



1,7-Electrocyclisations of stabilised azomethine ylides

Miklós Nyerges,* Andrea Virányi, Áron Pintér and László Tőke

Research Group of the Hungarian Academy of Sciences, Department of Organic Chemical Technology,
Technical University of Budapest, H-1521 Budapest P.O.B. 91, Hungary

Received 3 September 2002; revised 6 November 2002; accepted 22 November 2002

Abstract—Ester-stabilised $\alpha,\beta,\gamma,\delta$ -unsaturated azomethine ylides **9** were generated by the deprotonation method from isoquinolinium salts **8**. 1,7-Electrocyclisation of these dipoles followed by a 1,5-hydrogen shift, gives tetrahydro[5,6]azepino[2,1-*a*]isoquinolines **10**. © 2003 Elsevier Science Ltd. All rights reserved.

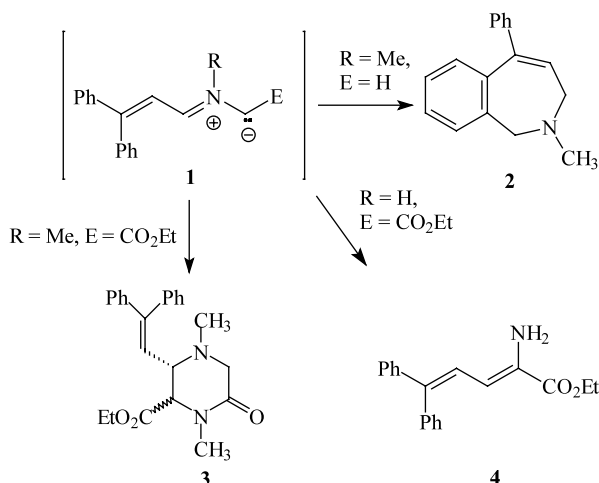
The 1,3-dipolar cycloaddition reaction is one of the most simple approaches for the construction of five membered heterocyclic rings.¹ There are, however, many other synthetically useful reactions of these dipoles, including the 1,5- and 1,7-electrocyclic^{2,3} ring closure of appropriately substituted dipolar systems. Recently, we⁴ and others⁵ published the first examples of the 1,7-electrocyclisation of non-stabilised azomethine ylides with $\alpha,\beta,\gamma,\delta$ -unsaturation.

During these and further studies we have found significant differences between the reactivity of $\alpha,\beta,\gamma,\delta$ -unsaturated, non-stabilised ylides **1** ($E=H$) and $\alpha,\beta,\gamma,\delta$ -

unsaturated, ester-stabilised azomethine ylides **1** ($E=CO_2Et$). The former dipoles react via a 1,7-electrocyclisation,⁶ followed by a [1,5]-hydrogen shift to give dihydrobenzazepines **2**, whilst the latter give other products (**3** or **4**) via novel rearrangements (Scheme 1).^{7,8} As a continuation of these studies we have now examined the reactivity of some $\alpha,\beta,\gamma,\delta$ -unsaturated, ester-stabilised azomethine ylides **9** generated by deprotonation of iminium salts **8**. The synthesis of precursors was carried out according to the Bischler–Napieralski procedure⁹ by cyclisation of amide **6** in the presence of $POCl_3$ that resulted in the formation of 3,4-dihydroisoquinolines **7**. Subsequent reaction with bromoalkyl derivatives in anhydrous ether gave the quaternary salts **8a–d** (Scheme 2).

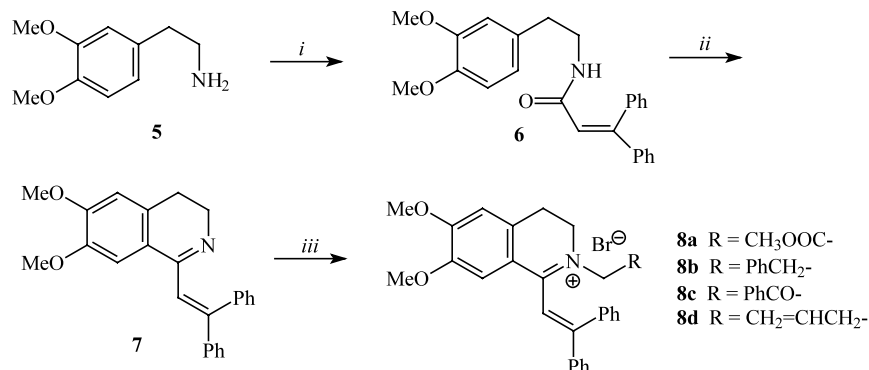
Reacting the isoquinolinium salts **8a–d** with triethylamine at ambient temperature, in dry ethanol leads to the formation of 8-substituted-2,3-dimethoxy-13-phenyl-5,6,8,14a-tetrahydro[5,6]azepino[2,1-*a*]isoquinolines **11a–d** via azomethine ylide intermediates **9a–d**. Compounds could be isolated by a simple filtration in moderate to good yields. In this reaction the azomethine ylides were produced by dehydrohalogenation of the isoquinolinium salts,¹⁰ which leads via a 1,7-electrocyclisation reaction to the azepine **10** and finally by a [1,5]-hydrogen shift to the products **11a–d**. The stereochemistry of tetracycles **11a–d** was deduced by NOE studies (Scheme 3).

