This article was downloaded by: [New York University] On: 07 October 2014, At: 04:35 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

Application of Carbodiimide in Heterocyclic Synthesis: New Facile Synthesis of 2-Aminoimidazolinone Derivatives

Ming-Wu Ding^a, Hai-Yang Tu^a & Zhao-Jie Liu^a ^a Institute of Organic Synthesis, Central China Normal University, Wuhan, China, 430070 Published online: 22 Aug 2006.

To cite this article: Ming-Wu Ding , Hai-Yang Tu & Zhao-Jie Liu (1997) Application of Carbodiimide in Heterocyclic Synthesis: New Facile Synthesis of 2-Aminoimidazolinone Derivatives, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 27:20, 3657-3662, DOI: 10.1080/00397919708007088

To link to this article: http://dx.doi.org/10.1080/00397919708007088

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and

are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

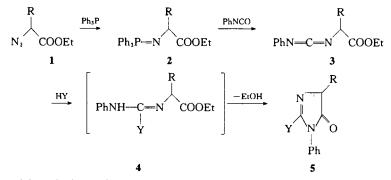
APPLICATION OF CARBODIIMIDE IN HETERO-CYCLIC SYNTHESIS:NEW FACILE SYNTHESIS OF 2-AMINOIMIDAZOLINONE DERIVATIVES

Ming-Wu Ding[•], Hai-Yang Tu and Zhao-Jie Liu Institute of Organic Synthesis. Central China Normal University, Wuhan, China, 430070

Abstract: A new facile synthesis of 2-amino-4H-imidazolin-4-ones 5 by using carbodiimides 3 which were obtained from aza-Wittig reaction of iminophosphoranes 2 with isocyanate is described.

4H-Imidazolin-4-ones are important heterocycles bearing good biological activities. Some of them have shown high herbicidal activities, such as imazamethabez and imazethapyr. They are also involved in some alkloids^[1]. There are many known methods for the synthesis of imidazolinones^[2] including the intramolecular aza-Wittig reaction^[3], however, 2-amino substituted imidazolinones were not easily accessible by currently existing routes. In our work searching for biologically active heterocycles^[4], we developed a new facile synthetic method to 2-amino-4H-imidazolin-4-one derivatives by using carbodiimides as synthons which were obtained from aza-Wittig reactions of iminophosphoranes with phenyl isocyanate.

The easily accessible azides 1 reacted with triphenylphosphine to create the iminophosphoranes 2, which were allowed to react with phenyl isocyanate to produce carbodiimide 3 via the intermolecular aza-Wittig reaction. Then a nucleophile was added to yield the imidazolinone 5 through the cyclization of 4. This approach supplies an easily accessible route to 4H-imidazolin-4-ones with various substituents. The reaction can be carried out either by one-pot version from the azides 1 or by isolation of the carbodiimides 3.



Although the carbodiimide has been widely used in heterocyclic synthesis^[5], the stability of some carbodiimides is uncertained. In preparation of carbodiimides **3**, the reaction temperature must be maintained below 30°C (the best temperature is 0°C) otherwise a deep dark solution will result. However, with the careful control of the reaction condition, **3** can be easily isolated in $75 \sim 80\%$ yields and be stored for about a month in room temperature, which providing a versatile synthon for imidazolinones. Attempts to isolate the guanidine intermediate **4** were not successful probably due to the rapid cyclization under room temperature.

The reactivity of carbodiimide 3 to various nucleophiles was researched, the results are listed in Table 1 and summarized as follows:

1. When HY is a secondary amine $(HY=HNR^{1}R^{2})$, the carbodiimides 3 react smoothly with HY in room temperature to give imidazolinones 5, however, when R^{1} , R^{2} are bulky $(R^{1}, R^{2}=i-Pr, c-hexyl)$, the reaction must be carried out in refluxing methylene dichloride, whereas when R^{1} , R^{2} are aryl $(R^{1}, R^{2}=Ph)$, the reaction didn't take place even after prolonged heating in methylene dichloride.

2. When HY is a heterocycles (HY=imidazole, benzotriazole), the reaction can take place in room temperature, however, the resulted imidazolinones 5 were unstable and getting dark when stored in room temperature.

