Synthesis of Chiral Macrocycles by Cyclodimerization of Diamines with Stepwise Nucleophilic Aromatic Substitution of 1,5-Difluoro-2,4dinitrobenzene

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Abstract: Chiral cyclophanes were synthesized by cyclodimerization of 2,7-diaza-1,2,3,6,7,8-hexahydropyrene N-acylated with Lamino acids through stepwise nucleophilic aromatic substitution of 1,5-difluoro-2,4-dinitrobenzene.

Key words: macrocycles, chiral, nucleophilic aromatic substitution, cyclophanes, diazapyrene

The study of supramolecular properties of macrocyclic molecules has been particularly successful over the years in providing detailed insight into the mechanisms of molecular recognition. This is largely due to that macrocycles are inherently more or less fairly rigid and preorganized and can, if properly designed, despite their relatively small size, present organized grooves or cavities for the binding of other small(er) molecules. Within this context, cyclophanes^{1–8} appear attractive as their aromatic nature allow for interactions with guest molecules e.g. via π - π , hydrophobic, π -hydrogen bonding, and cation- π forces. Although nature's receptors are chiral and have often been mimicked for molecular recognition studies, relatively few chiral macrocycles have been reported as model systems for biomimetic molecular recognitions.^{2,9-18} Small macrocycles are typically more flexible than protein binding sites unless the building blocks used for the macrocycle assembly are designed to confer conformational rigidity. The conformational flexibility of a macrocycle intended to mimic protein binding sites has to be lowered in order not to loose entropy upon ligand binding, yet high enough to adjust to the ligand structure, i.e. allow for induced fit. One way of modulating the flexibility of macrocycles is to vary the number and the structure of small rings (aromatic or aliphatic) in the macrocycle. Herein, we disclose a macrocyclization reaction sequence, which results in the introduction of two dinitroaniline rings into chiral cyclophanes. The reaction sequence involves cyclodimerizations of L-amino acid-functionalized 2,7-diaza-1,2,3,6,7,8-hexahydropyrenes through stepwise nucleophilic aromatic substitution of 1,5-difluoro-2,4dinitrobenzene.

The synthesis of 2,7-diaza-1,2,3,6,7,8-hexahydropyrene **2** from 1,4,5,8-naphthalenetetracarboxylic diimide¹⁹ (**1**) has

SYNLETT 2004, No. 14, pp 2517–2520 Advanced online publication: 10.11.2004 DOI: 10.1055/s-2004-835633; Art ID: D23004ST © Georg Thieme Verlag Stuttgart · New York recently been reported (Scheme 1).²⁰ Upon following the reported synthesis, we found that the time-consuming Soxhlet extraction for purification of crude 2 could be replaced by a considerably faster flash chromatography. The diamine 2 has a limited stability and is not suited for long-term storage. Consequently, the subsequent acylation of diamine with Boc-protected amino acids produce better yields of 3 with a freshly prepared 2.



Scheme 1

Initially, acylation of **2** was found to be difficult due to poor solubility of **2** in solvents normally used in peptide couplings (i.e. CH_2Cl_2 , DMF and THF). The diamine **2** dissolves well in a mixture of CH_2Cl_2 and MeOH, but this solvent mixture did however produce large amounts of amino acid methyl esters upon attempted acylation of **2** with amino acids. A reliable procedure for acylating the diamine **2** was finally found by using regular coupling agents, such as EDC or DIC, in chloroform. The bis-acylated compounds **3a–c** were obtained in good yields and were also found to be stable upon long-term storage. Deprotection of **3a–c** to diamines **4a–c** proceeded smoothly with HCl or TFA (Scheme 2). The hydrochloride salts corresponding to **4a–c** showed poor solubility, which limited their use as nucleophiles in subsequent reactions. Treatment of the crude bis-ammonium TFA salts 4a-c with 1,5-difluoro-2,4-dinitrobenzene in the presence of cesium carbonate in THF afforded 5a-c doubly equipped with monofluorobenzene moieties ready for cyclodimerization in a second nucleophilic aromatic substitution with 4a-c. Substitution of the second fluoride required harsher conditions. Refluxing equimolar amounts of 4 and 5 and 6 equivalents of cesium carbonate in THF for 20 hours furnished the target macrocycles 6a-f in relatively good yields of 15-25% after purification with flash chromatography followed by precipitation. Attempts to increase the cyclodimerization yields and to shorten reactions times by adding copper iodide to the reaction mixtures failed.²¹ The purities of **6a-f** were analyzed with RP-HPLC and macrocycles 6a,b,e,f yielded chromatograms with one peak corresponding to a purity of >95%. Macrocycles 6c,d were of lower purity and gave unfortunately less satisfactory chromatograms, though the identity of 6c-d were confirmed with FAB-HRMS. FAB-HRMS and ¹³C-NMR analysis of the macrocycles 6a-f confirmed their structures, whereas ¹H NMR analyses of **6a**-**f** proved difficult due to poor resolution presumably caused by slow conformational exchange. Hightemperature NMR experiments in deuterated DMSO were unsuccessful as the macrocycles 6a-f decomposed spontaneously at elevated temperatures.



