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## Efficient Synthesis of Fused Bicyclic Glutarimides. Its Application to (±)-Alloyohimbane and Louisianin D

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## ABSTRACT

The reaction of  $\alpha$ -sulfonyl acetamide 1 with various cyclic unsaturated esters 2 to fused bicyclic glutarimides is reported. Syntheses of  $(\pm)$ -alloyohimbane (4) and louisianin D (5) have been accomplished.

Bicyclic pyridines, piperidines,  $\delta$ -lactams, and 2-pyridones are important core structures that are found in numerous biologically active compounds.<sup>1</sup> Although many methods have been reported for the synthesis of such compounds,<sup>2</sup> we envisioned that our previously developed [3+3] annulation of  $\alpha$ -sulfonyl acetamide with  $\alpha,\beta$ -unsaturated esters to give polysubstituted glutarimides<sup>3</sup> would be ideal for

constructing fused bicyclic glutarimides which could be further converted to nitrogen-containing polycyclic alkaloids. <sup>2b,4</sup>

Thus, the reaction of  $\alpha$ -sulfonyl acetamide 1 with various cyclic unsaturated esters 2 was investigated. The results are shown in Table 1. It is interesting to note that 3a and 3b are both cis-fused bicyclic compounds. The structures of 3a and 3b were unequivocally established by single-crystal X-ray

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Table 1. Formation of Fused Bicyclic Glutarimides

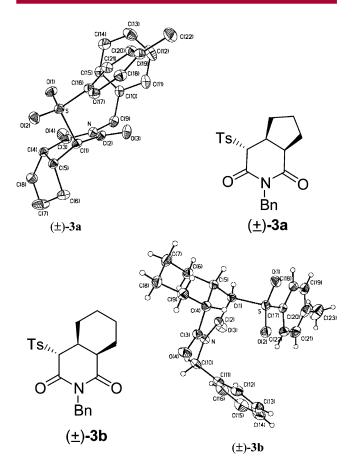
	Ŕ <b>1</b>	2	R 3
Entry	α-sulfonyl amide	Michael acceptor	product (yleid %)
1	Ts NH Bn	EtOO	Ts., 3 4 5 5 N O Bn
2	Ts NH Bn 1a	2a Eto O	(±)-3a(88%) <sup>a</sup> Ts.,, O  Bn (±)-3b (83%) <sup>a</sup>
3	Ts ONH Bn	EtOO	Ts <sub>(i)</sub>
4	Ts NH PMB	2c N Bn MeO O	Ts <sub>//,</sub> PMB (±)-3d (64%) <sup>a</sup>
5	Ts N O NH	EIO 0 2e	Ts <sub>//,</sub> N O N O (±)-3e (35%) <sup>9</sup>

 $^{a}$  All yields were based on  $\alpha$ -toluenesulfonyl acetamide.

analysis (Figure 1). The stereochemistries of 3c-e were determined by comparing their <sup>1</sup>H NMR spectra with those of 3a and 3b.

To demonstrate the utility of this one-pot process, the formal synthesis of  $(\pm)$ -alloyohimbane (4) was investigated. As shown in Scheme 1, regioselective reduction of 3e by sequential addition of triethylamine in THF and LAH reduction at refluxing temperature furnished 6. Treatment of 6 with sodium amalgam gave 4,5-annulated lactam 7. The spectral data of 7 were in agreement with those reported in the literature.  $^{4a}$  Lactam 7 has been converted to alloyohimbane (4).  $^{4a,5}$  Thus, the formal synthesis of alloyohimbane (4) was accomplished.

For the synthesis of louisianin D (5)<sup>4b</sup> produced by a species of *Streptomyces*,<sup>6</sup> glutarimide **3a** was chosen as the



**Figure 1.** X-ray structures of  $(\pm)$ -3a and  $(\pm)$ -3b.

starting material. Following the procedure developed in our laboratory, <sup>7</sup> **3a** was reduced regioselectively to the corresponding hydroxylactam **8**. Treatment of **8** with boron triflouride furnished enlactam **9**. Allylation of **9** followed by dehydrosulfonation produced double-bond migrated 2-pyridone **10**. To accomplish the synthesis of louisianin D, **10** was first converted to the corresponding 2-chloropyridine **11**, which was then reduced to bicyclic pyridine **12** by treatment of **11** with zinc in acetic acid. <sup>8</sup> Regioselective hydroxylation of **12** with LHMDS and oxygen yielded **13**, <sup>9</sup> which was then further oxidized with the swern-oxidation

**Scheme 1.** Formal Synthesis of 
$$(\pm)$$
-Alloyohimbane

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reagent to afford **5** (Scheme 2). The spectral data of **5** were in agreement with those reported in the literature.<sup>6</sup>

In conclusion, we have developed a one-pot reaction procedure to cis-fused bicyclic glutarimides. Syntheses of  $(\pm)$ -alloyohimbane (4) and louisianin D (5) were reported. Further application of fused bicyclic glutarimides to more complicated pentacyclic indole alkaloids is underway in our laboratory.

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**Supporting Information Available:** Additional spectroscopic data for all new compounds (<sup>1</sup>H NMR in CDCl<sub>3</sub>) and X-ray crystallographic data in CIF format. This material is available free of charge via the Internet at http://pubs.acs.org. OL060958K

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