Electronic effects on the *cis/trans* selectivity in formation of isoxazolidine-fused eight-membered ring *via* an intramolecular nitrone-alkene cycloaddition

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An intramolecular nitrone-alkene cycloaddition involving *in situ* generated nitrones demonstrated reaction profiles different from those previously reported for pyrimidine system. Tuning the electron density of the benzene ring had a significant effect on *cis/trans* selectivity. These reactions were useful for the synthesis of novel tricyclic hexahydrobenzo[*b*]isoxazolo[3,4-*f*][1,4]diazocin-4(1*H*)-ones and hexahydroisoxazolo[3,4-*f*]pyrido[3,2-*b*][1,4]diazocin-4(1*H*)-one under mild reaction conditions in good yields.

Keywords: isoxazolidines, tricyclic compounds, *cis/trans* selectivity, eight-membered rings, electronic effects, fused-ring systems, nitrone-alkene cycloaddition.

Eight-membered heterocycles, which often occur in various natural products and pharmacologically active substances, have attracted considerable attention because of their unusual structural features and interesting biological effects.¹ For example, balasubramide (1) was isolated from *Clausena indica* that grows in the central mountain rainforests of Sri Lanka.² (–)-Heliannuol A (2) was isolated from aqueous extracts of fresh sunflower leaves (*Helianthus annuus* L. var. SH-222) and exhibits strong allelopathic activity.³ Benzolactam-V8 (3) is a potent activator of protein kinase C.⁴ Nefopam (4) is a centrally-acting nonopioid analgesic drug.⁵ Buflavine (5) possesses interesting adrenolytic and antiserotonin activities.⁶ Pyrimido[4,5-*b*]-[1,5]oxazocine derivative **6** has been observed to be a neurokinin-1 (NK₁) receptor antagonist⁷ (Fig. 1).

1,3-Dipolar cycloadditions of nitrones are powerful synthetic methodologies and have often been utilized in the synthesis of natural products and biologically active molecules.⁸ The intramolecular 1,3-dipolar cycloaddition between nitrones and alkenes can form two new rings simultaneously, providing direct access to fused or bridged polycyclic isoxazolidines of considerable complexity with high levels of regio- and stereocontrol.^{8,9} Thus far, the intramolecular nitrone-alkene cycloaddition reaction have been successfully used for the construction of isoxazoli-



Figure 1. Natural and synthetic examples of 8-membered heterocycles.



 $\mathbb{R}^{3} \mathbb{R}^{1} \stackrel{O}{\longrightarrow} \mathbb{R}^{4} \xrightarrow{\mathbb{R}^{5} \mathbb{N} \mathbb{H} \mathbb{O} \mathbb{H} \mathbb{H} \mathbb{H}}_{\operatorname{NaOAc, CH_2Cl_2}} \mathbb{N} \xrightarrow{\mathbb{R}^{3} \mathbb{R}^{1} \stackrel{O}{\longrightarrow} \mathbb{H}}_{\mathbb{R}^{5} \mathbb{R}^{1} \mathbb{H} \mathbb{R}^{4}}$

 R^1 = Me, Bn, allyl; R^2 = Me, *cyclo*-Hex, Bn; R^3 = Cl, pyrrolidinyl, 4-MeOC₆H₄S; R^4 = H, Me, Bn; R^5 = Me, *cyclo*-Hex, Bn, Ph

dines fused with 4- to 7-membered rings.¹⁰ However, few examples have been reported where the synthesis of isoxazolidines fused with 8-membered rings has been achieved by intramolecular nitrone-alkene cycloaddition reactions.¹¹

Recently, we communicated an intramolecular nitronealkene cycloaddition in pyrimidine derivatives leading to a novel cis-fused tricyclic compound containing an 8-membered ring and isoxazolidine system (Scheme 1).¹² It is worth noting that the desired tricyclic product was not obtained when N-allylamine analog was used. The results revealed that the electronic effects seem to be important for enhanced reactivity. Although the preliminary scope of the intramolecular nitrone-alkene cycloaddition was studied in the pyrimidine system, the electronic effects of other aromatic rings remained to be investigated. During our studies, we noticed that benzene derivatives demonstrated reaction profiles different from those of the pyrimidine system, and the tuning of electron density in the benzene ring had a significant effect on the *cis/trans* selectivity (Scheme 2). Herein, the details of these studies are presented.

Scheme 3

Scheme 2



Several key precursors **12a–c** were prepared in 37–66% overall yields from compounds **9a–c** as depicted in Scheme 3. Treatment of derivatives **9a–c**,**e** with 2-(benzyl-amino)ethan-1-ol followed by reduction yielded compounds **10a–c**,**e**. Compound **10d** was obtained by reacting nitroaniline **9d** with 2-(benzylamino)ethan-1-ol. Reductive amination of compounds **10a–c** resulted in substituted amines **11a–c**. Direct acylation of compounds **11a–c** afforded products **12a–c**. However, a direct acylation of compound **11d** proved to be problematic.¹² The selective protection of the hydroxyl group in compounds **11d,e** with TMSCl, subsequent acylation of the amino group with acryloyl chloride and TFA removal of TMS gave compounds **12d,e**.

Oxidation of compounds 12a-c under the Parikh–Doering conditions cleanly generated the respective aldehydes 7a-c, which were used immediately without further purification (Table 1). Initially, the cyclization reaction of crude aldehyde 7a with *N*-methylhydroxylamine hydrochloride was investigated under our previously reported reaction



Table 1. Synthesis of benzo[b]isoxazolo[3,4-f][1,4]diazocin-4(1H)-ones 8*



Entry	Х	R	Time, h	cis-8 isomer (Yield, %)	trans-8 isomer (Yield, %)	<i>cis-</i> 8 : <i>trans-</i> 8 ratio
1	СН	CO ₂ Me	10	a (62)	a (5)	93:7
2	CH	Н	24	b (39)	b (28)	58:42
3	CH	OMe	28	c (20)	c (35)	36:64
4	CH	NO_2	8	d (71)	d (0)	100:0
5	Ν	Н	3	e (65)	e (0)	100:0

* Overall yield (three steps) from compound 12 and isolated using flash chromatography.



