FLAVONDIDS, 40.⁺ SYNTHESIS OF 3-ALKYL- AND —ARYLTHIOFLAVANONES AND THEIR TRANSFORMATIONS INTO SULFUR-CONTAINING FLAVONOIDS[≠]

TAMÁS PATONAY*, ERZSÉBET PATONAY-PÉLI and GYÖRGY LITKEI

Department of Organic Chemistry, Kossuth Lajos University, H-4010 Debrecen 20, Hungary

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Abstract - trans-3-Mesyloxyflavanones 1 were converted into cis- and trans--3-(alkylthio)- and -(phenylthio)flavanones 2-4 by nucleophilic substitution reaction with thiols or thiolates. Flavanones 2-4 were useful intermediates in the synthesis of various sulfur-containing derivatives; flavanones, flavones and dihydrochalcones possessing alkyl- or arylthio, -sulfinyl and -sulfonyl group. Oxidation of cis- and trans- isomers of 4 led to completely different products.

In preceding parts of this series we have reported on reactions of 3 -alkyl- and -arylsulfonyloxyflavanones with various nucleophiles demonstrating their usefulness in the synthesis of 3-substituted-flavonoids¹⁻³. As a continuation of this work we now present our results on the reaction of 3-mesyloxyflavanones with thiols.

The reaction of <u>trans-3-mesyloxyflavanone</u> ($\underline{1}\underline{9}$) with aliphatic thiols (ethanethiol, propanethiol) in the presence of either phase-transfer catalyst (K₂ CO₃/Adogen-464/PhH; a modified <u>Lissel</u> procedure⁴) or excess of triethylamine (<u>Simons method</u>⁵) afforded the mixture of the corresponding <u>cis-</u> and <u>trans-3-</u>-(alkylthio)flavanones ($\underline{2}\underline{9}$, $\underline{3}\underline{9}$)⁶, and the minor product flavone ($\underline{5}\underline{9}$) formed in a concurrent elimination reaction¹⁻³. The former system provided better yield but it was found to be responsive to the reaction time. After the disappearence of $\underline{1}\underline{9}$ a secondary transformation of $\underline{2}\underline{9}$, $\underline{3}\underline{9}$ into 1-(2-hydroxyphenyl)-2-(alkylthio)-3-phenyl--1-propanones [2'-hydroxy- α -(alkylthio)dihydrochalcones] ($\underline{6}\underline{9}$, $\underline{7}\underline{9}$) was observed. Dihydrochalcones <u>69</u> and <u>79</u> were prepared as sole products from <u>19</u> using longer reaction period and a large excess of thiol.

These products are presumed to form by a base-catalyzed ring-opening of $\underline{2}\underline{a}, \underline{3}\underline{a}$ into 2'-hydroxy- α -(alkylthio)chalcones (A) and the reduction of the intermediates A with the excess thiol. Similar secondary transformation has also been observed in the syntesis of 3-(acetylthio)flavanones¹.

Treatment of <u>trans</u>-3-mesyloxy-4'-R¹-flavanones ($\underline{1}\underline{a}-\underline{e}$) with benzenethiol under phase-transfer conditions furnished the expected <u>cis</u>- and <u>trans</u>-3-(phenylthio)-4'--R¹-flavanones ($\underline{4}\underline{a}-\underline{e}$) but better yield was achieved in benzene solution using thiolate anion as the nucleophile. Application of more polar solvent resulted in lower yield (Table 1).

⁺Part 39. see ref. 1.

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By-products were 4'-R¹-flavones ($\underline{5}\underline{a}-\underline{e}$), 1-(2-hydroxyphenyl)-2-(phenylthio)-3---(4-R¹-phenyl)-1-propanones ($\underline{8}\underline{c}-\underline{e}$), 2'-hydroxy-4-R¹-chalcones ($\underline{9}\underline{a}-\underline{e}$), 3-hydroxy-4'--R¹-flavanones ($\underline{1}\underline{0}\underline{a}-\underline{e}$) and diphenyl disulfide ($\underline{1}\underline{1}$)(Table 1).



