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**Cheol Hee Hwang, You Hoon Chong, Sue Yeon Song, Hyo Shin Kwak and Eun Lee\*** School of Chemistry and Molecular Engineering, Seoul National University, Seoul 151-747, Korea.

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E-mail: eunlee@snu.ac.kr; Fax: 82-2-889-1568; Tel: 82-2-880-6646

## Chiral secondary alcohols may be prepared from primary alcohols via asymmetric C–H insertion reactions of $\alpha'$ -alkoxy- $\alpha$ -diazoketones catalyzed by rhodium(II) (2*R*,3*R*)-3-phenylcholes-tane-2-carboxylate.

Rhodium(II)-catalyzed C–H insertion reactions of  $\alpha'$ -alkoxy- $\alpha$ diazoketones are well known to yield 3(2*H*)-furanones as the insertion occurs at the C–H bonds adjacent to the ether oxygens.<sup>1–4</sup> The reaction proceeds with retention of configuration, and efficient conversion of secondary alcohols to tertiary alcohols was realized *via* oxidative transformations. This way, chiral tertiary alcohols may be prepared from chiral secondary alcohols (Scheme 1).<sup>2</sup>

Preparation of chiral secondary alcohols from achiral primary alcohols presents a completely different and more difficult problem. Asymmetric C–H insertion reaction of  $\alpha'$ -alkoxy- $\alpha$ diazoketones prepared from primary alcohols requires developing appropriate chiral catalysts capable of discriminating two prochiral hydrogens on the carbinol carbon (Scheme 2).

The substrate  $\alpha'$ -octyloxy- $\alpha$ -diazoacetone (**3a**) was prepared from octanol (**1a**) *via* octyloxyacetic acid (**2a**). Insertion reactions were carried out and the product 3(2*H*)-furanone **4a** was converted into the cyclic acetal **5a** by treatment with *m*-chloroperoxybenzoic acid.<sup>2</sup> Methanolysis of **5a** under acidic conditions afforded methyl 3-hydroxydecanoate (**6a**) as the final product. Enantiomeric excess was calculated for each reaction by converting **6a** into the (*S*)-(*O*)acetylmandelate mixture (**7a** and **8a**) and analyzing the <sup>1</sup>H-NMR spectrum (Scheme 3).

Enantioselective processes for reactive carbenoids derived from diazoketones have proved to be problematic. Among acceptorsubstituted carbenoids, carbenoids derived from  $\alpha$ -diazoketones are more reactive than those derived from  $\alpha$ -diazoacetates and  $\alpha$ diazoacetamides, and C–H activation reactions with enantiopure rhodium(II) carboxylate or carboxamidate catalysts were reported to generate little asymmetric induction.<sup>5</sup> In our case, the insertion reaction of **3a** in dichloromethane in the presence of Rh<sub>2</sub>[*S*-DOSP]<sub>4</sub> at room temperature proceeded to give the final product **6a** in



Scheme 1 Chiral tertiary alcohol preparation via C-H insertion reaction.



Scheme 2 Chiral secondary alcohols from primary alcohols

relatively low yield (28%),<sup>6</sup> and the level of asymmetric induction was low (13% e.e.).<sup>7</sup> Use of  $Rh_2[S$ -TBSP]<sub>4</sub> did not much improve the chemical yield or the asymmetric induction. Use of  $Rh_2(5S-MEPY)_4$  in dichloromethane required heating, which resulted in a low level of asymmetric induction (8% e.e.) (Table 1).

There was clearly a need for a new chiral catalyst system. We considered the chiral *trans*-2-phenylcyclohexanecarboxylate motif for construction of chiral rhodium( $\pi$ ) carboxylate, and decided to investigate the efficacy of rhodium( $\pi$ ) (2*R*,3*R*)-3-phenylcholestane-2-carboxylate (**12**, Rh<sub>2</sub>(PCC)<sub>4</sub>).

Synthesis of **12** started from the known alcohol **10**<sup>8</sup> prepared from cholesterol (**9**). PCC oxidation afforded the corresponding ketone, which was mainly converted into the 3*S*-carboxaldehyde *via* methylenation, hydroboration, and oxidation. The requisite 3*R*-carboxaldehyde was obtained under basic equilibrating conditions, and ruthenium catalyzed oxidation provided the carboxylic acid **11**. The catalyst **12** was then synthesized *via* ligand exchange reaction with rhodium(II) acetate (Scheme 4).