Figure 2. X-ray crystallography data for compounds *cis*-8a and *trans*-8a with atoms represented by thermal vibration ellipsoids of 30% probability.

conditions using NaOAc in CH₂Cl₂, giving 62% isolated yield of ester *cis*-**8a** ($J_{3a,11a} = 8.4$ Hz) and 5% isolated yield of ester *trans*-**8a** ($J_{3a,11a} = 6.9$ Hz) (entry 1). This result was quite different from the previously reported case when only *cis*-isomer was obtained from analogous pyrimidine derivative. To unambiguously determine the configuration of both cycloadducts, X-ray crystallographic analyses of compounds *cis*-**8a** and *trans*-**8a** were performed (Fig. 2). The electronic effects on the *cis/trans* selectivity were further explored by using benzene derivatives with substituents of different electronic profiles or a pyridine derivative, and the results are summarized in Table 1.

As shown in Table 1, good overall yields (55-71%) of the desired products 8 were obtained for the three reaction steps (oxidation of alcohols to aldehydes, formation of nitrones, and cycloaddition). The reaction was sensitive to electronic factors of the benzene ring substituents. The reaction proceeded faster when electron-withdrawing group was attached to benzene ring (entries 1 and 4 vs entries 2 and 3). It is noteworthy that tuning the electron density of the benzene ring had a significant effect on the cis/trans selectivity. The presence of an electron-withdrawing group favored the exo adducts cis-8 (entries 4 and 1), while an electron-donating group favored the endo adducts trans-8 (entry 3), which indicated that decreasing the electron density of the aromatic ring leads to a higher amount of the exo diastereomer being formed. As expected, only the product cis-8e was obtained in 65% yield when precursor 12e was used (entry 5). These results are in line with those previously reported for the intermolecular nitrone-alkene cycloaddition.¹³ Minor variations in the electronic properties of dipolarophiles led to important changes in the ratio of the *cis/trans* isomers formed, possibly through secondary orbital interactions that may occur in the transition state.8g,13c

In conclusion, novel tricyclic isoxazolidine-fused eightmembered rings were successfully constructed *via* an intramolecular nitrone-alkene cycloaddition. The benzene derivatives used as substrates demonstrated reaction profiles different from those previously reported for pyrimidine substrates, and the tuning of electron density in the benzene ring had a significant effect on the *cis/trans* selectivity of these cycloaddition reactions.

Experimental

¹H and ¹³C NMR data were obtained on a Varian Mercury 300 NMR spectrometer (300 and 75 MHz, respectively) with TMS as internal standard and CDCl₃ as solvent unless otherwise stated. HRMS analysis was performed on an Agilent 1290-micrOTOF-Q II mass spectrometer. Melting points were determined with a Beijing Keyi Electro-Optical Factory XT5 melting point apparatus and are uncorrected. Dichloromethane was dried over CaH₂ and distilled prior to use. DMSO was dried over 4 Å molecular sieves. All other commercial reagents were used as received without additional purification.

Methyl 3-amino-4-[benzyl(2-hydroxyethyl)amino]benzoate (10a). Anhydrous K₂CO₃ (8.292 g, 60 mmol) was added to a stirred solution of methyl 4-fluoro-3-nitrobenzoate (9a) (3.983 g, 20 mmol) and BnNH(CH₂)₂OH (4.536 g, 30 mmol) in THF (20 ml). The resulting solution was stirred for 5 h at 60°C. The precipitate was collected by filtration. The resulting clear solution was concentrated in vacuo and CH₂Cl₂ (300 ml) was added. The organic layer was washed with water, dried over MgSO₄, and concentrated in vacuo. Purification by flash chromatography (petroleum ether-EtOAc, 4:1 to 1:1 v/v) afforded methyl 4-(benzyl(2-hydroxyethyl)amino)-3-nitrobenzoate (6.440 g), which was dissolved in EtOH (150 ml). The resulting solution was hydrogenated over Pd/C catalyst (1.288 g, 10% w/w) by stirring for 2 h under a hydrogen atmosphere (balloon). The Pd/C catalyst was then filtered off. The resulting clear solution was concentrated in vacuo. Purification by flash chromatography (petroleum ether-EtOAc, 1:1 to 1:5 v/v) afforded the desired product 10a. Yield 4.866 g (81%), pale-yellow oil. ¹H NMR spectrum, δ, ppm (J, Hz): 3.16 (2H, t, J = 5.4, CH₂OH); 3.57 (2H, t, J = 5.4, NCH₂CH₂OH); 3.87 (3H, s, CH₃); 4.18 (2H, s,

PhCH₂); 7.01–7.07 (1H, m, H Ar); 7.19–7.31 (5H, m, H Ar); 7.38–7.42 (2H, m, H Ar). ¹³C NMR spectrum, δ , ppm: 52.0; 55.3; 58.1; 59.9; 116.7; 120.3; 123.0; 126.9; 127.4; 128.4; 128.9; 137.8; 141.2; 142.8; 167.3. Found, *m/z*: 301.1554 [M+H]⁺. C₁₇H₂₁N₂O₃. Calculated, *m/z*: 301.1547.

2-[(2-Aminophenyl)(benzyl)amino]ethan-1-ol (10b). A stirred solution of 1-fluoro-2-nitrobenzene (9b) (7.055 g, 50 mmol) in pyridine (5 ml, 62 mmol) was treated by the addition of BnNH(CH₂)₂OH (9.374 g, 62 mmol). The resulting solution was stirred for 2 h at reflux. The solvent was removed in vacuo and CH₂Cl₂ (500 ml) was added. The organic layer was washed with water, dried over MgSO₄, and concentrated *in vacuo*. Purification by flash chromatography (4:1 to 1:1 petroleum ether-EtOAc, v/v) afforded 2-[benzyl(2-nitrophenyl)amino]ethan-1-ol (12.030 g), which was dissolved in EtOH (250 ml). The resulting solution was hydrogenated over Pd/C catalyst (2.406 g, 10% w/w) by stirring for 2.5 h under a hydrogen atmosphere (balloon). The Pd/C catalyst was filtered off. The resulting clear solution was concentrated in vacuo. Purification by flash chromatography (petroleum ether-EtOAc, 1:1 to 1:5 v/v) afforded the desired product 10b. Yield 8.723 g (72%), pale-yellow oil. ¹H NMR spectrum, δ, ppm (J, Hz): 3.12 (2H, t, J = 5.4, CH₂OH); 3.49 (2H, t, J = 5.4, NCH₂CH₂OH); 4.10 (2H, s, PhCH₂); 6.71–6.79 (2H, m, H Ar); 6.98 (1H, td, J = 7.5, J = 1.5, H Ar); 7.09 (1H, dd, J = 7.8, J = 1.5, H Ar); 7.23-7.29 (5H, m, H Ar).¹³C NMR spectrum, δ, ppm: 56.6; 60.0; 60.2; 115.9; 119.2; 124.0; 126.0; 127.2; 128.4; 129.0; 136.8; 138.5; 143.6. Found, m/z: 243.1498 $[M+H]^+$. C₁₅H₁₉N₂O. Calculated, m/z: 243.1492.