Table 1. Product-ratio data of the reaction of trans-3-mesyloxyflavanones with PhSX

Starting	Method ^a	Reaction time	Yield (%)						
mat.			1 ^b	4	5	<u>8</u>	2	10	Ph2S2 (11)
<u>l</u> g	Α	46 h	ø	84.7	tr	ø	6.8	tr	38.4
	С	26 h	ø	90.6	tr	Ø	< 1.0	ø	2.9
	Cl	lh	10.2	72.0	tr	Ø	4.3	5.0	8.9
	C_2	72 h	30.3	44.6	12.0	Ø	3.9	Ø	16.0
	ເງັ	35 min	1.0	77.5	4.3	Ø	17.5	Ø	11.8
	C,	30 min	1.5	73.3	tr	ø	17.6	Ø	17.7
	C ₅	20 min	Ø	30.5	tr	ø	46.9	ø	64.2
Ĩ₽	A	90 h	ø	62.6	3.3	Ø	5.6	5.7	43.4
	С	30 h	5.0	77.1	2.1	ø	пі	Ø	4.6
Ì⊊	А	90 h	Ø	53.5	ø	ni	14.2	7.8	39.1
	С	7.5 h	6.9	84.1	3.2	Ø	2.2	2.1	5.7
ī₫	А	25.5 h	tr	73.6	ø	8.1	6.5	ni	34.6
	С	2.5 h	6.1	85.4	2.0	Ø	1.9	1.3	6.9
<u>l</u> e	А	21.5 h	Ø	32.2	9.0	18.45	4.1	11.2	50.5
	С	36.h	11.7	65.4	4.5	Ø	2.85	Ø	5.5

^aDetails are given in Experimental Section

^bUnreacted starting material

ni: not isolated product

tr: traces

Chalcones $\frac{9}{2}$ are supposed to form from dihydrochalcones $\frac{9}{2}$ in a reaction involving elimination of benzenethiol. This pathway is supported by the co--occurence of $\frac{9}{2}$ and $\frac{9}{2}$ when R¹ is an electron-attracting substituent (see Table 1) as well as by the fact that treatment of either $\frac{4}{2}$ or $\frac{6}{2}$ with bases gave $\frac{9}{2}$ (or an equilibrium mixture of $\frac{9}{2}$ and its cyclic isomer flavanone ($\frac{1}{2}$)).

3-(Ethylthio) flavone ($\underline{1}\underline{3}$) and 3-(phenylthio)flavone ($\underline{1}\underline{4}$) were prepared from $\underline{2}\underline{a}$ and $\underline{4}\underline{a}$ (<u>cis-trans</u> mixture), respectively, by means of a 2,3-dichloro-5,6-dicy-ano-1,4-benzoquinone (DDQ) dehydrogenation process^{1,3,7}. Oxidation of $\underline{1}\underline{4}$ with hydrogen peroxide in acetic acid gave 3-(phenylsulfonyl)flavone ($\underline{1}\underline{5}$). These transformations have practical importance providing new route to flavones claimed



<u>16 a - d</u>

T. PATONAY et al.

to posses biological activity. Earlier 3-(methylthio)flavones have been reported as CNS-depressants^{8,9} whereas for 6-methyl-3-(arylsulfonyl)-2-aryl- or -heteroarylchromones, synthesized from 2'-hydroxy-5'-methyl-2-(arylsulfonyl)acetophenones via α -(arylsulfonyl)chalcones, antiallergic effect¹⁰ was described.

Oxidation of cis-4a-d with m-chloroperbenzoic acid (MCPBA) (2 eq) resulted in cis-3-(phenylsulfonyl)-4'- R^{\perp} -flavanones (<u>l6a</u>-d) with excellent yields. From \underline{cis} - $\underline{4a}$ with MCPBA (1 eq) \underline{cis} -3-(phenylsulfinyl)flavanone ($\underline{17}$) was synthesized under controlled conditions¹¹. At the same time, similar reaction of $\frac{1}{1}$ led exclusively to 5g and no traces of any sulfine or sulfone could be detected in the mother liquor.

Treatment of a cis-trans mixture of 2g or 4g with hydrogen peroxide afforded only 18 and 16a, respectively, besides 5a. Spontaneous transformation of the primary product trans-3-(ethyl- or phenylsulfinyl)flavanone into 5g can be rationalized in terms of the cis-elimination of sulfenic acid being activated by the presence of the aryl group $\overline{12}, 13$.

Umino et al.¹⁴ patented the formation of a small amount of trans-3-(methylsulfinyl)flavanones besides the <u>cis</u> isomers by condensing 2'-hydroxy-2-(methylsulfinyl)acetophenone with araldehydes, whereas Strandtmann et al.¹⁵ obtained only 5a in a similar reaction.

Thermolysis of 17 in refluxing benzene gave 52, as the sole product, but in hot acetone both 5g and (presumably via a <u>Pummerer</u>-type reaction) 14 were formed. This reaction sequence was supported by the treatment of $\frac{1}{2}$ with acetic anhydride. This reaction gave rise to the "regular"^{12,16} Pummerer product 3-acetoxy-3--(phenylthio)flavanones (12,20), as minor components, as well as to 14 (formed directly from the α -thiocarbonium ion 12,16,17,21) and 52 (direct elimination product); the 14:5a ratio being dependent on the presence or absence of acid catalyst. Similar acid-catalyzed Pummerer reaction have also been observed with 3-(methylsulfinyl)flavanones^{9,18}.

