Rhodium(III) complexes with a bidentate *N***-heterocyclic carbene ligand** bearing flexible dendritic frameworks[†]‡

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Received 24th January 2007, Accepted 6th March 2007 First published as an Advance Article on the web 20th March 2007 DOI: 10.1039/b701104k

Rh(III) complexes with a bidentate *N*-heterocyclic carbene ligand bearing flexible dendritic frameworks have been synthesized and fully characterized by X-ray crystallographic analysis, NMR measurements and theoretical calculations.

Much attention has been paid to dendrimers with organometallic centers since the well-defined hyperbranched frameworks will bring about unique chemical properties including novel catalysis.^{1,2} We have reported various dendritic ligands with pyridine,³ phosphine⁴ and *N*-heterocyclic carbene (NHC)^{5,6} functionalities at the core and used as transition-metal-catalyzed reactions. Recently, we have synthesized a series of Rh(1) complexes with a *mono*dentate NHC ligand bearing Fréchet-type⁷ flexible dendritic frameworks. In the hydrosilylation of ketones catalyzed by them, yields of the corresponding alcohols increased with higher dendrimer generation: a positive dendrimer effect was evident.⁵ Furthermore, Rh(1) complexes with a NHC ligand having a rigid 2,3,4,5-tetraphenylphenyl and its higher dendritic frameworks were efficient catalysts in the hydrosilylation of α , β -unsaturated ketones.⁶

The NHC ligands have been widely used in the field of homogeneous catalysis and organometallic chemistry.⁸ Formation of stronger bonds to the metals associated with NHC ligands as compared with phosphines diminish their dissociation from metal centers. Among them, *bi*dentate NHC ligands are of special interest since stability of their metal complexes is considerably enhanced by the chelating effect.⁹ However, examples of the *bi*dentate NHC ligands are very limited as compared to those of *mono*dentate NHC cases. As for Rh(III) complexes,¹⁰ the complexes are efficient catalysts in hydrosilylation^{10a} and transfer hydrogenation.^{10b} In this study, novel Rh(III) complexes with a bidentate NHC ligand having Fréchet-type⁷ flexible polybenzylether dendrimers were synthesized and fully characterized. Then, these complexes were employed as catalysts in the hydrosilylation of 2-cyclohexen-1-one and good 1,4-selectivity was observed.

A series of imidazolium salts, which are precursors for the desired bidentate NHC ligands, were synthesized in two steps. At first, *N*-substituted imidazoles bearing the Fréchet-type⁷ polybenzylether dendrimers (**1a–c**) were synthesized by the reaction of imidazole with one equivalent of the corresponding

dendrimer-bromides, G_n -Br (n = 0, 1 and 2) as shown in eqn (1).¹¹ Subsequently, the reaction of **1a**-**c** with CH₂I₂ in toluene afforded the corresponding imidazolium salts $[(G_0)(C_3H_3N_2)-CH_2(C_3H_3N_2)(G_0)]I_2$ (**2a**),^{9a} $[(G_1)(C_3H_3N_2)CH_2(C_3H_3N_2)(G_1)]I_2$ (**2b**), and $[(G_2)(C_3H_3N_2)CH_2(C_3H_3N_2)(G_2)]I_2$ (**2c**) (eqn (2)). The new compounds (**1b**, **1c**, **2b** and **2c**) were fully characterized by elemental analysis, FD-MS spectra and NMR measurements.

Rh(III) complexes with the NHC ligand $RhI_2(OAc)[(G_0) (C_3H_2N_2)CH_2(C_3H_2N_2)(G_0)$] (3a), $RhI_2(OAc)[(G_1)(C_3H_2N_2) CH_2(C_3H_2N_2)(G_1)$] (3b), and $RhI_2(OAc)[(G_2)(C_3H_2N_2)CH_2 (C_3H_2N_2)(G_2)$] (3c) were synthesized by the reaction of 2a-2c with [RhCl(COD)]₂ in the presence of KI and KOAc (eqn (3)).^{10a} In addition, a methyl derivative RhI₂(OAc)[Me(C₃H₂N₂)CH₂- $(C_3H_2N_2)Me$] (3d) was also synthesized with the same method using $[Me(C_3H_3N_2)CH_2(C_3H_3N_2)Me]I_2$.¹² The new complexes (3a-3d) were also fully characterized. In ¹³C{¹H} NMR spectra of 3a-3d, the coordinated carbone carbons appeared as sharp doublets at $\delta = 151.5$ ppm ($J_{Rh-C} = 41.6$ Hz), 155.0 ppm (41.6 Hz), 154.8 ppm (41.7 Hz), and 151.3 ppm (43.1 Hz), respectively. The isolated complexes were highly stable both to air and moisture in the solid state. The ¹H NMR spectra of **3a–3d** did not change over a week, indicating these complexes are also considerably stable in solution.



