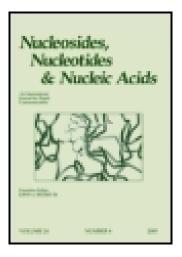
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A NEW AND SHORT CONVERGENT SYNTHETIC STRATEGY TO CARBOCYCLIC NUCLEOSIDES

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□ An efficient synthesis for racemic cyclopent-3-en-1-yl nucleoside analogues has been developed starting from cyclopentadiene. The key step is the regioselective hydroboration of a mixture of intermediate alkylatede cyclopentadienes to give one cyclopentenol.

Keywords Carbocyclic nucleosides; cyclopentenyl nucleosides; Mitsunobu coupling; purines; pyrimidines

INTRODUCTION

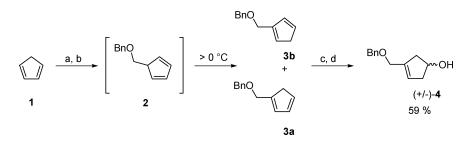
Carbocyclic nucleosides are compounds in which the furan ring has been replaced by a carbocyclic system. These nucleoside analogues have attracted considerable interest due to their important biological activity.^[1] Various structural modifications were introduced in both the heterocyclic base and the sugar moiety.^[2] Such compounds are stable to hydrolysis by phosphorylases that cleave the glycosidic bond in conventional nucleosides.^[3] The bioactivity of the naturally occurring carbocyclic nucleosides (-)-aristeromycin and (-)-neplanocin A led to an interest in this class of compounds, resulting in the generation of a number of antiviral carbocyclic nucleosides. In the past, carbocyclic nucleoside analogues like Abacavir^[4] (Ziagen) and Entecavir^[5] (Baraclude) showed encouraging in vitro and in vivo activities. Recently, Abacavir was approved as a HIV-drug for clinical application. Moreover, Entecavir was approved by FDA in 2005 for the treatment of chronic HBV infections.

RESULTS AND DISCUSSION

Recently, we published a new procedure for the preparation of racemic cyclopent-3-en-1-yl nucleoside analogues.^[7] Cyclopentenol **4** was prepared

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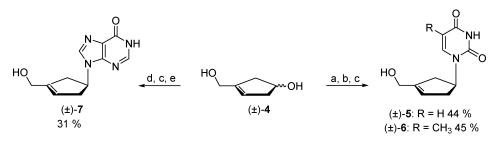
for a convergent approach to carbocyclic nucleosides. This cyclopentenol 4 was developed from cyclopentadiene 1 that was deprotonated and alkylated with benzyloxymethyl chloride to give the diene 2. This material undergoes isomerization into *two* thermodynamically more stable cyclopentadienes 3a,b. These two dienes were hydroborated with 9-BBN as a sterically demanding hydroboration reagent to (\pm) -3-[(benzyloxy)methyl]cyclopent-3-en-1-ol 4 in 42% overall yield. Next, the diene mixture was treated with 0.3 equiv. of BH₃ in THF. Surprisingly, also in this case cyclopentenol 4 was isolated as the sole product in 59% overall yield after alkaline work-up. (Scheme 1).



SCHEME 1 a)NaH, 0°C, 0.5 hours, THF; b) benzyloxymenlthyl chloride, -50° C, 2 hours, rt, 12 hours, THF; c) BH₃ *THF, 24 hours, THF; d) NaOH/H₂O₂ 12 hours, THF.

To prove the suitability of cyclopentenol to act as starting material for the synthesis of carbocyclic nucleosides, pyrimidine heterocycles were introduced using the modified Mitsunobu reaction conditions as reported previously.^[8,11] Thus, N3-benzoylthymine and N3-benzoyluracil were condensed successfully with cyclopentenol 4. Alkaline treatment of the crude reaction product led to O-benzylcyclopentenyl nucleosides. In the case of 3-benzovlthymine the $N1/O^2$ -regioselectivity was found to be 5:1 while a 9:1-regioselectivity was found for 3-benzoyluracil. Debenzylation of the nucleosides proceeded in 76-82% yield using BCl₃ at -78°C to give cyclopentenyl nucleoside analogues, respectively.^[12] Mitsunobu coupling of precursors (\pm) -4 with 6-chloropurine led to the formation of the protected purine nucleosides in 54% yield, respectively.^[7] The N9/N7-regioselectivity was 9:1 and the two isomers were separated by column chromatography. Finally, removal of the O-benzyl group of the nucleoside proceeded in 82% yield using BCl_3 at $-78^{\circ}C$ to give cyclopentenyl nucleoside analogues 5,6. In addition, the purine analogue was converted to the inosine nucleoside analogue (\pm) -7 by treatment of the chloropurine derivative with sodium methanolate and 2-mercaptoethanol in 68% yield (Scheme 2).

In summary, we have elaborated an interesting new pathway towards the synthesis of carbocyclic pyrimidine and hypoxanthine nucleoside analogues. The method starts from cheap cyclopentadiene and is a powerful addition



SCHEME 2 a) DIAD, PPh₃, N3-^{Bz}Ura/N3-^{Bz}Thy, -40°C, 2 hours, CH³ CN; b) NaOH/CH₃ OH 1%, 12 hours; c) BCl₃ 5 hours, -78°C, CH₂Cl₂; d) DIAD, PPh₃, 6-chloropurine, 24 hours, THF; e)2-mercaptoethanol, NaOMe, MeOH, 60°C, 4 hours.

to our previously reported methodology. The biological evaluation of the compounds is under investigation.

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