

Available online at www.sciencedirect.com



Tetrahedron Letters 46 (2005) 5239-5242

Tetrahedron Letters

A new synthetic approach to enantiomerically enriched dihydrobenzofurans: use of a hydrolytic kinetic resolution and an intramolecular epoxide ring opening protocol using 1-benzyloxy-2-oxiranylmethylbenzenes

Umadevi Bhoga*

Organic Division, Indian Institute of Chemical Technology, Hyderabad 500 007, India Received 7 January 2005; revised 29 March 2005; accepted 15 April 2005

Abstract—A novel approach towards the preparation of racemic 1-benzyloxy-2-oxiranylmethylbenzenes using dimethyldioxirane and their hydrolytic kinetic resolution using (R, R)(Salen)Co(III)(OAc) (Jacobsen's catalyst) to afford the (R)-epoxides and (S)-1,2-diols, enantioselectively, is described. The (R)-1-benzyloxy-2-oxiranylmethylbenzenes were then cyclized via an intramolecular epoxide opening reaction to give (S)-2-hydroxymethyl-2,3-dihydrobenzofurans. © 2005 Published by Elsevier Ltd.

Naturally occurring 2-substituted-2,3-dihydrobenzofurans are an important class of biologically active oxygen-containing heterocycles.^{1–6} Natural products possessing the dihydrobenzofuran nucleus, exhibit a wide range of biological activities. For example, arthrographol is an anti-fungal,⁷ megapodiol is an antileukaemic,⁸ 2-(3,4-dimethoxyphenyl)-3-hydroxymethyl-5-(3-hydroxypropyl)-2,3-dihydrobenzofuran-7-ol⁹ and conocarpan are anti-cancer agents,¹⁰ tremetone causes milk sickness in cattle¹¹ and furaquinocines are antibiotics.¹² There are only a few methods, to the best of our knowledge, established for the synthesis of enantiomerically enriched dihydrobenzofurans.^{7–12}

Herein, we report a new synthesis of enantiomerically enriched dihydrobenzofuran derivatives and naphtho[1,2-*b*]furans. *o*-Allylphenols $2\mathbf{a}-\mathbf{c}^{13-15}$ were converted to the corresponding benzylethers $3\mathbf{a}-\mathbf{c}$,¹⁶ in order to be able to isolate the epoxy derivatives in the subsequent step. The resulting *o*-allylbenzylethers $3\mathbf{a}-\mathbf{c}$ were reacted with dimethyldioxirane^{17–20} generated in situ from Oxone and acetone to furnish racemic 1-benzyloxy-2-oxiranylmethylbenzenes $4\mathbf{a}-\mathbf{c}$.²¹ Next, the solvent-free hydrolytic kinetic resolution (HKR) of the racemic 1-benzyloxy-2-oxiranylmethylbenzenes $4\mathbf{a}-\mathbf{c}$,

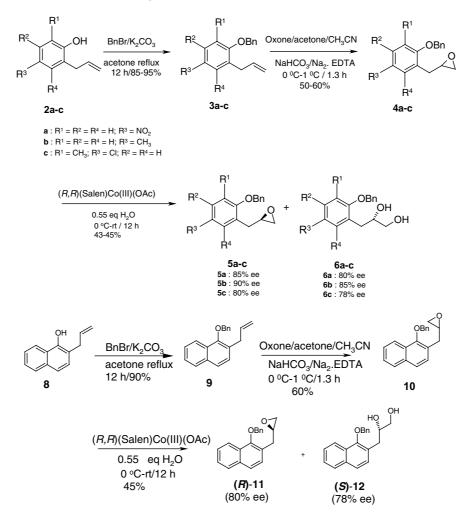
0040-4039/\$ - see front matter @ 2005 Published by Elsevier Ltd. doi:10.1016/j.tetlet.2005.04.120

using the chiral salen cobalt complex, (R,R)(Salen)-Co(III)(OAc),^{22,23} gave the optically pure (*R*)-epoxides **5a–c** and (*S*)-1,2-diols **6a–c** in 80–90% ee's and 78–85% ee's, respectively (Scheme 1).²⁴

Similarly, *o*-allylnaphthol **8** was converted to benzyl ether **9** on reaction with benzyl bromide and anhydrous potassium carbonate in acetone. Benzyl ether **9** was subsequently treated with dimethyldioxirane to afford racemic epoxide **10**, which underwent solvent-free HKR using Jacobsen's catalyst to give (R)-epoxide **11** and (S)-1,2-diol **12** in 80% ee and 78% ee, respectively (Scheme 1).

