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### NOVEL REDUCTION OF ARYLSULFONYL CHLORIDES TO DISULFIDES WITH Sm/NiCl<sub>2</sub>/KI SYSTEM

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Abstract: Arylsulfonyl chlorides can be readily reduced to corresponding disulfides with  $Sm/NiCl_2/KI$  system in moderate to good yields at 60 °C.

Since arylsulfonyl chlorides are easily prepared by the chlorosulfonation of aromatic compounds with chlorosulfonic acid<sup>(1)</sup>, their conversion to other organic sulfur compounds with sulfur in the lower oxidation states is synthetically useful. Among these, organic disulfides are a class of useful synthetic intermediates because of its use in a variety of chemical transformations<sup>(2,3)</sup>. Many reagents have been used to reduce the sulfonyl halides to the corresponding disulfides, such as sodium tellurated borohydride NaBH<sub>2</sub>Te<sub>3</sub><sup>(4)</sup>, piperidinium tetrathiotungstate ( $NH_2$ )<sub>2</sub>WS4<sup>(5)</sup>, sodium cyanoborohydride NaBH<sub>3</sub>CN<sup>(6)</sup>, etc. Herein we wish to report a new method for the reduction of arylsulfonyl chlorides using Sm/NiCl<sub>2</sub>/KI system, we found that arylsulfonyl

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chlorides can be readily reduced to diaryldisulfides in moderate to good yields in HMPA at 60 °C. Some results were summarized in the Table.

$$ArSO_2Cl \xrightarrow{Sm/NiCl_2/KI}{60^{\circ}C, 3h} ArSSAr$$

# Table Reduction of arylsulfonyl chlorides to disulfides with Sm/NiCl<sub>2</sub>/KI system

Entry	Substrate	Product •	Yield**%	m. p. 'C (lit) <sup>[7]</sup>
1	⟨SO₂Cl	(	70	58-60(61)*
2	CH <sub>3</sub>	$( \bigcirc S )_{2} $ CH <sub>3</sub>	60	35—37(38) <sup>b</sup>
3	H <sub>3</sub> C-	( H <sub>3</sub> CS ) <sub>2</sub>	62	43-45(46)°
4	ClSO2Cl	( Cl	74	68-70(71) <sup>d</sup>
5	BrSO2Cl	( Br	68	89-91(93.5)°
6	Cl-SO <sub>2</sub> Cl CH <sub>3</sub>	( Cl-S) <sub>2</sub> CH <sub>3</sub>	74	50-52(48-50)[8]
7	H <sub>3</sub> C-CH <sub>3</sub> H <sub>3</sub> C-CH <sub>3</sub>	$(H_3C - S)_2$	68	122-125(125)
8	H <sub>3</sub> CO-CI-SO <sub>2</sub> CI	( H <sub>3</sub> CO	63	41-43(44) <sup>8</sup>

\* All products gave satisfactory IR and <sup>1</sup>H-NMR spectra

\* \* Isolated yields

In view of the easily available starting materials, the chemoselectivity, good yield, neutral condition as well as the simple operation, we think that the present procedrue provides a useful method for the conversion of arylsulfonyl chlorides to disulfides.

### **Experimental Section**

Melting points were uncorrected. HMPA was dried by  $CaH_2$  and was then distilled in vacuo. <sup>1</sup>H NMR spectra were recorded with a PMX-60 spectrometer, using TMS as internal standard. IR spectra were determined on PE-683 spectrometer. Benzenesulfonyl, 2-toluenesulfonyl and 4-toluenesulfonyl chlorides are commercially available, other sulfonyl chlorides were prepared according to literature<sup>[1,8]</sup>.

### General procedure

Under an inert atmosphere of nitrogen, to a mixture of samarium (2 mmol), NiCl<sub>2</sub>(0.8 mmol) and KI (4 mmol) was added a solution of arylsulfonyl chloride (1 mmol) in 4 ml HMPA. The slurry was stirred magnetically for 3 h at 60 °C and cooled to room temperature. 20 ml ether was added. organic layer was separated and aqueous layer was extracted with ether (2×20 ml). The combined organic layers were washed with water (30 ml × 3). After the solution was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, the solution was removed under reduced pressure. The residue was then purified by preparative TLC on silica gel (cyclohexane as eluent).

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