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Traceless Photolabile Linker Expedites Chemical Synthesis of Complex Oligosaccharides by Automated Glycan Assembly

Kim Le Mai Hoang,[†] Alonso Pardo-Vargas,^{†,‡} Yuntao Zhu,[†] Yang Yu,^{†,‡} Mirco Loria,[‡] Martina Delbianco,[†] and Peter H. Seeberger^{*,†,‡}.

[†]Department of Biomolecular Systems, Max-Planck-Institute of Colloids and Interfaces, Am Mühlenberg 1, 14476 Potsdam, Germany

[‡]Institute of Chemistry and Biochemistry, Freie Universität Berlin, Arnimallee 22, 14195 Berlin, Germany

ABSTRACT: Automated glycan assembly (AGA) aims at accelerating access to synthetic oligosaccharides to meet the demand for defined glycans as tools for molecular glycobiology. The linkers used to connect the growing glycan chain to the solid support play a pivotal role for the synthetic strategy as they determine all chemical conditions used during the synthesis and the form of the glycan obtained at the end of it. Here, we describe a traceless photolabile linker to prepare carbohydrates with a free reducing end. Modification of the *o*-nitrobenzyl scaffold of the linker is key to high yields and compatibility with the AGA workflow. **The assembly of an asymmetrical bi-antennary *N*-glycan from oligosaccharide fragments prepared by AGA** and linear as well as branched β -oligoglucans are described to illustrate the power of the method. These substrates will serve as standards and biomarkers to examine the unique specificity of glycosyl hydrolases.

INTRODUCTION

The procurement of complex oligosaccharides in satisfactory quantity and quality as tools for glycobiology remains challenging. Harvesting from natural sources is untenable, because the post-translational glycosylation pathway is heavily influenced by minute changes in the environment of the living cell, leading to multiple glycoforms.¹ At present, chemical and enzymatic approaches, or combinations thereof, are often the only reliable methods to access pure glycans.² Efforts to reduce the time and resources spent on traditional chemical syntheses have focused mainly on automated glycan assembly (AGA)³ and computer-assisted one-pot synthesis.⁴

AGA relies on the delivery of building blocks to a suspension of solid support that is equipped with a linker, followed by addition of a suitable activating solution using either home-built instruments or the commercial Glycconeer 2.1 glycan synthesizer.⁵ Precise control of important reaction parameters such as temperature and concentration under an inert atmosphere allows for near quantitative coupling of building blocks to the resin. Repetitive cycles consisting of glycosylation, capping and selective deprotection steps extend the glycan chain to the desired structure. Different linkers for AGA were explored (Figure 1), including the metathesis-labile linker **1**,⁶ the base-labile linker **2**,⁷ and the photocleavable linkers **3** and **4**.⁸ Photocleavable linkers such as **3** and **4** are often referred to as “traceless” since the on-resin removal of photosensitive groups only require exposure to light to release the product into the solution. They have proven stable under both acidic and basic conditions and are

compatible with a wide range of protecting groups such as carbonates (9-fluorenylmethoxycarbonyl, Fmoc), esters (levulinoyl, Lev and benzoate, Bz) as well as ethers (benzyl, Bn). These linkers are cleaved to afford glycans carrying an aminoalkyl spacer at the reducing end. The resulting oligosaccharides, after global deprotection, are ready for conjugation to carrier proteins to give glycoconjugate vaccine candidates,⁹ or to be immobilized on microarrays for high-throughput analysis of protein-carbohydrate interactions.¹⁰

Here, we report the design and use of the spacer-free photolabile linker **5** that retains the excellent chemical orthogonality and enables access to protected glycans with a free reducing end. These in turn can be transformed into the corresponding glycosyl fluorides, phosphates, and trichloroacetimidates. Oligosaccharide fragments produced in this way can be united in block couplings to prepare more complex polysaccharides. Furthermore, free reducing glycans are valuable analytical standards to inspect fragmentation patterns in tandem mass spectrometry,¹¹ or **for glycosynthase assays that demand linker-free glycan epitopes as precursors**. Photolabile linker **5** offers flexibility compared to previous methods,¹² which prepared glycans with a benzyl group at the reducing terminus. The new linker **5** is illustrated for the AGA of a penta- and a trisaccharide that are combined in a [5+3] glycosylation strategy to obtain an asymmetric bi-antennary *N*-glycan fragment. Additionally, several structurally defined linear and branched laminarin β -(1,3)-oligoglucans are prepared, which will be instrumental to elucidate the **hydrolytic mechanism of various hydrolases**

