

N-Allyl-N-tert-butyldimethylsilylamine for chiral ligand-controlled asymmetric conjugate addition to *tert*-butyl alkenoates†

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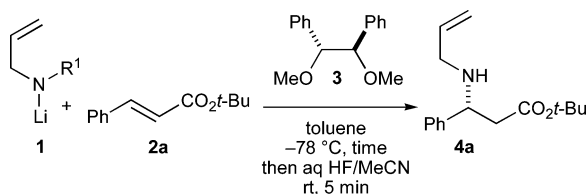
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The chiral ligand controlled asymmetric conjugate addition reaction of lithium *N*-allyl-*N*-(*tert*-butyldimethylsilyl)amide to alkenoates proceeded smoothly to give, after protodesilylation, the corresponding 3-allylaminoalkanoates with high enantioselectivities in high yields. The allyl group on the nitrogen atom was easily removable to afford 3-aminoalkanoates.

Asymmetric conjugate addition of chiral lithium amides to alkenoates has been one of the most powerful methods for the synthesis of chiral 3-aminoalkanoates as has been extensively studied by Davies and his group.^{1,2} A chiral ligand-controlled asymmetric reaction³ of an achiral lithium amide has been recently reported⁴ as a new entry to this interesting field. However, this method relies on the use of *N*-benzy-*N*-trimethylsilylamine as a nitrogen source that requires hydrogenolysis for removal of the benzyl group. We describe herein that allylamine is also a good source of a nitrogen nucleophile in the asymmetric addition to alkenoates, and the allyl group is easily removable by the isomerisation reaction with a rhodium catalyst.^{5,6}

We began our studies with the reaction of allylamine (Scheme 1). The lithium amide **1a** ($R^1 = H$) was *in situ* prepared from allylamine with butyllithium in the presence of chiral ligand **3** in toluene and was then treated with *tert*-butyl cinnamate **2a**. Although the reaction proceeded within a half hour at $-78^\circ C$ to give **4a** in 64% isolated yield, the enantioselectivity was unfortunately almost marginal (Table 1, entry 1).

The reaction of lithium *N*-allyl-*N*-trimethylsilylamine **1b** ($R^1 = TMS$, 3 equiv) was effected by **3** (3.6 equiv) at $-78^\circ C$ to give, after purification by silica gel column chromatography, **4a** with 69% ee in 94% yield (entry 2). The absolute configuration and enantioselectivity of **4a** were determined by converting to **5** with the



Scheme 1 Asymmetric reactions of lithium allylamides **1** with *tert*-butyl cinnamate **2a** by the mediation of **3**.

Table 1 Asymmetric addition of lithium allylamides **1** to cinnamate **2a** in toluene at $-78^\circ C$ giving **4a**

Entry	1	R^1	time (h)	Yield (%)	Ee (%)
1	a	H	0.5	64	5
2	b	TMS	0.5	94	69
3	c	TES	1	91	83
4	d	TBS	1	95	86
5 ^a	d	TBS	1.5	95	89
6	e	TIPS	5	0	

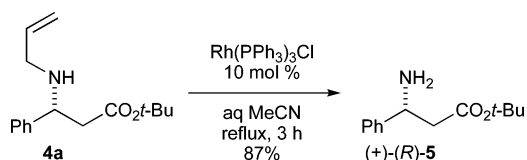
^a Performed at $-95^\circ C$.

† Electronic supplementary information (ESI) available: general procedure for addition reaction, deallylation, silylation of allylamine and data for compounds. See <http://www.rsc.org/suppdata/cc/b4/b405347h/>

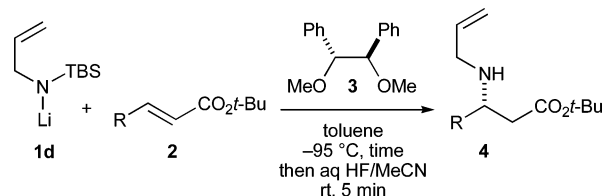
established absolute configuration as shown below. Dramatic improvement of enantioselectivity was observed by changing a TMS group to TES (triethylsilyl) (**1c**: $R^1 = TES$) giving **4a** with 83% ee in 91% yield (entry 3). A TES group was not removed on silica gel column chromatography, but easily protodesilylated with aqueous HF in acetonitrile at room temperature for 5 min to afford **4a**.⁷ TBS (*tert*-butyldimethylsilyl) amide **1d** ($R^1 = TBS$)⁸ was much more effective to give **4a** with 86% ee in 95% yield (entry 4). The enantioselectivity was improved to 89% ee by conducting the reaction at $-95^\circ C$ (entry 5). TIPS (triisopropylsilyl) amide **1e** ($R^1 = TIPS$) was a poor nucleophile not to give a conjugate addition product probably because of a too much steric hindrance (entry 6).

The allyl group of **4a** was easily removed with rhodium chloride⁹ in refluxing aqueous acetonitrile for 3 h to give (+)-(*R*)-**5**¹⁰ in 87% isolated yield without any racemisation (Scheme 2). The stereochemistry of the newly created stereogenic center was thus established to be (*R*).

The asymmetric addition reactions of **1d** with other acyclic and cyclic alkenoates **2** were also controlled by **3** at $-78^\circ C$ or $-95^\circ C$ in toluene to afford **4** with up to 94% ee (Scheme 3). Crotonate **2b** ($R = Me$) was converted to **4b** in 90% yield (Table 2, entry 1). The ee of **4b** was determined to be 90% by 1H NMR using (*S*)-(-)-1,1'-bi-2,2'-naphthol as a chiral shift reagent.¹¹ The reaction of



Scheme 2 Removal of the allyl group of **4a** to give **5** with the established absolute configuration.



