Found: C, 69.43; H, 7.44.

Titanocene(II)-Promoted Reaction of 7 with Ethylene. Cp₂TiCl₂ (224 mg, 0.9 mmol), magnesium turnings (26 mg, 1.1 mmol; purchased from Nacalai Tesque Inc. Kyoto, Japan), and finely powdered molecular sieves 4 A (90 mg) were placed in a flask and dried with heating in vacuo. After cooling, THF (1.8 mL) and P(OEt)₃ (0.31 mL, 1.8 mmol) were added successively with stirring at 25 °C under ethylene. After 3 h, a THF (0.9 mL) solution of 7 (83 mg, 0.3 mmol) was added to the mixture. The mixture was stirred for 1 h at 25 °C and then refluxed for 1 h. The reaction was quenched by addition of 1 M NaOH, and the resulting insoluble materials were filtered off through Celite and washed with ether. The organic materials were extracted with ether and dried. After removal of the solvent, the residue was purified by PTLC (hexane) to give a mixture of 1-methyl-1-(5-phenylpent-1-yn-1-yl)cyclopropane (9) and 1-(3-phenylpropyl)-1-(prop-1-yn-1-yl)cyclopropane (10) (34 mg, 57%, 9:10 = 79:21). A mixture of 9 and 10: IR (neat) 3084, 3025, 2933, 2859, 1603, 1496, 1454, 1379, 1337, 1080, 1020, 745, 699 cm $^{-1}$; ¹H NMR δ 0.48 (dd, J=4.0, 6.4 Hz, 0.42H), 0.53 (dd, J = 4.0, 6.4 Hz, 1.58 H), 0.79 (dd, J = 4.0, 6.4 Hz,0.42H), 0.82 (dd, J = 4.0, 6.4 Hz, 1.58H), 1.25 (s, 2.37H), 1.72-1.83 (m, 2.42H), 1.76 (s, 0.63H), 2.14 (t, J=7.1 Hz, 1.58H), 2.64 (t, J = 7.2 Hz, 0.42H), 2.69 (t, J = 7.5 Hz, 1.58H), 7.14-7.32 (m, 5H); $^{13}{\rm C}$ NMR δ 3.6, 6.7, 11.8, 14.9, 16.1, 18.2, 24.6, $29.5,\, 30.7,\, 34.8,\, 35.6,\, 38.1,\, 71.9,\, 75.0,\, 83.9,\, 86.8,\, 125.6,\, 125.8,\, 125.6,\, 125.8,\, 125.6,\, 125.8$ 128.2, 128.3, 128.4, 128.5, 141.8, 142.6. Anal. Calcd for C₁₅H₁₈: C, 90.85; H, 9.15. Found: C, 90.38; H, 9.23.

Titanocene(II)-Promoted Cyclopropanation of Terminal Olefins 8 with the α -(Trialkylsilyl)- β , γ -acetylenic **Thioacetals 2.** To a THF (1.2 mL) solution of **1**, prepared from Cp₂TiCl₂ (149 mg, 0.6 mmol), magnesium turnings (17 mg, 0.7 mmol), molecular sieves 4 A (60 mg), and P(OEt)₃ (0.21 mL, 1.2 mmol) in the presence of 8c (0.36 mL, 2.4 mmol), was added a THF (0.9 mL) solution of 2a (100 mg, 0.3 mmol) at 25 °C under argon. The mixture was stirred for 1 h at 25 °C. The workup and purification described above gave 11c (73 mg, 67%). 11c: IR (neat) 3063, 3025, 2938, 2858, 2156, 1603, 1495, 1454, 1248, 1031, 840, 758, 698, 635 cm $^{-1};$ $^{1}\mathrm{H}$ NMR δ 0.13 (s, 5.49H), 0.16 (s, 3.51H), 0.24 (dd, J = 4.3, 6.5 Hz, 0.61H), 0.58(dd, J = 3.8, 5.4 Hz, 0.39H), 0.67 - 0.76 (m, 0.78H), 1.04 (dd, J)= 4.2, 8.8 Hz, 0.61H), 1.14-1.53 (m, 3.61H), 1.66-1.95 (m,3H), 2.56-2.86 (m, 4H), 7.15-7.30 (m, 10H); 13 C NMR $\delta -0.03$, $0.00,\ 16.4,\ 18.2,\ 21.4,\ 21.6,\ 25.5,\ 26.7,\ 29.0,\ 29.2,\ 30.4,\ 31.2,$ 32.6, 35.1, 35.2, 35.3, 35.5, 37.7, 79.6, 82.7, 109.3, 112.9, 125.3, 125.4, 127.86, 127.88, 127.90, 127.94, 128.01, 128.07, 128.12,

141.7, 142.1, 142.18, 142.23. Anal. Calcd for C₂₅H₃₂Si: C, 83.27; H, 8.94. Found: C, 83.57; H, 9.21.

The cyclopropanes 11a,b,d-k were also obtained by similar reactions of **2** with terminal olefins **8**.

