

# Synthesis of Bowl-Shaped Polycyclic Aromatic Hydrocarbons via Palladium-Catalyzed Intramolecular Arylation Reactions

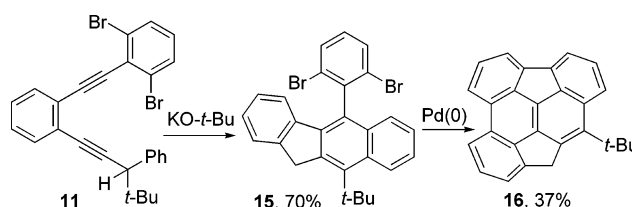
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## ABSTRACT



Treatment of the benzannulated enediyne **11** with potassium *tert*-butoxide in refluxing toluene for 12 h produced **15** via a cascade sequence of cyclization reactions. Two subsequent palladium-catalyzed intramolecular arylation reactions then afforded the bowl-shaped polycyclic aromatic hydrocarbon **16**. The X-ray structures of **16** and two closely related molecules show the presence of significant curvatures.

Bowl-shaped polycyclic aromatic hydrocarbons (buckybowls) have received considerable attention in recent years.<sup>1</sup> This is due in part to the possibility of using these curved hydrocarbons as building blocks for the construction of fullerenes.<sup>2</sup> In addition, the chemical reactivities of the more accessible concave side of the interior carbon atoms of buckybowls could mimic the endohedral chemistry of fullerenes.<sup>3</sup> Furthermore, buckybowls are of interest themselves for a variety of reasons. They provide a platform for the study of the effect of pyramidalization of sp<sup>2</sup>-hybridized carbons on aromaticity.<sup>4</sup> On exposure to lithium metal,

corannulene (**1**), a C<sub>20</sub>H<sub>10</sub> buckybowl, was found to form a fascinating tetraanion.<sup>5</sup> Reports of complexation of buckybowls with transition metals have also begun to emerge.<sup>6</sup>

The first synthesis of corannulene was achieved via a multistep synthetic sequence.<sup>7</sup> The use of flash vacuum pyrolysis to connect distantly separated carbon atoms in

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(1) For reviews, see: (a) Rabideau, P. W.; Sygula, A. *Acc. Chem. Res.* **1996**, *29*, 235–242. (b) Scott, L. T. *Pure Appl. Chem.* **1996**, *68*, 291–300. (c) Mehta, G.; Rao, H. S. P. *Tetrahedron* **1998**, *54*, 13325–13370. (d) Scott, L. T.; Bronstein, H. E.; Preda, D. V.; Ansems, R. B. M.; Bratcher, M. S.; Hagen, S. *Pure Appl. Chem.* **1999**, *71*, 209–219. (e) Scott, L. T. *Angew. Chem., Int. Ed.* **2004**, *43*, 4994–5007.

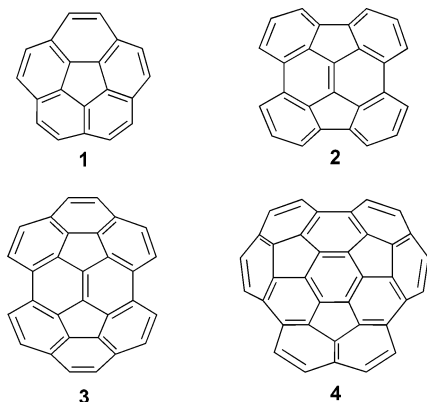
(2) (a) Diederich, F.; Rubin, Y. *Angew. Chem., Int. Ed. Engl.* **1992**, *31*, 1101–1123. (b) McElvany, S. W.; Ross, M. M.; Goroff, N. S.; Diederich, F. *Science* **1993**, *259*, 1594–1596. (c) Sastry, G. N.; Jemmis, E. D.; Mehta, G.; Shah, S. R. *J. Chem. Soc., Perkin Trans. 2* **1993**, 1867–1871.

