Structure Revision and Syntheses of Epohelmins A and B

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ABSTRACT



Epohelmins A (24) and B (26) have been reassigned as pyrrolizidin-1-ols, rather than the proposed 9-oxa-4-azabicyclo[6.1.0]nonane structures 1 and 2, respectively. Syntheses of epohelmin A (24) (eight steps, 52% overall yield) and epohelmin B (26) (11 steps, 43% overall yield) have been achieved starting from *N*-Cbz-(*S*)-prolinal (9) and ortho ester ketone 17 using a stereoselective aldol reaction and a stereoselective reductive cyclization as the key steps.

Shibuya, Ebizuka, and co-workers recently reported the isolation of the novel lanosterol synthase inhibitors epohelmins A (1) and B (2) from a fungal strain FKI-0929.¹ The structures were determined by detailed spectroscopic analysis and proposed to be novel 9-oxa-4-azabicyclo[6.1.0]-nonanes (see Figure 1).





There were both chemical and spectroscopic grounds to question these structure assignments. 9-Oxa-4-azabicyclo-[6.1.0]nonanes cyclize readily to give pyrrolizidin-1-ols.^{2,3} Such a cyclization of any of the four isomers of **3** would give pyrrolizidin-1-ols **4** (see eq 1). The protons assigned to

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the epoxy groups of epohelmins A and B absorb in the range δ 4–4.5 and the carbons in the range δ 70–74 (see Figure 1). The analogous protons and carbons of *trans*- and *cis*-epoxy cyclooctanes (**5** and **6**) absorb at much higher field than those of the epohelmins with the protons at δ 2.8 and 2.9 and the carbons at δ 59.6 and 55.6, respectively (see Figure 2).⁴

On the other hand, the methine hydrogens and carbons of *trans*- and *cis*-pyrrolizidin-1-ols (7 and 8) have similar chemical shifts to the epohelmins as shown.⁵ Although

^{(1) (}a) Sakano, Y.; Shibuya, M.; Yamaguchi, Y.; Masuma, R.; Tomada, H.; Ômura, S.; Ebizuka, Y. *J. Antibiot.* **2004**, *57*, 564–568. (b) Ômura, S.; Kyoda, H.; Masuma, R.; Ebizuka, Y.; Shibuya, M. Japan Kokai Tokkyo Koho JP 2004196680, 2004; *Chem. Abstr.* **2004**, *141*, 122412m.

⁽²⁾ Glass, R. S.; Deardorff, D. R.; Gains, L. H. Tetrahedron Lett. 1978, 2965–2968.

⁽³⁾ Subramanian, T.; Lin, C.-C.; Lin, C.-C. *Tetrahedron Lett.* **2001**, *42*, 4079–4082.

⁽⁴⁾ Poon, T. H. W.; Pringle, K.; Foote, C. S. J. Am. Chem. Soc. 1995, 117, 7611-7618.



Figure 2. NMR data of epoxycyclooctanes (5 and 6) and pyrrolizidin-1-ols (7 and 8).

pyrrolizidin-1-ols **4** look quite different than 9-oxa-4azabicyclo[6.1.0]nonanes **3**, the only difference is the formation of a C-N bond and cleavage of a C-O bond so that the 2D NMR correlations used to assign the structure of epohelmins A and B as **1** and **2** will be equally applicable to the stereoisomers of **4**.

This analysis suggests that epohelmins A and B are two of the four stereoisomers of **4**. If this is the case, the stereochemical analysis of **1** and **2** is based on the wrong skeleton and may not be applicable to the stereochemistry of **4**. We decided that the stereochemistry was best determined by the synthesis of model 3-alkylpyrrolizidin-1-ols such as **11**. Aldol addition of the kinetic enolate of 2-undecanone to *N*-Cbz-phenylalaninal followed by reductive pyrrolidine formation and methylation led to all four isomers of preussin (2-benzyl-1-methyl-5-(nonyl)pyrrolidin-3-ol).⁶ This short sequence seemed ideal since at this time we did not know the stereochemistry of the epohelmins.