Scheme 2

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Monte-Carlo conformational searches²² of **6a** (side-chains protected as methyl ethers in order to simplify calculations) gave the one dominating group of closely related conformers (150-200 similar conformers within 5 kcal/ mol) with cleft-like collapsed structures in which the two naphthalene units were face-to-face stacked (Figure 1). The two dinitroaniline units were oriented perpendicular to the naphthalene units in opposite directions. The main group conformers differed by small changes in the angle between the naphthalene and dinitroaniline units and the conformations of the flexible amino acid side chain, which suggests that the macrocycles indeed are conformationally well-defined. A few minor conformers with either a naphthalene sandwiched between the dinitroanilines or a dinitroaniline sandwiched between two naphthalenes were found, however, with higher energies than the main group conformers.



Figure 1 Stereo representation of the global energy minimum conformation of an analog of **6a** with side-chains protected as methyl ethers in order to simplify calculations

In summary, we have developed a short synthesis of cyclophanes via step-wise nucleophilic aromatic substitution of 1,5-difluoro-2,4-dinitrobenzene with diamines obtained by acylation of 2,7-diaza-1,2,3,6,7,8hexahydropyrene with amino acids. The method is versatile as variation of the diamine component opens up for the construction of a range of related 1,5-difluoro-2,4dinitrobenzene derived cyclophanes as receptors with potential of mimicking biomolecule recognition.

2,7-Diaza-1,2,3,6,7,8-hexahydropyrene (2):

1,4,5,8-Naphthalenetetracarboxylic diimide (1, 3.00 g, 11.3 mmol) was suspended in dry THF (75 mL, dried over MS 4Å) in an ovendried and nitrogen-flushed 250 mL three-necked flask. To this mixture BF3·THF (5.0 mL, 45 mmol) was added, followed by BH₃·DMS in THF (70 mL, 2 M). The reaction was refluxed for 64 h, cooled to 0 °C and slowly quenched with MeOH (15 mL) while immersed in an ice-bath. (CAUTION: Excessive foaming may occur at this moment). Then, 6 M HCl (25 mL) was added and the reaction was refluxed for another 3 h. After cooling to ambient temperature, the reaction was adjusted to pH 13 with 50% NaOH. The organic solvents were evaporated and the remaining aqueous suspension was filtered. Filtered solids were dried in vacuo and further purified by flash chromatography (SiO₂, CH₂Cl₂-MeOH-Et₃N $20:0:1 \rightarrow 20:1:1 \rightarrow 20:2:1$ gradient) yielding 1.55 g (65%) of **2** as a pale solid. TLC (EtOAc-MeOH-Et₃N 16:4:1): $R_f = 0.2$. ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3): \delta = 7.15 \text{ (s, 4 H)}, 4.20 \text{ (s, 8 H)}.$

Typical Acylation of 2: [*S*,*S*]-*N*,*N'-Bis*-[3-benzyloxy-2-(*N*-*t*-but-oxycarbonylamino)-propionyl]-2,7-diaza-1,2,3,6,7,8-hexahy-dropyrene (3a):

Compound 2 (0.60 g, 2.9 mmol) was suspended and sonicated for 5 min at ambient temperature in CHCl₃ (150 mL, purified on basic Al₂O₃). To the suspension was added BOC-Ser(OBn)-OH (2.53 g, 8.6 mmol), EDC (2.19 g, 11.4 mmol), HOBt (0.88 g, 5.7 mmol), DMAP (1.44 g, 11.4 mmol) and NaHCO₃ (0.96 g, 11.4 mmol) and the resulting mixture was stirred for 18 h. The reaction was washed with 5% HCl (2 \times 200 mL), sat. NaHCO₃ (3 \times 200 mL), and sat. NaCl (3 \times 200 mL). The organic phase was dried (MgSO₄), filtered, and evaporated into a pale solid (2.50 g), which was purified by flash chromatography (SiO₂, heptane-EtOAc 1:1) to give 3a (1.57 g, 72%). A sample of analytical purity was prepared by precipitation from MeOH–H₂O (9:1). TLC (heptane–EtOAc 1:1): $R_f = 0.25$. $[\alpha]_{D}^{23}$ –3.9 (c 1, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃): δ = 7.30– 7.00 (m, 14 H), 5.50 (d, 2 H, J = 8 Hz), 5.17–4.92 (m, 10 H), 4.37 (s, 4 H), 3.67–3.52 (m, 4 H), 1.41 (s, 18 H). ¹³C NMR (100 MHz, CDCl₃): δ = 169.4, 155.3, 137.7, 130.5, 130.2, 128.4, 127.7, 127.6, 122.7, 122.5, 122.1, 121.8, 80.0, 77.4, 73.3, 71.2, 50.5, 48.9, 45.4, 28.5. FAB-HRMS: m/z calcd for $C_{44}H_{53}N_4O_8$ [M + H]⁺: 765.3864. Found: 765.3859.