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[‡] Electronic supplementary information (ESI) available: Synthetic method and analytical data of 1, 2 and 3. See DOI: 10.1039/b701104k



The molecular structures of **3a**, **3b** and **3d** have been successfully determined by X-ray diffraction study. Fig. 1 shows the ORTEP drawing of **3a**, **3b** and **3d**.§¶ In the structures, each rhodium atom has distorted octahedral coordination geometry. The average Rh–C bond lengths of **3a**, **3b** and **3d** were 1.962(4), 1.969(4) and 1.965(3) Å, respectively. These bond lengths are similar to those of other Rh(III)–NHC complexes.¹⁰ Fig. 1(b) showed that in the solid state the benzylether moiety on the NHC ligand spatially spreads out and covers the wide area from the Rh center.



Fig. 1 ORTEP drawings of (a) **3a**, (b) **3b** and (c) **3d** with thermal ellipsoids at 50% probability levels.

Fig. 2 showed the ¹H NMR spectra of **3a–3c** in CDCl₃. All the ¹H and ¹³C NMR resonances were unambiguously assigned with the aid of H-H COSY and HSQC spectroscopy. The ¹H NMR spectrum of 3a exhibited the NHC ring proton resonances as two doublet peaks at $\delta = 6.98$ and 6.82 ppm (J = 1.8 Hz; signals a and b in Fig. 2(a)). A bridging methylene proton resonance was observed at $\delta = 6.05$ ppm (signal c in Fig. 2(a)). Similarly, the ¹H NMR spectrum of 3b showed signals of two NHC ring protons at $\delta = 6.92$ and 6.84 ppm, and the bridging methylene proton at $\delta = 5.95$ ppm (Fig. 2(b)). However, the ¹H NMR spectrum of the second generation (G_2) complex 3c displayed the NHC ring proton resonances at $\delta = 6.68$ and 6.62 ppm, and a bridging methylene proton at $\delta = 5.58$ ppm (Fig. 2(c)). In the spectrum, one of the NHC ring proton resonance (signal a) and the bridging methylene proton resonance (signal c) showed considerable upfield shifts. The latter peak (signal c) overlapped with the methylene resonance of the benzyl moiety on the NHC ring (signal d in Fig. 2(c)). Interestingly, these shifts were not observed in the ¹H NMR spectrum of the corresponding imidazolium salt 2c. Although no noe interaction between the aromatic ring protons



Fig. 2 ¹H NMR spectra of (a) 3a, (b) 3b and (c) 3c in CDCl₃.

and the NHC ring protons was observed, such up-field shifts might be caused by the ring-current effect of aromatic rings of the dendritic framework: the second generation dendritic moieties would then come closer to the NHC center. Actually, ONIOM¹³ (B3LYP/LANL2DZ:UFF)-CAChe¹⁴/CONFLEX¹⁵/MM3¹⁶ calculations afforded the significantly folded structure shown in Fig. 3. It is noteworthy that this structure is quite different from the open structure in Fig. 1(b) determined by X-ray crystallography.



Fig. 3 An optimized structure of 3c calculated by ONIOM (B3LYP/LANL2DZ:UFF)-CAChe/CONFLEX/MM3. The yellow part indicates the $RhI_2(OAc)[(C_3H_2N_2)CH_2(C_3H_2N_2)]$ core.