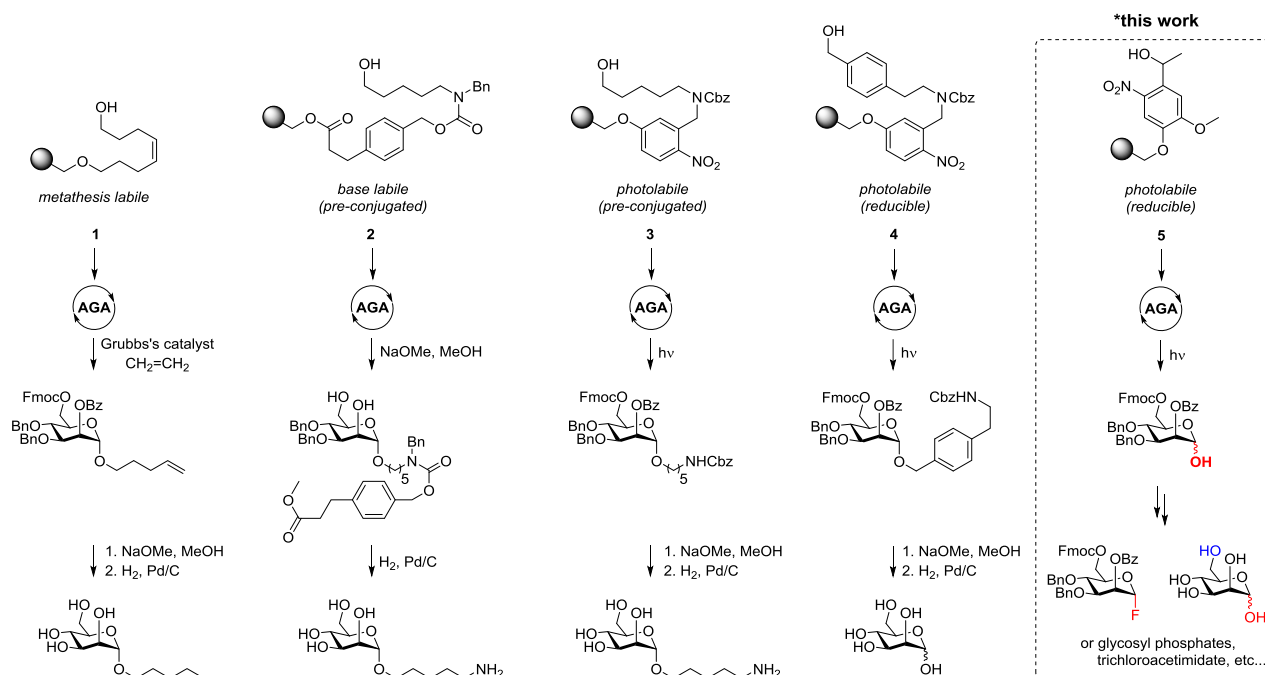


Figure 1. Linkers for AGA.

from marine bacteria. Investigations into these highly specific processes are hampered by the lack of high purity β -glucans from seawater that contain heterogenous glycans released by different microbial species.

RESULTS AND DISCUSSION

Design and Synthesis of Photolabile Linkers. One of the most frequently used photolabile groups¹³ to date is the methyl-6-nitroveratryl (MeNV) that introduces two modifications to the *o*-nitrobenzyl (oNB) scaffold: the α -methyl group at the benzylic carbon and two methoxy groups on the aromatic ring. The methyl group was thought to improve the chemical yield by the release of a ketone byproduct instead of an aldehyde, which is less prone to side-reactions.¹³ Meanwhile, the introduction of electron donating substituents on the aromatic ring increases the absorption coefficient at longer wavelengths, to match the peak emission at 366 nm from a mercury UV lamp.¹⁴ A design based on the MeNV motif should improve the photocleavage process. To provide an anchor point to the solid support, one of the methoxy groups of the MeNV was substituted. Three different MeNV-type linkers **5**, **6** and **7** (Figure 2) were prepared starting from apocynin, an abundant plant extract. After photocleavage, the resin equipped with linker **5** will produce spacer-free reducible glycans while linker **6** returns glycans with the aminoalkyl spacer and linker **7** produces benzyl-protected glycans at the reducing end. For a direct comparison, the oNB-type linkers **8** and **3** were prepared to provide the same products as linkers **5** and **6**. Resin loading was determined for each batch of resin by glycosylation of 50 mg resin with 6-O-Fmoc mannose building block **A** on the synthesizer (Scheme 2). Cleavage of the Fmoc group present on the solid support was quantified by UV-vis spec-

trophotometry¹⁵ and revealed loading values ranging from 0.32 to 0.36 mmol/g.

Linker **8** was prepared by reduction of 5-hydroxy-2-nitrobenzaldehyde **9** and immobilization on Merrifield resin by heating with Cs_2CO_3 in the presence of catalytic amounts of TBAI (Scheme 1A). Linker **3** was prepared starting from the reductive condensation of **9** with 5-amino-pentan-1-ol to yield **11** in 85% yield.⁸ Addition of Cbz to the secondary amine was followed by immobilization onto Merrifield resin to obtain the linker-equipped resin **3** (Scheme 1B).