In sum, mesylates 1 were found to show high reactivity toward thiols and thiolates. This nucleophilic substitution reaction offers a convenient method for the synthesis of various 3-(alkylthio)- or -(arylthio)flavanones, precursors of other sulfur-containing flavonoids.



EXPERIMENTAL SECTION

 $\begin{array}{l} \mbox{EXPERIMENTAL SECTION} \\ \mbox{Mp's were determined with a Kofler apparatus and are uncorrected. IR spectra were record with a Perkin-Elmer 283 instrument in KBr discs unless otherwise stated. H-NMR spectra were measured on a Bruker WP 200 SY (200 MHz) spectrometer in CDC1_{3}solutions (internal standard TMS, <math>\delta = 0$ ppm). Mass spectra were recorded with a VG-7035 GC-MS system (EI). Kieselgel 40 (Merck, 0.063-0.2 mm) was used for column chromatography, eluent was PhH unless otherwise specified. TLC was performed on Kieselgel 60 F_{254} using 4:1 PhMe-EtOAc or PhH as developing system. All the isolated flavones ($5\underline{2}\underline{2}\underline{2}\underline{2}$), 3-hydroxyflavones ($1\underline{2}\underline{0}\underline{2}\underline{2}\underline{2}$), and 2'-hydroxychalcones ($\underline{9}\underline{a}, \underline{c}\underline{-e}$) were identified by means of mp, mixed mp, TLC comparison and/or IR spectra. $\frac{3-(Ethylthio)flavanone(2\underline{2}\underline{2})}{16.3 mmol}, and K_{2}CO_{3}$ (4.34 mmol), Adogen-464 (0.1 mmol) and abs. PhH (16 ml) was stirred at room temp. When the reaction was complete (TLC monitoring) the insoluble material was filtered off, washed with PhH. The combined PhH fractions were washed with H_{2}O, dried (MgSO_{4}), the concentrated residue was fractionated by the solution of th

column chromatography and 132 mg (14.7 %) of $\underline{6a}$, 622 mg (68.7 %) of $\underline{2a}$ and 110 mg (15.8 %) of $\underline{5a}$ was obtained. The eluted $\underline{2a}$ was found to be a 3:1 mixture of \underline{cis} and \underline{trans} isomers by H-NMR, attempts to separate the isomers remained unsuccesful. $\underline{2a}$, \overline{oil} ; IR (film) 2963 (CH₃), 2921 (CH₂), 2864 (CH₃ + CH₂S), 1680br (C=0), 1302, 1224, 972sh, 952 (flavanone skeleton) M5 284 (M⁺, 13 %), 224 (89), 223 (59), 221 (9), 164 (100), 151 (29), 147 (34), 135 (76.5), 134 (30), 121 (63), 1120 (32), 105 (20.5), 103 (25.5), 102 (20), 92 (46), 91 (76), 77 (64.5), \underline{cis} - $\underline{2a}$, H-NMR 7.99 (dd, H-5), 5.70(d, H-2), 3.61 (d, H-3) ($J_{23} = 2.1$ Hz), 2.43 (q, \overline{CH}_2), 1.08 (t, \overline{CH}_3). \underline{trans} - $\underline{2a}$, H-NMR 7.84 (dd, H-5), 5.66 (d, H-2), 3.96 (d, H-3) ($J_{23} = 4.5$ Hz), 2.56 (\overline{q} , \overline{CH}_2), 1.25 (t, \overline{CH}_3). Method B. A soln of $\underline{1a}$ (3.14 mmol), EtSH (32.6 mmol), Et₃N (32.5 mmol) in abs. Method B. A soln of $\underline{1a}$ (3.14 mmol), EtSH (32.6 mmol), Et₃N (32.5 mmol) in abs. Method B. A soln of $\underline{1a}$ (3.14 mmol) (H-2). The dried and evaporated extract was fractionated by column chromatography (4:1) petroleum ether-EtDAc) to give 18 mg (2.0 %) of $\underline{6a}$, 511 mg (57.2 %) of $\underline{2a}$ ($\underline{cis}/\underline{trans} = 7:3$; H-NMR), 35 mg (3.5 %) of unreacted $\underline{1a}$ and 130 mg (18.6 %) of $\underline{5a}$.