The insertion reaction of **3a** in dichloromethane at room temperature in the presence of  $Rh_2[PCC]_4$  (**12**) proceeded to yield the product **6a** in higher enantiomeric excess (80% yield, 37% e.e.) (Table 1). Different solvent systems were tested aiming at more efficient asymmetric induction using the catalyst **12**, but the situation did not improve in dichloromethane–pentane (1 : 10, -78 °C), in fluorobenzene (-40 °C), and in pentane (r.t.). The insertion reaction in pentane at -45 °C was found to yield the product **6a** in 71% e.e. Further lowering the temperature was not practical as the



**Scheme 3** Insertion reaction of the substrates. *Reagents and conditions*: 1) NaH, THF; ClCH<sub>2</sub>CO<sub>2</sub>Na, HMPA, reflux; 2) (COCl)<sub>2</sub>, benzene; 3) CH<sub>2</sub>N<sub>2</sub>, ether; 4) see Table 1; 5) H<sub>2</sub>, Pd/C (this step is omitted for **4a** and **4d**. R' = R in **5a–8a** and **5d–8d**, and R' =  $n-C_{17}H_{35}$  in **5b–8b** and **5c–8c**); 6) mCPBA, DCM; 7) *p*-TsOH, MeOH; 8) (*S*)-(*O*)-acetylmandeloyl chloride, pyridine, DCM.

a R=n-C7H15

Table 1 Results of insertion reactions

Substrate	Catalyst (mol%)	Solvent <sup>a</sup>	Temp.	Yield (%) <sup>b</sup>	e.e. (%)
3a	Rh <sub>2</sub> [S-DOSP] <sub>4</sub> (2)	DCM	r.t.	28	13( <i>R</i> )
	$Rh_2[S-TBSP]_4(2)$	DCM	r.t.	31	14( <i>R</i> )
	$Rh_{2}[5S-MEPY]_{4}(2)$	DCM	reflux	84	8( <i>R</i> )
	$Rh_{2}[PCC]_{4}(12)(2)$	DCM	r.t.	80	37( <i>R</i> )
	(1)	DCM-pentane (1:10)	−78 °C	87	36( <i>R</i> )
	(1)	PhF	−40 °C	65	30( <i>R</i> )
	(2)	pentane	r.t.	79	33( <i>R</i> )
	(2)	pentane	−45 °C	72	71( <i>R</i> )
3b	$Rh_2[S-DOSP]_4(1)$	DCM	0 °C	31	23( <i>R</i> )
	$Rh_2[S-PTPA]_4(1)$	DCM	0 °C	67	16( <i>R</i> )
	$Rh_{2}[PCC]_{2}(12)(1)$	pentane	0 °C	60	47( <i>R</i> )
	(1)	pentane	−78 °C	62	83( <i>R</i> )
3c	(1)	pentane	−78 °C	71	80( <i>R</i> )
3d	(1)	pentane	−78 °C	52	67( <i>R</i> )

<sup>*a*</sup> Slow addition of the substrate *via* syringe pump for 4–6 h into the solution containing the catalyst (1 mol% at 0.2–0.5 mM or 2 mol% at 0.8–0.9 mM). <sup>*b*</sup> Three-step yield of **6a** from **3a** and **6d** from **3d**. Yield of **4b** from **3b** and **4c** from **3c**.



Scheme 4 Synthesis of  $Rh_2(PCC)_4$ . Reagents and conditions: 1) PCC, 4 Å MS, DCM; 2)  $Cp_2TiMe_2$ , THF, reflux; 3)  $BH_3$ ·THF;  $H_2O_2$ , NaOH; 4) PCC, 4 Å MS, DCM; 5) 2 M NaOH, THF, reflux; 6) 5 mol% RuO<sub>2</sub>, NaIO<sub>4</sub>, MeCN–CCl<sub>4</sub>–H<sub>2</sub>O (2 : 2 : 3); 7)  $Rh_2(OAc)_4$ , PhCl, reflux, 48 h, Soxhlet (Na<sub>2</sub>CO<sub>3</sub> trap).

substrate diazoketone **3a** crystallized out in pentane at lower temperature.

A new  $\alpha$ -diazoketone substrate **3b** was prepared from oleyl alcohol (**1b**), and the C–H insertion product **4b** was analyzed after conversion to **6b** *via* hydrogenation, mCPBA oxidation, and methanolysis. The result obtained from the insertion reaction in the presence of Rh<sub>2</sub>[*S*-DOSP]<sub>4</sub> in dichloromethane at 0 °C was comparable to the result obtained for the reaction of **3a**. Insertion reaction in the presence of Rh<sub>2</sub>[*S*-PTPA]<sub>4</sub> in dichloromethane at 0 °C was comparable to the result obtained for the reaction of **3a**. Insertion reaction in the presence of Rh<sub>2</sub>[*S*-PTPA]<sub>4</sub> in dichloromethane at 0 °C gave a higher yield of **4b**, but the enantiomeric excess did not improve. The reaction of **3b** in the presence of Rh<sub>2</sub>[PCC]<sub>4</sub> (**12**) in pentane at 0 °C proceeded to yield **4b** in 47% e.e., and eventually at -78 °C in 83% e.e.

In the presence of the catalyst **12** in pentane at -78 °C, the furanone **4c** was obtained from the  $\alpha$ -diazoketone substrate **3c** (prepared from linoleyl alcohol (**1c**)) in 80% e.e. The C–H insertion reaction of the substrate **3d** prepared from cyclohexanemethanol (**1d**) proceeded in 67% e.e.

The present studies show that  $Rh_2(PCC)_4$  is an efficient chiral catalyst in asymmetric C–H insertion reaction of  $\alpha'$ -alkoxy- $\alpha$ -diazoketones. Due to its lipophilic character, we were able to carry out insertion reactions in non-polar solvents at low temperature. Further applications of this catalyst in related carbenoid reactions will be the subject of future communications.

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