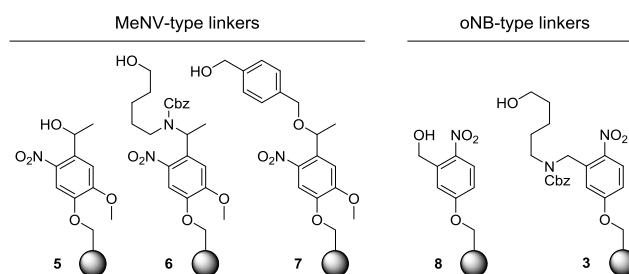
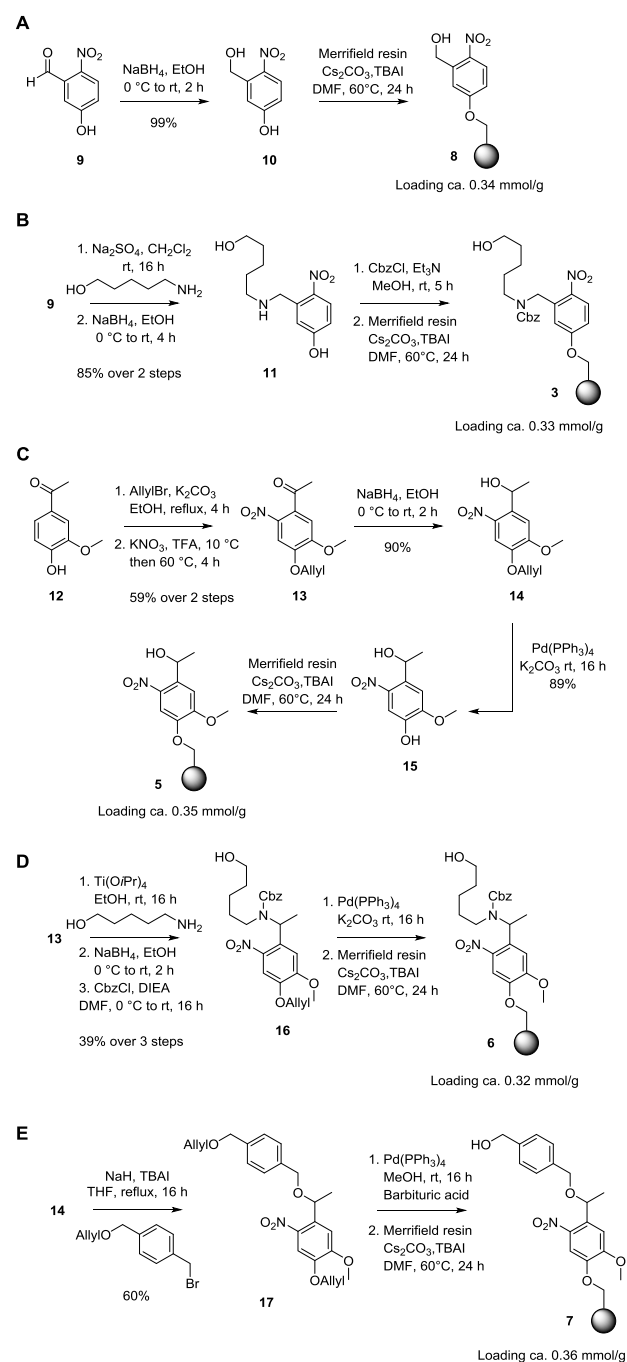


Figure 2. Merrifield resins functionalized with photolabile linkers for AGA.

Syntheses of linkers containing the MeNV scaffold started from allyl-protected apocynin **12** that was selectively nitrated with KNO_3 in TFA. This reaction is very sensitive to the initial temperature¹⁶ as di-nitrated products were formed when a slight excess of KNO_3 was introduced at 25 °C. Reactions initiated at 0 °C produced only trace amounts of **13**. The reaction proceeded best when KNO_3 was added portion-wise at exactly 10 °C, followed by heating at 60 °C for four hours to give mono-nitro **13** selectively (59% conversion starting from **12**). Next, reduction of the ketone and cleavage of the allyl group employ-

ing $\text{Pd}(\text{PPh}_3)_4$ proceeded smoothly to provide precursor **15**, ready for conjugation with the resin to give linker **5** (Scheme 1C). Utilizing intermediates **13** and **14**, the *N*-linked and *O*-benzylether-linked variations of linker **5** were prepared. The reductive amination of ketone **13** was known for its poor yield.¹⁷ Reasonable conversion was achieved by using $\text{Ti}(\text{OiPr})_4$ in an one-pot transformation of **13** to the Cbz-protected compound **16** (39% over three steps).¹⁸ Removal of the allyl group and on-resin functionalization afforded linker **6** (Scheme 1D). Double deallylation of **17** to linker **7** proceeded best when 1,3-dimethylbarbituric acid replaced K_2CO_3 as scavenger, as the latter failed to completely remove the protecting groups (Scheme 1E).

