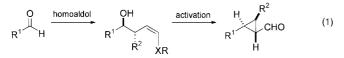
Cyclopropanes

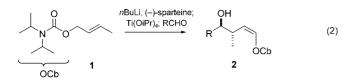
Enantioselective Synthesis of Cyclopropanes by Aldehyde Homologation**

Christina A. Risatti and Richard E. Taylor*

The prevalence of cyclopropane-containing compounds with biological activity, whether isolated from natural sources or rationally designed pharmaceutical agents, has inspired chemists to find novel and diverse approaches to their synthesis.^[1] We recently reported an efficient two-step homoaldol/activation sequence that resulted in the isolation of racemic 1,2-disubstituted cyclopropyl aldehydes.^[2] Herein we demonstrate the successful preparation of enantiomerically enriched and diastereomerically pure 1,2,3-trisubstituted cyclopropyl aldehydes by asymmetric metalation chemistry. In essence, this two-step procedure provides a three-carbon homologation of an aldehyde to give a nonracemic, stereo-chemically rich cyclopropyl aldehyde, [Eq. (1)].



Hoppe et al. have elegantly demonstrated that homoaldol adducts are readily available from (*E*)-butenyl *N*,*N*-diisopropylcarbamates **1** through deprotonation with nBuLi/(-)-sparteine complex and transmetalation with $Ti(OiPr)_4$.^[3] Condensation with an aldehyde was shown to provide *O*-enecarbamates **2** with high diastereoselectivity and good enantioselectivity [Eq. (2)].



Carbamate 1 is readily prepared through condensation of N,N-diisopropyl carbamyl chloride with crotyl alcohol in refluxing pyridine. A series of aldehydes were applied to the described homoaldol conditions using the crotyl substrate as a

| [*] | C. A. Risatti, Prof. R. E. Taylor |
|------|---|
| [] | Department of Chemistry and Biochemistry |
| | · · · · · · · · |
| | University of Notre Dame |
| | 251 Nieuwland Science Hall, Notre Dame, IN 46556-5670 (USA) |
| | Fax: (+1) 574-631-6652 |
| | E-mail: taylor.61@nd.edu |
| [**] | Support of this work by the National Science Foundation (CHE-02 |

^[**] Support of this work by the National Science Foundation (CHE-02-10918) is gratefully acknowledged. C.A.R. thanks Eli Lilly for support through an ACS Division of Organic Chemistry Graduate Fellowship.

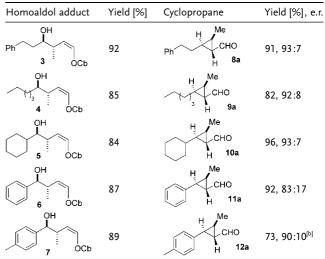
Supporting information for this article is available on the WWW under http://www.angewandte.org or from the author.

Angew. Chem. Int. Ed. 2004, 43, 6671–6672

DOI: 10.1002/anie.200461106

representative example. Aliphatic aldehydes, including sterically hindered substrates, resulted in high yields of (*Z*)homoaldol adducts **3–5** with high *anti* selectivity as expected (Table 1). In addition aromatic aldehydes, including benzaldehyde and tolualdehyde, performed well in the coupling reaction to provide **6** and **7**, respectively (Table 1). In each case, the diastereoselectivity was >95:5 by crude ¹H NMR analysis.

Table 1: Trisubstituted cyclopropanes from O-enecarbamates.^[a]

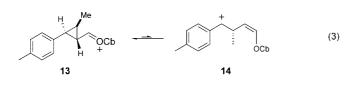


[[]a] Reaction conditions: Tf₂O, 2,6-lutidine, toluene, -78 °C to -50 °C. [b] Two minor diastereomers were also observed in this case (5:1:1 ratio).

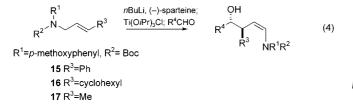
Exposure of homoaldol adducts 3-7 to triflic anhydride and 2,6-lutidine provided good to excellent yields of 1,2,3trisubstituted cyclopropyl aldehydes 8a-12a (Table 1). In contrast to our previous study^[2] these cyclization substrates were more sensitive to the reaction conditions. When our standard activation conditions were employed, in CH₂Cl₂, multiple products were observed. However, when the reaction solvent was conducted in toluene, the inherent rate and alternative reaction pathways could be reduced such that the reaction was driven toward the formation of the desired cyclopropyl aldehyde products 8a-12a. The enantiomeric purity of the cyclopropane derivatives was determined by reduction of the aldehyde with lithium aluminum hydride and analysis of the Mosher esters.^[4] The observed enantiomer ratios corresponded well with expected enantioselectivity of the starting homoaldol adducts,^[3] suggesting the cyclization step proceeds in a stereospecific manner.

The aliphatic systems provided high yields and diastereoselectivity for the cyclopropyl aldehyde products 8a-10a. In each case, the ring closure takes place with inversion of configuration at the alcohol center.^[5] In contrast, reduced diastereoselectivity was observed upon activation of homoaldol adduct 7. The inductively donating methyl group appears to promote equilibration of the oxonium ion intermediate 13 through the homoallylic cation 14 [Eq. (3)]. Under these conditions no products resulting from elimination were observed.