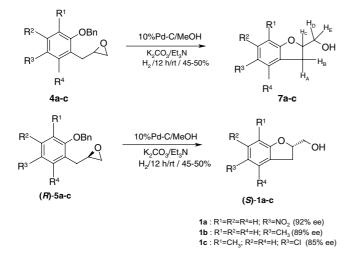
The racemic epoxides **4a**–**c** and **10** and (*R*)-epoxides **5a**–**c** and **11** were debenzylated and cyclized in situ via an intramolecular epoxide opening reaction. This was accomplished using 10% Pd–C in methanol under a H₂ atmosphere and adding a catalytic amount of triethylamine and anhydrous potassium carbonate at room temperature²⁵ to furnish the desired products, racemates **7a–c** and **13** and enantiomerically enriched **1a–c** and **13a** in 85–92% ee's and 75% ee, respectively,²⁶ via a S_N2 mechanism (Schemes 2 and 3). The $[\alpha]_D^{25}$ of **1b** was +20.0 (*c* 1.5, CHCl₃), which compared favourably with that published, (lit. +21.0, *c* 1.0, CHCl₃).²⁷ The formation of (*S*)-5-methyl-2-hydroxymethyl-2,3-dihydrobenzofuran **1b** from (*R*)-epoxide **5b** was confirmed by

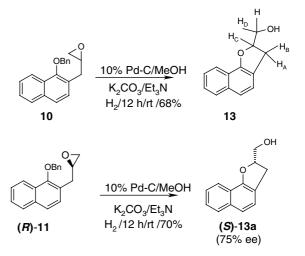
^{*}E-mail: ubhoga@yahoo.co.in



Scheme 1.

Scheme 2.





Scheme 3.

comparison with a sample prepared from enzymatic kinetic resolution by a standard procedure²⁷ using a lipase.

In summary, a new synthesis of optically pure 2hydroxymethyl-2,3-dihydrobenzofurans and a naphtho[1,2-*b*]furan in 85–92% ee's and 75% ee, respectively, was accomplished by debenzylation and in situ cyclization via an intramolecular epoxide ring opening reaction of **5a–c** and **11**, which in turn were prepared from HKR in 80–90% ee's and 78% ee, respectively.

Acknowledgements

The author thanks Dr. J. S. Yadav, Director, IICT, Hyderabad, Dr. B. V. Rao and the late Dr. A. K. Singh for their constant encouragement and fruitful discussions during this work. U.B. also thanks CSIR, New Delhi for financial support and for a Senior Research Fellowship.

References and notes

- 1. Nascimento, I. R.; Lopes, L. M. X. Phytochemistry 1999, 52, 345.
- Benevides, P. J. C.; Sartorelli, P.; Kato, M. J. *Phytochem*istry **1999**, 52, 339.
- 3. Ishii, H.; Ishikawa, T. Tetrahedron Lett. 1982, 23, 4345.
- Bohlmann, F.; Scheidges, C.; Zdero, C.; King, R. M.; Robinson, H. *Phytochemistry* 1984, 23, 1109.
- Lau, C. K.; Belanger, P. C.; Dufresne, C.; Scheigetz, J.; Therien, M.; Fitzsimmons, B.; Young, R. N.; Ford-Hutchinson, A. W.; Riendeau, D.; Denis, D.; Guay, J.; Charleson, C.; Piechuta, H.; McFarlane, C. S.; Lee Chiu, S. H.; Eline, D.; Alvaro, R. F.; Miwa, G.; Walsh, J. L. J. Med. Chem. 1992, 35, 1299.
- Hellberg, M. R.; Namil, A.; Delgado, P.; David, K. C.; Kessler, T. L.; Graff, G.; Haggard, K. S.; Nixon, J. C. J. Med. Chem. 1999, 42, 267.
- Pfefferle, W.; Anke, H.; Bross, M.; Steffan, B.; Vianden, R.; Steglich, W. J. Antibiot. 1990, 43, 649.
- Jarvis, B. B.; Pena, B. N.; Comezoglu, N. S.; Rao, M. M. Phytochemistry 1986, 25, 533.
- Pieters, L.; Bruyne, T. D.; Claeys, M.; Vlietinck, A. J. Nat. Prod. 1993, 56, 899.
- Achenbach, H.; Usubillaga, A.; Utz, W.; Rodriguez, H. A. Phytochemistry 1991, 30, 3753.
- Bonner, W. A.; Burke, N. I.; Fleck, W. E.; Hill, R. K.; Joule, J. A.; Sjoberg, B.; Zalkow, J. H. *Tetrahedron* 1964, 20, 1419.
- Saito, T.; Suzuki, T.; Morimoto, M.; Akiyama, C.; Ochiai, T.; Takeuchi, K.; Matsumoto, T.; Suzuki, K. J. Am. Chem. Soc. 1998, 120, 11633.
- White, W. N.; Gwynn, D.; Schlitt, R.; Girard, C.; Fife, W. J. Am. Chem. Soc. 1958, 80, 3271.
- 14. Svanholm, U.; Parker, V. D. J. Chem. Soc., Perkin Trans. 2 1974, 169.
- 15. Lasek, W.; Makosza, M. Synthesis 1993, 8, 780.
- 16. General procedure for the synthesis of 1-benzyloxy-2-allyl-4-nitrobenzene 3a. A mixture of 2-allyl-4-nitrophenol (350 mg, 1.95 mmol) and anhydrous K_2CO_3 (540 mg, 4 mmol) in dry acetone (10 mL) was allowed to stir for 30 min at room temperature. To this was added benzyl bromide (334 mg, 1.95 mmol) dropwise and the resulting mixture refluxed for 12 h at 60 °C. After completion of the reaction (TLC), the solvent was evaporated under reduced pressure to afford a residue, which was extracted with dichloromethane (5×10 mL). The combined organic layer was washed with water and dried over anhydrous Na₂SO₄. Evaporation of the solvent gave a residue, which was purified by column chromatography over silica gel using pet. ether as eluent to furnish **3a** as a semi-solid (490 mg, 94%), (3b,c and 9 were prepared similarly). ¹H NMR (200 MHz, CDCl₃): δ 3.41 (m, 2H) 5.20 (m, 4H), 5.99 (m, 1H), 6.89 (m, 1H), 7.40 (m, 5H), 8.00 (m, 2H); MS (EI): m/z 269 (M⁺, 100%).
- Adam, W.; Hadjiarapoglou, L. P.; Curci, R.; Mello, R.. In Organic Peroxides; Ando, W., Ed.; Wiley: New York, 1992; Chapter 4, p 195.