Treatment of 2-heptanone with LDA in THF at -78 °C provided the kinetic enolate, which was treated with N-Cbz-(S)-prolinal $(9)^7$ to afford 10 with excellent stereoselectivity in 95% yield (see Scheme 1). Hydrogenolysis (1 atm) of 10 over Pd(OH)₂ in MeOH for 12 h liberated the secondary amine, which reacted with the ketone to form an iminium salt or enamine, which was reduced to give pyrrolizidinol 11 in 71% yield. Several minor byproducts were formed that may include diastereomers of 11 other than epohelmin B model 16. Similar results were obtained with other Pd catalysts. As expected, the ¹H and ¹³C NMR spectra of 11 vary as a function of pH. The epohelmins were isolated using an eluent containing HOAc, suggesting that the natural products were isolated as acetate salts. We were pleased to find that the ¹H and ¹³C NMR spectra of a CDCl₃ solution of 11 containing 0.85-0.90 equiv of HOAc corresponded precisely with those of the ring portion of epohelmin A.⁸

We had not expected either step in the formation of **11** to be stereoselective. Kitahara obtained mixtures in the addition



of lithium enolates to N-Cbz-phenylalaninal; the zinc enolate added stereoselectively.6 Addition of zinc and lithium enolates of ethyl acetate to N-Boc-prolinal afforded 2:1 to 4:1 mixtures of isomers favoring the Felkin-Anh product.9 We decided to prepare oxazolidinone 14 to prove the stereochemistry of aldol product 10. Treating 10 with K₂-CO₃ did not form the expected oxazolidinone but instead resulted in hydrolysis and double dehydration to give pyrrole **12**.¹⁰ Reduction of the ketone with NaBH₄ in MeOH followed by treatment with K₂CO₃ in 1:1 2-propanol/water for 18 h at 70 °C provided oxazolidinone 13 in 71% overall yield as a mixture of stereoisomers. Dess-Martin oxidation afforded the desired oxazolidinone ketone 14 in 99% yield. The coupling constant between the methine hydrogens of 14 is 7.3 Hz as in related compounds indicating that the hydrogens are cis; in similar compounds with anti hydrogens the coupling constant is 4.0 Hz.¹¹ This established that the enolate of 2-heptanone added to 9 with high Felkin-Anh selectivity to give 10.

The stereochemistry at C_3 of epohelmin A model **11** was established by the NOE between H_1 and H_3 . We had not expected the reductive cyclization to be stereoselective because reductive cyclization of related pyrrolidine ketones lacking the hydroxyl group gave 1:1 mixtures of 3-alkylpyrrolizidines.¹² Presumably, the hydroxyl group blocks the bottom face so that hydrogenation occurs preferentially from the top face to give **11**.

We now turned our attention to the preparation of an epohelmin B model. As suggested by Ebizuka and Shibuya, we suspected that epohelmins A and B differed in the alcohol, not alkyl, stereochemistry. Changing the stereochemistry at the alkyl group should change the ¹³C NMR

⁽⁵⁾ Christine, C.; Ikhiri, K.; Ahond, A.; Mourabit, A. A.; Poupat, C.; Potier, P. *Tetrahedron* **2000**, *56*, 1837–1850.

^{(6) (}a) Okue, M.; Watanabe, H.; Kitahara, T. *Tetrahedron* **2001**, *57*, 4107–4110. (b) Okue, M.; Watanabe, H.; Kasahara, K.; Yoshida, M.; Horinouchi, S.; Kitahara, T. *Biosci. Biotechnol. Biochem.* **2002**, *66*, 1093–1096.