In a similar manner were prepared:

 $\label{eq:spinor} \begin{array}{l} \textbf{[S,S]-N,N'-Bis-[3-(benzyloxycarbonyl)-2-(N-t-butoxycarbonyl-amino)-propionyl]-2,7-diaza-1,2,3,6,7,8-hexahydropyrene (3b):} \\ TLC (heptane-EtOAc 1:1): R_f = 0.17. [α]_0^{21} -4.6 (c 1, CH_2Cl_2$). ^1H \\ NMR (400 MHz, CDCl_3): δ = 7.35-7.17 (m, 14 H), 5.42 (d, 2 H, J = 7 Hz$), 5.23-4.60 (m, 14 H), 2.95-2.75 (m, 2 H), 2.67-2.59 (br, 2 H), 1.42 (s, 18 H). ^{13}C NMR (100 MHz, CDCl_3$): δ = 170.9, 169.4, 155.0, 135.8, 130.7, 130.2, 129.9, 128.7, 128.4, 128.3, 127.7, 127.4, 122.5, 122.0, 80.4, 77.4, 66.8, 47.9, 45.5, 38.0, 28.5. FAB-HRMS: m/z calcd for $C_{46}H_{53}N_4O_{10}$ [M+H]^+: 821.3762$. Found: 821.3759$. } \end{array}$

 $\label{eq:spinor} \begin{array}{l} \textbf{[S,S]-N,N'-Bis-[4-(benzyloxycarbonyl)-2-(N-t-butoxycarbonyl-amino)-butanoyl]-2,7-diaza-1,2,3,6,7,8-hexahydropyrene (3c): \\ TLC (heptane–EtOAc 1:1): R_f = 0.20. [α]_D^{21} +41 (c 1, CH_2Cl_2$). 1H NMR (400 MHz, CDCl_3$): δ = 7.45–7.20 (m, 14 H), 5.50 (d, 2 H, J = 7 Hz), 5.35–5.15 (m, 8 H), 5.05–4.80 (m, 6 H), 2.65–2.50 (br, 2 H), 2.45–2.30 (br, 2 H), 2.10–1.90 (br, 2 H), 1.65–1.50 (br, 2 H), 1.45 (s, 18 H). ^{13}C NMR (100 MHz, CDCl_3$): δ = 173.2, 170.7, 156.0, 136.2, 130.8, 130.4, 130.2, 129.8, 128.8, 128.4, 127.9, 122.7, 122.7, 122.5, 79.9, 66.6, 49.8, 48.3, 45.4, 29.5, 29.0, 28.5. FAB-HRMS: m/z calcd for $C_{48}H_{56}N_4O_{10}Na $[M+Na]^+: 871.3894$. Found: 871.3907. \\ \end{array}$

Typical Synthesis of *Bis*-fluorophenyls 5a-c: [*S*,*S*]-*N*,*N*'-*Bis*-[3-benzyloxy-2-(*N*-5-fluoro-2,4-dinitrophenylamino)-propionyl]-2,7-diaza-1,2,3,6,7,8-hexahydropyrene (5a):