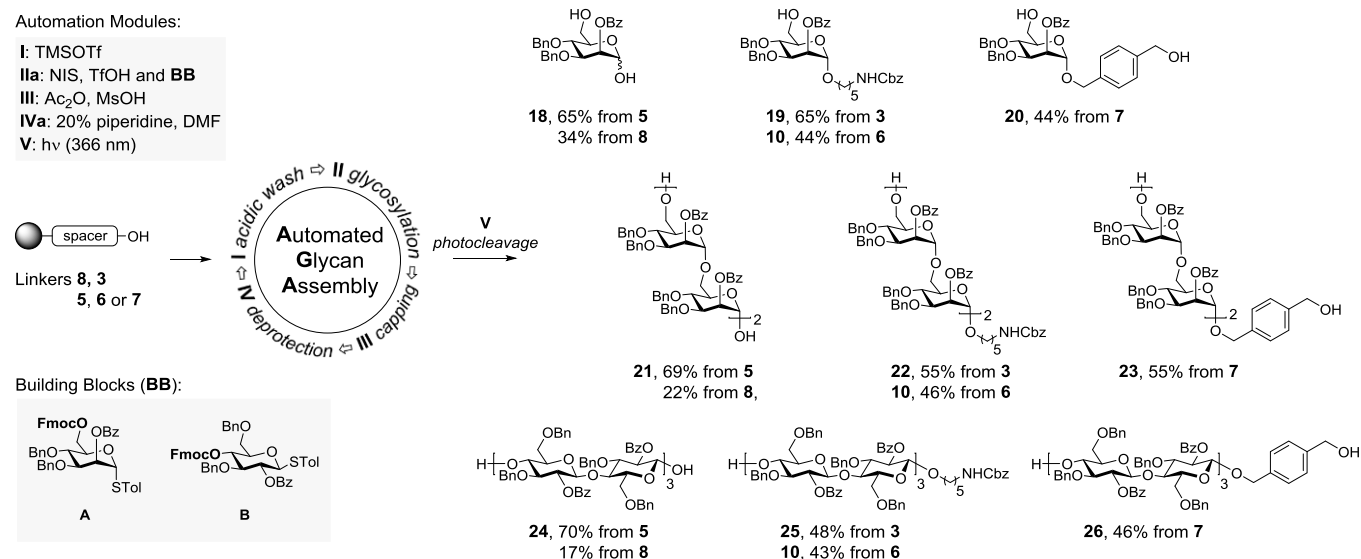
Scheme 1. Synthesis of photolabile linkers.



Evaluation of Photolabile Linkers for AGA. With the photocleavable linkers in hand, evaluation of the support-bound photolytic process was based on the quantity of fully protected oligosaccharides released after cleavage (Scheme 2). Recovery of cleaved product is an important parameter that emphasizes the practical utility of each functionalized resin. The automated synthesis of tetra-mannose **22** proceeded as follows: functionalized resin **3** (40 mg = 0.013 mmol) was placed in the reaction vessel before the acidic wash module **I** was executed (TMSOTf in CH_2Cl_2 at -20°C for 3 min) to remove any residual base and water from the resin. Next, module **IIa** delivered 6.5 equivalents of mannose **A** and a solution of activator (NIS/TfOH in CH_2Cl_2 /dioxane, -20°C for 5 min, then 0°C for 20 min) to the reaction vessel. Capping module **III** (MsOH in $\text{Ac}_2\text{O}/\text{CH}_2\text{Cl}_2$, 20 min) masked any unreacted hydroxyl groups, followed by cleavage of the Fmoc carbonate on the C-6 hydroxyl group with module **IVa** (20% piperidine in DMF, 5 min). The module sequence **I-IIa-III-IVa** was repeated four times to obtain the resin-bound tetra-mannose **22**. Cleavage of **22** from the solid support was achieved using a continuous flow photoreactor.¹⁹ Mono-mannose, tetra- α -(1,6)-mannose and hexa- β -(1,4)-glucose were assembled on resins equipped with different linkers. All linkers proved to be compatible with all standardized AGA protocols. No trace of deletion sequences was observed in any of the HPLC chromatograms of the crude products after photocleavage. Thus, differences in the isolated yield should correlate exclusively with the photolytic sensitivity of each linker. Linkers **5** and **8**, for example, both delivered the free reducing end sugars, but differed significantly in their photolytic efficiency. For oNB-type linker **8**, the highest conversion observed was monomer **18** at only 34%, with even lower yields for tetra-mannose (**21**, 22%) and hexa-glucose (**24**, 17%). In contrast, MeNV-type linker **5** consistently delivered the same products in 65-70% yield. *O*-linked resin **5** performed best in this study, surpassing both the *N*-linked resin **6** and *O*-Bn-linked resin **7**.