As a probe reaction to investigate the catalytic activity of **3**, hydrosilylation of 2-cyclohexen-1-one¹⁷ was carried out (Scheme 1).* As catalysts, **3a**, **3b** and **3c** showed high catalytic activity to afford mixtures of cyclohexanone (1,4-adduct: **4**) and 2-cyclohexen-1-ol (1,2-adduct: **5**) in 91–99% yields with the 1,4adduct as a major product (79–81% selectivities). Although the yield and 1,4-regioselectivity decreased, the corresponding methyl derivative (**3d**) also showed moderate catalytic activity. Recently, we reported that the 1,2-adduct (**5**) was obtained as the major product (61–69% selectivities in 62–83% total yields) in the same hydrosilylation of 2-cyclohexen-1-one catalyzed by Rh(1) complexes having the *mono*dentate NHC ligand: RhCl(COD)-[(G_n)₂(C₃H₂N₂)] (G_n: n = 0–3).⁶ With the bidentate NHC ligand,



3 (0.5 mol%)

the dendrimer moiety of higher generation must affect the catalytic environment as shown in Fig. 2 and 3. However, almost no effect of the dendrimer generation was observed in the hydrosilylation as shown in Scheme 1. In the catalytic reaction, the robust chelating structures of 3 are more important and successfully realized the good 1,4-regioselectivity.

K₂CO₃ -

1.4-Adduct

(4)

79/21

81/19

80/20

65/35

OH

1,2-Adduct

(5)

Notes and references

 $Crystal data for RhI_2(OAc)[(G_0)(C_3H_2N_2)CH_2(C_3H_2N_2)(G_0)] \cdot C_4H_8O$ $(3a \cdot THF)$: C₂₇H₃₁I₂N₄O₃Rh, M = 816.28, T = 113 K, monoclinic, space group $P2_1/c$ (No. 14), a = 12.126(5), b = 14.786(6), c = 15.987(6) Å, $\beta =$ 92.446(6)°, U = 2863(2) Å³, Z = 4, μ (Mo K α) = 2.78 cm⁻¹, observed reflections 5511 ($I > 2\sigma(I)$), R1, wR2 (all data) = 0.042, 0.098. For $RhI_2(OAc)[(G_1)(C_3H_2N_2)CH_2(C_3H_2N_2)(G_1)] \cdot CH_3CN$ (**3b** · CH₃CN): $C_{53}H_{50}I_2N_5O_6Rh$, M = 1209.72, T = 113 K, triclinic, space group $P\bar{1}$ (No. 2), a = 9.338(4), b = 13.972(6), c = 20.243(8) Å, a = 20.243(8), $\beta = 85.57(1)$, $\gamma = 78.379(9)^{\circ}$, U = 2461(2) Å³, Z = 2, μ (Mo K α) = 1.66 cm⁻¹, observed reflections 8789 ($I > 2\sigma(I)$), R1, wR2 (all data) = 0.043, 0.102. For $RhI_2(OAc)[Me(C_3H_2N_2)CH_2(C_3H_2N_2)Me]$ (3d): $C_{11}H_{15}I_2N_4O_2Rh$, M = 591.98, T = 113 K, monoclinic, space group Cc (No. 5), a = 9.685(2), b = 13.321(3), c = 12.691(3) Å, $\beta = 94.340(2)^{\circ}, U = 1632.6(7)$ Å³, Z = 4, μ (Mo K α) = 4.83 cm⁻¹, Flack parameter = 0.222(16), observed reflections $2074 (I > 2\sigma(I)), R1, wR2 (all data) = 0.019, 0.059.$

¶CCDC reference numbers 633257, 633258 and 633259. For crystallographic data in CIF or other electronic format see DOI: 10.1039/b701104k * General procedure for hydrosilylation of 2-cyclohexen-1-one. A rhodium catalyst 3a (3.7 mg, 0.005 mmol) was placed in a 10-cm³ Schlenk tube under an Ar atmosphere. Anhydrous THF (1.0 cm³), 2-cyclohexen-1-one $(9.7 \times 10^{-2} \text{ dm}^3, 1 \text{ mmol})$, and bibenzyl (45.6 mg, 0.25 mmol as an internal standard) were added under an Ar flow and the solution was stirred for 5 min. Ph₂SiH₂ (0.22 dm³, 1.2 mmol) was added via a syringe and the reaction mixture was stirred at room temperature (22 \pm 1 $^{\circ}\breve{C)}$ for 20 h. The desilylation was performed by adding K₂CO₃ (ca. 1 mg) and MeOH (1 cm³), then the mixture was stirred for 30 min. A total yield (99%) and a ratio (79/21) of cyclohexanone and 2-cyclohexen-1-ol were determined by a GC (Shimadzu CPB10 column, 25 m length, 0.25 mm i.d.).

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