Scheme 3 Asymmetric addition of **1d** to **2** by the mediation of a chiral ligand **3**.

Table 2 Asymmetric addition of *N*-TBS-allylamine **1d** to **2** in toluene at $-95^\circ C$ giving **4**[†]

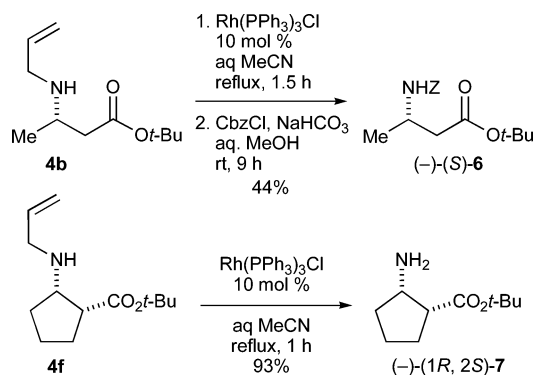
Entry	2	R	time (h)	4	Yield (%)	Ee (%)
1	b	Me	1	b	90	90
2 ^a	c	<i>i</i> -Pr	5	c	84	76
3 ^a	d	(<i>E</i>)-MeCH=CH	18	d	88	75
4	e	1-Naph	3	e	89	94
5	f		1.5	<i>cis</i> - f	82	82
				<i>trans</i> - f	11	84

^a Performed at $-78^\circ C$.

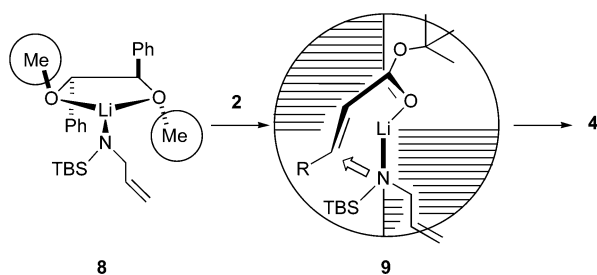
alkenoate **2c** having an isopropyl terminal group ($R = i\text{-Pr}$) gave **4c** with 76% ee in 84% yield (entry 2). Sorbate **2d** ($R = (E)\text{-MeCH=CH}$) was regioselectively converted to 1,4-addition product **4d** with 75% ee in 88% yield (entry 3). The reaction of **2e** ($R = 1\text{-naphthyl}$) gave **4e** with 94% ee in 89% yield (entry 4). Cyclopentenecarboxylate **2f** was also applicable in the reaction to give diastereoselectively *cis*-**4f** with 82% ee as a major product in 82% yield (entry 5). The minor *trans*-**4f** with 84% ee was also produced in 11% yield. It is important to note that chiral ligand **3** was recovered in high yield and was re-used in the asymmetric reaction without loss of selectivity.

Deallylation of **4b** and *cis*-**4f** was achieved by treatment with Wilkinson's catalyst in refluxing aqueous acetonitrile to give (–)-(*S*)-**6** in 44% overall yield after conversion to a Cbz derivative,¹² and (–)-(*1R,2S*)-**7**¹³ in 93% yield without any racemisation. It is important to note that allylamine attacks to the bottom face of linear and cyclic alkenoates shown as **2** giving **4**.

The stereocontrolled formation of chiral **4** is predictable by using a model (Scheme 5). Coordination of the carbonyl oxygen atom of **2** to a lithium atom in a chelate **8** may be the first event for the reaction. Two etheral methyl groups are fixed in *trans* relationship to the adjacent two phenyl groups. It is reasonable to speculate that coordination of **2** takes place in the less hindered region avoiding steric repulsion by the methyl groups of **3** as shown in **9**.¹⁴ The intra-complex attack of the nitrogen atom to the *si*-face (for example $R = \text{Me}$) at the 3-position of *s-cis*-**2** provides chiral **4** with the observed absolute configuration.



Scheme 4 Removal of the allyl groups of **4** giving **6** and **7** with the established absolute configurations.



Scheme 5 Plausible stereoselection for the production of **4** from **9**.

In summary, an external chiral ligand-controlled asymmetric conjugate addition of allylamine to alkenoates was developed. Since both of enantiomers **3** are available, either enantiomer of **4** is accessible. Synthetic utility of the products is the next target.

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Notes and references

‡ *Procedure* (Table 2, entry 1): *n*-BuLi (3.0 mmol) was added to **1d** (3.0 mmol) in toluene (8 mL) at $-78\text{ }^{\circ}\text{C}$ over 5 min. After 0.5 h, **3** (3.6 mmol) in toluene (6 mL) was added. The mixture was stirred for 0.5 h at $-78\text{ }^{\circ}\text{C}$, then cooled to $-95\text{ }^{\circ}\text{C}$. A toluene (2 mL) solution of **2b** (1.0 mmol) was added over 5 min. The mixture was stirred at $-95\text{ }^{\circ}\text{C}$ for 1.5 h and quenched with satd. NH_4Cl (3 mL). After addition of satd. NaHCO_3 (4.5 mL), the whole was extracted with AcOEt. The organic layer was washed with brine, dried, and concentrated. The crude product in MeCN (30 mL) was treated with 46% HF (3 mL) at rt for 5 min. After addition of satd. NaHCO_3 , the mixture was extracted with AcOEt. The organic layer was washed with brine and dried. Concentration and chromatography (AcOEt/hexane = 1/1) gave **3** (quant. recovery) and **4b** with 90% ee (determined by ^1H NMR using (S)-(–)-1,1'-bi-2,2'-naphthol as a chiral shift reagent).

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