 $== \frac{3-(\text{Propylthio})\text{flavanone} (3\underline{a}). \text{ According to Method A, using 6.6 mmol of PrSH.} \\ \text{Column chromatographic separation afforded 827 mg (88.3 %) of 3a (cis/trans = 3:1), and 34 mg (4.9 %) of 5\underline{a}. 3\underline{a}$ oil; IR (film) 2956 (CH₃), 2923 (CH₂), 2865 (CH₃ + CH₂S), 1682 (C=0), 1320sh, 1300, 1224, 951 (flavanone skeleton).² MS 298 (M⁺⁻³, 9), 224 (100), 223 (66), 221 (8), 207 (7), 178 (56), 165 (22,), 147 (41), 135 (33), 121 (45), 120 (26), 104 (9.5), 103 (16), 102 (8), 93 (8), 92 (13), 91 (27.5), 77 (23.5). cis-3\underline{a}, H-NMR 8.03 (dd, H-5), 5.69 (d, H-2), 3.58 (d, H-3) (J₂3 = 2.2 Hz), $\approx 2.4 \text{ (m, SCH}_2), 1.4 \text{ (m, CH}_2\text{CH}_3), 0.79 \text{ (t, CH}_3), \frac{\text{trans}-3\underline{a}, H-NMR 7.88 \text{ (dd, H-5)}, 5.66 (d, H-2), 3.93 (d, H-3)^2 (J₂3 = 4.5 Hz), <math>\approx 2.45 \text{ (m, SCH}_2), 1.6 \text{ (m, CH}_2), 0.95 \text{ (t, CH}_3).$

3-(Phenylthio)-4'- R^{\perp} -flavanones ($4\underline{a}$ - \underline{e}) and 1-(2-hydroxyphenyl)-2-(phenylthio)-

-3-(4-R1-phenyl)-1-propanones (gg,g). 3 mmol of la-e² and 4.25 mmol of PhSH were reacted and worked up according to Method A. Results are given in the Table. <u>Method C.</u> A mixture of la-e (3 mmol), PhSNa (3.1 mmol) and abs. PhH (20 ml) was stirred at room temp. When the reaction completed the insoluble salts were filtered off and washed with PhH. The combined PhH fractions were washed with H₀, dried and concentrated, the oily residue was separated by column chromatography. The filter-cake was dissolved in H₂O, extracted with CH₂Cl₂, dried and evaporated to give pure 10.

Grief the filter-cake was dissolved in H₂C, extracted with CH₂Cl₂, dried and evaporated to give pure <u>10</u>. Starting from <u>1a</u> some modified experiments were performed as follows: Method C₁: 25 mol % Adogen-464 was added as catalyst. Method C₂: 25 mol % <u>1a</u> B-Crown-6 was added as catalyst. Method C₂: 25 mol % <u>1a</u> B-Crown-6 was added as catalyst. Method C₂: 25 mol % <u>1a</u> B-Crown-6 was added as catalyst. Method C₂: 25 mol % <u>1a</u> B-Crown-6 was added as catalyst. Method C₂: 25 mol % <u>1a</u> B-Crown-6 was added as catalyst. Method C₂: MeCN was used instead of PhH. Method C₂: Me₂CO was used i

T. PATONAY et al.

 $\frac{1}{1}$ 182 (15), 147 (65), 121 (58), 120 (62), 110 (100), 109 (26), 102 (34), 93 (13). H-NMR 11.9 (s, 2'-OH), 7.68 (dd, H-6'), 4.86 (ABX \rightarrow "A₂X", H-2), 3.61 (ABX \rightarrow "A₂X",

Be, mp. 94-95.5 ^OC (petroleum ether-EtOAc); IR 1642 (chelated C=O), 1518, 1349, $\overline{1106}$, 862, 854 (NO₂), 1290 (Ar-CO), 1203 (Ar-OH). MS 379 (M⁺, 5), 269 (69), 268 (40), 252 (10), 222 (10.5), 176 (7), 165 (7), 147 (96), 130 (6.5) 121 (100), 120 (35), 110 (100), 109 (26) 102 (12.5), 93 (17), 92 (11). H-NMR 11.9 (s, 2'-OH), 8.12 (dd, H-3",5"), 7.73 (dd, H-6'), 4.95 (ABX \rightarrow "A₂X", H-2), 3.70(ABX \rightarrow "A₂X", H-3).

