A Concise Synthesis of (+)-SCH 351448

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ABSTRACT





Our interest in naturally produced small molecules of structural novelty and potential biological significance led us to take note of SCH 351448 (1, Figure 1)-a hexanesoluble material purified from a Micromonospora sp. fermentation broth.¹ The reported ability of natural **1** to activate transcription from the low-density lipoprotein (LDL) receptor promoter is of mechanistic and biomedical importance.² The single-crystal X-ray structure of 1 shows a remarkable topology. The ionized and un-ionized carboxy groups are intramolecularly hydrogen-bonded and together with the phenol and hydroxy groups accommodate a heptacoordinate sodium ion in the interior cavity of a hydrophobic globular structure.¹ At this point, it remains unclear whether 1 functions by mediating sodium or other ion transport across membranes or whether the role of the chelated sodium ion is structural; i.e., does 1 behave as a hydrophobic small molecule ligand for a cellular receptor? To answer these and related questions, we initiated a synthetic program to provide

material for future structural, physicochemical, and biological studies. Herein, we communicate a short and scalable synthesis of SCH 351448.^{3,4}

Taking maximal advantage of the dimeric nature of **1**, we hoped to identify viable esterification/lactonization strategies to combine two identical or closely related C_1-C_{29} fragments (Figure 1). Recently, we reported that photolysis of 2-Ph-benzo[1,3]dioxinones (**I**, **R** = Ph) in the presence of alcohols is a powerful method for the synthesis of otherwise difficult to access sterically hindered salicylate esters (eq 1).⁵ These studies revealed that the corresponding 2-Me-benzodioxinones (**I**, **R** = Me) do not photolyze to the powerful acylating quinoketene intermediate **II**.⁵



In an ideal approach, dilactones would be accessible directly from photochemical homodimerization of a photo-

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Figure 1. Symmetry-based approach for the synthesis of SCH 351448.

responsive fragment with a free C₁₁-alcohol (A1)—a reasonable prospectus based on our finding that photolysis of benzodioxinone I (R = Ph, R' = $-O(CH_2)_7OH$) yielded dilactone 2 as the major product besides monolactone (eq 1).⁶ Alternatively, a hydroxy-protected photoactive fragment (A2) would be photolyzed in the presence of an orthogonal photosilent acyl-acceptor fragment (A3) to yield an ester dimer, necessitating a subsequent nonphotochemical lactonization.

To fully explore the above-mentioned dimerization strategies, we prepared a series of differentially protected fragments as shown in Scheme 1. Our point of departure was the β -hydroxyketone **3**, prepared in multigram quantities as described previously.⁴ Anti-selective reduction with Me₄N(AcO)₃BH⁷ provided an *anti*-diol (82%)⁴ which was differentially protected by treatment with TBSC1 (80%)⁴ followed by TMSCl (88%) to yield bis-silyl ether **4**. Alternatively, a C₉-OMOM-protected fragment **5** was synthesized by Evans–Tishchenko reduction of **3** (MeCHO and SmI₂, 82%),⁸ followed by protection (\rightarrow C₉–OMOM, 95%),

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acetate removal (\rightarrow C₁₁-OH), and TMS protection (\rightarrow 5, 70% for two steps). The terminal olefin of fragments 4/5 served as a handle for C-C bond formation with aryl triflates **6a,b**. Thus, palladium-catalyzed cross-coupling of in situ prepared *B*-alkyl derivatives of 4/5 (9-BBN, THF, 23 °C) with aryl triflates **6a,b** under Suzuki-Miyaura conditions (aq K₃PO₄, cat. PdCl₂dppf, DMF, 23 °C) provided products **7b,d,f,g** after removal of the C₁₁-TMS ether with HF•pyridine.^{9,10}

Next (Scheme 2), we performed a series of experiments designed to obtain dimeric lactones directly by ultraviolet irradiation (300 nm, 0.1 M in CH₂Cl₂, 1 h) of unprotected (C₁₁-OH) photoactive monomers **7b**, **7f**, or **7c** (from **7b** by prolonged exposure to HF•pyridine, Scheme 1). Unfortunately, intramolecular lactonization was the dominant pathway, providing C₁₁-lactone **8f** (34% yield) from alcohol **7f** (16% recovered) and a separable 2:1 mixture of regioisomeric lactones **8c** and **9c** (38% yield) from diol substrate **7c** (30% recovered). Interestingly, photolysis of the monohydroxy

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⁽¹⁰⁾ TMS protection gave optimal results for this fragment coupling.



substrate **7b** was accompanied by silyl migration to yield two regioisomeric lactones **8b** and **9b** in a ratio of 5:1 (54% yield).

The intrinsic propensity for intramolecular cyclization of **7b,c,f** necessitated a circumvention to stage an enforced heterodimerization of an orthogonal photosilent/photoactive pair. Unfortunately, irradiation (300 nm, CH₂Cl₂, 1 h) of a mixture of **7a** (1 equiv) and **7d** (3 equiv) did not provide the desired ester dimer but yielded a TBS-transposed monolactone **9b** (~10%) from intramolecular cyclization of **7a**, unreacted **7d** (10–15%) and a diol **7h** (from **7d**, 65%) as shown in Scheme 2. Based on this result, we (1) exchanged the labile C₁₁-OTMS in the photoreactive partner for a trifluoroacetate (**7e**, 92% from **7b**, Scheme 1) to eliminate intramolecular cyclization, and (2) flanked the C₁₁-alcohol in the acyl-acceptor fragment with a C₉-OMOM (i.e., **7g**) to decrease steric hindrance around the C₁₁-alcohol and increase stability.

As shown in Scheme 3, irradiation (300 nm, CH₂Cl₂, 1 h) of a mixture of **7e** (0.2 M, 1 equiv) and **7g** (2 equiv) did yield the desired ester **10** in 60% yield based on consumed **7e** (**7e**,**g** recovered in 50% and 80% respectively). With this key event accomplished, lactone **12** was obtained in 50% yield by treatment of alcohol **11** (from **10** by trifluoroacetate removal, 94%), with NaHMDS (THF, -78 °C).^{3,11} Finally, hydrogenolytic debenzylation gave diacid **13**, which upon



treatment with 48% aq HF in MeCN, followed by hexane extraction from aq HCl (4 N, saturated with NaCl),³ yielded **1** contaminated with a minor byproduct (80%).¹² HPLC purification yielded pure material with ¹H and ¹³C NMR and mass spectral data in complete agreement with those reported for natural¹ and synthetic³ **1**.¹³

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Supporting Information Available: Experimental procedures, characterization data, and copies of NMR spectra for new compounds, synthetic **1**, and natural **1**. This material is available free of charge via the Internet at http://pubs.acs.org.

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^{(12) &}lt;sup>1</sup>H NMR shifts corresponding to this byproduct were also visible in the spectrum of the natural product (from ref 3). Comparison ¹H NMR spectra are provided in the Supporting Information.

⁽¹³⁾ Our material was destrorotatory: $[\alpha]^{20}{}_{D} = +22.4$ (*c* 0.10, CHCl₃), similar to synthetic **1** of ref 3: $[\alpha]^{13}{}_{D} = +31.2$ (*c* 0.73, CHCl₃). The specific rotation and absolute configuration of the natural sample were not reported.