2-[(2-Amino-4-methoxyphenyl)(benzyl)amino]ethan-1-ol (10c). A stirred solution of 1-fluoro-4-methoxy-2-nitrobenzene (9c) (8.555 g, 50 mmol) in pyridine (5 ml, 62 mmol) was treated by the addition of BnNH(CH₂)₂OH (9.374 g, 62 mmol). The resulting solution was stirred for 4 h at reflux. The solvent was removed in vacuo and CH₂Cl₂ (500 ml) was added. The organic layer was washed with water, dried over MgSO₄, and concentrated in vacuo. Purification by flash chromatography (4:1 to 1:1 petroleum ether-EtOAc, v/v) afforded 2-[benzyl(4-methoxy-2-nitrophenyl)amino]ethan-1-ol (12.990 g), which was dissolved in EtOH (250 ml). The resulting solution was hydrogenated over 10% Pd/C catalyst (2.598 g) by stirring for 3 h under a hydrogen atmosphere (balloon). The Pd/C catalyst was filtered off. The resulting clear solution was concentrated in vacuo. Purification by flash chromatography (1:1 to 1:5 petroleum ether-EtOAc, v/v) afforded the desired product 10c. Yield 8.850 g (65%), pale-yellow oil. ¹H NMR spectrum, δ , ppm (*J*, Hz): 3.06 (2H, t, *J* = 5.4, CH₂OH); 3.45 (2H, t, J = 5.4, NCH₂CH₂OH); 3.74 (3H, s, CH₃); 4.03 (2H, s, PhCH₂); 6.25 (1H, d, *J* = 2.7, H Ar); 6.33 (1H, dd, J = 8.7, J = 2.7, H Ar); 7.01 (1H, d, J = 8.7, H Ar); 7.20-7.30 (5H, m, H Ar). ¹³C NMR spectrum, δ, ppm: 55.2; 57.1; 60.2; 60.6; 101.2; 104.3; 124.8; 127.1; 128.2; 129.0; 129.9; 138.7; 144.9; 157.8. Found, *m*/*z*: 273.1607 [M+H]⁺. C₁₆H₂₁N₂O₂. Calculated, *m*/*z*: 273.1598.

2-[(2-Amino-4-nitrophenyl)(benzyl)amino]ethan-1-ol (10d). A stirred solution of 2-fluoro-5-nitroaniline (9d) (7.805 g, 50 mmol) in pyridine (5 ml, 62 mmol) was treated by the addition of BnNH(CH₂)₂OH (9.374 g, 62 mmol). The resulting solution was stirred for 6 days at reflux. The solvent was removed *in vacuo* and CH₂Cl₂ (400 ml) was added. The organic layer was washed with water, dried over MgSO₄, and concentrated *in vacuo*. Purification by flash chromatography (petroleum ether–EtOAc, 4:1 to 1:1 v/v) afforded the desired product **10d**. Yield 7.155 g (50%), yellow oil. ¹H NMR spectrum, δ , ppm (*J*, Hz): 3.21 (2H, t, J = 5.4, CH₂OH); 3.67 (2H, t, J = 5.4, NCH₂CH₂OH); 4.26 (2H, s, PhCH₂); 7.01 (1H, d, J = 9.0, H Ar); 7.19–7.31 (5H, m, H Ar); 7.52–7.56 (2H, m, H Ar). ¹³C NMR spectrum, δ , ppm: 54.5; 56.7; 59.8; 109.8; 113.5; 122.4; 127.5; 128.5; 128.6; 137.4; 142.9; 143.0; 144.5. Found, *m*/*z*: 288.1351 [M+H]⁺. C₁₅H₁₈N₃O₃. Calculated, *m*/*z*: 288.1343.

2-[(3-Aminopyridin-2-yl)(benzyl)amino]ethan-1-ol (10e). A stirred solution of 2-chloro-3-nitropyridine (9e) (7.925 g, 50 mmol) and BnNH(CH₂)₂OH (8.316 g, 55 mmol) in *n*-BuOH (100 ml) was treated by the addition of Et₃N (8.36 ml, 60 mmol). The resulting solution was stirred for 4 h at reflux. The solvent was removed in vacuo and CH₂Cl₂ (500 ml) was added. The organic layers were washed with saturated NaHCO₃, dried over anhydrous MgSO₄, and concentrated in vacuo. Purification by flash chromatography (petroleum ether-EtOAc, 4:1 to 1:1 v/v) afforded 2-[benzyl(3-nitropyridin-2-yl)amino]ethan-1-ol (11.9 g), which was dissolved in EtOH (300 ml). The resulting solution was hydrogenated over 10% Pd/C catalyst (2.38 g) and stirred for 3 h under a hydrogen atmosphere (balloon). The Pd/C catalyst was filtered off. The resulting clear solution was concentrated in vacuo. Purification by flash chromatography (petroleum ether-EtOAc, 1:1 to 1:5 v/v) afforded the desired product 10e. Yield 9.611 g (79%), white solid, mp 61–63°C. ¹H NMR spectrum, δ , ppm (*J*, Hz): 3.40 (2H, t, J = 4.8, CH₂OH); 3.74 (2H, t, J = 4.8, NCH₂CH₂OH); 4.37 (2H, s, PhCH₂); 6.86 (1H, dd, *J* = 7.8, J = 4.8, H Ar); 6.97 (1H, dd, J = 7.8, J = 1.5, H Ar); 7.24– 7.41 (5H, m, H Ar); 7.74 (1H, dd, *J* = 4.8, *J* = 1.5, H Ar). ¹³C NMR spectrum, δ, ppm: 51.8; 52.8; 59.1; 119.8; 122.6; 127.2; 127.6; 128.7; 136.4; 136.6; 138.6; 150.0. Found, m/z: 244.1452 $[M+H]^+$. C₁₄H₁₈N₃O. Calculated, *m/z*: 244.1444.