l-(2-Hydroxyphenyl)-2-(ethylthio)-3-phenyl-l-propanone (6a). A mixture of la

(9.4 mmol), EtSH (40.8 mmol) anh. K CO₃ (13 mmol), Adogen-464 (0.3 mmol) and abs. PhH (45 ml) was kept at room temp for 15 days and then worked up according to rnn (42 mi) was kept at room temp for 15 days and then worked up according to Method A. The concentrated residue was crystallized from abs. EtOH-petroleum ether to yield 1.83 g (68.4 %) of $\underline{6}_{2}$, mp. 76-78 C.-IR 2954 (CH₃), 2923 (CH₂), 2872 (CH₃ + CH₂S), 1641 (chelated C=0), 1306 (Ar-CO), 1210 (Ar-OH). MS 286⁻(M⁺, 21), 268⁻(3), 226 (38), 224 (17), 223 (22), 207 (7), 151 (36), 147 (20.5), 121 (100), 120 (17), 104 (19), 93 (20), 91 (6), 77 (14), 65 (33.5). H-NMR 12.1 (s, 2'-OH), 7.76 (dd, H-6'), 4.59 (ABX → "A₂X", H-2), 3.57 (ABX → "A₂X", H-3), 2.37 (dq, CH₂CH₃), 1.18 (t, CH₃). Found: C, 70.88; H, 6.04; S, 10.98. $C_{17}H_{18}O_2S$ requires: C, 71.30; H,6.33; S, 11.20.

1-(2-Hydroxyphenyl)-2-(propylthio)-3-phenyl-1-propanone (7g). In a same

manner as described for <u>6a</u>, reaction time 117 hrs. Yield: 84.6 %, mp. 42.5-43.5 $^{\text{O}}$ C (EtOH). IR (CCl₄) 2958 (CH₃), 1641 (chelated C=O), 1308, 1286 (Ar-CO), 1204 (Ar-OH). H-NMR 12.1 (s, 2'-OH), 7.77 (dd, H-6'), 4.53 (ABX - "A₂X", H-2), 3.56 (ABX - "A₂X", H-3), 2.33 (m, CH₂CH₂CH₃), 1.54 (m, CH₂CH₂CH₃), 0.91 (t, CH₃). Found: C, 71.55; H, 6.84; S, 10,93. C₁₈H₂₀O₂S requires: C, 71.97; H, 6.71; S, 10.67. 10.67.

2'-Hydroxychalcone ($\underline{9}\underline{a}$) from $\underline{4}\underline{a}$. A soln of $\underline{4}\underline{a}$ (0.87 mmol); <u>Cis/trans</u> = 7:3) and PhSNa (0.91 mmol) in abs. MeCN (10 ml) was stirred for 3 hrs at room temp, then poured into H_2O , extracted with CH_2CL_2 . The dried extract was concentrated in vacuo and separated by column chromatography to afford 51 mg (26.1 %) of $\underline{9a}$ besides 128 mg (44.1 %) of unreacted $\underline{4a}$, 121 mg (62.1 %) of $\underline{11}$ and 5 mg (2.6 %)

of $\frac{5}{29}$. From $\frac{69}{29}$. A mixture of $\frac{69}{29}$ (0.7 mmol), anh. K₂CO₃ (1.45 mmol) and PhH (10 ml) was refluxed for 18 hrs, then worked up according to Method A. The residue was crystallized from petroleum ether to give 87 mg (55.6 %) of $\frac{9}{2}$ Presence of flavanone ($\frac{1}{2}$) was detected in the mother liquor by TLC.

When the reaction was repeated in the presence of Adogen-464 (0.2 mmol) (24 hrs, room temp), column chromatography afforded 57 mg (36.4 %) of $\frac{9}{2}$ and 24 mg (15.3 %) of $\frac{1}{2}$.

<u>cis</u>-3-(Ethylsulfonyl)flavanone (<u>l</u>§). A soln of <u>2</u>g (0.9 mmol; <u>cis</u>/<u>trans</u> = 7:3) and of 30 % aq H_2O_2 (0.5 ml) in AcOH (5 ml) was allowed to stand for 44 hrs at and of 50 % aq H₂O₂ (0.5 ml) in Actin (5 ml) was allowed to statution 44 mls at room temp, then poured into H₂O and extracted with Et₂O. The ethereal extract was washed with sat NaHCO₂ soln, dried and concentrated, the residue was crystallized from petroleum ether-EtoAc to give 174 mg (62.1 %) of 19_1 mp. 93-95 °C.IR 1667 (C=O), 1317, 1141 (SO₂) 1210, 946 (flavanone skeleton). H-NMR 7.86 (dd, H-5), 6.53 (d, H-2), 4.30 (d, H-3) (J₂ = 1.1 Hz), 3.29 (q, CH₂), 1.48 (t, CH₃). Found: C, 65.00; H, 4.89; S, 10.18. C₁₇H₁O₄S requires: C, 64.54; H, 5.10; S,10.13. Only 59 but no trans-isomer of 18 was detected in the mother liguor by TLC.