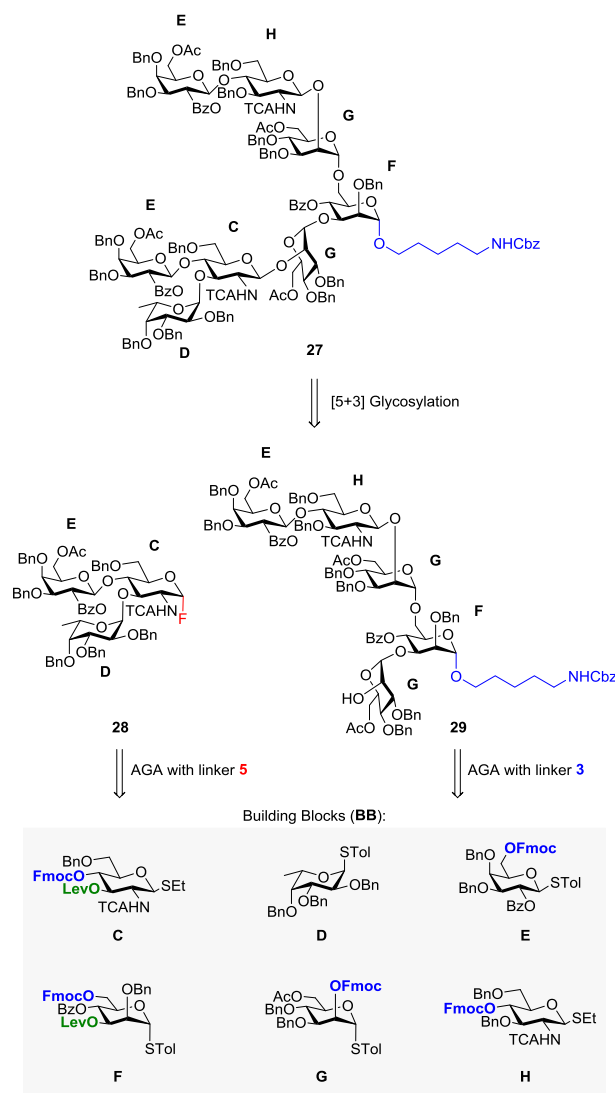
The resin equipped with *N*-linked oNB **3** outperformed the *N*-linked MeNV-type resin **6** in all experiments to procure 5-aminopentyl mannoside **19**, tetra-mannoside **22** and hexa-glucoside **25**. Still, the resin with linker **6** released more product than with linker **8**. The photolytic sensitivity of *O*-Bn-linked resin **7** was found to be between that of the *N*-linked resins **3** and **6**. The following order of photocleavage performance was observed: **5** > **3** > **7** > **6** > **8** (direct *O*-linked MeNV-type > *N*-linked oNB-type > *O*-Bn-linked MeNV-type > *N*-linked MeNV-type > direct *O*-linked oNB-type). This seemingly counterintuitive series highlights one of the caveats when picking a suitable photocleavable scaffold for a specific leaving group. The photolytic process depends both on nature of the leaving groups and the photo-labile core.²⁰ Hence, to prepare reducible glycans by AGA, the use of MeNV-type linkers **5** or **7** is most efficient. For conjugation-ready oligosaccharides carrying a terminal amine, the best option is oNB-type linker **3**.

Scheme 2. Comparison of Photolabile Linkers.

Assembly of an Asymmetric *N*-glycan Fragment.

AGA of asymmetrically branched octasaccharide **27**, present in complex-type *N*-glycans served to illustrate the utility of the new linker (Figure 3). The complexity of *N*-glycans and their biological importance has rendered them attractive targets to test chemical²⁵⁻²⁸ and chemoenzymatic synthesis methods.²⁹⁻³² Following the convergent strategy employed by most solution-phase syntheses of bi-antennary *N*-glycans, we embarked on a [5+3] glycosylation strategy. Target oligosaccharide **27** was constructed by union of trisaccharide donor **28** and pentasaccharide acceptor **29**, that were both prepared by AGA (Figure 3). Resins with photocleavable linkers **5** and **3** and six commercially available building blocks (**C-H**) were employed in the automated syntheses.

The automated synthesis of donor **28** commenced with glucosamine **C** (Figure 4) and the orthogonal removal of levulinoyl ester at C-3 position was achieved by executing module **IVb** (N₂H₄.HOAc in pyridine). Stereoselective α -fucosylation was secured by running the glycosylation module **IIa** twice for fucose **D** at lower temperature (-40 °C for 5 min, then -20 °C for 20 min). The Fmoc group at the C-4 position of **C** was removed by module **IVa** (20% piperidine in DMF). Next, the modules **I-IIa-IVa-III** were executed to introduce galactose **F** as the third sugar. By capping after the deprotection module, the Fmoc group at C-6 of galactose **E** was converted to the more resilient acetate group. At any point during the automated process, a small amount of resin could be extracted, irradiated using a portable UV lamp (6 W, 366 nm) for ten minutes to release the glycans from the solid support. This convenient real time monitoring of the reaction progress via HPLC and MALDI analysis helped to optimize the reaction conditions and identify potential problems. Photocleavage from the solid support afforded trisaccharide **30** as a mixture of α/β -anomers in 52% yield. Fluorination of **30** by Deoxo-Fluor yielded glycosyl fluoride **28**

Figure 3. Retrosynthetic analysis of octasaccharide **27**.

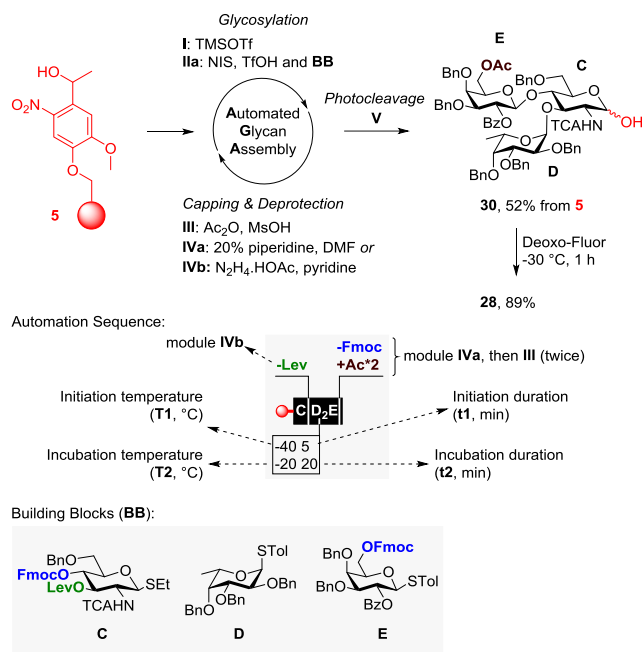


Figure 4. Automated synthesis of trisaccharide donor **28**. A typical glycosylation cycle included the module sequence I-IIa-III-IVa, represented by the letter of the building block being used, e.g. C. Changes in the sequence or parameters were indicated with break line and box.