Synthesis of compounds 11a-e (General method). Sodium methoxide (6.482 g, 120 mmol) was added to a stirred solution of the appropriate compound 10a-e (20 mmol) and paraformaldehyde (3.6 g, 120 mmol) in MeOH (320 ml). The resulting solution was stirred for 2 h at reflux. When the solution was cooled to 0°C, NaBH₄ (4.536 g, 120 mmol) was added portionwise to the mixture within 10 min. The resulting solution was then stirred for 1 h at reflux. When the solution was cooled to 0°C, the reaction mixture was treated with saturated aqueous citric acid to adjust pH to 7. The solvent was removed in vacuo and water (100 ml) was added. Then the aqueous phase was extracted with CH₂Cl₂ (3×100 ml). The combined CH₂Cl₂ layer was dried over anhydrous MgSO₄ and concentrated in vacuo. Purification by flash chromatography (petroleum ether–EtOAc, 1:1 to 1:5 v/v) afforded the desired product 11a-e.

Methyl 4-[benzyl(2-hydroxyethyl)amino]-3-(methylamino)benzoate (11a). Yield 5.407 g (86%), pale-yellow oil. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.82 (3H, s, NCH₃); 3.11 (2H, t, *J* = 5.4, CH₂OH); 3.51–3.54 (2H, m, NCH₂CH₂OH); 3.88 (3H, s, CO₂CH₃); 4.11 (2H, s, PhCH₂); 7.01 (1H, d, *J* = 8.1, H Ar); 7.15–7.18 (2H, m, H Ar); 7.20–7.29 (4H, m, H Ar); 7.36 (1H, dd, *J* = 8.1, *J* = 1.8, H Ar). ¹³C NMR spectrum, δ , ppm: 30.5; 51.9; 55.1; 58.2; 59.8; 110.9; 118.4; 122.2; 127.1; 127.3; 128.3; 128.8; 137.6; 140.8; 145.4; 167.7. Found, *m/z*: 315.1709 [M+H]⁺. C₁₈H₂₃N₂O₃. Calculated, *m/z*: 315.1703.

2-{Benzyl[2-(methylamino)phenyl]amino}ethan-1-ol (**11b**). Yield 4.613 g (90%), pale-yellow oil. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.73 (3H, s, NCH₃); 3.09 (2H, t, *J* = 5.1, CH₂OH); 3.45 (2H, t, *J* = 5.1, NCH₂CH₂OH); 4.05 (2H, s, PhCH₂); 6.63 (1H, dd, *J* = 8.4, *J* = 1.5, H Ar); 6.73 (1H, td, *J* = 7.8, *J* = 1.5, H Ar); 7.06–7.12 (2H, m, H Ar); 7.19–7.31 (5H, m, H Ar). ¹³C NMR spectrum, δ , ppm: 30.7; 56.5; 60.2; 60.3; 110.6; 117.3; 123.4; 126.4; 127.3; 128.3; 129.1; 136.3; 138.4; 146.5. Found, *m/z*: 257.1650 [M+H]⁺. C₁₆H₂₁N₂O. Calculated, *m/z*: 257.1648.

2-{Benzyl[4-methoxy-2-(methylamino)phenyl]amino}-ethan-1-ol (11c). Yield 5.269 g (92%), pale-yellow oil. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.68 (3H, s, NCH₃); 3.05 (2H, t, *J* = 4.5, CH₂OH); 3.43 (2H, t, *J* = 5.1, NCH₂CH₂OH); 3.78 (3H, s, OCH₃); 3.99 (2H, s, PhCH₂); 6.15 (1H, d, *J* = 2.7, H Ar); 6.25 (1H, dd, *J* = 8.4, *J* = 2.7, H Ar); 6.99 (1H, d, *J* = 8.4, H Ar); 7.20–7.31 (5H, m, H Ar). ¹³C NMR spectrum, δ , ppm: 30.5; 55.3; 57.1; 60.3; 61.0; 97.3; 101.2; 124.2; 127.2; 128.3; 129.1; 129.6; 138.6; 147.9; 158.6. Found, *m/z*: 287.1759 [M+H]⁺. C₁₇H₂₃N₂O₂. Calculated, *m/z*: 287.1754.

2-{Benzyl[2-(methylamino)-4-nitrophenyl]amino}ethan-1-ol (11d). Yield 5.664 g (94%), yellow oil. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.89 (3H, s, NCH₃); 3.17 (2H, t, *J* = 5.4, CH₂OH); 3.64 (2H, t, *J* = 5.4, NCH₂CH₂OH); 4.21 (2H, s, PhCH₂); 6.98 (1H, d, *J* = 8.7, H Ar); 7.16–7.19 (2H, m, H Ar); 7.24–7.29 (3H, m, H Ar); 7.40 (1H, d, *J* = 2.7, H Ar); 7.51 (1H, dd, *J* = 8.7, *J* = 2.7, H Ar). ¹³C NMR spectrum, δ , ppm: 30.3; 54.6; 56.8; 59.7; 104.0; 111.8; 121.6; 127.5; 128.4; 128.6; 137.2; 142.5; 145.2; 145.4. Found, *m*/*z*: 302.1502 [M+H]⁺. C₁₆H₂₀N₃O₃. Calculated, *m*/*z*: 302.1499.

2-{Benzyl[3-(methylamino)pyridin-2-yl]amino}ethan-1-ol (11e). Yield 4.682 g (91%), white solid, mp 105–107°C. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.80 (3H, s, NCH₃); 3.35 (2H, t, *J* = 4.8, CH₂OH); 3.72 (2H, t, *J* = 4.8, NCH₂CH₂OH); 4.29 (2H, s, PhCH₂); 6.84 (1H, dd, *J* = 7.8, *J* = 1.5, H Ar); 6.97 (1H, dd, *J* = 7.8, *J* = 4.8, H Ar); 7.26– 7.37 (5H, m, H Ar); 7.68 (1H, dd, *J* = 4.8, *J* = 1.5, H Ar). ¹³C NMR spectrum, δ , ppm: 30.4; 52.0; 53.3; 59.0; 116.4; 120.4; 127.2; 127.8; 128.7; 134.2; 138.6; 139.5; 149.9. Found, *m/z*: 258.1606 [M+H]⁺. C₁₅H₂₀N₃O. Calculated, *m/z*: 258.1601.

