

VITAMIN D C-22 ALDEHYDES. NEW KEY INTERMEDIATES FOR THE SYNTHESIS
OF SIDE CHAIN MODIFIED VITAMIN D ANALOGUES

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Summary: Vitamin D C-22 aldehyde and 1 α -hydroxyvitamin D C-22 aldehyde were efficiently synthesized starting from the readily available 22,23-bisnorcholeic acid. The usefulness of the compounds as common intermediates for the synthesis of side chain modified analogues of vitamins D₂ and D₃ has been demonstrated.

Steroid-type aldehydes have long been used in this² and other laboratories³ as intermediates for the synthesis of side chain modified analogues of vitamins D₂ and D₃. In the commonly used methodology the coupling of the steroid aldehyde with the respective side chain fragment was followed by the elaboration of the vitamin D triene system and regio- and stereoselective 1 α -hydroxylation⁴ of the resulting vitamin. The main disadvantage of this approach is that the multi-step procedure of the generation of the vitamin D triene system and the 1 α -hydroxylation process is to be repeated every time when synthesizing any single analogue. Recently, a 1 α -hydroxyvitamin D C-22 alcohol⁵ was prepared via the SO₂ adduct of the C-22 aldehyde. However, in this method⁵, regeneration of the vitamin D (5Z,7E)-triene system required additional photoisomerization.

As a part of our systematic studies on the biological action of side chain modified analogues of vitamin D₂ and D₃, we developed a new synthetic strategy based on the use of the new vitamin D C-22 aldehydes 1 and 2.

In this paper we describe an efficient synthesis of aldehydes 1 and 2 as well as some representative examples demonstrating the usefulness of these compounds as new key intermediates for the synthesis of vitamin D₂ and D₃ analogues of our current interest.

The synthesis of aldehydes 1 and 2 started from the commercially available 3 β -acetoxy-22,23-bisnor-5-choleic acid 3 (Steraloids, Inc., Wilton, NH). Hydrolysis of the acetate group of 3 followed by the protection of both functional groups afforded silyl ester 4. Allylic bromination (dibromantoin anhydride, NaHCO₃ in hexane, reflux, 30 min) and dehydrobromination⁶ (Bu₄NBr, Bu₄NF, s-collidine, THF, 4 hrs) with simultaneous desilylation gave the 5,7-diene 5 in 48% overall yield. Diene 5 gave after UV irradiation (Hanovia 608A36 lamp, Vycor filter, benzene-ethyl ether 1:4 v/v) the respective previtamin⁷. The previtamin (not shown) and the starting diene 5 were separated by silica gel flash chromatography. The

vitamin D-type ester 6 was obtained by the usual isomerization of the previtamin (36% yield from 5). Ester 6 was then stereoselectively hydroxylated at C-1 by SeO_2 catalyzed t-BuOOH oxidation⁴ of the 3,5-cyclovitamin intermediate (not shown) to give 1 α -hydroxy ester 7 (19% overall yield). The ester was partially hydrolyzed at C-3 position (0.1N KOH in MeOH, ether) and then both hydroxyls were silylated (TBDMSCl, imidazole, DMF, 55°C, 15 min) to form silyl ester 10 in nearly quantitative yield. Title aldehyde 1 was obtained by LiAlH_4 reduction of ester 10 (92%) followed by Swern oxidation (DMSO, oxalyl chloride, CH_2Cl_2 , TEA, -61°C, 62%) of alcohol 11. Alternatively, aldehydes 1 and 2⁸ were also obtained by DIBAL reduction⁹ of the respective esters 6 and 7 (54 and 57%). Alcohols 8 and 9, respectively, were formed as by-products of the reduction (29 and 32%). Alcohol 11 can also serve as intermediate in the synthesis of vitamin D_3 analogues (with a single 22-23 bond).

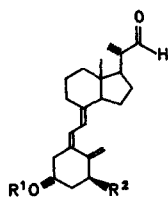
Our approach to the chiral synthon 14 utilizes the now available methyl R(-)-3-hydroxy-2-methylpropionate (Aldrich, Milwaukee, WI). The ester was converted to the sulfone 13 ($[\alpha]_{\text{D}}^{22} = +27.9^\circ$ c 1.3 CHCl_3) using standard procedure (MeMgBr, Et_2O -THF, 57%; TsCl, Py, 90%; PhSH, t-BuOK, DMF, 95%; m-CPBA, CH_2Cl_2 , 74%). The tertiary hydroxyl in 13 was protected with 2,3-dihydrofuran to form 14 (76%). For the synthesis of vitamin D_2 analogue 16 aldehyde 12 was condensed with sulfone 14 (LDA, THF, -75°C, 65%). Dehydroxy-desulfonylation of the condensation product (not shown) with sodium amalgam in buffered methanol provided the vitamin D_2 analogue 15 (55%) with the desired E stereochemistry¹⁰ of the side chain olefin and the retained natural configuration at C-20¹¹. Deprotection of the 25-hydroxyl in 15 (PPTS, MeOH, r.t., 20 min, 95%) followed by the cleavage of the silyl ethers ($[\text{Bu}_4\text{N}]\text{F}$, THF, 55°C, 15 min, 80%) afforded (24R)-1,25-dihydroxyvitamin D_2 (16). The vitamin D_2 analogue 16 obtained by the new procedure described here was identical in all respects¹² (HPLC, UV, NMR and MS spectra) with an authentic sample obtained previously in this laboratory^{2b}. It was found that the vitamin D triene system of both aldehydes 1 and 2 was not affected under the above conditions of Swern oxidation and the condensation reaction. Thus, the methodology we developed clearly demonstrates that the protection of the vitamin D triene system, commonly used by other authors, is not necessary for the side chain constructions exemplified in this paper.

