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Convergent enantioselective syntheses of two potential C25–C40 subunits of (–)-caylobolide A

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ABSTRACT

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Marine sources continually provide the synthetic community a variety of structurally challenging natural products.¹ One aspect that makes the total synthesis of natural products so challenging is the uncertain configuration of multiple stereogenic centers within a target compound. It is well known that diastereomeric possibilities increase exponentially by a factor of 2^n , where *n* is equal to the number of chiral centers. Thus, the greater number of undefined stereogenic centers can rapidly complicate a synthetic approach. One such natural product, termed caylobolide A (1), was isolated via bioassay-guided purification in 2002 from the marine cyanobacteria Lyngbya majuscule collected at Cay Lobos, Bahamas by Molinski and MacMillan.² As shown in Scheme 1, caylobolide A contains eight undefined stereocenters, thus 256 diastereomeric structural possibilities. Another interesting feature of 1 is the repeating 1,5-diol motif present along the 36-membered lactone core. In addition to its intriguing macrocyclic structure, caylobolide A has shown cytotoxic properties against the human colon tumor cell line HCT 116 (IC₅₀ = $9.9 \,\mu$ M). Based on the limited biological data and very unique and challenging structural features of caylobolide A, we sought to undertake the synthesis of 1. Herein, we disclose our synthetic approach to the C25-C40 subunit of 1.

The retrosynthetic strategy of caylobolide centered on a highly convergent approach is highlighted in Scheme 1. Thus, we envisaged esterification at C35 (either standard or macrocyclic) and olefin metathesis (either cross or ring-closing) process at C23–C24 to forge the 36-membered ring of **1**. We initially decided to focus on constructing the C25–C40 segment of **1** due to the defined stereocenters (relative configuration via the Kishi universal NMR database and absolute configuration by Mosher analysis) at C25, C27, C29, and C33.^{3,4}

The convergent syntheses of two possible diastereomers of the C25-C40 subunit resident in (-)-caylobo-

lide A have been accomplished. The key reaction featured a chemoselective Ru-catalyzed cross-metath-

esis between a fully elaborated type I and two functionalized type II α,β -unsaturated ketones.

With this idea in mind, the two potential C25–C40 segments **2** and **3** would be synthesized via a cross-metathesis of homoallylic alcohol **4** and the two α , β -unsaturated ketones **5** and **6**. We arbitrarily chose the *S*-configuration at C36 for compounds **5** and **6** to simply illustrate the synthetic approach to **2** and **3**.

With the initial synthetic plan in hand, our focus was turned to completing the homoallylic diol **4**. As shown in Scheme 2, treatment of the known TBS protected aldehyde **7**⁵ with (+)-Ipc₂Ballyl under the standard reaction conditions as pioneered by Brown furnished the requisite homoallylic alcohol.⁶ An ensuing protection of the free hydroxyl group with BnBr, NaH, and Bu₄NI afforded benzyl ether **8** in 69% yield over two steps from **7**. Oxidative cleavage with O₃ and PPh₃ mediated reductive quench of the olefin moiety resident in **8** was readily accomplished and provided the aldehyde **9** in 60% yield. With the required aldehyde in hand, we envisioned a chelation-controlled allylation with an appropriate stannane utilizing the benzyl directing group and TiCl₄.⁷ Much to our delight, treatment of **9** with TiCl₄ and Bu₃Snallyl at $-78 \,^{\circ}$ C furnished the desired *anti*-homoallylic alcohol with excellent diastereoselectivity (~12:1 dr) as deduced by ¹H NMR.

An ensuing protection of the free hydroxyl moiety as a TBS ether was readily accomplished under standard reaction conditions (TBSCl and imidazole) to afford the protected diol **4** in 75% yield over two steps from $\mathbf{9.}^8$





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Scheme 1. Retrosynthetic analysis of caylobolide (1).



Scheme 2. Synthesis of intermediate **4.** Reagents and conditions: (a) (+)-Ipc₂BOMe (1.6 equiv), allyIMgBr (1.5 equiv) Et₂O, 0 °C to -78 °C to rt, 6 h, 83%; (b) BnBr (1.1 equiv), NaH (2.0 equiv), Bu₄NI (0.1 equiv), DMF, rt, 24 h, 83%; (c) CH₂Cl₂, -78 °C, then PPH₃ (4.0 equiv), rt, 6 h, 60%; (d) allyISnBu₃ (2.0 equiv), TiCl₄ (1.0 equiv), CH₂Cl₂, -78 °C, 8 h, 81%; (e) TBSCI (2.0 equiv), imidazole (4.0 equiv), DMAP (0.2 equiv), DMF, rt, 24 h, 92%.