Synthesis of compounds 12a–c (General method). DIPEA (0.784 ml, 4.5 mmol) was added to a stirred solution of the appropriate compound 11a–c (3.0 mmol) in CH₂Cl₂ (30 ml). The resulting solution was cooled to 0°C, and a solution of acryloyl chloride (0.292 ml, 3.6 mmol) in anhydrous CH₂Cl₂ (30 ml) was added dropwise over 2 h. The reaction mixture was washed with saturated aqueous Na₂CO₃ (50 ml). The water layer was extracted with CH_2Cl_2 (4×20 ml). The combined organic layers were dried over anhydrous MgSO₄ and concentrated *in vacuo*. Purification by flash chromatography (petroleum ether– EtOAc, 3:1 to 1:4 v/v) afforded the desired product **12a–c**.

Methyl 4-[benzyl(2-hydroxyethyl)amino]-3-(N-methyl-acrylamido)benzoate (12a). Yield 884 mg (80%), paleyellow oil. ¹H NMR spectrum, δ , ppm (*J*, Hz): 3.23–3.30 (2H, m, CH₂OH); 3.35 (3H, s, NCH₃); 3.67 (2H, t, *J* = 5.7, NCH₂CH₂OH); 3.89 (3H, s, CO₂CH₃); 4.33–4.44 (2H, m, PhCH₂); 5.58 (1H, dd, *J* = 10.2, *J* = 2.1) and 6.46 (1H, dd, *J* = 16.8, *J* = 1.8, COCH=CH₂); 6.12 (1H, dd, *J* = 16.8, *J* = 10.2, COCH=CH₂); 7.09 (1H, d, *J* = 8.7, H Ar); 7.12–7.20 (2H, m, H Ar); 7.24–7.32 (3H, m, H Ar); 7.75 (1H, d, *J* = 2.1, H Ar); 7.91 (1H, dd, *J* = 8.7, *J* = 2.1, H Ar). ¹³C NMR spectrum, δ , ppm: 35.8; 52.1; 52.8; 56.3; 59.8; 121.1; 123.4; 127.5; 127.9; 128.0; 128.3; 128.5; 130.0; 131.9; 134.4; 137.1; 151.1; 166.1; 166.2. Found, *m/z*: 369.1821 [M+H]⁺. C₂₁H₂₅N₂O₄. Calculated, *m/z*: 369.1809.

N-{2-[Benzyl(2-hydroxyethyl)amino]phenyl}-*N*-methylacrylamide (12b). Yield 838 mg (90%), pale-yellow oil. ¹H NMR spectrum, δ, ppm (*J*, Hz): 3.08–3.20 (2H, m, C<u>H</u>₂OH); 3.38 (3H, s, NCH₃); 3.61 (2H, t, J = 5.4, NC<u>H</u>₂CH₂OH); 4.19 (2H, s, PhCH₂); 5.55 (1H, dd, J = 10.2, J = 2.1) and 6.45 (1H, dd, J = 16.8, J = 2.1, COCH=C<u>H</u>₂); 6.13 (1H, dd, J = 16.8, J = 10.2, COC<u>H</u>=CH₂); 7.08–7.17 (4H, m, H Ar); 7.22–7.32 (5H, m, H Ar). ¹³C NMR spectrum, δ, ppm: 36.4; 53.0; 57.7; 59.8; 123.0; 123.8; 127.6; 128.2; 128.4; 128.6; 128.8; 129.0; 130.2; 137.2; 137.4; 147.6; 166.1. Found, *m*/*z*: 311.1762 [M+H]⁺. C₁₉H₂₃N₂O₂. Calculated, *m*/*z*: 311.1754.

N-{2-[Benzyl(2-hydroxyethyl)amino]-5-methoxyphenyl}-*N*-methylacrylamide (12c). Yield 960 mg (94%), paleyellow oil. ¹H NMR spectrum, δ, ppm (*J*, Hz): 3.02–3.08 (2H, m, C<u>H</u>₂OH); 3.36 (3H, s, NCH₃); 3.56 (2H, t, *J* = 5.4, NC<u>H</u>₂CH₂OH); 3.79 (3H, s, OCH₃); 4.05 (2H, s, PhCH₂); 5.53 (1H, dd, *J* = 10.2, *J* = 2.1) and 6.44 (1H, dd, *J* = 16.8, *J* = 2.1, COCH=C<u>H</u>₂); 6.12 (1H, dd, *J* = 16.8, *J* = 10.2, COC<u>H</u>=CH₂); 6.68 (1H, d, *J* = 3.0, H Ar); 6.85 (1H, dd, *J* = 9.0, *J* = 3.0, H Ar); 7.07 (1H, d, *J* = 9.0, H Ar); 7.13– 7.16 (2H, m, H Ar); 7.22–7.31 (3H, m, H Ar). ¹³C NMR spectrum, δ, ppm: 36.6; 54.0; 55.7; 58.9; 59.7; 114.1; 115.2; 124.6; 127.5; 128.0; 128.4 (2C); 129.1; 137.4; 139.0; 140.9; 156.3; 166.0. Found, *m*/*z*: 341.1868 [M+H]⁺. C₂₀H₂₅N₂O₃. Calculated, *m*/*z*: 341.1860.

Synthesis of compounds 12d,e. A solution of trimethylsilyl chloride (0.573 ml, 4.5 mmol) in anhydrous CH_2Cl_2 (4.5 ml) was added dropwise to a stirred solution of compound 11d,e (3.0 mmol) and DIPEA (0.784 ml, 4.5 mmol) in anhydrous CH_2Cl_2 (9 ml) at 0°C. The resulting solution was allowed to warm to 25°C. After 2 h, DIPEA (0.836 ml, 4.8 mmol) and DMAP (73 mg, 0.6 mmol) were added, the resulting solution was cooled to 0°C, and a solution of acryloyl chloride (0.39 ml, 4.8 mmol) in anhydrous CH_2Cl_2 (9 ml) was added dropwise. The reaction mixture was then allowed to warm to 25°C. After 2 h, a solution of TFA (2.23 ml, 30 mmol) in CH_2Cl_2 (4.5 ml) was added dropwise at 0°C. The resulting solution was stirred for 30 min at 0°C. The reaction mixture was then diluted with CH_2Cl_2 (30 ml), washed with saturated aqueous Na₂CO₃ (50 ml). The water layer was extracted with CH_2Cl_2 (4×20 ml). The combined organic layers were dried over anhydrous MgSO₄ and concentrated *in vacuo*. Purification by flash chromatography (3:1 to 1:4 petroleum ether–EtOAc, v/v) afforded the desired product **12 d**,e.