For the preparation of vitamin D_3 analogues, a series of intermediates described in this paper may be employed, including aldehydes 1, 2, and 12, esters 6 and 7 and alcohols 8 and 9. As a representative example from the D_3 series, the new analog 17 was easily prepared (91% yield) by treating ester 7 with the respective Grignard reagent.

Analogue 17 with the tertiary hydroxyl in the side chain (as in the natural hormone) is being evaluated in this laboratory for its biological activity.

The synthetic scheme outlined in this letter represents, to the best of our knowledge, the most versatile and straightforward synthetic strategy for the preparation of side chain modified analogues of 1 α -hydroxyvitamin D_2 and 1 α -hydroxyvitamin D_3 .

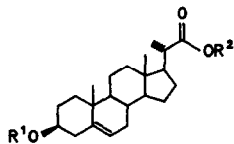
Acknowledgment. The authors are indebted to Drs. S. Yamada and Hans Reich for their helpful suggestions. The assistance of Rowland Randall and Charles Delwiche in taking mass spectra and Ann Cavaiani and Milo Westler in recording the NMR spectra is gratefully acknowledged. The work was supported by grants nos. AM-32701 and DK-14481 from the National Institutes of Health and by the Harry Steenbock Research Fund of the Wisconsin Alumni Research Foundation.



1, R¹, R² = H

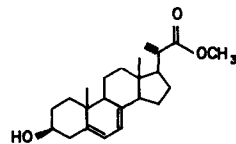
2, R¹ = H; R² = OH

12, R¹ = TBDMS; R² = OTBDMS

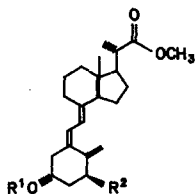


3, R¹ = Ac; R² = H

4, R¹ = TBDMS; R² = CH₃



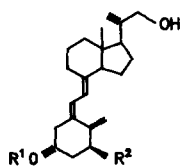
5



6, R¹, R² = H

7, R¹ = Ac; R² = OH

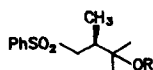
10, TBDMS; R² = OTBDMS



8, R¹ = R² = H

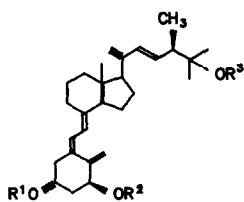
9, R¹ = H; R² = OH

11, R¹ = TBDMS; R² = OTBDMS



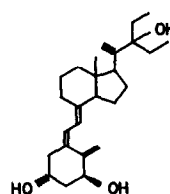
13, R = H

14, R = THF



15, R¹ = R² = TBDMS; R³ = THF

16, R¹ = R² = R³ = H



17

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- Analytical data are as follows:
1 MS: 328 (M^+ , 29), 310 (5), 295 (31), 269 (11), 253 (6), 136 (47), 118 (86), 29 (100);
 1H -NMR ($CDCl_3$) δ : 0.59 (3H, s, 18- CH_3), 1.14 (3H, d, $J=7$ Hz, 21- CH_3), 4.0 (1H, m, 3-H), 4.81 (1H, d, $J=1.2$ Hz, 19E-H), 5.05 (1H, d, $J=1.2$ Hz, 19Z-H), 6.05 (1H, d, $J=11$ Hz, 7-H), 6.23 (1H, d, $J=11$ Hz, 6-H), 9.58 (1H, d, $J=3.8$ Hz, 22-H).
2 MS: 344 (M^+ , 22), 326 (13), 311 (2), 285 (4), 269 (4), 152 (29), 134 (100);
 1H -NMR ($CDCl_3$) δ : 0.59 (3H, s, 18- CH_3), 1.15 (3H, d, $J=7$ Hz, 21- CH_3), 4.2 (1H, m, 3-H), 4.4 (1H, m, 1-H), 4.99 (1H, d, $J=1.2$ Hz, 19Z-H), 5.31 (1H, d, $J=1.2$ Hz, 19E-H), 6.02 (1H, d, $J=11$ Hz, 7-H), 6.36 (1H, d, $J=11$ Hz, 6-H), 9.56 (1H, d, $J=4$ Hz, 22-H).
- Single isomer of natural C-20 configuration was exclusively formed as determined by 1H -NMR.
- The reductive elimination of α -hydroxysulfones have been known to generate exclusively trans olefins. P. J. Kocienski, B. Lythgoe and I. Waterhouse, J. Chem. Soc. Perkin I, 1980, 1045 and references cited therein.
- Epimerization at C-20 has not been observed at the condensation of C-22 steroid aldehydes with various carboanions. S. C. Eyley and D. H. Williams, J. Chem. Soc. Perkin I, 1976, 727 and 731; see also reference 1a.
- See ref. 2b for HPLC, UV and MS spectra.
 1H -NMR ($CDCl_3$) δ : 0.54 (3H, s, 18- CH_3), 0.97 (3H, d, $J=6.9$ Hz, 28- CH_3), 1.01 (3H, d, $J=6.6$ Hz, 21- CH_3), 1.10 and 1.15 (3H and 3H, each s, 26- CH_3 and 27- CH_3), 4.21 (1H, m, 3-H), 4.41 (1H, m, 1-H), 4.97 (1H, brs, 19Z-H), 5.29 (2H, m, 22-H and 23-H), 5.31 (1H, brs, 19E-H), 5.99 (1H, d, $J=11.2$ Hz, 7-H), 6.35 (1H, d, $J=11.2$ Hz, 6-H).

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