- Curci, R. In Advances in Oxygenated Processes; Baumstark, A., Ed.; JAI: Greenwich, CT, 1990; Vol. 2, Chapter 1, p 1.
- 19. (a) Murray, R. W. Chem. Rev. 1989, 89, 1187; (b) Dimethyldioxirane is a selective, reactive oxidizing agent, capable of epoxidation of alkenes, arenes, imines and sulfides. It can be prepared by portionwise addition of solid Oxone (potassium monoperoxysulfate) to a vigorously stirred solution of NaHCO₃ in a mixture of reagent grade acetone and distilled water at 0 °C. Here, Oxone serves as a stoichiometric oxidizing agent under a variety of conditions. Therefore, an aqueous solution of Oxone can be used to perform oxidations in homogeneous solution and in biphasic systems using an immiscible cosolvent and a phase transfer catalyst at pH 7 to prevent hydrolysis of the epoxide.
- 20. Adam, W.; Edwards, J. O.; Curci, R. Acc. Chem. Res. 1989, 22, 205.
- 21. Typical procedure for the synthesis of (\pm) -1-benzyloxy-2oxiranylmethyl-4-nitrobenzene 4a. To an acetonitrile solution (54 mL) of olefin **3a** (1.74 g, 7.3 mmol) was added Na₂. EDTA solution (36.3 mL, 4×10^{-4} M) and the resulting homogeneous solution was cooled to 0 °C. To this was added precooled acetone (7.3 mL) using a precooled syringe. After 10 min, a mixture of NaHCO₃ (4.75 g, 56 mmol) and Oxone (22.5 g, 36 mmol) was added in portions over a period of 1 h while maintaining a pH of 7. The reaction was complete in 1.3 h by TLC. The reaction mixture was poured into water (100 mL) and extracted with CH₂Cl₂ (250 mL). The combined extracts were washed with brine and then dried over anhydrous Na2SO4. Evaporation of the solvent gave a residue, which was purified by column chromatography using pet. ether to afford 4a as an oil (50%). Similarly, 4b,c and 10 were prepared in 50-60% yields, respectively. ¹H NMR (200 MHz, CDCl₃): δ 2.50 (m, 1H), 2.69 (m, 1H), 2.99 (m, 2H), 3.20 (m, 1H), 5.20 (s, 2H), 6.99 (d, J = 10.0 Hz, 1H), 7.40 (m, 5H), 8.20 (m, 2H); MS (EI): *m/z* (M⁺, 285, 100%).
- Jacobsen, E. N.; Kakiuchi, F.; Konsler, R. G.; Larrow, J. F.; Tokunaga, M. *Tetrahedron Lett.* 1997, 38, 773.
- 23. Larrow, J. F.; Schaus, S. E.; Jacobsen, E. N. J. Am. Chem. Soc. 1996, 118, 7420.
- 24. General procedure for the hydrolytic kinetic resolution of (\pm) -1-benzyloxy-2-oxiranylmethyl benzenes. To the neat epoxide **4a** (270 mg, 0.95 mmol) was added 0.2 mol % of (R,R)(Salen)Co(III)(OAc) catalyst (50 mg), portionwise, over a period of 15 min. To the reaction mixture was added 9.30 µL of water (0.55 equiv), dropwise, using a Hamilton syringe over a period of 10 min at 0 °C. The resulting slurry was allowed to stir for 12 h at room temperature, then the reaction mixture was filtered and the residue washed with chloroform (5 mL). The filtrate was evaporated to afford a residue, which was chromatographed over neutral alumina using 1:19 ethyl acetate-pet. ether to furnish (*R*)-epoxide **5a** and (*S*)-1,2-diol **6a** in 85% ee and 80% ee, respectively.