(3) Faust, R.; Vollhardt, K. P. C. *J. Chem. Soc., Chem. Commun.* **1993**, 1471–1473.

(4) (a) Ferrer, S. M.; Molina, J. M. *J. Comput. Chem.* **1999**, *20*, 1412–1421. (b) Steiner, E.; Fowler, P. W.; Jenneskens, L. W. *Angew. Chem., Int. Ed.* **2001**, *40*, 362–366. (c) Poater, J.; Fradera, X.; Duran, M.; Solà, M. *Chem. Eur. J.* **2003**, *9*, 1113–1122.

(5) (a) Janata, J.; Gendell, J.; Ling, C.-Y.; Barth, W.; Backes, L.; Mark, H. B., Jr.; Lawton, R. G. *J. Am. Chem. Soc.* **1967**, *89*, 3056–3058. (b) Ayalon, A.; Rabinovitz, M.; Cheng, P.-C.; Scott, L. T. *Angew. Chem., Int. Ed. Engl.* **1992**, *31*, 1636–1637. (c) Ayalon, A.; Sygula, A.; Cheng, P.-C.; Rabinovitz, M.; Rabideau, P. W.; Scott, L. T. *Science* **1994**, *265*, 1065–1067. (d) Aprahamian, I.; Preda, D. V.; Bancu, M.; Belanger, A. P.; Sheradsky, T.; Scott, L. T.; Rabinovitz, M. *J. Org. Chem.* **2006**, *71*, 290–298.

(6) (a) Seiders, T. J.; Baldrige, K. K.; O'Connor, J. M.; Siegel, J. S. *J. Am. Chem. Soc.* **1997**, *119*, 4781–4782. (b) Vecchi, P. A.; Alvarez, C. M.; Ellern, A.; Angelici, R. J.; Sygula, A.; Sygula, R.; Rabideau, P. W. *Angew. Chem., Int. Ed.* **2004**, *43*, 4497–4500. (c) Petrukhina, M. A.; Andreini, K. W. *Organometallics* **2005**, *24*, 1394–1397. (d) Stoddart, M. W.; Brownie, J. H.; Baird, M. C.; Schmider, H. L. *J. Organomet. Chem.* **2005**, *690*, 3440–3450. (e) Petrukhina, M. A.; Scott, L. T. *J. Chem. Soc., Dalton Trans.* **2005**, *18*, 2969–2975. (f) Vecchi, P. A.; Alvarez, C. M.; Ellern, A.; Angelici, R. J.; Sygula, A.; Sygula, R.; Rabideau, P. W. *Organometallics* **2005**, *24*, 4543–4552.