<u>cis</u>-3-(Phenylsulfonyl)-4'-R¹-flavanones (<u>l</u><u>é</u>a॒-dֲ). MCPBA oxidation. To a

stirred soln of <u>cis-4a-d</u> (0.45 mmol) in CHCl₂ (10 ml) was added dropwise a soln of MCPBA (1.16 mmol) in CHCl₂ (10 ml). After stirring for 18-21 hrs (TLC) the mixture was washed with sat MaHCO₃ soln, dried, the solvent was removed <u>in vacuo</u> and the residue was recrystallized. <u>16a</u>, yield: 85.1 %, mp. 153-154 °C (petroleum ether-EtOH). IR 1678 (C=O), ≈1311, II49, 1140 (SO₂), ≈1305, 1214, 935 (flavanone skeleton). H-NMR 7.89 (dd, H-2",6"), 7.76 (dd, H-5), 6.49 (d, H-2), 5.34 (d, H-3) (J₂₃ = 1.45 Hz). Found: C, 69.53; H, 4.53; S, 8.79. C₂₁H₁O₄S requires: C, 69.21; H, 4.43; S, 8.80. <u>165</u>, yield: 78.9 %, mpl 121-123 °C (EtOH). IR 2830 (MeO), 1682 (C=O), 1307br (SO₂ + flavanone skeleton), 1249, 1035 (Ar-O-Me), 1205, 925 (flavanone skeleton). ¹H NMP (DMSD) 7.87 (dd, H-2", 6") 7.74 (dd, H-5) 6.88 (d, H-3', 5') 6.41 (d, H-2') $\begin{array}{c} (302 + 114 \text{ diversion of skeleters}, 1277, 1077 (\text{dd}, H-5), 6.88 (\text{d}, H-3', 5'), 6.41 (\text{d}, H-2), \\ 1 \text{H-NMR} (DMSO) 7.87 (\text{dd}, H-2", 6"), 7.74 (\text{dd}, H-5), 6.88 (\text{d}, H-3', 5'), 6.41 (\text{d}, H-2), \\ 5.30 (\text{d}, H-3) (J_{23} = 1.3 \text{ Hz}), 3.69 (\text{s}, MeO). Found: C, 67.20; H, 4.60; S, 8.05 \\ C_{22}H_{18}O_{5}S \text{ requires: } C, 66.99; H, 4.60; S, 8.13. \\ 1 \frac{6}{26}C, \text{ yield: } 86.9 \text{ %, mp. 170-172 } C (EtOH). \text{ IR 1683 } (C=0), 1313, 1149 } (SO_{2}), \\ 1 306, 1 202, 930 (\text{flavanone skeleton}). \\ 1 \text{H-NMR 7.84 } (\text{dd}, \text{H-2",6"}), 7.72 & (\text{dd}, \text{H-5}), \\ 6.55 & (\text{d}, \text{H-2}), 4.37 & (\text{d}, \text{H-3}) & (J_{23} = 1.35 \text{ Hz}). \\ 1 \text{Found: C1, } 8.98; S, 7.80. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.89; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.99; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.99; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.99; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.99; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.99; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.99; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.99; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.99; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.90; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.90; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.90; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.90; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.90; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.00; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.00; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.00; S, 8.04. \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.00; S, 8.00; \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.00; S, 8.00; \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.00; \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.00; \\ 1 \frac{1}{21}C_{10}O_{4}S \text{ requires: C1, } 8.00; \\ 1 \frac{1}{20}C_{10}O_{10}O_{10}O_{10}O_{10}O_{10}O_{10}O_{10}O_{10}O_{10}O_{10}O$

16d yield: 80.7 %, mp. 170-172 °C (EtOH). IR 1682 (C=O), 1314, 1149 (SO), 1306, 1201, 927 (flavanone skeleton). ¹H-NMR 7.83 (dd, H-2",6"), 7.72 (dd, H-5),

1832

6.53 (d, H-2), 4.37 (d, H-3) (J₂₃ = 1.5 Hz). Found: Br, 18.20; S, 7.20. $C_{21}H_{15}Br0_{4}S$ requires: Br, 18.02; S, 7.23. H_{202} oxidation. 4a (1.32 mmol; cis/trans = 7:3) was treated with H₂O₂ as described for the prepr of 18 to afford 118 mg (24.7 %) of 16a. Presence of 5a and 10a was detected in the mother liquor by TLC.

