## TOTAL SYNTHESIS OF CYCLO-L-RHAMNOHEXAOSE BY A STEREOSELECTIVE THERMAL GLYCOSYLATION

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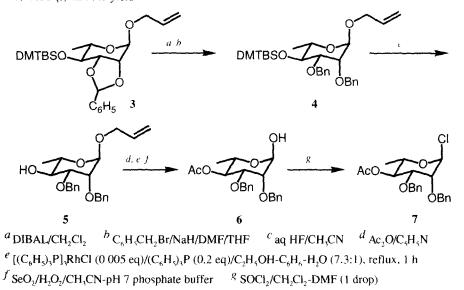
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**Abstract**: The first cyclooligosaccharide of L series, cyclo-L-rhamnohexaose, have been synthesized from L-rhamnose by  $\alpha$ -selective thermal glycosylation and PhSeOTf promoted cycloglycosylation.

Significant attentions have been focused on the inclusion chemistry of cyclodextrins and the related compounds,<sup>1</sup> and synthetic studies of cyclooligosaccharides are continued intensively. For example, Ogawa and co-workers reported the total synthesis of cyclodextrins<sup>2,4</sup> and a manno isomer of cyclodextrins<sup>5,7</sup> by means of phenylselenyl triflate promoted cycloglycosylations. Synthesis of 1-3 $\beta$  linked cycloglucohexaose was reported by Collins' group.<sup>8</sup> Modification of  $\alpha$ -cyclodextrin into trimethyl,<sup>9</sup> 1,3-anhydro,<sup>10</sup> and a chimera analog<sup>11</sup> have also reported recently. However, cyclooligosaccharides available today are only limited to D series. As we have developed a novel thermal glycosylation procedure,<sup>12,13</sup> which can be applied to rhamnosylation with high  $\alpha$ -selectivity,<sup>14</sup> we designed the synthesis of cyclo-L-rhamnohexaose (1), the first cyclooligosaccharide of L series. Herein described is the total synthesis of 1, which would open a new dimension of the inclusion chemistry.

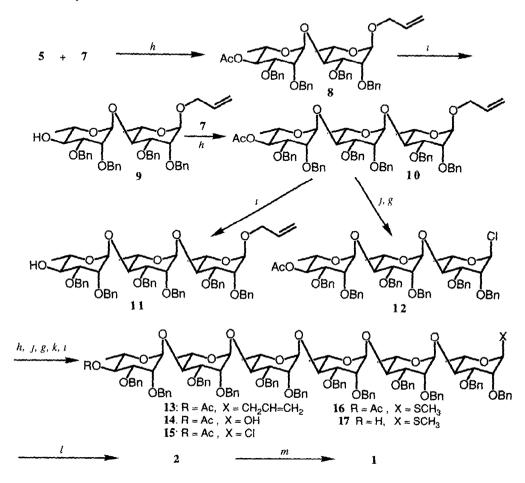
L-Rhamnose monohydrate was successively protected by allyl alcohol (1-OH), <sup>15</sup> benzaldehyde dimethylacetal (2,3-OH), and *tert*-butyldimethylsilyl chloride (4-OH), to give **3** in 54% overall yield. DIBAL reduction of **3**<sup>16</sup> in dichloromethane afforded a mixture of 2- and 3-benzylethers in a ratio of 5:3, which was then

benzylated by benzyl bromide and NaH in DMF-THF (1·3) to give allyl 2.3-*O*-benzyl-4-*O*-tert-butyldimethyl-silyl- $\alpha$ -L-rhamnopyranoside (4),  $[\alpha]_D^{23}$ -44 6° ( $\epsilon$  1.5, CHCl<sub>4</sub>), in 88% yield. The silyl group of 4 was cleaved by aqueous hydrogen fluoride in acetonitrile to give the alcohol 5,  $[\alpha]_D^{23}$ -2 6° ( $\epsilon$  2 1, CHCl<sub>3</sub>), in 92% yield After acetylation, allyl group was removed in two steps to the hemiacetal 6 in 97% yield Rhodium complex catalyzed isomerization of double bond into 1,2-position followed by an oxidative hydrolysis using H<sub>2</sub>O<sub>2</sub>-SeO<sub>2</sub> in acetonitrile and phosphate buffer. The chlorination of 6 with SOCl<sub>2</sub> afforded the rhamnosyl chloride 7,  $[\alpha]_D^{17}$ -75.6° ( $\epsilon$  4 8, CHCl<sub>3</sub>), in 97% yield