Compound 3a (100 mg, 0.13 mmol) was dissolved in CH₂Cl₂ (5 mL) and triethylsilane (85 µL, 0.50 mmol) was added, followed by TFA (290 µL, 3.9 mmol). After 16 h, the solution was concentrated, the residue dissolved in THF (5 mL, dried over MS), and Cs_2CO_3 (256 mg, 0.80 mmol) and 1,5-difluoro-2,4-dinitrobenzene (214 mg, 0.50 mmol) were added. The resulting suspension was stirred for 2 h and then concentrated. Flash chromatography (SiO₂, heptane-EtOAc 1:1) yielded product 5a as a yellow solid (102 mg, 84%). TLC (heptane–EtOAc 1:1): $R_f = 0.23$. $[\alpha]_D^{24} + 24$ (c 0.5, CH_2CI_2). ¹H NMR (400 MHz, CDCl₃): $\delta = 9.62-9.55$ (br, 2 H), 8.90 (d, 2 H, J = 13 Hz), 7.29–7.10 (m, 14 H), 6.72 (d, 2 H, J = 13 Hz), 5.15–4.87 (m, 10 H), 4.43 (s, 4 H), 3.82–3.66 (m, 4 H). $^{13}\mathrm{C}$ NMR (100 MHz, $CDCl_3$): $\delta = 167.0, 161.0, 158.4, 148.0, 147.9, 136.7, 130.6, 129.2,$ 128.7, 128.4, 127.9, 127.8, 127.6, 126.0, 125.9, 122.8, 122.4, 102.7, 102.4, 73.9, 70.9, 55.2, 48.9, 46.1. FAB-HRMS: m/z calcd for $C_{46}H_{39}F_2N_8O_{12}$ [M + H]⁺: 933.2656. Found: 933.2657.

In a similar manner were prepared:

[*S*,*S*]-*N*,*N*'-Bis-[3-(benzyloxycarbonyl)-2-(*N*-5-fluoro-2,4-dinitrophenylamino)-propionyl]-2,7-diaza-1,2,3,6,7,8-hexahydropyrene (5b):

TLC (heptane–EtOAc 1:1): $R_f = 0.21$. $[a]_D^{25} + 35$ (*c* 0.5, CH₂Cl₂). ¹H NMR (400 MHz, CDCl₃): $\delta = 9.34-9.20$ (br, 2 H), 8.94–8.86 (br, 2 H), 7.32–7.07 (m, 10 H), 6.64 (m, 2 H), 5.25–4.93 (m, 14 H), 3.00–2.87 (m, 2 H), 2.85–2.74 (m, 2 H). ¹³C NMR (100 MHz, CDCl₃): $\delta = 170.6$, 167.9, 147.8, 147.7, 135.2, 135.19, 131.10, 130.0, 129.12, 129.06, 128.9, 128.3, 128.1, 127.9, 126.5, 123.7, 123.2, 122.5, 121.9, 102.3, 102.0, 68.0, 51.2, 51.1, 49.3, 46.6, 38.5. FAB-HRMS: *m*/z calcd for C₄₈H₃₉F₂N₈O₁₄ [M + H]⁺: 989.2555. Found: 989.2556.

[*S*,*S*]-*N*,*N*'-*Bis*-[4-(benzyloxycarbonyl)-2-(*N*-5-fluoro-2,4-dinitrophenylamino)-butanoyl]-2,7-diaza-1,2,3,6,7,8-hexahydropyrene (5c):

TLC (heptane–EtOAc 1:1): $R_f = 0.16. [a]_D^{21} +58 (c 0.5, CH_2Cl_2).$ ¹H NMR (400 MHz, CDCl₃): $\delta = 9.10 (dd, 2 H, J = 7, 23 Hz), 8.98$ (d, 2 H, J = 8 Hz), 7.34–7.20 (m, 14 H), 6.43 (t, 2 H, J = 14 Hz), 5.19–4.91 (m, 14 H), 2.49–2.39 (br, 4 H), 2.25–2.10 (br, 2 H), 1.98– 1.85 (br, 2 H). ¹³C NMR (100 MHz, CDCl₃): $\delta = 173.5, 173.4,$ 168.6, 168.5, 161.4, 158.7, 148.5, 148.4, 135.4, 131.0, 130.3, 130.1, 129.3, 128.9, 128.63, 128.56, 128.2, 127.9, 126.3, 126.2, 123.5, 123.0, 122.5, 122.0, 101.6, 101.4, 67.3, 52.7, 48.8, 46.2, 29.0, 28.0. FAB-HRMS: m/z calcd for $C_{50}H_{42}F_2N_8O_{14}Na [M + Na]^+$: 1039.2687. Found: 1039.2675.