^{(7) (}a) Langley, D. R.; Thurston, D. E. J. Org. Chem. 1987, 52, 91–97.
(b) Corey, E. J.; Shibata, S.; Bakshi, R. K. J. Org. Chem. 1988, 53, 2861–2863.
(c) Dolbeare, K.; Pontoriero, G. F.; Gupta, S. K.; Mishra, R. K.; Johnson, R. L. Bioorg. Med. Chem. 2003, 11, 4103–4112.

⁽⁸⁾ We thank Prof. M. Shibuya for a complete set of 1D and 2D NMR spectra of epohelmins A and B.

 ^{(9) (}a) Andrés, J. M.; Pedrosa, R.; Pérez, A.; Pérez-Encabo, A. Tetrahedron 2001, 57, 8521–8530. (b) Hanson, G. J.; Baran, J. S.; Lindberg, T. Tetrahedron Lett. 1986, 27, 3577–3580.

⁽¹⁰⁾ For a similar compound see: Pommelet, J. C.; Jourdain, F.; Dhimane, H. *Molecules* **2000**, *5*, 1130–1138.

^{(11) (}a) Parker, K. A.; O'Fee, R. J. Am. Chem. Soc. 1983, 105, 654–655. (b) Kiyooka, S.-i.; Nakano, M.; Shiota, F.; Fujiyama, R. J. Org. Chem. 1989, 54, 5409–5411. (c) St-Denis, Y.; Chan, T.-H. J. Org. Chem. 1992, 57, 3078–3085.

⁽¹²⁾ Provot, O.; Célérier, J.-P.; Lhommet, G. J. Heterocycl. Chem. 1998, 35, 371–376.



spectrum of the lower portion of the molecule,¹³ whereas changing the stereochemistry at the alcohol should change the ¹³C NMR spectrum of the top portion of the molecule as is observed. Inversion of *trans*-pyrrolizidin-1-ol stereochemistry has been addressed by Chamberlin, who found that this was best accomplished by oxidation to the ketone and reduction with L-Selectride.¹⁴ Swern oxidation of 11 afforded the unstable ketone 15 in 97% yield. Reduction with L-Selectride in THF provided epohelmin B model 16 in 69% yield. The ¹H and ¹³C NMR spectra of the acetate salt of **16** corresponded precisely with those of the ring portion of epohelmin B. The stereochemistry of the pentyl side chain was confirmed by the NOE between H₃ and H₄, which established that H₃ is on the concave face and the pentyl group is on the convex face. Reduction of 15 with NaBH₄ in MeOH gave a 25:1 mixture of 11 and 16 analogous to that observed by Chamberlin in related systems.¹⁴ Although reduction of pyrrolizidin-1-ones with NaBH(OMe)3 in benzene has been reported to give *cis*-pyrrolizidin-1-ols,¹⁵ no reaction occurred with 15 under these conditions.

We now turned our attention to adapting this synthesis to accommodate an enone in the side chain. Ortho ester ketone **17** was prepared from 4-acetylbutyric acid by the literature procedure.¹⁶ Conversion of **17** to the kinetic enolate with LDA and addition of prolinal **9** afforded aldol product **18** (see Scheme 3). Hydrolysis of the ortho ester in 4:2:1 AcOH/ THF/H₂O and transesterification with K₂CO₃ in MeOH afforded methyl ester **19** in 75% overall yield from **17**.¹⁷ Protection of the hydroxyl group facilitated purification, the preparation of the keto phosphonate, and the Wittig reaction. Treatment of **19** with excess ethyl vinyl ether and catalytic PPTS in CH₂Cl₂ provided 97% of **20**. Hydrogenolysis and reductive cyclization with H₂ (1 atm) over Pd(OH)₂ in MeOH for 12 h afforded 93% of the crude protected pyrrolizidine methyl ester **21**.