3. When HY is an alcohol (HY=MeOH, EtOH), only traces of products were detected and the carbodiimides 3 were recovered even after heating in CH_2Cl_2 .

	R	Y	Reaction Temperature	Reaction Time	Yield(%)
5a	н	-NEt ₂	r. t.	2hr	85
5b	Н	-N	40 C	6hr	40
5c	Н		40 C	8hr	45
5d	Н	—N(Ph)Me	r. t.	4hr	63
5e	н	-N	r. t.	2hr	78
5f	н	-N 0	r.t.	2hr	80
5g	Н		r. t.	2hr	75
5h	н	NN	r. t.	2hr	75
5i	Н	-NPh ₂	40 C	18hr	0
5j	Me	-NEt ₂	r. t.	2hr	86
5 k	Н	-OCH ₃	40 C	5hr	trace
51	Н	—OEt	40 C	18hr	trace

Table 1. Preparation of 4H-Imidazolin-5-one from the Carbodiimides 3

* isolated yield based on carbodiimide 3.

Attempts to extend this method to prepare the carbodiimides 8 and 11. which might be synthons for imidazoles or 1,2.4-triazin-5-ones. were unsuccessful. In these cases, the carbodiimides might produced but dimerization or polymerization might take place quickly^[6].

$$N_{3}CH_{2}CN \xrightarrow{Ph_{3}P} Ph_{3}P = NCH_{2}CN \xrightarrow{PhNCO} PhN = C = NCH_{2}CN$$

$$6 \qquad 7 \qquad 8$$

$$N_{2}CH_{2}COOEt \xrightarrow{Ph_{3}P} Ph_{3}P = N - N = CHCOOEt \xrightarrow{PhNCO} PhN = C = N - N = CHCOOEt$$

$$9 \qquad 10 \qquad 11$$

EXPERIMENTAL

NMR were taken on a Varian XL-200 spectrometer. IR were recorded on a Shimadzu IR-408 infrared spectrometer. MS were measured on a HP 5988A spectrometer.

General procedure for the preparation of 4H-imidazolin-4-ones 5:

A solution of triphenylphosphine (2.62 g, 10 mmol) in dry methylene dichloride (15 ml) was added dropwise under nitrogen at room temperature to a well-stirred solution of the azide 1 (10 mmol) in dry methylene dichloride (10 ml). The reaction mixture was stirred at room temperature for 0.5 h, then phenyl isocyanate (1.19 g, 10 mmol) was added at 0°C. After the reaction mixture was stirred at room temperature for 2 h, the solvent was removed off under reduced pressure and the residual material was eluted with ether/petroleum ether (1 : 10) through a short silica gel column to give the carbodiimde 3 (yields $75 \sim 80\%$).

A solution of the nucleophile HY (2 mmol) in $CH_2Cl_2(5 \text{ ml})$ was added to 3 (2 mmol) in $CH_2Cl_2(10 \text{ ml})$, the mixture was stirred for the time shown in Table 1 at room temperature or at refluxing temperature, the solvent was removed and the residual material was eluted with ethyl estate to give the imidazolinones.

¹HNMR, IR and MS for some compounds 5:

- 5a: ¹HNMR (CDCl₃, 200MHz) δ 7. 49~7.27 (m, 5H), 4. 26(s, 2H), 3. 04 (q, 4H, J=7.0Hz), 0. 99(t, 6H, J=7.0Hz). IR(cm⁻¹), 1740, 1620, 1440, 1275, 1100. MS(m/z), 231(M⁺, 17.57%), 202(39.72%), 188 (11.83%), 145(42.71%), 112(51.13%), 77(100%).
- 5d: 1 HNMR (CDCl₃, 200MHz) δ 7. 44 ~ 6. 64 (m, 10H), 4. 28(s, 2H), 3. 22(s, 3H). IR (cm⁻¹), 1742, 1625, 1590, 1495. MS (m/z), 265 (M⁺, 10. 15%), 131(10. 43%), 77(100%).
- 5e: 1 HNMR(CDCl₃,200MHz) δ 7. 50 \sim 7. 14(m,5H),4. 15(s,2H),3. 08 \sim 2. 88 (m,4H),1. 64 \sim 1. 32(m,6H). IR(cm⁻¹),1738,1624,1420,1338,1280. MS(m/z),243(M⁺,2. 30%),214(1. 40%),131(8. 53%),77(100%).
- 5f: ¹HNMR(CDCl₃, 200MHz)δ7. 50~7. 12(m, 5H), 4. 14(s, 2H), 3. 60~3. 44