In summary, we have demonstrated the first 1,7-electrocyclisation of electron-withdrawing group stabilised azomethine ylides (**11a** and **11c**), while the explored reaction sequence provides a useful route to the tetrahydro[5,6]azepino[2,1-*a*]isoquinoline ring system.

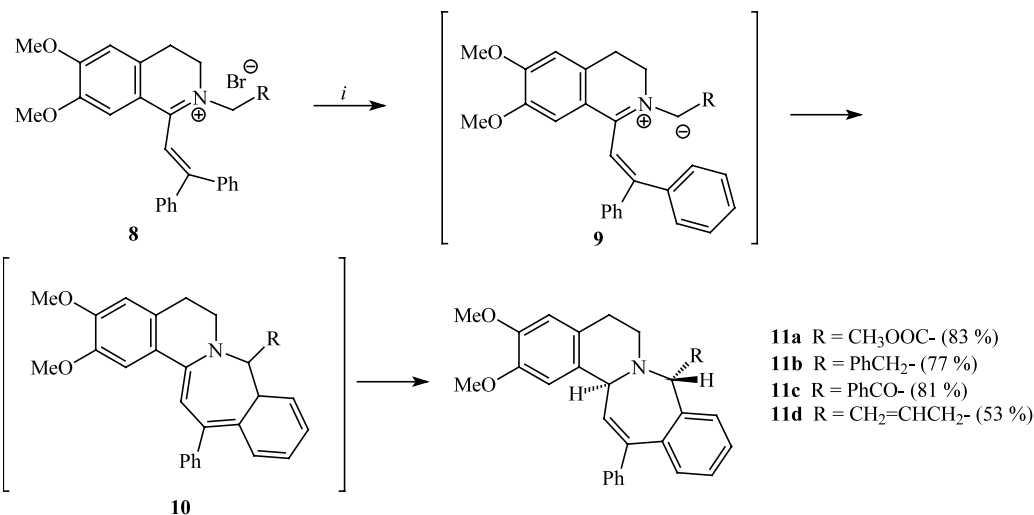


Scheme 1.

* Corresponding author. Fax: (+361)463-3648; e-mail: mnyerges@mail.bme.hu



Scheme 2. Reagents and conditions: (i) Ph₂C=CHCOCl, NaOH, Et₂O, H₂O, rt (88%); (ii) POCl₃, toluene, reflux (72%); (iii) RCH₂Br, Et₂O, rt (95–100%).



Scheme 3. Reagents and conditions: (i) Et₃N, EtOH, rt.

Acknowledgements

This work was financially supported by the National Found for Science and Research, Hungary (OTKA Project No. T 032221). N.M. thanks the Ministry of Education for a György Békésy fellowship.

References

1. Tsuge, O.; Kanemasa, S. In *Advances in Heterocyclic Chemistry*; Katritzky, A., Ed.; Academic Press, 1989; Vol. 45, pp. 232–349.
2. Huisgen, R. *Angew. Chem., Int. Ed. Engl.* **1980**, *19*, 947.
3. Groundwater, P. W.; Nyerges, M. *Adv. Heterocyclic Chem.* **1999**, *73*, 97.
4. Arany, A.; Groundwater, P. W.; Nyerges, M. *Tetrahedron Lett.* **1998**, *39*, 3267.
5. Marx, K.; Eberbach, W. *Tetrahedron* **1997**, *53*, 14687.
6. Arany, A.; Bendell, D.; Groundwater, P. W.; Garnett, I.; Nyerges, M. *J. Chem. Soc., Perkin Trans. 1* **1999**, 2605.
7. (a) Groundwater, P. W.; Sharif, T.; Arany, A.; Hibbs, D. E.; Hursthouse, M. B.; Nyerges, M. *Tetrahedron Lett.* **1998**, *38*, 1433; (b) Groundwater, P. W.; Sharif, T.; Arany, A.; Hibbs, D. E.; Hursthouse, M. B.; Garnett, I.; Nyerges, M. *J. Chem. Soc., Perkin Trans. 1* **1998**, 2837.
8. Nyerges, M.; Arany, A.; Fejes, I.; Groundwater, P. W.; Zhang, W.; Bendell, D.; Anderson, R. J.; Tőke, L. *Tetrahedron* **2002**, *58*, 845.
9. Cortes, É. C.; Romere, E. C.; Ramirez, G. F. *J. Heterocyclic Chem.* **1994**, *31*, 1425.
10. (a) Tischer, T.; Tőke, L.; Tóth, G. *Acta Chim. Hung.* **1990**, *127*, 171; (b) Janke, F.; Himmelreich, U.; Tóth, G.; Tischer, T.; Bende, Z.; Tőke, L. *J. Heterocyclic Chem.* **1991**, *28*, 867.