<u>cis</u>-3-(Phenylsulfinyl)flavanone (<u>17</u>). A soln of MCPBA (0.94 mmol) in CHCl₃

(10 ml) was added dropwise to a stirred and cooled (-18 $^{\circ}$ C) soln of cis-4a (0.9 mmol) during a 20 min period. After 70 min the mixture was washed with sat NaHCO₃ soln (2x45 ml), dried and concentrated <u>in vacuo</u>. The sticky residue was kept in the refrigerator with a mixture of getroleum ether (8 ml) and CCl₄ (1 ml) to give 255 mg (81.1 %) of <u>17</u>, mp. 100-105 C (dec.). IR 1681 (C=0), 1304, 1223, 950 (flavanone skeleton), 1037 (SO). H-NMR 8.26 (dd, H-5), 5.98 (d, H-2), 3.80 (d, H-3) (J₂₃ = 2.65 Hz). Found: C, 72.27; H, 4.73; S, 9.03. C₂₁H₁₆O₃S requires: C, 72.39; H, 4.63; S, 9.03.

Flavone (<u>5</u>a). Treatment of <u>trans</u>-<u>4a</u> (0.45 mmol) with MCPBA in a same manner as described for 17 afforded 71 mg (71.0 %) of 5a as the only product.

3-(Ethylthio)flavone (13). A soln of 2a (1.25 mmol; <u>cis/trans</u> = 7:3) and DDQ (3.08 mmol) in dioxane (12 ml) was refluxed for 15 hrs,concentrated <u>in vacuo</u> and the dark residue was purified by column chromatography (4:1 petroleum ether--t0Ac) to afford 185 mg (52.3 %) of 13, mp. 90-91 °C (petroleum ether). IR 2980 (CH₃), 2922 (CH₂), 1642 (C=0), 1353, I227 (flavone skeleton). H-NMR 8.28 (dd, H-5), 2.88 (q, CH₂), 1.12 (t, CH₃). Found: C, 72.90; H, 504; S, 11.06. $C_{17}H_{14}O_2S$ requires: C, 72.56; H, 5.00; S, 11.36.

3-(Phenylthio)flavone (14).In the same manner as described for the prepn of 13, 1.83 mmol of 4a (cis/trans = 7:3) gave 177 mg (29.2 %) of 14,mp. 131-133 °C (petroleum ether-EtOH). IR 1650 (C=O), 1355, 1210 (flavone skeleton), 1077 (S-Ph). Found: C, 77.05; H, 4.36; S, 9.78. $C_{21}H_{14}O_2S$ requires: C, 76.34; H, 4.27; S, 9.70.

3-(Phenylsulfonyl)flavone (15). A soln of 14 (0.26 mmol) and 30 % aq H₂O₂ (0.3 ml) in AcOH (3 ml) was allowed to react at room temp for 72 hrs and then poured into H_0. The precipitate was filtered of to give 86 mg (90.1 %) of 15, mp. 224.5-227 C (petroleum ether-EtOAc). IR 1656, 1651 (C=0), 1350, 1211 (flavanone skeleton), 1330, 1319, 1164, 1145 (SO₂). Found: C, 69.13; 3.98; S, 8.94. $C_{21}H_{14}O_4S$ requires: C, 69.60; H, 3.89; S, 8.85.

Thermolysis of 17 in PhH. A soln of 17 (0.40 mmol) in abs. PhH (15 ml) was refluxed for 16 hrs, concentrated in vacuo and the residue was crystallized from petroleum ether to yield 59 mg (66.1 %) of $\frac{5}{2}$ as the sole product. Thermolysis of $\frac{17}{17}$ in Me₂CO. A soln of $\frac{17}{12}$ (0.38 mmol) in abs. Me₂CO (10 ml)

was refluxed for 64 hrs, the solvent was evaporated and the residue was fractionated by column chromatography to give 33 mg (26.4 %) of $\frac{1}{4}$ and 44 mg (51.8 %) of $\frac{5}{4}$.

Pummerer reaction of 17 in the presence of acid. Five drops (cca 0.06 ml) cc. H₂SO₄ was added to a soln of 17 (1.28 mmol) in Ac₂O (18 ml) at room temp. After 24 hts the mixture was poured into H₂O, extracted with CH₂Cl₂. The dried and concentrated extract was fractionated by column chromatography to afford 41 mg (8.2 %) of 19, mp. 180-181.5 (petroleum ether). IR 1756 (OAC), 1694 (C=O), 1293, 1226, 1023 (flavanone skeleton), 1206, 1041 (C-O-Ac). MS 390 (M⁺⁺, 4), 348 (12.5), 330 (5), 281 (19), 239 (100), 223 (7), 211 (74), 199 (28), 181 (20), 165 (11), 152 (13), 133 (43), 121 (31), 118 (16.5), 110 (37), 105 (60.5), 77 (23.5). H-NMR 7.90 (dd, H-5), 6.50 (s, H-2), 2.12 (s, OAC); a NOE of + 41 % between the H-2 and OAc was registered irradiating δ = 6.50 ppm signal. Further elution yielded 260 mg (61.5 %) of $\frac{14}{2}$, 8 mg (2.6 %) of $\frac{1}{2}$ O and 20 mg (7.0 %) of $\frac{5}{2}$.