N-{2-[Benzyl(2-hydroxyethyl)amino]-5-nitrophenyl}-N-methylacrylamide (12d). Yield 842 mg (79%), yellow solid, mp 89–90°C. ¹H NMR (6.10 ppm/6.66 ppm) indicated the ratio of two rotamers of compound 12d to be 4:1. ¹H NMR spectrum, δ , ppm (*J*, Hz): 3.33 (2.61H, s, NCH₃); 3.35–3.44 (1.95H, m, CH₂OH); 3.47 (0.67H, s, NCH₃); 3.66-3.80 (1.94H, m, NCH₂CH₂OH); 4.21-4.43 (0.36H, m, PhCH₂); 4.45–4.58 (1.64H, m, PhCH₂); 5.62 (0.81H, dd, J = 10.2, J = 1.8) and 6.48 (0.82H, dd, J = 16.8, J = 16.8)J = 1.8, COCH=CH₂); 5.89 (0.20H, dd, J = 9.9, J = 1.8) and 6.55 (0.18H, d, *J* = 2.1, COCH=CH₂); 6.10 (0.80H, dd, $J = 16.8, J = 10.2, \text{COCH}=\text{CH}_2$; 6.66 (0.20H, dd, J = 16.5, J = 16.5*J* = 10.2, COC<u>H</u>=CH₂); 7.07–7.18 (3.38H, m, H Ar); 7.28– 7.34 (2.75H, m, H Ar); 7.94 (0.56H, d, *J* = 2.7, H Ar); 8.07 (0.17H, d, J = 2.4, H Ar); 8.10 (0.82H, dd, J = 9.0, J = 2.7,H Ar). ¹³C NMR spectrum (major rotamer), δ , ppm: 35.7; 53.0; 56.0; 60.0; 120.4; 124.3; 126.7; 127.7; 127.8; 128.1; 128.8; 129.4; 133.3; 136.7; 140.6; 152.7; 166.1. Found, m/z: $356.1620 [M+H]^+$. C₁₉H₂₂N₃O₄. Calculated, *m/z*: 356.1605.

N-{2-[Benzyl(2-hydroxyethyl)amino]pyridin-3-yl}-*N*-methylacrylamide (12e). Yield 859 mg (92%), paleyellow oil. ¹H NMR spectrum, δ, ppm (*J*, Hz): 3.16 (3H, s, NCH₃); 3.58–3.79 (4H, m, NC<u>H₂CH₂OH</u>); 4.49–4.64 (2H, m, PhCH₂); 5.43 (1H, dd, J = 10.2, J = 2.1) and 6.30 (1H, dd, J = 16.8, J = 2.1, COCH=C<u>H₂</u>); 5.71 (1H, dd, J = 16.8, J = 10.2, COC<u>H</u>=CH₂); 6.85 (1H, dd, J = 7.5, J = 4.8, H Ar); 7.14–7.17 (2H, m, H Ar); 7.19–7.29 (4H, m, H Ar); 8.20 (1H, dd, J = 4.8, J = 1.8, H Ar). ¹³C NMR spectrum, δ, ppm: 35.6; 53.2; 53.8; 60.7; 116.1; 127.0; 127.2; 127.7; 127.9 (2C); 128.4; 138.3; 138.9; 146.5; 156.1; 166.0. Found, *m/z*: 312.1715 [M+H]⁺. C₁₈H₂₂N₃O₂. Calculated, *m/z*: 312.1707.

Synthesis of compounds 8a-e (General method). Compound 12a-e (0.50 mmol) in CH₂Cl₂ (20 ml) was treated with DIPEA (0.697 ml, 4.0 mmol) and DMSO (0.709 ml, 10 mmol). The reaction mixture was cooled to 0-2°C and SO₃-pyridine (320 mg, 2.0 mmol) was added in two portions. The resulting solution was stirred for 2 h at 0-2°C. The reaction mixture was then diluted with CH_2Cl_2 (20 ml), washed with 0.2 M aqueous KHSO₄ (20 ml), water (20 ml), and brine (20 ml), dried over MgSO₄, filtered, and diluted with CH_2Cl_2 to give a solution of crude aldehyde 7a-e in CH₂Cl₂ (50 ml). NaOAc (49 mg, 0.60 mmol) and N-methylhydroxylamine hydrochloride (50 mg, 0.60 mmol) were added to the above solution. The mixture was stirred for the corresponding time at 25°C, then washed with saturated aqueous Na₂CO₃ (25 ml), dried over MgSO₄, and concentrated in vacuo. Purification by flash chromatography (petroleum ether-EtOAc, 5:1 to 1:2 v/v) afforded the desired product 8a-e.

Methyl $(3aR^*,11aR^*)$ -10-benzyl-1,5-dimethyl-4-oxo-1,3,3a,4,5,10,11,11a-octahydrobenzo[b]isoxazolo[3,4-f]-[1,4]diazocine-7-carboxylate (*cis*-8a). Yield 122 mg (62%), white solid, mp 133–134°C. ¹H NMR spectrum, δ, ppm (*J*, Hz): 2.44 (3H, s, ONCH₃); 2.51 (1H, br. s, CH₃NC<u>H</u>); 3.07 (1H, d, J = 14.4) and 3.58 (1H, dd, J = 14.7, J = 9.9, BnNC<u>H</u>₂CH); 3.17 (3H, s, CONCH₃); 3.51 (1H, dd, J = 17.1, J = 8.4, COCH); 3.89 (3H, s, CO₂CH₃); 3.95 (1H, t, J = 8.4) and 4.39 (1H, t, J = 8.1, NOCH₂); 4.51–4.69 (2H, m, PhCH₂); 7.14 (1H, d, J = 8.7, H Ar); 7.23–7.26 (2H, m, H Ar); 7.30–7.39 (3H, m, H Ar); 7.76 (1H, d, J = 2.1, H Ar); 7.89 (1H, dd, J = 8.7, J = 2.1, H Ar). ¹³C NMR spectrum, δ, ppm: 37.0; 43.5; 49.8; 52.1; 52.8; 59.4; 66.7; 70.2; 119.8; 122.7; 127.5; 128.0; 129.0; 130.6; 131.5; 132.2; 137.3; 150.8; 166.1; 170.3. Found, *m/z*: 396.1924 [M+H]⁺. C₂₂H₂₆N₃O₄. Calculated, *m/z*: 396.1918.