Deuteration of the Organotitanium Intermediate. To a THF (1.8 mL) solution of 1, prepared from Cp₂TiCl₂ (224 mg, 0.9 mmol), magnesium turnings (26 mg, 1.1 mmol), molecular sieves 4 A (90 mg), and P(OEt)₃ (0.31 mL, 1.8 mmol), was added a THF (0.9 mL) solution of 2a (100 mg, 0.3 mmol) at 25 °C under argon. After the solution was stirred for 10 min, D₂O (0.3 mL) was added and the mixture was further stirred for 30 min. The usual workup and purification afforded a mixture of the dideuterated 1- and 2-alkynylsilanes 16a (>95% D) and **16b** (>95% D) (43 mg, 61%, **16a:16b** = 15:85). A mixture of **16a** and **16b**: ${}^{1}H$ NMR δ 0.11 (s, 7.56H), 0.15 (s, 1.35H), 1.50-1.57 (m, 0.3H), 1.67-1.72 (m, 0.3H), 1.78 (tt, J= 7.3, 7.3 Hz, 1.7H, 2.17 (t, J = 6.9 Hz, 1.7H), 2.62 (t, J = 7.6 (t, J = 7.6 (t, J = 1.8 (t, J = 1.8Hz, 0.3H), 2.72 (t, J = 7.7 Hz, 1.7H), 7.13–7.32 (m, 5H); ¹³C NMR (major isomer) δ –2.24, 0.00, 18.3, 31.0, 34.7, 77.8, 78.2, 125.6, 128,1, 128.4, 142.1.

Reaction of 2 with Dimethylphenylsilane. To a THF (1.2 mL) solution of 1, prepared from Cp2TiCl2 (224 mg, 0.9 mmol), magnesium turnings (26 mg, 1.1 mmol), molecular sieves 4 A (90 mg), and P(OEt)₃ (0.31 mL, 1.8 mmol), was added a THF (0.9 mL) solution of 2a (100 mg, 0.3 mmol) at 25 °C under argon. After the solution was stirred for 10 min, dimethylphenylsilane (0.54 mL, 3.6 mmol) was added and the mixture was stirred for 3 h. The usual workup and purification afforded 3-(dimethylphenylsilyl)-6-phenyl-1-(trimethylsilyl)hex-1-yne (18a) (53 mg, 48%) and 1-(dimethylphenylsilyl)-2-(3-phenylpropyl)-1-(trimethylsilyl)hex-1,2-diene (19) (12 mg, 11%). **18a**: IR (neat) 2957, 2157, 1427, 1249, 1115, 840, 646 cm⁻¹; ¹H NMR δ 0.13 (s, 9H), 0.37 (s, 6H), 1.41 (dt, J = 7.5, 7.5 Hz, 2H), 1.59-1.98 (m, 2H), 1.93 (t, J = 7.3 Hz, 1H), 2.46-1.08 (m, 2H)2.66 (m, 2H), 7.11-7.57 (m, 10H); ${}^{13}C$ NMR $\delta -5.2$, -4.5, 0.3, 20.8, 28.5, 31.0, 35.2, 85.5, 109.2, 125.6, 127.6, 128.2, 128.4, 129.3, 134.1, 136.8, 142.5. Anal. Calcd for $C_{23}H_{32}Si_2$: C, 75.75; H, 8.84. Found: C, 75.72; H, 9.03. **19**: IR (neat) 2955, 1919, 1248, 1111, 896, 839, 805, 698 cm⁻¹; ¹H NMR δ 0.04 (s, 9H), 0.44 (s, 6H), 1.72 (tt, J = 7.8, 7.5 Hz, 2H), 2.04 (dt, J = 6.9, 7.5 Hz, 2H), 2.68 (t, J = 7.8 Hz, 2H), 4.52 (t, J = 6.9 Hz, 1H),7.21-7.25 (m, 3H), 7.30-7.41 (m, 5H), 7.33-7.57 (m, 2H); 13 C NMR δ -1.39, -1.35, 0.04, 27.1, 32.0, 35.5, 76.0, 125.6, 127.6, 128.2, 128.4, 128.9, 133.9, 139.2, 142.6, 211.9. Anal. Calcd for C₂₃H₃₂Si₂: C, 75.75; H, 8.84. Found: C, 75.74; H, 8.71.

3-(Dimethylphenylsilyl)-1-(trimethylsilyl)but-1-yne (18b) was also obtained by the reactions of both 2c and 5 with dimethvlphenvlsilane.