Pummerer reaction of 12 without acid catalyst. A soln of 12 (0.29 mmol) in Ac 0 (5 ml) was allowed to stand at room temp for 95 hrs, the solvent was distilled off under reduced pressure. Column chromatography of the residue afforded 17 mg (54.3 %) of 11. Next fraction was a yellow oil (18 mg, 16.1 %) having IR and MS spectra identical with those of 19. H-NMR indicated the presence of 19 as the majorcomponent, spectral data of the minor product 20 : 7.80 (dd, H-5), 6.53 (s, H-2), 1.92 (s, OAc); estimated 19/20 ratio = 2:1. Further elution using a PhH--EtOAc mixture with increasing amount of EtOAc yielded 10 mg (10.6 %) of 14 and AT mg (73.7 %) of 5a 47 mg (73.7 %) of 5a.

REFERENCES

- 1. T. Patonay, G. Litkei and R. Bognár, <u>Tetrahedron</u>, <u>40</u>, 3425 (1984).
- 2. T. Patonay, M. Rákosi, G. Litkei and R. Bognár, Liebigs Ann. Chem.,

- T. Patonay, G. Litkei and R. Bognár, <u>Acta Chim. Acad. Sci. Hung.</u>, <u>108</u>, 135 (1981).
- 4. M. Lissel, <u>J. Chem. Res.</u> (S), 286 (1982); (M), 2946 (1982).
- 5. S.S. Simons, M. Pons and D.F. Johnson, <u>J. Org. Chem.,</u> 45, 3084 (1980).
- The synthesis of 3-(ethylthio)flavanone(22)from 2'-hydroxy-2-(ethylthio)acetophenone via α-(ethylthio)chalcone was reported by Fujita et al. (<u>Tetrahedron Lett.</u>, 4115 (1978)) without any physical, spectroscopic or stereochemical data of the product.
- S. Matsuura, M. Iinuma, K. Ishikawa and K. Kagei, <u>Chem. Pharm. Bull.</u> <u>26</u>, 305 (1978).
- N. Umino, N. Ito and R. Ishida, <u>Japan. Pat.</u> 75 62.976 (1975); <u>Chem. Abs.</u>, <u>83</u>, 193 085 b (1975).
- N. Umino, N. Ito and R. Ishida; ibid., 75 64.272 (1975); <u>Chem. Abs.</u>, ₿Ѯ, 114 206 r (1976).
- 10. K.P. Jadhav and D.B. Ingle, <u>Indian J. Chem.</u>, <u>228</u>, 150 (1983).
- 11. G.A. Russel and L.A. Ochrymovycz, <u>J. Org. Chem.</u>, <u>35</u>, 2106 (1970).
- S. Dae, <u>Organic Chemistry of Sulfur</u> (Ed. S. Dae), 383, Plenum, New York -- London, 1977.
- 13. ⁸B.M. Trost, <u>Acc. Chem. Res.</u>, <u>1</u><u>1</u>, 453 (1978); ^bibid., <u>Chem. Rev.</u>, <u>7</u><u>8</u>, 363 (1978).
- N. Umino, N. Ito and R. Ishida, <u>Japan. Pat.</u> 75 62.975 (1975); <u>Chem. Abs.</u>, <u>83</u>,178 826 a (1975).
- M.v. Strandtmann, S. Klutchko, M.P. Cohen and J. Shavel, <u>J. Heterocycl. Chem.</u>, <u>9</u>, 171 (1972).
- 16. G. Kresze, <u>Methoden der Organischen Chemie</u> (Houben-Weyl), Band E 11 (Ed. D. Klamann), 669, Thieme, Stuttgart New York (1985).
- 17. S. Wolfe and P.M. Kazmaier, <u>Can. J. Chem., 57</u>, 2388, 2397 (1979).
- N. Umino, N. Ito and R. Ishida, <u>Japan Pat.</u> 75 62.977 (1975); <u>Chem. Abs.</u>, <u>83</u>, 178 825e (1975).
- Atlas of Spectral Data and Physical Constants for Organic Compounds (Ed. J.G. Grasseli and W.M. Ritchey), 2nd Ed., Vol. 3., 208, CRC, Cleveland (1975).