Methyl (3aR*,11aS*)-10-benzyl-1,5-dimethyl-4-oxo-1,3,3a,4,5,10,11,11a-octahydrobenzo[b]isoxazolo[3,4-f]-[1,4]diazocine-7-carboxylate (trans-8a). Yield 9 mg (5%), pale-yellow solid, mp 125–126°C. ¹H NMR spectrum, δ, ppm (J, Hz): 2.57 (3H, s, ONCH₃); 2.81 (2H, dd, J = 18.6, J = 6.9, BnNCH₂CH); 3.13 (1H, ddd, J = 15.0, $J = 8.4, J = 6.3, CH_3NCH$; 3.25 (3H, s, CONCH₃); 3.23 (1H, dd, J = 10.2, J = 6.9, COCH); 3.78 (1H, t, J = 8.4) and 4.42 (1H, t, J = 7.2, NOCH₂); 3.93 (3H, s, CO₂CH₃); 4.20 (2H, s, PhCH₂); 7.18–7.26 (4H, m, H Ar); 7.27–7.31 (1H, m, H Ar); 7.61 (1H, d, J = 8.4, H Ar); 7.88 (1H, d, J = 2.1, H Ar); 8.01 (1H, dd, J = 8.4, J = 2.1, H Ar). ¹³C NMR spectrum, δ, ppm: 37.3; 44.2; 51.2; 52.5; 56.9; 61.1; 68.3; 72.7; 127.8; 128.1; 128.6; 128.7; 128.8; 129.2; 130.1; 138.0; 142.0; 153.1; 165.9; 170.6. Found, m/z: 396.1924 [M+H]⁺. C₂₂H₂₆N₃O₄. Calculated, *m/z*: 396.1918.

(3a*R**,11a*R**)-10-Benzyl-1,5-dimethyl-3,3a,5,10,11,11ahexahydrobenzo[*b*]isoxazolo[3,4-*f*][1,4]diazocin-4(1*H*)one (*cis*-8b). Yield 66 mg (39%), pale-yellow solid, mp 97– 98°C. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.47 (1H, br. s, CH₃NC<u>H</u>); 2.52 (3H, s, ONCH₃); 3.00 (1H, d, *J* = 13.2) and 3.42–3.47 (1H, m, BnNC<u>H</u>₂CH); 3.09 (3H, s, CONCH₃); 3.36–3.39 (1H, m, COCH); 3.95 (1H, t, *J* = 8.4) and 4.38 (1H, t, *J* = 8.1, NOCH₂); 4.36 (1H, d, *J* = 15.3) and 4.55 (1H, d, *J* = 15.0, PhCH₂); 6.97–7.08 (2H, m, H Ar); 7.19–7.25 (4H, m, H Ar); 7.28–7.36 (3H, m, H Ar). ¹³C NMR spectrum, δ , ppm: 36.9; 43.7; 50.0; 53.4; 60.3; 67.0; 70.0; 122.9; 123.1; 127.6; 127.9; 128.8 (2C); 129.2; 135.5; 138.2; 147.0; 170.7. Found, *m*/*z*: 338.1871 [M+H]⁺. C₂₀H₂₄N₃O₂. Calculated, *m*/*z*: 338.1863.

(3a*R**,11a*S**)-10-Benzyl-1,5-dimethyl-3,3a,5,10,11,11ahexahydrobenzo[*b*]isoxazolo[3,4-*f*][1,4]diazocin-4(1*H*)one (*trans*-8b). Yield 47 mg (28%), pale-yellow solid, mp 109–110°C. ¹H NMR spectrum, δ, ppm (*J*, Hz): 2.55 (3H, s, ONCH₃); 2.72–2.83 (2H, m, BnNC<u>H</u>₂CH); 3.13–3.21 (1H, m, CH₃NC<u>H</u>); 3.22 (3H, s, CONCH₃); 3.24–3.31 (1H, m, COCH); 3.78 (1H, t, *J* = 8.1) and 4.42 (1H, t, *J* = 8.4, NOCH₂); 4.18 (2H, s, PhCH₂); 7.17–7.25 (4H, m, H Ar); 7.26–7.31 (3H, m, H Ar); 7.33–7.39 (1H, m, H Ar); 7.55 (1H, dd, *J* = 8.1, *J* = 1.5, H Ar). ¹³C NMR spectrum, δ, ppm: 37.2; 44.2; 51.0; 56.8; 61.4; 68.3; 73.0; 126.4; 127.2; 127.5; 127.7; 128.5; 128.7; 129.2; 138.6; 141.8; 148.8; 170.8. Found, *m*/*z*: 338.1873 [M+H]⁺. C₂₀H₂₄N₃O₂. Calculated, *m*/*z*: 338.1863.

(3a*R**,11a*R**)-10-Benzyl-7-methoxy-1,5-dimethyl-3,3a,5,10,11,11a-hexahydrobenzo[*b*]isoxazolo[3,4-*f*][1,4]diazocin-4(1*H*)-one (*cis*-8c). Yield 37 mg (20%), paleyellow oil. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.48 (1H, br. s, CH₃NC<u>H</u>); 2.55 (3H, s, ONCH₃); 2.92–2.99 (1H, m) and 3.30–3.37 (1H, m, BnNC<u>H₂</u>CH); 2.99 (3H, s, CONCH₃); 3.26–3.29 (1H, m, COCH); 3.79 (3H, s, OCH₃); 3.97 (1H, t, *J* = 8.7) and 4.42 (1H, t, *J* = 8.1, NOCH₂); 4.19 (1H, d, *J* = 14.4,) and 4.35 (1H, d, *J* = 14.1, PhCH₂); 6.60 (1H, d, *J* = 3.0, H Ar); 6.85 (1H, dd, *J* = 9.0, *J* = 3.0, H Ar); 7.19–7.33 (6H, m, H Ar). ¹³C NMR spectrum, δ , ppm: 37.0; 44.0; 50.0; 54.2; 55.7; 61.3; 67.2; 69.7; 112.6; 115.0; 126.4; 127.5; 128.4; 128.6; 138.0; 138.7; 139.9; 156.4; 170.7. Found, *m*/*z*: 368.1977 [M+H]⁺. C₂₁H₂₆N₃O₃. Calculated, *m*/*z*: 368.1969.