Titanocene(II)-Promoted Olefination of Ketones 23a-i with 2. To a THF (1.8 mL) solution of 1, prepared from Cp₂-TiCl₂ (224 mg, 0.9 mmol), magnesium turnings (26 mg, 1.1 mmol), molecular sieves 4 A (90 mg), and P(OEt)₃ (0.31 mL, 1.8 mmol), was added a THF (0.9 mL) solution of 2f (82 mg, 0.3 mmol) at 25 °C. After 10 min, 1,5-diphenylpentan-3-one (23b) (86 mg, 0.36 mmol) in THF (0.9 mL) was added dropwise over 5 min and the mixture was stirred for 1 h. The usual workup and purification gave 1-(tert-butyldimethylsilyl)-3methyl-4-phenethyl-6-phenylhex-3-en-1-yne (22f) (83 mg, 71%). **22f**: IR (neat) 2928, 2137, 1604, 1496, 1454, 1249, 697 cm⁻¹; ¹H NMR δ 0.14 (s, 6H), 0.96 (s, 9H), 1.79 (s, 3H), 2.40 (dd, J= 6.1, 10.4 Hz, 2H), 2.63-2.81 (m, 6H), 7.16-7.32 (m, 10H);

¹³C NMR δ -4.4, 16.7, 18.4, 26.2, 34.1, 34.2, 34.7, 37.5, 94.0, 107.4, 114.1, 125.8, 126.0, 128.3, 128.36, 128.41, 141.7, 142.1, 147.9. Anal. Calcd for C₂₇H₃₆Si: C, 83.44; H, 9.34. Found: C, 83.74; H, 9.24.

The enynes 22a-e,g-p and their isomers 24a-e were also obtained by similar reactions of 2 with ketones 23.

Titanocene(II)-Promoted Olefination of Esters and Lactones 23j-n with 2. To a THF (1.8 mL) solution of 1, prepared from Cp₂TiCl₂ (224 mg, 0.9 mmol), magnesium turnings (26 mg, 1.1 mmol), molecular sieves 4 A (90 mg), and P(OEt)₃ (0.31 mL, 1.8 mmol), was added a THF (0.9 mL) solution of 2d (113 mg, 0.3 mmol) at 25 °C. After 10 min, γ -butyrolactone (23k) (31 mg, 0.36 mmol) in THF (0.9 mL) was added and the mixture was stirred for 3 h. The usual workup and purification by chromatography on Al₂O₃ (eluted with 1% triethylamine in hexane) gave 2-[1-(tert-butyldimethylsilyl)-6-phenylhex-1-yn-3-ylidene]oxorane (22r) (72 mg, 70%). 22r: IR (neat) 2929, 2129, 1655, 1459, 1251, 1177, 827, 776, 699 cm⁻¹; ¹H NMR δ 0.11 (s, 6H), 0.95 (s, 9H), 1.82 (tt, J = 7.7, $7.7~{\rm Hz},\,2{\rm H}),\,2.02~({\rm tt},\,J=7.2,\,7.2~{\rm Hz},\,2{\rm H}),\,2.18~({\rm t},\,J=7.3~{\rm Hz},\,2.1)$ 2H), 2.63 (t, J = 7.8 Hz, 2H), 2.76 (t, J = 7.6 Hz, 2H), 4.18 (t, J = 6.8 Hz, 2H), 7.12–7.30 (m. 5H); 13 C NMR δ –4.2, 16.7, 24.5, 26.2, 28.0, 29.6, 30.0, 35.4, 72.1, 91.7, 92.4, 107.1, 125.4,128.1, 128.5, 142.9, 164.8. Anal. Calcd for C₂₂H₃₂OSi: C, 77.59; H, 9.47. Found: C, 77.40; H, 9.76.

The enynes $\mathbf{22q}_{,\mathbf{s}}\mathbf{-u}$ were also obtained by similar reactions of 2 with the esters 23j,m,n and lactone 23l.

Protodesilylation. To the enyne **22j** (35 mg, 0.1 mmol) was added TBAF (1 M in THF, 1 mL, 1 mmol) at 0 °C. After being stirred for 2 h, the reaction was quenched by addition of H₂O. The organic materials were extracted with ether, successively washed with 1 M HCl and H₂O, and dried over Na₂SO₄. The solvent was removed under reduced pressure and the crude product was purified by PTLC (hexane) to give (6-phenylhex-1-yne-3-ylidene)cyclohexane (**26f**) (19 mg, 80%). **26f**: IR (neat) 3305, 3026, 2927, 2854, 2087, 1604, 1496, 1452, 1030, 745, 699cm⁻¹; ¹H NMR δ 1.45–1.66 (m, 6H), 1.83 (tt, J = 7.7, 7.7 Hz, 2H), 2.16-2.24 (m, 4H), 2.43-2.54 (m, 2H), 2.62 (t, J = 7.9Hz, 2H), 3.01 (s, 1H), 7.14–7.30 (m, 5H); 13 C NMR δ 26.5, 27.9, 30.0, 30.5, 30.9, 34.0, 35.3, 79.0, 84.5, 112.8, 125.6, 128.2, 128.4, 142.5, 149.7. Anal. Calcd for C₁₈H₂₂: C, 90.70; H, 9.30. Found: C, 90.49; H, 9.34.

The alkenylcyclopropanes 25a-c and enynes 26a-e,g,h were also obtained by a similar procedure.

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Supporting Information Available: Characterization data for the compounds 4b,c, 2a-c,e,f, 11a,b,d-k, 13, 18b, 22a-e,g-q,s-u, 24a-e, 25a-c, and 26a-e,g,h. This material is available free of charge via the Internet at http:// pubs.acs.org.

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