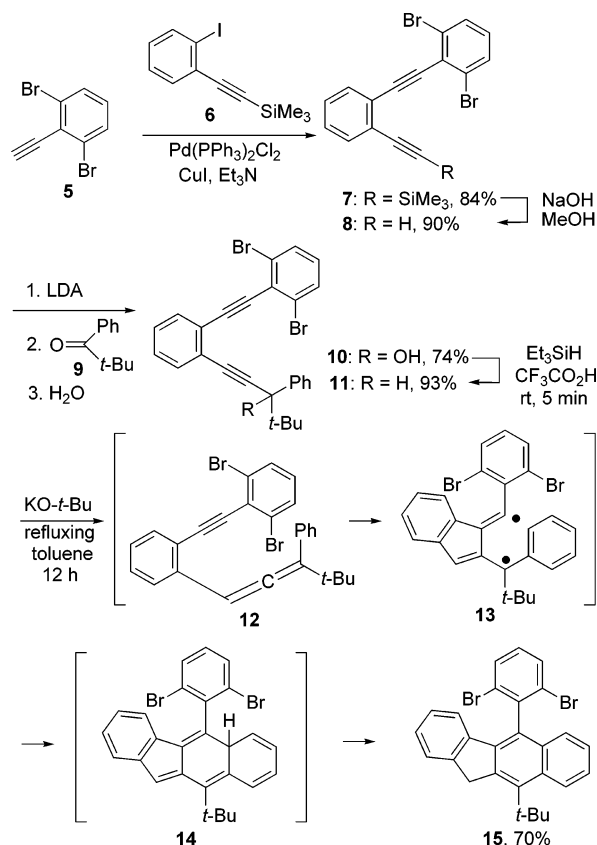


planar polycyclic aromatic precursors provided a more direct route to a host of buckybowl,<sup>1</sup> including corannulene,<sup>8</sup> diindeno[1,2,3,4-*defg*;1',2',3',4'-*mnop*]chrysene (**2**),<sup>9</sup> semi-buckminsterfullerene **3**,<sup>10</sup> and circumtrindene (**4**),<sup>11</sup> culminating in the successful synthesis of C<sub>60</sub> from a C<sub>60</sub>H<sub>27</sub>Cl<sub>3</sub> molecule.<sup>12</sup> Several efficient nonpyrolytic methods for buckybowl were also reported,<sup>13</sup> including the titanium-, vanadium-, and nickel-mediated intramolecular reductive coupling of benzylic and benzylidene bromides,<sup>13a-d,j</sup> the intramolecular carbenoid coupling of dibromomethyl groups,<sup>13g-i</sup> and the palladium-catalyzed intramolecular arylation of aryl halides.<sup>13e,f</sup> We recently reported an efficient route to 5-phenyl-11*H*-benzo[*b*]fluorenes involving condensation between benzannulated enediynes and aryl ketones to produce benzannulated enediynyl propargylic alcohols followed by reduction and a sequence of cascade cyclization reactions.<sup>14</sup> We have successfully adopted this synthetic method to prepare the corresponding 5-(2,6-dibromophenyl)-

11*H*-benzo[*b*]fluorenes for the subsequent palladium-catalyzed intramolecular arylation reactions leading to buckybowl.

The requisite (2,6-dibromophenyl)ethyne (**5**) was prepared by the Sonogashira reaction between 1,3-dibromo-2-iodobenzene and (trimethylsilyl)ethyne followed by desilylation as reported previously.<sup>15</sup> A second Sonogashira reaction between **5** and [(2-iodophenyl)ethynyl]trimethylsilane (**6**)<sup>16</sup> then led to **7**, which was readily desilylated to afford the benzannulated enediyne **8** (Scheme 1). Condensation between

Scheme 1



(7) Barth, W. E.; Lawton, R. G. *J. Am. Chem. Soc.* **1966**, *88*, 380–381. (b) Barth, W. E.; Lawton, R. G. *J. Am. Chem. Soc.* **1971**, *93*, 1730–1745.

(8) (a) Scott, L. T.; Hashemi, M. M.; Meyer, D. T.; Warren, H. B. *J. Am. Chem. Soc.* **1991**, *113*, 7082–7084. (b) Borchardt, A.; Fuchicello, A.; Kilway, K. V.; Baldrige, K. K.; Siegel, J. S. *J. Am. Chem. Soc.* **1992**, *114*, 1921–1923. (c) Zimmermann, G.; Nuechter, U.; Hagen, S.; Nuechter, M. *Tetrahedron Lett.* **1994**, *35*, 4747–4750. (d) Liu, C. Z.; Rabideau, P. W. *Tetrahedron Lett.* **1996**, *37*, 3437–3440. (e) Scott, L. T.; Cheng, P.-C.; Hashemi, M. M.; Bratcher, M. S.; Meyer, D. T.; Warren, H. B. *J. Am. Chem. Soc.* **1997**, *119*, 10963–10968. (f) Mehta, G.; Panda, G. *Tetrahedron Lett.* **1997**, *38*, 2145–2148.

(9) (a) Hagen, S.; Nuechter, U.; Nuechter, M.; Zimmermann, G. *Polycyclic Aromat. Compd.* **1995**, *4*, 209–217. (b) Bronstein, H. E.; Choi, N.; Scott, L. T. *J. Am. Chem. Soc.* **2002**, *124*, 8870–8875.