Scheme 3. Synthesis of intermediate **13.** Reagents and conditions: (a) NaHMDS (1.2 equiv), Mel (2.5 equiv), THF, -78 to -20 °C, 2 h, 81%; (b) LiBH₄ (3.0 equiv), MeOH (3.0 equiv), Et₂O, rt, 6 h, 93%; (c) TEMPO (0.15 equiv), PhI(OAc)₂ (1.1 equiv), CH₂Cl₂, rt, 20 h, 82%.

Our outline to the C25–C40 segment of **1** required the synthesis of chiral aldehyde **13** as delineated in Scheme 3. Hence, an asymmetric alkylation of the known chiral oxazolidinone **10** with NaH-MDS and MeI provided **11** in 81% yield with excellent diastereoselectivity (>15:1 dr).⁹ Subsequent reduction of the oxazolidinone moiety with LiBH₄ readily afforded the chiral primary alcohol **12** in 93% yield. Final oxidation of the hydroxyl group of **12** with TEMPO and PhI(OAc)₂ furnished the desired aldehyde **13** in 82% yield and set the stage for building the subunits **5** and **6**.¹⁰

With the chiral aldehyde 13 readily in hand and in multi-gram quantities, we next focused our effort on the completion of the cross-metathesis precursor ketones 5 and 6 as delineated in Schemes 4 and 5. Thus, treatment of the chiral aldehyde 13 with (+)-Ipc₂Ballyl provided the corresponding homoallylic alcohol with a respectable 66% yield and dr of \sim 10:1. Ensuing protection of the free hydroxyl group with TESCI, DMAP, and imidazole furnished the triethylsilyl ether 14 in 73% yield. Ozonolysis of the terminal olefin followed by the addition of PPh₃ resulted in the formation of the requisite aldehyde 15 in 63% yield. In order to complete the cross-metathesis coupling type II olefin 5, a vinyl addition and subsequent oxidation to furnish the α,β -unsaturated ketone 5 were required. Thus, the addition of vinyl Grignard reagent to 15 provided the allylic alcohol as an extraneous mixture of diastereomers and ensuing oxidation of the hydroxyl moiety with Dess-Martin periodinane (DMP) afforded ketone 5 in 38% yield over two steps from 15.¹¹ Access to ketone 6 followed a very similar synthetic pathway, but differed only in utilizing the (-)-Ipc₂Ballyl reagent as shown in Scheme 5. All of the other steps (b-e) were



Scheme 4. Synthesis of intermediate **5.** Reagents and conditions: (a) (+)-Ipc₂BOMe (1.6 equiv), allyIMgBr (1.5 equiv), Et₂O, -78 °C to rt, 3 h 66%; (b) TESCI (1.5 equiv), imidazole (3.0 equiv), DMAP (0.1 equiv), CH₂Cl₂, rt, 12 h, 73%; (c) CH₂Cl₂ -78 °C, 1 h, then PPh₃ (4.0 equiv), rt, 6 h, 63%; (d) vinyIMgBr, (2.0 equiv), Et₂O, -78 °C, 4 h, 60%; (e) DMP (2.0 equiv), pyridine (5.0 equiv), CH₂Cl₂, rt, 15 h, 63%.



Scheme 5. Synthesis of intermediate **6.** Reagents and conditions: (a) (-)-lpc₂BOMe (1.6 equiv), allylMgBr (1.5 equiv), Et₂O, -78 °C to rt, 3 h 86%; (b) TESCI (1.5 equiv), imidazole (3.0 equiv), DMAP (0.1 equiv), CH₂Cl₂, rt, 12 h, 75%; (c) CH₂Cl₂, -78 °C, 1 h, then PPh₃ (4.0 equiv), rt, 6 h, 60%; (d) vinylMgBr, (2.0 equiv), Et₂O, -78 °C, 4 h, 75%; (e) DMP (2.0 equiv), pyridine (5.0 equiv), CH₂Cl₂, rt, 15 h, 70%.