(3a*R**,11a*S**)-10-Benzyl-7-methoxy-1,5-dimethyl-3,3a,5,10,11,11a-hexahydrobenzo[*b*]isoxazolo[3,4-*f*][1,4]diazocin-4(1*H*)-one (*trans*-8c). Yield 64 mg (35%), white solid, mp 124–125°C. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.56 (3H, s, ONCH₃); 2.70–2.81 (2H, m, BnNC<u>H</u>₂CH); 3.13–3.18 (1H, m, CH₃NC<u>H</u>); 3.19 (3H, s, CONCH₃); 3.23 –3.27 (1H, m, COCH); 3.74–3.79 (1H, m) and 4.41 (1H, t, *J* = 6.9, NOCH₂); 3.80 (3H, s, OCH₃); 4.15 (2H, s, PhC<u>H₂); 6.66 (1H, d, *J* = 3.0, H Ar); 6.90 (1H, dd, *J* = 9.0, *J* = 3.0, H Ar); 7.17–7.25 (4H, m, H Ar); 7.27–7.30 (1H, m, H Ar); 7.45 (1H, d, *J* = 9.0, H Ar). ¹³C NMR spectrum, δ , ppm: 37.0; 44.2; 51.1; 55.6; 57.1; 61.6; 68.3; 73.2; 111.0; 114.9; 127.4; 128.3; 128.4; 128.7; 138.8; 141.5; 142.6; 158.1; 170.8. Found, *m*/*z*: 368.1977 [M+H]⁺. C₂₁H₂₆N₃O₃. Calculated, *m*/*z*: 368.1969.</u>

(3aR*,11aR*)-10-Benzyl-1,5-dimethyl-7-nitro-3,3a,5,10,11,11ahexahydrobenzo[b]isoxazolo[3,4-f][1,4]diazocin-4(1*H*)one (*cis*-8d). Yield 136 mg (71%), yellow solid, mp 169– 170°C. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.41 (3H, s, ONCH₃); 2.52 (1H, br. s, CH₃NC<u>H</u>); 3.14 (1H, d, *J* = 15.3) and 3.67 (1H, dd, *J* = 15.0, *J* = 9.6, BnNC<u>H₂</u>CH); 3.21 (3H, s, CONCH₃); 3.51–3.60 (1H, m, COCH); 3.98 (1H, t, *J* = 8.4) and 4.41 (1H, t, *J* = 8.1, NOCH₂); 4.63 (1H, d, *J* = 15.9) and 4.70 (1H, d, *J* = 16.2, PhCH₂); 7.15 (1H, d, *J* = 9.3, H Ar); 7.24–7.26 (2H, m, H Ar); 7.31–7.42 (3H, m, H Ar); 8.01 (1H, d, *J* = 2.7, H Ar); 8.11 (1H, dd, *J* = 9.3, *J* = 2.7, H Ar). ¹³C NMR spectrum, δ , ppm: 37.3; 43.4; 49.7; 52.8; 59.5; 66.6; 70.2; 118.9; 125.1; 126.4; 127.4; 128.3; 129.2; 131.0; 136.3; 140.1; 152.4; 170.1. Found, *m/z*: 383.1715 [M+H]⁺. C₂₀H₂₃N₄O₄. Calculated, *m/z*: 383.1714.

(3a*R**,11a*R**)-10-Benzyl-1,5-dimethyl-3,3a,5,10,11,11ahexahydroisoxazolo[3,4-*f*]pyrido[3,2-*b*][1,4]diazocin-4(1*H*)one (*cis*-8e). Yield 110 mg (65%), white solid, mp 202– 203°C. ¹H NMR spectrum, δ , ppm (*J*, Hz): 2.22 (3H, s, ONCH₃); 2.26 (1H, br. s, CH₃NC<u>H</u>); 3.13 (1H, d, *J* = 15.3) and 3.52 (1H, dd, *J* = 15.3, *J* = 9.6, BnNC<u>H₂</u>CH); 3.19 (3H, s, CONCH₃); 3.43–3.49 (1H, m, COCH); 3.89 (1H, t, *J* = 8.4) and 4.37 (1H, t, *J* = 7.2, NOCH₂); 4.67 (1H, d, *J* = 14.7) and 5.18 (1H, d, *J* = 14.7, PhCH₂); 6.80 (1H, dd, *J* = 7.5, *J* = 4.8, H Ar); 7.27–7.36 (4H, m, H Ar); 7.37–7.43 (2H, m, H Ar); 8.25 (1H, dd, *J* = 4.8, *J* = 1.8, H Ar). ¹³C NMR spectrum, δ , ppm: 37.2; 43.2; 49.8; 50.0; 55.2; 66.5; 70.3; 115.2; 125.2; 127.6; 128.7 (2C); 137.8; 138.7; 147.7; 155.8; 170.5. Found, *m*/*z*: 339.1823 [M+H]⁺. C₁₉H₂₃N₄O₂. Calculated, *m*/*z*: 339.1816.

X-ray structural study of compounds *cis*-8a and *trans*-8a. Crystals of compound *cis*-8a suitable for X-ray

structural analysis were obtained by slow evaporation from CH₂Cl₂/petroleum ether. Crystals of compound trans-8a suitable for X-ray structural analysis were obtained by slow evaporation from EtOAc/petroleum ether. The X-ray structural study was performed on an automated Rigaku R-AXIS RAPID IP diffractometer (graphite monochromator, λ (MoK α), ω -scanning). Empirical accounting for absorption and the correction of systematic errors were performed with the SADABS software. The structure was solved by a direct method and refined by full matrix method of least squares by F^2_{hkl} with anisotropic thermal parameters for all non-hydrogen atoms, using the SHELXTL package.¹⁴ The hydrogen positions at the carbon atoms were calculated. The complete crystallographic dataset was deposited at the Cambridge Crystallographic Data Center (the deposit numbers are CCDC 1010955 (compound cis-8a) and CCDC 1010956 (compound trans-8a)).

The Supplementary information file containing NMR spectra of the synthesized compounds is available at http://link.springer.com/journal/10593.

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