(10) (a) Rabideau, P. W.; Abdourazak, A. H.; Folsom, H. E.; Marcinow, Z.; Sygula, A.; Sygula, R. *J. Am. Chem. Soc.* **1994**, *116*, 7891–7892. (b) Hagen, S.; Bratcher, M. S.; Erickson, M. S.; Zimmermann, G.; Scott, L. T. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 406–408. (c) Mehta, G.; Panda, G. *Chem. Commun.* **1997**, 2081–2082.

(11) Ansems, R. B. M.; Scott, L. T. *J. Am. Chem. Soc.* **2000**, *122*, 2719–2724.

(12) Scott, L. T.; Boorum, M. M.; McMahon, B. J.; Hagen, S.; Mack, J.; Blank, J.; Wegner, H.; de Meijere, A. *Science* **2002**, *295*, 1500–1503.

(13) (a) Seiders, T. J.; Baldrige, K. K.; Siegel, J. S. *J. Am. Chem. Soc.* **1996**, *118*, 2754–2755. (b) Sygula, A.; Rabideau, P. W. *J. Am. Chem. Soc.* **1998**, *120*, 12666–12667. (c) Sygula, A.; Rabideau, P. W. *J. Am. Chem. Soc.* **1999**, *121*, 7800–7803. (d) Seiders, T. J.; Elliott, E. L.; Grube, G. H.; Siegel, J. S. *J. Am. Chem. Soc.* **1999**, *121*, 7804–7813. (e) Reisch, H. A.; Bratcher, M. S.; Scott, L. T. *Org. Lett.* **2000**, *2*, 1427–1430. (f) Wang, L.; Shevlin, P. B. *Org. Lett.* **2000**, *2*, 3703–3705. (g) Sygula, A.; Marcinow, Z.; Fronczek, F. R.; Guzei, I.; Rabideau, P. W. *Chem. Commun.* **2000**, 2439–2440. (h) Sygula, A.; Rabideau, P. W. *J. Am. Chem. Soc.* **2000**, *122*, 6323–6324. (i) Sygula, A.; Xu, G.; Marcinow, Z.; Rabideau, P. W. *Tetrahedron* **2001**, *57*, 3637–3644. (j) Sygula, A.; Karlen, S. D.; Sygula, R.; Rabideau, P. W. *Org. Lett.* **2002**, *4*, 3135–3137.

**8** and pivalophenone (**9**) then produced enediynyl propargylic alcohol **10**. Treatment of **10** with triethylsilane in the presence of trifluoroacetic acid then afforded **11**. On exposure to potassium *tert*-butoxide in refluxing toluene for 12 h, **11** was converted to 5-(2,6-dibromophenyl)-10-(1,1-dimethylethyl)-11*H*-benzo[*b*]fluorene (**15**) in a single operation. The transformation from **11** to **15** presumably proceeded through an initial 1,3-protropic rearrangement to form the corresponding benzannulated enyne-allene **12**. A Schmitt cyclization reaction<sup>17</sup> to generate biradical **13** for an intramolecular radical–radical coupling to afford, in situ, **14** followed by a

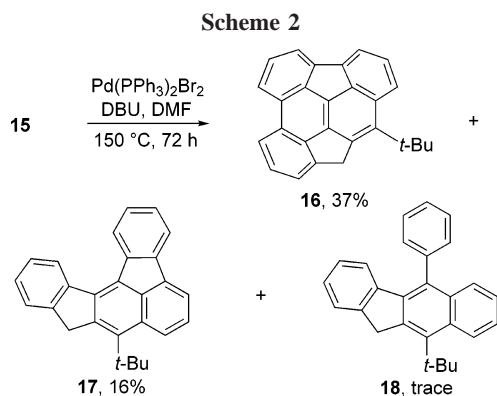
(14) Li, H.; Zhang, H.-R.; Petersen, J. L.; Wang, K. K. *J. Org. Chem.* **2001**, *66*, 6662–6668.