Communications



Ideally one would like access to both enantiomers of the cyclopropyl aldehyde products. In fact, Beak et al. have demonstrated that *N*-(*tert*-butoxycarbonyl)-*N*-(*p*-methoxyphenyl)cinnamyl amine **15** and *N*-(*tert*-butoxycarbonyl)-*N*-(*p*-methoxyphenyl)-3-cyclohexylallylamine **16** undergo asymmetric deprotonation by an *n*BuLi/(–)-sparteine complex [Eq. (4)].^[6] However, in this case, transmetalation with Ti(O*i*Pr)₃Cl followed by quenching with an aldehyde furnished the enantiomeric, complementary homoaldol adducts.



The aldehydes used previously to generate the *O*-enecarbamates were subjected to the Beak homoaldol protocol. To complement the earlier work, the substrates were preformed with crotylamine **17**, and the yields and selectivities corresponded nicely with those observed by Beak for the cyclohexyl and cinnamyl systems. The (*Z*)-*N*-enecarbamates **18–22** were isolated with predominantly *anti* stereochemistry and the diastereoselectivity was > 95:5 by crude ¹H NMR analysis (Table 2). Homoaldol adducts **18–22** were activated with triflic anhydride and 2,6-lutidine to provide aldehydes

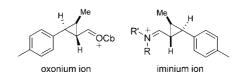
Table 2: Trisubstituted cyclopropanes from N-enecarbamates.[a]

| Homoaldol adduct | Yield [%] | Cyclopropane | Yield [%], e.r. |
|--|-----------|----------------------|-----------------|
| Ph 18 Ph NR ¹ R ² | 69 | OHC H 8b | 91, 90:10 |
| QH (1) 19 NR ¹ R ² | 66 | OHC H H 9b | 70, 94:6 |
| | 73 | OHC H H 10b | 86, 91:9 |
| | 67 | OHC H H 11b | 96, 94:6 |
| 0H 22 NR ¹ R ² | 68 | H OHC H 12b | 70, 92:8 |

[a] Reaction conditions: Tf₂O, 2,6-lutidine, toluene, -78 °C to -50 °C. R¹ = *para*-methoxyphenyl, R² = *tert*-butoxycarbonyl.

8b–12b in good to excellent yield. Again, the enantioselectivity of the cyclopropane products corresponded well with the expected enantiomer ratios from the homoaldol coupling, suggesting the cyclization step proceeds with complete stereospecificity.

In this system, cyclization of the aliphatic substrates 18-20 proceeded cleanly in CH₂Cl₂. However, care was taken with the activated aryl systems, and toluene was used as the reaction solvent. It is notable that, in contrast to *O*-enecarbamate substrate 7, activation of *N*-enecarbamate 22 provided 12b as a single diastereomer. The iminium ion intermediate, present in solution prior to quenching, is apparently more stable than the corresponding oxonium ion intermediate, and therefore, an equilibrium with its homoallylic cation is not relevant (Figure 1). Although the method





has been applied to 1,2,3-trisubstituted cyclopropyl aldehydes containing a methyl substituent, alternative allylamine derivatives are expected to provide diversely substituted cyclopropanes and broaden the scope of this two-step three-carbon homologation procedure.

In conclusion we have demonstrated the stereospecific cyclization of enantiomerically enriched homoaldol adducts. This chemistry is a remarkably efficient method for the homologation of readily available aldehydes to give non-racemic, stereochemically rich 1,2,3-trisubstituted cyclo-propyl aldehydes.^[7] Application of this method to the preparation of biologically relevant cyclopropanes is currently underway in our laboratories and will be reported in due course.

Received: June 28, 2004

Keywords: cations · cyclization · cyclopropanes · small ring systems · synthetic methods

- H. Lebel, J.-F. Marcoux, C. Molinaro, A. B. Charette, *Chem. Rev.* 2003, 103, 977.
- [2] R. E. Taylor, C. A. Risatti, F. C. Engelhardt, M. J. Schmitt, Org. Lett. 2003, 5, 1377.
- [3] For a lead reference, see: K. R. K. Prasad, D. Hoppe, Synlett 2000, 1067.
- [4] J. S. Dale, H. S. Mosher, J. Am. Chem. Soc. 1973, 95, 512.
- [5] R. E. Taylor, F. C. Engelhardt, M. J. Schmitt, H. Yuan, J. Am. Chem. Soc. 2001, 123, 2964.
- [6] a) M. C. Whisler, L. Vaillancourt, P. Beak, Org. Lett. 2000, 2, 2655;
 b) M. C. Whisler, P. Beak, J. Org. Chem. 2003, 68, 1207.
- [7] During review of this manuscript we learnt of a similar investigation. See the preceding Communication in this issue:
 R. Kalkofen, S. Brandau, B. Wibbeling, D. Hoppe, *Angew. Chem.* 2004, *116*, 6836; *Angew. Chem. Int. Ed.* 2004, *43*, 6667.