Scheme 6. Synthesis of intermediate **2.** Reagents and conditions: (a) **5** (2.0 equiv), **18** (0.10 equiv), CH₂Cl₂, 40 °C, 24 h, 75%; (b) $[(PhP_3)CuH]_6$ (0.10 equiv) PhSiH₃ (1.5 equiv), toluene, rt, 24 h, 70%; (c) PPTS (0.2 equiv), MeOH, CH₂Cl₂ rt, 0.5 h, 80%; (d) Et₂BOMe (1.1 equiv), NaBH₄ (1.1 equiv), THF, MeOH, -78 °C, 3 h, 90%; (e) 2,2-DMP (2.0 equiv), CSA (0.10 equiv), CH₂Cl₂, rt, 5 h, 90%.

carried out in an analogous fashion and the isolated yields were quite comparable.

With all the three precursors in hand (type I (**4**) and type II (**5** and **6**) olefins), we proceeded to converge the aliphatic subunits via a Ru-catalyzed cross-metathesis reaction and complete the C25–C40 segment of **1** as delineated in Schemes 6 and 7.

Inspired by Grubbs' report on selective cross-metathesis, we were hopeful that the treatment of **4** (type I) with an excess of **5** (type II) in the presence of Grubbs' second generation catalyst (**18**) would chemoselectively couple the two subunits.^{12,13} In this specific example, the stereochemistry of the corresponding olefin product would be inconsequential, as it would be reduced at a later stage. Hence, the cross-metathesis of **4** with 2.0 equiv of **5** and 10 mol % of **18** readily proceeded to provide **19** in 75% yield with high levels of selectivity for the *E*-isomer. Subsequent conjugate reduction of the α , β -unsaturated carbonyl by means of Stryker's reagent and PhSiH₃ afforded the fully saturated compound **20** in 70% yield.¹⁴ A chemoselective desilylation of the TES ether resident in **20** by utilizing PPTS in MeOH furnished β -hydroxy ketone **21** and set the stage for the introduction of the final chiral center in the C25–C40 fragment of **1**.

Thus, a directed *syn*-reduction was readily accomplished upon treatment of ketone **21** with the obligatory reagents (Et₂BOMe and NaBH₄) as described by Prasad and afforded diol **22** in 90% with high levels of diastereoselectivity (>20:1 by ¹H NMR).¹⁵ Final protection of the diol moiety resident in **22** utilizing 2,2-dimethoxy propane and a catalytic amount of CSA (10 mol %) furnished acetonide **2** in 90% yield and completed one of the C25–C40 diastereomers of **1**.¹⁶



Scheme 7. Synthesis of intermediate **3.** Reagents and conditions: (a) **6** (2.0 equiv), **18** (0.10 equiv), CH_2Cl_2 , 40 °C, 24 h, 77%; (b) [(PhP₃)CuH]₆ (0.10 equiv), PhSiH₃ (1.5 equiv), toluene, rt, 24 h, 69%; (c) PPTS (0.20 equiv), MeOH, CH_2Cl_2 , rt, 0.5 h, 80%; (d) Me₄NBH(OAc)₃ (6.0 equiv), MeCN, AcOH, -20 °C, 48 h, 70%; (e) 2,2-DMP (2.0 equiv), CSA (0.10 equiv), CH_2Cl_2 , rt, 5 h, 90%.

Similar to that of ketone **6** in Scheme 5, access to the other C25–C40 diastereomer **3** followed a very similar synthetic pathway to that of **2**.¹⁷ It did, however, differ by utilizing the diastereomeric coupling partner **6** (instead of **5**) and a subsequent Me₄B(OAc)₃H directed *anti*-reduction of the β -hydroxy ketone **25** as shown in Scheme 7. The remaining steps were carried out in a similar manner and the isolated yields were quite comparable.

In conclusion, we have completed the convergent syntheses of two possible diastereomers of the C25–C40 subunit resident in (–)-caylobolide A. One of the key reactions utilized a β -hydroxy directed *anti*-stannane allylation of aldehyde **9** to provide the necessary stereochemistry for the type I olefin coupling partner **4**. In addition, a chemoselective cross-metathesis of **4** with the two diastereomeric α , β -unsaturated ketones **5** and **6** in the presence of catalyst **18** allowed for the completion of the two C25–C40 subunits **2** and **3**. While the remaining stereochemistry of **1** is currently unknown, other natural products suggest that the stereochemistry of the 1,5-diol subunit might be *syn* based on biosynthetic polyketide assembly.¹⁸ However, it is worth noting that the 1,5-diol subunit has been shown to be *anti* in some natural products, that is, marinisporolide A.¹⁸ Studies toward the total synthesis of **1** are ongoing and will be reported in due course.