(15) Bradshaw, J. D.; Guo, L.; Tessier, C. A.; Youngs, W. J. *Organometallics* **1996**, *15*, 2582–2584.

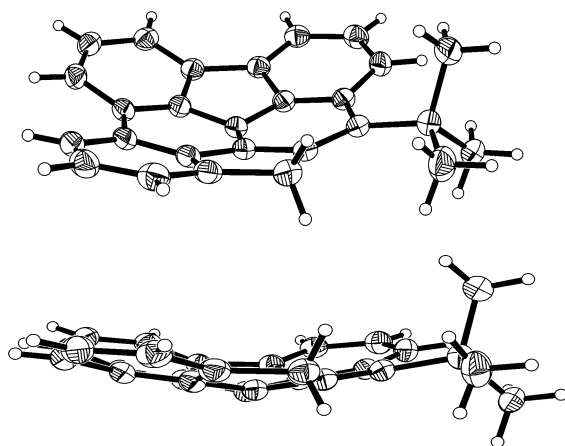
(16) Grilli, S.; Lunazzi, L.; Mazzanti, A.; Pinamonti, M. *Tetrahedron* **2004**, *60*, 4451–4458.

second prototropic rearrangement to regain aromaticity then furnished **15** as proposed previously.<sup>14</sup>

It was gratifying to observe that treatment of **15** with 10 mol % of Pd(PPh<sub>3</sub>)<sub>2</sub>Br<sub>2</sub> in the presence of DBU under the conditions reported by Scott et al.<sup>13e</sup> produced buckybowl **16** (Scheme 2). The structure of **16** was confirmed by X-ray



structure analysis (Figure 1). Similar to what was observed

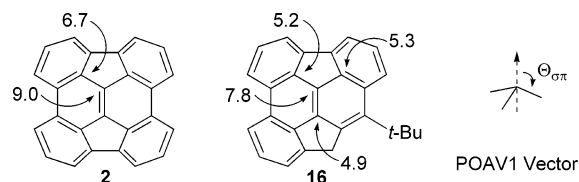


**Figure 1.** X-ray crystal structure of **16**.

previously,<sup>13e</sup> reductive debromination either on one side or on both sides also occurred, giving rise to the monoclosed hydrocarbon **17** as a minor product and a small amount of **18**, respectively. The structure of **17** was tentatively assigned based on the assumption that the first cyclopalladation reaction of **15** occurred preferentially with the naphthyl moiety of **15** producing a five-membered ring as reported previously.<sup>18</sup> The resulting brominated fluoranthene intermediate could undergo either a second cyclopalladation reaction to produce **16** or a reductive debromination reaction

to give **17**. Analysis of the crude reaction mixture by GC/MS before purification by silica gel chromatography also suggests the presence of a small amount of **18**.

The X-ray structure of **16** indicates the presence of a significant curvature. Compared to **2**, the structure of **16** appears to be less strained with one less ring due to the absence of three sp<sup>2</sup>-carbons in the southeastern corner. The pyramidalization angle, defined as  $\Theta_{\sigma\pi} - 90$  using the  $\pi$ -orbital axis vector analysis (POAV1),<sup>19</sup> of the central ethylene carbon atoms of **2** was reported to be 9.0°, whereas the pyramidalization angle of the four carbon atoms attached to the central double bond was determined to be 6.7° (Figure 2).<sup>9b</sup> The POAV1 angles of these carbon atoms



**Figure 2.** POAV1 pyramidalization angles ( $\Theta_{\sigma\pi} - 90$ ) of **2** and **16**.

are clearly larger than those of the corresponding carbon atoms of **16** based on the geometry obtained from the X-ray analysis.