(a) (*R*)-1-Benzyloxy-2-oxiranylmethyl-4-nitrobenzene **5a**. Yield: 45%, $[\alpha]_D^{25}$ +12.2 (*c* 1.0, CHCl₃), HPLC {column chiracel OD, 0.5:9.5 'PrOH/*n*-hexane, flow rate: 1 mL/min, ee: 85%}.

(b) (*S*)-1-Benzyloxy-2-(2',3'-dihydroxy)propanyl-4-nitrobenzene **6a**. Yield: 43%, ¹H NMR (200 MHz, CDCl₃): δ 2.40 (br s, 1H), 2.82 (m, 2H), 3.30 (m, 3H), 4.10 (m, 1H), 5.20 (s, 2H), 6.90 (d, *J* = 10.0 Hz, 1H), 7.40 (m, 5H), 8.10 (m, 2H); $[\alpha]_{D}^{25}$ +10.1 (*c* 1.0, CHCl₃), HPLC {column chiracel OD, 0.5:9.5 'PrOH/*n*-hexane, flow rate: 1 mL/min, ee: 80%}.

 Rao, B. V.; Sitharamaiah, D. Unpublished results, Indian Institute of Chemical Technology, Hyderabad, India, 1998. 26. Typical procedure for the preparation of (±)-and (S)-2-hydroxymethyl-2,3-dihydrobenzofurans and naphtho-furans: A mixture of racemic epoxide 4a (280 mg, 1.10 mmol), 10% Pd–C (20 mg), a catalytic amount of triethylamine and anhydrous potassium carbonate (5 mg) in methanol (10 mL) was stirred under H₂ balloon pressure for 12 h at room temperature. After the reaction was complete (TLC), the catalyst and the solid impurities were removed by filtration and washed with methanol (15 mL). The combined filtrate was evaporated under reduced pressure to furnish a residue, which was purified by column chromatography using 1:19 ethyl acetate–pet. ether as eluent to afford 7a. Compounds 1a–c, 7b,c, 13 and 13a were prepared similarly.

(a) (S)-5-Nitro-2-hydroxymethyl-2,3-dihydrobenzofuran 1a. Yield: 50%; ¹H NMR (200 MHz, CDCl₃): δ 1.80 (s, 1H, OH), 3.20 (m, 1H, H_A), 3.40 (m, 1H, H_B), 3.80 (m, 1H, H_D), 3.95 (m, 1H, H_E), 5.10 (m, 1H, H_C), 6.80 (d, J =10.0 Hz, 1H), 8.10 (m, 2H); MS (EI): m/z 195 (M⁺, 100%), 149 (M⁺-NO₂, 80%), 164 (55), 141 (34), 118 (25), 89 (10); $[\alpha]_D^{25}$ +14.00 (c 1.5, CHCl₃); HPLC {column chiracel OD, 0.5:9.5 ⁱPrOH/*n*-hexane, flow rate: 1 mL/min, ee: 92%}. (b) (S)-2-Hydroxymethyl-2,3-dihydronaphtho[1,2-*b*]furan **13a**. Yield: 70%; ⁱH NMR (200 MHz, CDCl₃): δ 2.00 (br s, 1H, OH), 3.25 (m, 1H, H_A), 3.50 (m, 1H, H_B), 3.80 (m, 1H, H_D), 3.90 (m, 1H, H_E), 5.10 (m, 1H, H_C), 7.00 (m, 1H), 7.30 (m, 1H), 7.45 (m, 1H), 7.52 (m, 1H), 7.65 (m, 1H), 7.80 (m, 1H); MS (EI): m/z 200 (M⁺, 100%); $[\alpha]_D^{25}$ +38.0 (c 1.0, CHCl₃); HPLC {column chiracel OD, 0.5:9.5 ⁱPrOH/*n*-hexane, flow rate: 1 mL/min, ee: 75%}.

Ramdas, S.; Krupadanam, D. G. L. Tetrahedron: Asymmetry 2000, 11, 3375.