Acknowledgments

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References and notes

- 1. Bugni, T. S.; Ireland, C. M. Nat. Prod. Rep. 2004, 21, 143.
- 2. MacMillan, J. B.; Molinski, T. F. Org. Lett. 2002, 4, 1535.
- (a) Lee, J.; Kobayashi, Y.; Tezuka, K.; Kishi, Y. Org. Lett. **1999**, *1*, 2181; (b) Kobayashi, Y.; Tan, C.-H.; Kishi, Y. Helv. Chim. Acta **2000**, *83*, 2562.
 (a) Ohtani L. Kusumi T. Kashman, Y. Kakisawa H. J. Am. Chem. Soc. **1991**, *113*.
- (a) Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. J. Am. Chem. Soc. 1991, 113, 4092; Seco, J. M.; Quiñoa, E.; Riguera, R. Tetrahedron: Asymmetry 2000, 2781, 11; (c) Seco, J. M.; Martino, M.; Quiñoa, E.; Riguera, R. Org. Lett. 2000, 2, 3261.
 Miles, W. H.; Connell, K. B.; Ulas, G.; Tuson, H. H.; Dethoff, E. A.; Mehta, V.;
- Mines, W. H.; Connell, K. B.; Olas, G.; Iuson, H. H.; Detholl, E. A.; Melita, V.; Thrall, A. J. J. Org. Chem. 2010, 75, 6820.
 (a) Racherla U. S. Brown, H. C. L. Org. Chem. 1991, 56, 401: (b) ladbay, P. K.
- (a) Racherla, U. S.; Brown, H. C. J. Org. Chem. **1991**, 56, 401; (b) Jadhav, P. K.; Bhat, K. S.; Perumal, P. T.; Brown, H. C. J. Org. Chem. **1986**, 51, 432.
 (a) Reetz, M. T. Angew. Chem., Int. Ed. Engl. **1984**, 23, 556; (b) Reetz, M. T. Acc.
- (a) Reetz, M. 1. Angew. Chem., Int. Ed. Engl. 1984, 23, 556; (b) Reetz, M. 1. Acc. Chem. Res. 1993, 26, 462.
- 8. Data for 4: ¹H NMR (360 MHz, CDCl₃) δ 7.4 (m, 4H), 7.3 (m, 1H), 5.8 (m, 1H), 5.1 (m, 2H), 4.6 (d, *J* = 11.4 Hz, 1H), 4.5 (d, *J* = 11.4 Hz, 1H), 4.0 (m, 1H), 3.7 (m, 3H), 2.3 (m, 2H), 1.8 (m, 3H), 1.6 (m, 1H), 0.9 (m, 18H), 0.08 (m, 12H). ¹³C NMR (125 MHz, CDCl₃) δ 139.1, 134.7, 128.2, 127.4, 127.2, 116.8, 73.5, 70.6, 68.9, 59.5, 53.3, 42.5, 42.4, 37.7, 34.1, 25.9, 22.3, 18.2, 18.0, 13.9, -4.1, -4.5, -5.4. IR (CH₂Cl₂): 3072, 3031, 2949, 2927, 2857, 1468, 1431, 1383, 1353, 1253, 1209, 1093, 1004, 934, 912, 837, 804, 774 cm⁻¹. $R_{\rm f}$ = 0.2, 3% EtOAc in hexane. [α]² + 3.8° (c 0.034 g/mL, CH₂Cl₂). HRMS (EI) calculated for C₂₃H₄₁O₃Si₂ (M-C₄H₉): 421.2610 found: 421.2594.
- (a) Evans, D. A.; Ennis, M. D.; Mathre, D. J. J. Am. Chem. Soc. 1982, 104, 1737; (b) Decicco, C. P.; Grover, P. J. Org. Chem. 1996, 61, 3534.
- De Mico, A.; Margarita, R.; Parlanti, L.; Vescovi, A.; Piancatelli, G. J. Org. Chem. 1997, 62, 6974.
- 11. Dess, D. B.; Martin, J. C. J. Org. Chem. 1983, 48, 4155.