in 89% yield after HPLC purification. Glycosyl fluorides are glycosyl donors for both chemical³³ and enzymatic³⁴ trans-glycosylation reactions.

With glycosyl donor **28** in hand, the automated synthesis of glycosyl acceptor **29** was initiated (Figure 5). Branching at the C-6 position of mannose F was achieved via step-wise introduction of mannose G, glucosamine H, and galactose E with an on-resin capping of C-6 as acetyl ester (module IVa, then III). The Lev ester on mannose F was cleaved to reveal the C-3 hydroxyl group, ready for branching with mannose building block G. Fmoc removal on G concluded the automated sequence before photolytic release furnished acceptor **29** (Figure 5A). However, the building block sequence FGHEG was not crowned by success. Analysis of the crude mixture by HPLC and MALDI revealed the presence of several deletion sequences and regioisomeric products (Figure 5B). Deletion sequences resulting from incomplete couplings were avoided by performing the glycosylation modules twice on unreactive building blocks such as F and H. Fmoc cleavage using piperidine was previously found to be accompanied by partial migration of Lev to the liberated hydroxyl group.³⁵ This unwanted byproduct was not formed when Et₃N was used as base. Thus, we introduced the deprotection module IVc (20% Et₃N in DMF) to replace module IVa. Glycosylation module IIa was performed twice for building blocks F and H. The coupling time to introduce building block H and E was increased from 20 to 40 min. Gratifyingly, the refined building block sequence F₂GH₂EG significantly improved the synthesis, as evident from the HPLC traces of the crude product showing only two peaks (Figure 5C). Structural analysis of

each compound revealed that the main glycan was acceptor **29** and the minor its diastereoisomer having an aberrant Man-β-(1,3)-Man glycosidic bond with a ¹J_{C,H} value of 160 Hz at 95.8 ppm (Figure 5A, circled in red). To support this assignment, AGA of glycan sequence F₂GH₂E proceeded smoothly to afford the congener tetrasaccharide as the sole product (see ESI, compound S6). The poor stereo-selectivity (α/β = 3:2) of the last glycosylation cycle leading to **29** was unexpected since the preliminary screening³⁶ of all building blocks showed excellent selectivity. The “double stereodifferentiation” effect, also referred to as the “matched and mismatched” principle, may account for the stereochemical scrambling in the synthesis of β-(1,3)-glucans.³⁷ A similar mechanism could be responsible for the substantial ratio of β-isomer of **29**.

With both glycosyl donor **28** and glycosyl acceptor **29** in hand, the glycosylation reaction promoted by AgOTf/Cp₂HfCl₂ furnished octasaccharide **27** in 21% isolated yield (Scheme 3). An aliquot taken from the reaction vial after four hours was analyzed by analytical HPLC and showed near complete disappearance of donor **20**. Product **27** was

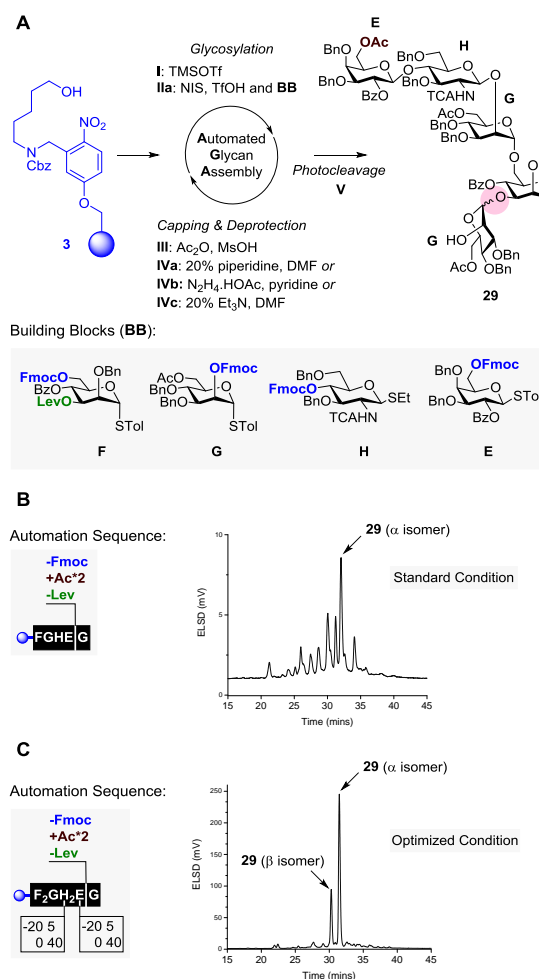
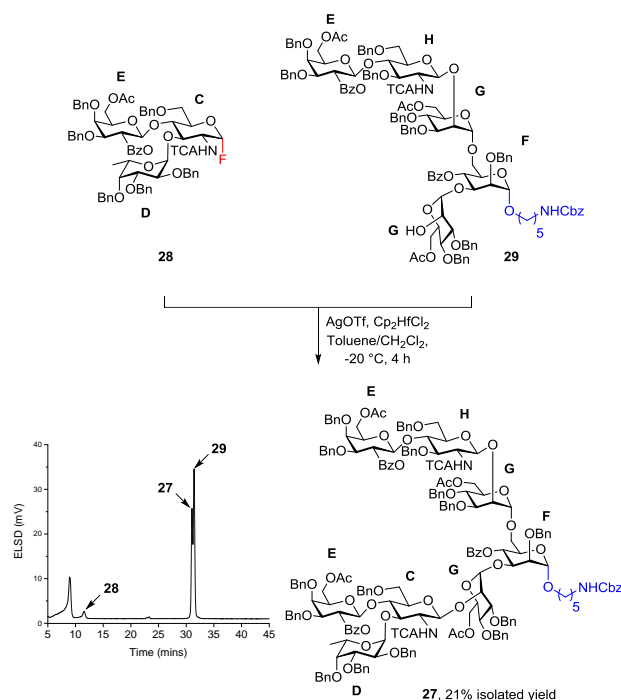