Typical Synthesis of Macrocycles 6a–f: Macrocycle 6a:

Compound **3a** (84 mg, 0.11 mmol) was dissolved in CH₂Cl₂ (5 mL). Triethylsilane (75 µL, 0.40 mmol) was added, followed by TFA (270 µL, 3.3 mmol). The solution was stirred at ambient temperature for 16 h and then evaporated into a syrup. This syrup was dissolved in THF (130 mL, dried over MS) to which was added Cs₂CO₃ (215 mg, 0.70 mmol) and compound **5a** (102 mg, 0.11 mmol). The resulting suspension was refluxed for 20 h and then evaporated. Flash chromatography (SiO₂, heptane-EtOAc 1:4) followed by precipitation from CHCl₃-MeOH (1:2) yielded 6a as a vellow solid (36 mg, 22%). TLC (heptane–EtOAc 1:4): $R_f = 0.10$. HPLC (Column: Supelcosil 250 mm \times 10 mm, 5 μ m. Eluent: H₂O-MeCN 1:0 over 5 min, $1:0 \rightarrow 0:1$ over 1 h, 0:1 over 5 min. Flow: 5 mL/min) $t_{\rm R} = 47$ min. $[\alpha]_{\rm D}^{21} - 38$ (c 0.5, CHCl₃). The ¹H NMR spectrum showed poorly resolved broad peaks. ¹³C NMR (125 MHz, CDCl₃): δ = 167.0, 147.4, 137.2, 130.6, 129.4, 128.7, 128.6, 128.3, 128.1, 127.7, 124.8, 123.5, 121.2, 93.0, 73.8, 67.7, 54.1, 48.7, 46.0. FAB-HRMS: m/z calcd for $C_{80}H_{73}N_{12}O_{16}$ [M + H]⁺: 1457.5268. Found: 1457.5251.

In a similar manner were prepared:

Macrocycle 6b:

Compound **4b** and **5b** gave **6b** in 20% yield. TLC (heptane–EtOAc 1:4): $R_f = 0.50$. HPLC: $t_R = 46$ min. $[\alpha]_D^{21} + 18$ (*c* 0.5, CHCl₃). ¹³C NMR (125 MHz, CDCl₃): $\delta = 171.0$, 168.8, 146.5, 135.5, 135.2, 131.4, 129.8, 128.82, 128.77, 128.71, 128.65, 128.61, 128.4, 128.3, 128.0, 127.9, 127.6, 127.4, 124.5, 122.7, 92.7, 67.1, 49.0, 46.5, 37.5. FAB-HRMS: *m*/z calcd for $C_{84}H_{72}N_{12}O_{20}Na$ [M + Na]⁺: 1591.4884. Found: 1591.4904.

Macrocycle 6c:

Compound **4c** and **5c** gave impure **6c**. FAB-HRMS: m/z calcd for $C_{88}H_{81}N_{12}O_{20}$ [M + H]⁺: 1625.5691. Found: 1625.5717.

Macrocycle 6d:

Compound **4a** and **5b** (or **4b** and **5a**) gave impure **6d**. FAB-HRMS: m/z calcd for $C_{82}H_{72}N_{12}O_{18}Na \ [M + Na]^+$: 1535.4986. Found: 1535.4996.

Macrocycle 6e:

Compound **4a** and **5c** (or **4c** and **5a**) gave **6e** in 18% yield. TLC (heptane–EtOAc 1:4): $R_f = 0.10$. HPLC: $t_R = 49$ min. $[\alpha]_D^{21} + 13$ (*c* 0.5, CHCl₃). ¹³C NMR (100 MHz, CDCl₃): $\delta = 173.7$, 168.5, 166.9, 147.4, 147.3, 136.9, 135.2, 130.6, 130.4, 129.6, 128.9, 128.7, 128.5, 128.2, 128.0, 127.9, 127.8, 124.8, 124.7, 123.8, 123.6, 122.1, 121.1, 92.8, 73.6, 68.4, 66.7, 54.1, 51.4, 48.9, 48.2, 45.7, 29.9, 28.7, 27.1. FAB-HRMS: m/z calcd for $C_{84}H_{77}N_{12}O_{18}$ [M + H]⁺: 1541.5480. Found: 1541.5490.

Macrocycle 6f:

Compound **4b** and **5c** (or **4c** and **5b**) gave **6b** in 20% yield. TLC (heptane–EtOAc 1:4): $R_f = 0.20$. HPLC: $t_R = 45$ min. $[\alpha]_D^{21} + 80$ (*c* 0.5, CHCl₃). ¹³C NMR (100 MHz, CDCl₃): $\delta = 173.3$, 172.2, 168.7, 167.8, 147.3, 146.4, 135.1, 134.6, 130.7, 130.4, 130.2, 129.8, 128.88, 128.83, 128.76, 128.73, 128.59, 128.49, 127.8, 127.6, 126.5, 124.64, 124.58, 123.8, 123.7, 122.1, 122.0, 95.6, 92.6, 68.2, 67.2, 67.0, 66.8, 66.3, 51.0, 50.0, 48.5, 45.8, 28.5, 27.3, 25.8. FAB-HRMS: m/z calcd for $C_{86}H_{76}N_{12}O_{20}Na$ [M + Na]⁺: 1619.5197. Found: 1619.5199.

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