The <sup>1</sup>H NMR signal of the diastereotopic methylene hydrogens of **16** occurs as a singlet at  $\delta$  4.75 ppm. Barring accidental isochrony, this observation suggests a rapid bowl-to-bowl inversion on the NMR time scale at room temperature, reminiscent of what was observed previously in substituted corannulene derivatives.<sup>13a,20</sup>

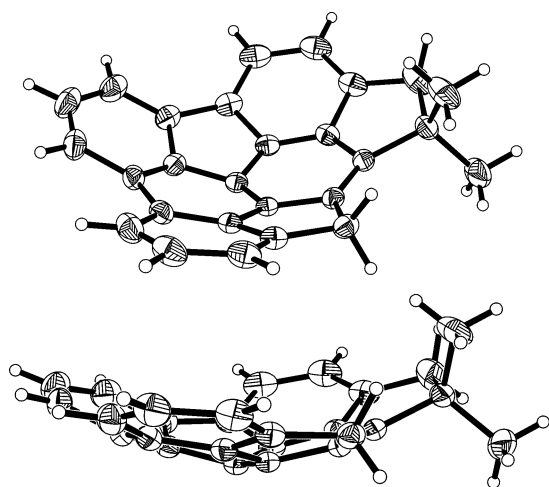
Similarly, by using aryl ketones **19a**<sup>14</sup> and **19b**<sup>14</sup> for condensation with the lithium acetylide of **8**, benzannulated enediynyl propargylic alcohols **20a** and **20b** were obtained, respectively. Reduction with triethylsilane in the presence of trifluoroacetic acid then gave benzannulated enediynes **21a** and **21b**. Treatment of **21a** and **21b** with potassium *tert*-butoxide in refluxing toluene for 12 h then furnished **22a** and **22b**, respectively. The palladium-catalyzed intramolecular arylation reactions of **22a** produced buckybowl **23a** and the monocyclized hydrocarbon **24a**. The X-ray structure of **23a** also indicates the presence of a significant curvature. Similarly, buckybowl **23b** along with the monocyclized hydrocarbon **24b** were likewise produced from **22b**. The structure of **23b** was also confirmed by X-ray structure analysis (Figure 3). The presence of an additional five-

(19) (a) Haddon, R. C.; Scott, L. T. *Pure Appl. Chem.* **1986**, *58*, 137–142. (b) Haddon, R. C. *J. Am. Chem. Soc.* **1987**, *109*, 1676–1685. (c) Haddon, R. C. *Science* **1993**, *261*, 1545–1550.

(20) (a) Scott, L. T.; Hashemi, M. M.; Bratcher, M. S. *J. Am. Chem. Soc.* **1992**, *114*, 1920–1921. (b) Sygula, A.; Rabideau, P. W. *THEOCHEM* **1995**, *333*, 215–226. (c) Biedermann, P. U.; Pogodin, S.; Agranat, I. *J. Org. Chem.* **1999**, *64*, 3655–3662. (d) Marcinow, Z.; Sygula, A.; Ellern, A.; Rabideau, P. W. *Org. Lett.* **2001**, *3*, 3527–3529. (e) Seiders, T. J.; Baldrige, K. K.; Grube, G. H.; Siegel, J. S. *J. Am. Chem. Soc.* **2001**, *123*, 517–525.

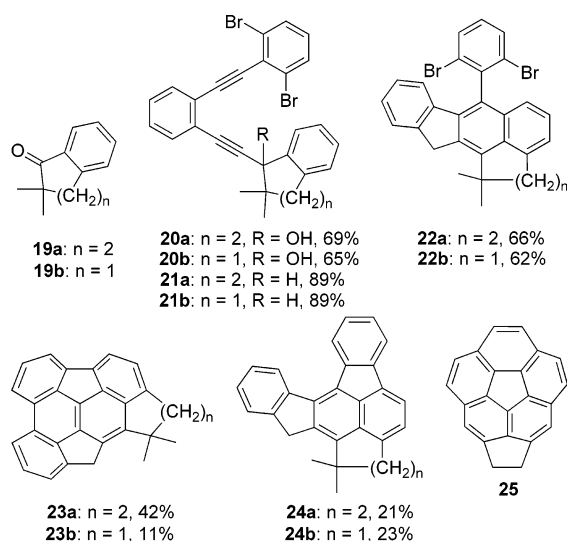
(17) (a) Schmittl, M.; Strittmatter, M.; Vollmann, K.; Kiau, S. *Tetrahedron Lett.* **1996**, *37*, 999–1002. (b) Schmittl, M.; Strittmatter, M.; Kiau, S. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 1843–1845.