Addition of crude **21** to a large excess of $LiCH_2PO(OMe)_2$ in THF at -78 °C and stirring for 12 h with warming to 25



°C gave keto phosphonate **22** in 81% yield after chromatography. Treatment of **22** with NaH in THF followed by addition of hexanal and stirring for 3 h afforded enone **23** in 95% yield. Deprotection by stirring in 10% aqueous HOAc for 4 h at 25 °C gave epohelmin A (**24**) quantitatively. The ¹H and ¹³C NMR spectra of **24** that is 85–90% protonated with HOAc are identical to those reported for epohelmin A, thereby unambiguously establishing the revised structure of epohelmin A. The optical rotation of **24** that is 85–90% protonated with HOAc, $[\alpha]^{22}_{D}$ +22, is identical to that of the natural product, indicating that the absolute configuration is as shown since **9** is prepared from (*S*)-proline.

Epohelmin B (26) was prepared from 24 by Swern oxidation to give the unstable dione followed by reduction with L-Selectride in THF to give diol 25 in 82% yield as a

⁽¹³⁾ For the assigned spectra of *cis*- and *trans*-3-methylpyrrolizidine, see: Skvortsov, I. M.; Antipova, I. V. J. Org. Chem. USSR **1979**, *15*, 777–783; *Zh. Org. Khim.* **1979**, *15*, 868–875.

⁽¹⁴⁾ Chamberlin, A. R.; Chung, J. Y. L. J. Org. Chem. 1985, 50, 4425-4431.

⁽¹⁵⁾ Gundermann, H.; Schnekenburger, J. Arch. Pharm. (Weinheim) 1998, 321, 925–929.

^{(16) (}a) Corey, E. J.; Raju, N. *Tetrahedron Lett.* 1983, 24, 5571–5574.
(b) Oku, A.; Numata, M. *J. Org. Chem.* 2000, 65, 1899–1906. (c) Mohapatra, S.; Capdevila, J. H.; Murphy, R. C.; Hevko, J. M.; Falck, J. R. *Tetrahedron Lett.* 2001, 42, 4109–4110.

⁽¹⁷⁾ Kwok, P.-Y.; Muellner, F. W.; Chen, C.-K.; Fried, J. J. Am. Chem. Soc. 1987, 109, 3684–3692.

mixture of diastereomers at the side chain alcohol. Oxidation of **25** with MnO₂ proceeded slowly and in modest yield. We had previously observed that the Dess–Martin periodinane was not effective for the oxidation of model **11** or epohelmin A (**24**). As expected from this observation, selective oxidation of diol **25** with Dess–Martin periodinane afforded epohelmin B (**26**) in 98% yield. The ¹H and ¹³C NMR spectra of the HOAc salt of **26** are identical to those reported for epohelmin B, thereby unambiguously establishing the revised structure of epohelmin B. The optical rotation of the acetate salt of **26**, $[\alpha]^{22}_{D}$ +29, is similar to that of the natural product, $[\alpha]^{22}_{D}$ +25, indicating that the absolute configuration is as shown.

A plausible biosynthesis of epohelmins A (24) and B (26) proceeds through the Claisen condensation of heptaketide 28 with activated proline 27 to give adduct 29. Hydrolysis and decarboxylation will give trione 30. Reductive cyclization and reduction of the ring ketone will lead to epohelmins A and B. If this is the biosynthetic pathway, the very efficient key reductive cyclization step leading to 11 and 21 is biomimetic.

In conclusion, we have reassigned the proposed 9-oxa-4azabicyclo[6.1.0]nonane structures of epohelmins A (1) and B (2) as pyrrolizidin-1-ols 24 and 26, respectively. We have developed syntheses from *N*-Cbz-(*S*)-prolinal (9) and ortho ester ketone 17 that provide epohelmin A (24) in eight steps and 52% overall yield and epohelmin B (26) in 11 steps and 43% overall yield using a stereoselective aldol reaction and a stereoselective reductive cyclization as the key steps.



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Supporting Information Available: Full experimental details and copies of ¹H and ¹³C NMR spectra. Comparison of NMR data of epohelmin A and **11**. NOE studies of **11** and **16**. This material is available free of charge via the Internet at http://pubs.acs.org.

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