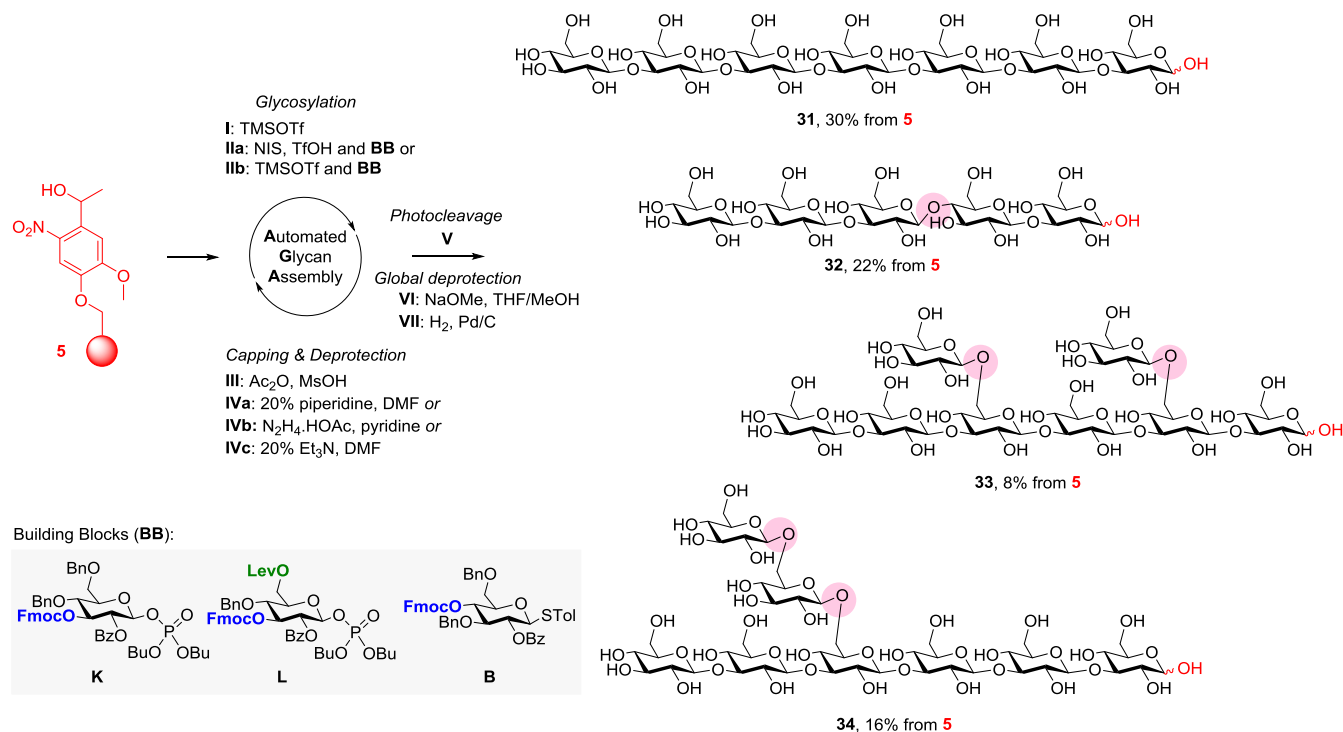
Figure 5. (A) Automated synthesis of pentasaccharide acceptor **29** using building blocks E-H. (B) Standard AGA condition led to complicated HPLC traces of crude products. (C) Optimized condition revealed glycan **29** as a mixture of diastereoisomers.

found to elute in close proximity to glycosyl acceptor **29**. The fraction containing only glycan **27** was collected and ^1H -, ^{13}C -, and coupled HSQC-NMR experiments confirmed the homogeneity of the sample, indicating a successful β -glucosamidation. The synthesis of the asymmetrical *N*-glycan octasaccharide **27** from glycan fragments prepared by AGA highlights the utility of the traceless photolabile linker **5** for the convergent synthesis of complex oligosaccharides.