(18) Rice, J. E.; Cai, Z.-W. *J. Org. Chem.* **1993**, *58*, 1415–1424.

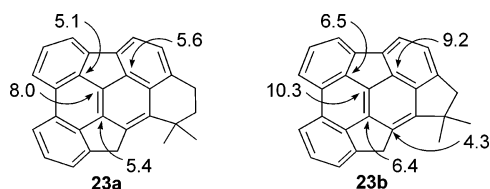


**Figure 3.** X-ray crystal structure of **23b**.

membered ring in **23b** appears to cause its structure to be more strained than those of **16** and **23a**. As a result, the transformation from **22b** to **23b** is less efficient.



Based on the geometry obtained from the X-ray analysis, one of the central carbon atoms of **23b** has a POAV1 angle of  $10.3^\circ$  (Figure 4), which is larger than those of **2** and



**Figure 4.** POAV1 pyramidalization angles of **23a** and **23b**.

atoms having a POAV1 angle of  $11.64^\circ$ . In addition, one of the four carbon atoms attached to the central double bond of **23b** has a POAV1 angle of  $9.2^\circ$ , which is also larger than those of the corresponding carbon atoms of **2**.

It was reported that, unlike corannulene, dihydrocyclopentacorannulene (**25**) with an additional five-membered ring “locked” the bowl configuration, preventing a rapid bowl-to-bowl inversion process.<sup>21</sup> However, no such effect was observed for **23b** with the two sets of the methylene hydrogens appeared as two sharp singlets at  $\delta$  4.41 and 3.29 ppm and the two groups of methyl hydrogens appeared as a sharp singlet at  $\delta$  1.63 ppm at  $25^\circ\text{C}$ . At  $-33^\circ\text{C}$ , the methylene signals, recorded on a 600-MHz NMR spectrometer, became two broad humps ranging from ca.  $\delta$  4.6 to 4.2 ppm and from ca.  $\delta$  3.4 to 3.1 ppm, and the methyl signal also appeared as a broad peak ranging from ca.  $\delta$  1.8 to 1.5 ppm. Further broadening but without the appearance of distinct signals for the diastereotopic hydrogens was observed at  $-40^\circ\text{C}$ , suggesting relatively fast bowl-to-bowl inversion on the NMR time scale even at  $-40^\circ\text{C}$ .

A new synthetic route to buckybowl was developed. The convergent assembly of the benzannulated enediyne propargylic alcohols and the efficient cascade cyclizations of the resulting enyne-allenes leading to polycyclic aromatic dibromides for subsequent palladium-catalyzed intramolecular arylation reactions are the key features of this synthetic pathway. The use of benzannulated enediyne propargylic alcohols to form the corresponding aromatic dibromides represents a new approach to the preparation of buckybowl precursors. Three new buckybowl precursors with their structures confirmed by X-ray structure analyses were thus synthesized.

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**Supporting Information Available:** Experimental procedures, spectroscopic data, and  $^1\text{H}$  and/or  $^{13}\text{C}$  NMR spectra of **7**, **8**, **10**, **11**, **15–17**, and **20a,b–24a,b**; ORTEP drawings for **16**, **23a**, and **23b**; and X-ray crystallographic data of **16**, **23a**, and **23b** (CIF). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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corresponds to 88% of that found for  $\text{C}_{60}$  with all carbon

(21) Abdourazak, A. H.; Sygula, A.; Rabideau, P. W. *J. Am. Chem. Soc.* **1993**, *115*, 3010–3011.