 $(m, 4H), 3.05 \sim 2.90(m, 4H)$. IR $(cm^{-1}), 1742, 1620, 1395, 1114$. MS $(m/z), 245(M^{\ddagger}, 6.58\%), 216(3, 00\%), 188(15, 13\%), 154(39, 83), 131(53, 93), 77(100\%)$.

5): 1 HNMR(CDCl₃, 200MHz) δ 7. 50 ~ 7. 12(m, 5H), 4. 14(q, 1H, J = 7. 2Hz), 2. 97(q, 4H, J = 7. 2Hz), 1. 42(d, 3H, J = 7. 2Hz), 0. 94(t, 6H, J = 7. 2Hz). IR(cm⁻¹), 1740, 1615, 1495, 1275, 1180. MS(m/z), 245(M[‡], 5. 04%), 216 (6. 44%), 145(15. 56%), 141(16. 59%), 119(19. 84%), 77(100%).

Acknowledgements

We gratefully acknowledge financial support of this work by the Dawn Plan of Science and Technology for Young Scientists of Wuhan City and the Natural Science Foundation of Hubei Province.

REFERENCES AND NOTES

- Guella, G.; Mancini, I.; Zibrowius, H.; Pietra, F. Helv. Chim. Acta., 1989, 72, 1444; Djura, P.; Faulkner, D. J. J. Org. Chem., 1980, 45; 735; Debitus, C.; Cesario, M.; Guilhem, J.; Pascard, C.; Pais, M. Tetrahedron Lett., 1989, 30, 1535.
- Lehr, H.; Karlan, S.; Goldberg, M. W., J. Am. Chem. Soc. 1953, 75, 3640; Cole, J. O.; Ronzio, A. R., J. Am. Chem. Soc., 1944, 66, 1584; Devasia, G. M.; Shafi, P. M., Indian J. Chem., 1979, 17B, 526; 1981, 20B, 657.; Kumar, P.; Makerjee, A. K., Indian J. Chem., 1979, 18B, 240.; Bonner, W.; Hermann, D., Ann. 1978, 1704.; Ashare, R.; Mukerjee, A. K., Indian J. Chem. Soc., Sect. B, 1986, 25B, 762.
- Takeuchi, H.; Hagiwara, S.; Eguchi, S., Tetrahedron, 1989, 45, 6375.
- Ding, M. W.; Shi, D. Q.; Xiao, W. J.; Huang, W. F.; Wu, T. J., Phosphorus, Sulfur and Silicon, 1995, 102, 59.; Chinese J. Applied Chem. 1995, 12, 9.; Chem. J. Chin. Univ. 1995, 16, 1396.
- For recent examples of application of carbodiimide in heterocyclic synthesis. see: Okawa, T.; Eguchi, S.; Kakehi, A. J. Chem. Soc., Perkin. Trans. I, 1996, 247.; Wamhoff, H.; Bamberg, C.; Herrmann, S.; Nieger, M. J. Org. Chem., 1994, 59, 3985.; Molina, P.; Alajarin, M.; Vidal, A., Tetrahedron, 1995, 51, 5351.; Molina, P.; Alias, A.; Bal-

ado, A.; Arques, A., *Liebigs Ann. Chem.* **1994**, 745.; Rodrigues, J. A. R.; Leiva, G. C; de Sousa, D. F., *Tetrahedron Lett.*, **1995**, 36, 59.; Molina, P.; Aller, E.; Ecija, M.; Lorenzo, A. *Synthesis*, **1996**, 690; or see the review: Molina, P.; Vilaplana, M. J. *Synthesis*, **1994**, 1197.

 Molina, P.; Alajarin, M.; Lopez-Leonardo, C., J. prakt. Chem. 1993, 335, 305.

(Received in The Netherlands 20 May 1997)