- For a very recent review, see: (a) Vougioukalakis, G. C.; Grubbs, R. H. Chem. Rev. 2010, 110, 1746; (b) Chatterjee, A. K.; Choi, T. L.; Sanders, D. P.; Grubbs, R. H. J. Am. Chem. Soc. 2003, 125, 11360.
- 13. Scholl, M.; Ding, S.; Lee, C. W.; Grubbs, R. H. Org. Lett. 1999, 1, 953.
- (a) Brestensky, D. M.; Huseland, D. E.; McGettigan, C.; Stryker, J. M. Tetrahedron Lett. **1988**, 29, 3749; (b) Mahoney, W. S.; Brestensky, D. M.; Stryker, J. M. J. Am. Chem. Soc. **1988**, 110, 291; (c) Brestensky, D. M.; Stryker, J. M. Tetrahedron Lett. **1989**, 30, 5677; (d) Koenig, T. M.; Daeuble, J. F.; Brestensky, D. M.; Stryker, J. M. Tetrahedron Lett. **1990**, 31, 3237.
- Chen, K.-M.; Hardtmann, G. E.; Prasad, K.; Repic, O.; Shapiro, M. J. *Tetrahedron Lett.* **1987**, *28*, 155.
- 16. Data for **2**: ¹H NMR (360 MHz, CDCl₃) δ 7.5 (m, 1H), 7.3 (m, 4H), 4.6 (d, J = 11.4 Hz, 1H), 4.5 (d, J = 11.4 Hz, 1H), 3.9 (m, 1H), 3.7 (m, 4H), 3.5 (m, 1H), 1.7 (m, 3H), 1.5 (s, 6H), 1.4 (m, 5H), 1.3 (m, 7H), 1.2 (m, 5H), 1.0 (m, 2H), 0.89 (m, 18H), 0.83 (d, J = 6.6 Hz, 3H), 0.05 (m, 12H). ¹³C NMR (125 MHz, CDCl₃) δ 139.1, 129.6, 128.3, 127.5, 98.2, 73.7, 72.7, 70.8, 69.5, 69.0, 59.6, 42.7, 37.8, 37.7, 36.9, 33.1, 32.0, 30.2, 29.1, 25.9, 25.8, 23.0, 20.2, 19.8, 18.2, 18.1, 18.0, 14.3, 14.1, -4.0, -4.3, -5.3. IR (CH₂Cl₂): 3051, 2984, 2852, 2252, 910, 729, 649 cm⁻¹. $R_{\rm f} = 0.84$ in 10% EtOAc in hexanes. $|z|_{2}^{2} 0.10$ (c 0.084 g/mL, CH₂Cl₂). HRMS (EI) calculated for C₃₅H₆₅O₅Si₂ (M-C₄H₉): 621.4375 found: 621.4371
- 17. Data for **3**: ¹H NMR (360 MHz, CDCl₃) δ 7.5 (m, 1H), 7.3 (m, 4H), 4.6 (d, J = 11.4 Hz, 1H), 4.5 (d, J = 11.4 Hz, 1H), 3.5 (m, 2H), 1.7 (m, 3H), 1.4 (m, 8H), 1.3 (s, 6H), 1.2 (m, 5H), 1.1 (m, 3H), 0.97 (m, 2H), 0.88 (m, 21H), 0.04 (m, 12H). ¹³C NMR (125 MHz, CDCl³) δ 139.0, 129.5, 128.3, 127.5, 100.8, 73.7, 70.8, 70.4, 69.5, 66.8, 59.6, 42.7, 37.7, 37.8, 37.9, 36.7, 36.3, 31.5, 30.8, 29.18, 25.9, 24.7, 24.4, 22.9, 20.8, 18.1, 15.2, 14.0, -4.0, -4.3, -5.3. IR (CH₂Cl₂): 3054, 2986, 2956, 2934, 2857, 2305, 1418, 1267, 893, 733 cm⁻¹. $R_{\rm f}$ = 0.77 in 10% EtOAc in hexanes. [x]_D² 3.80 (c 0.012 g/mL, CH₂Cl₂) HRMS (EI) calculated for C₃₈H₇₁0, Si₂ (M-CH₃): 663.4832 found: 663.4840.
- 18. Friestad, G. K.; Sreenilayam, G. Pure Appl. Chem. 2011, 83, 461.