Scheme 3. Assembly of octasaccharide **27**.



Scheme 4. Automated Synthesis of Oligo- β -Glucans.



Automated Synthesis of Oligo- β -Glucans. Marine algae are major carbon sink that convert carbon dioxide into carbohydrate materials such as laminarin, an oligosaccharide comprising of β -(1,3)-linked glucose with variable degrees of β -(1,4)- and β -(1,6)-glucose residues in the backbone and the branches. Studies concerning the laminarin hydrolase mechanism require homogenous β -(1,3)-glucans. Traceless resin **5** expedites the automated synthesis of oligosaccharides that serve as analytical standards and substrates to investigate the ecological roles of laminarin.

Convergent solution-phase syntheses of β -(1,3)-glucans were hampered by low yields and aberrant α -linkage formation due to “double stereo-differentiation” effects, as well as limitation in chain length, position and degree of branching. AGA of linear and branched β -(1,3)-glucans carrying an aminoalkyl spacer at the reducing end was achieved earlier using glycosyl phosphate building blocks and photolabile linker **3**.³⁸ The unprotected and spacer-free laminarin derivatives **31–34** were chosen as model substrates to compare the linker performance to the earlier study (Scheme 4). In addition to linear β -(1,3)-heptaglucose **31**, different branched β -glucans such as penta-glucose **32**, as well as octa-glucoses **33** and **34** were prepared (Scheme 4, circled in red). **The more reactive phosphate building blocks K and L required 4.0 equivalents for the AGA cycle.** Extension of the linear β -(1,3)-glucose backbone employed building block **K** bearing Fmoc group at the C-3 position. For structures with a β -(1,6)-branch, the linear backbone was first built up using the pre-defined sequence of glycosyl phosphates **K** and **L**. Building block **L** contains a C-3 Fmoc protected hydroxyl group and a Lev ester on the C-6 hydroxyl. Selective removal of the levulinoyl ester allowed for the extension of the β -(1,6)-branch. Module **IVa** (20% piperidine in DMF) was used to remove Fmoc, except when Lev was present,

in which case module **IVc** (20% Et₃N in DMF) was employed. Further optimization³⁶ resulted in a 20% reduction of building block consumption. The methanolysis of base-labile protecting groups directly on the solid support was more effective than in solution. The partially protected oligosaccharides were photocleaved from the resin and immediately debenzylated via hydrogenation using Pd/C catalyst (60 psi in under one hour, module **VII**). With this protocol, only a single purification with reverse-phase HPLC was needed to yield oligosaccharides **31** and **32** in 30% and 22% yield respectively. Branched octasaccharides **33** and **34** were isolated in lower quantity as these glycans unexpectedly fragmented during hydrogenation to create mixtures of truncated glycans.

CONCLUSION

Traceless photolabile linkers for automated glycan assembly were developed to prepare complex oligosaccharides with a free reducing end. For oligosaccharides with an aminoalkyl spacer at the reducing end, o-NB type linker **3** offers the best photocleavage efficiency, whereas for spacer-free hemiacetal glycans, MeNV-type linker **5** is the best choice. The new linker enabled the convergent synthesis of an asymmetrically branched *N*-glycan octasaccharide and of pure laminarins. Levulinoyl ester migration during Fmoc cleavage is suppressed by using Et₃N as base. AGA of four β-(1,3)-glucans was achieved using 20% less building block than previously. A new protocol for the global deprotection of oligosaccharides with free reducing end was developed. AGA using the new linkers enables access to glycans that can be converted into glycosylating agents to be used on block couplings. The glycans with a free reducing end are valuable standards for mass spectrometry and biological assays.

ASSOCIATED CONTENT

Supporting Information.

The supporting information is available free of charge via the Internet at <http://pubs.acs.org>. Synthetic procedures, AGA modules, and characterization data of glycans, including HPLC chromatograms and NMR spectra

AUTHOR INFORMATION

Corresponding Author

*Peter.Seeberger@mpikg.mpg.de

ORCID

Peter H. Seeberger: 0000-0003-3394-8466

Notes

The authors declare no competing financial interest.

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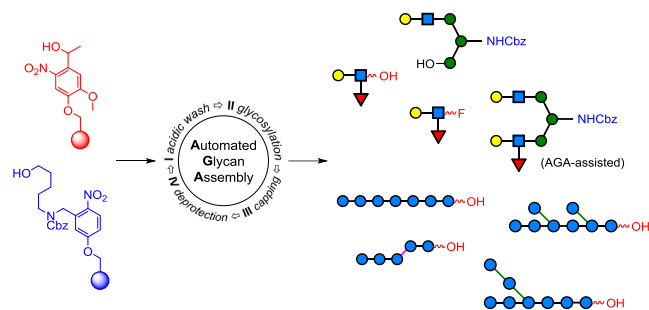


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