Thermal glycosylation of **5** with chloride **7** (1.4 equiv) was achieved by heating the mixture at 70°C for 15 h in the presence of  $\alpha$ -methylstyrene (3 equiv), the acid scavenger, giving rise to a single stereoisomer of the  $\alpha$ -1,4-disaccharide **8**,  $[\alpha]_0^{18}$  -15 7° ( $\epsilon$  2.6, CHCl<sub>3</sub>), in 60% yield <sup>17 18</sup> Hydrolysis of the acetyl group in **8** yielded the alcohol **9**,  $[\alpha]_0^{17}$  -5 1° ( $\epsilon$  1.7, CHCl<sub>3</sub>), which was again subjected to the thermal rhamnosylation with **7** (1.4 equiv) under the similar conditions (40 h),<sup>17</sup> to give the trimer **10**,  $[\alpha]_0^{18}$  -8 1° ( $\epsilon$  2.9, CHCl<sub>3</sub>), in 52% yield. While methanolysis of **10** afforded the trimer alcohol **11**,  $[\alpha]_0^{17}$  +4.0° ( $\epsilon$  1.2, CHCl<sub>3</sub>), in 78% yield, the PdCl<sub>2</sub> catalyzed hydrolysis and subsequent chlorination of **10** yielded the trimer chloride **12**,  $[\alpha]_0^{18}$  -25.1° ( $\epsilon$  0.7, CHCl<sub>3</sub>), in 80% yield. The thermal coupling of **11** and **12** (1.2 equiv) for 24 h at 70°C and an additional 1.5 h at 90°C in the presence of  $\alpha$ -methylstyrene (3 equiv) afforded the hexamer **13**,  $[\alpha]_0^{18}$  +2 5° ( $\epsilon$  0.6, CHCl<sub>3</sub>), in 60% yield. The hexamer **13** was successively subjected to the deprotection at the anomeric center by PdCl<sub>2</sub>/aq AcOH (to **14** in 76% yield), chlorination [to **15**,  $[\alpha]_0^{22}$ -7.4° ( $\epsilon$  0.5, CHCl<sub>3</sub>), in 73% yield], *S*-methylation [to **16** ( $\alpha/\beta$  2:3) in 88% yield], and hydrolysis of acetyl group to give hexamer alcohol **17**  $[\alpha]_0^{18}$  +20.3° ( $\epsilon$  1.1,

CHCl<sub>3</sub>), in 79% yield.



<sup>h</sup> α-methylstyrene (3 equiv) 70°C, 15-40h  $^{-1}$  CH<sub>3</sub>ONa/CH<sub>3</sub>OH  $^{-J}$  PdCl<sub>2</sub> (1.2 eq)/AcOH-H<sub>2</sub>O, (20:1) room temp, 9h.  $^{-k}$  (C<sub>4</sub>H<sub>9</sub>)<sub>3</sub>SnSCH<sub>2</sub>/BF<sub>3</sub>/molecular sieves AW-300/CH<sub>2</sub>Cl<sub>2</sub>, 0°C, 1 h.  $^{-J}$  C<sub>6</sub>H<sub>3</sub>SeOTf/molecular sieves-4A/(CH<sub>2</sub>Cl)<sub>2</sub>.  $^{-m}$  H<sub>2</sub>/Pd(OH)<sub>2</sub>/CH<sub>3</sub>OH-CH<sub>3</sub>CO<sub>2</sub>C<sub>2</sub>H<sub>3</sub>-H<sub>2</sub>O (12:1:1).

The hexamer alcohol 17 was treated at -20°C for overnight with phenylselenyl triflate in 1,2-dichloroethane in the presence of molecular sieves 4A,6 and the cyclization product 2,  $\{\alpha\}_D^{22} + 30.3^{\circ}(c\ 0.15, CHCl_3)$ , was isolated as colorless crystals, mp 139-141°C, in 23% yield after silica gel column chromatography. This product exhibited sharp monomer-like <sup>1</sup>H and <sup>13</sup>C NMR spectra in CDCl<sub>3</sub>. <sup>19</sup> Upon hydrogenolysis of 2 in the presence of 20%Pd(OH)<sub>2</sub>-C, the first cyclooligosaccharide of L-series,  $\alpha$ -cyclo-L-rhamnohexaose (1)  $\{\alpha\}_D^{23}$ -18.5° ( $c\ 0.085$ , CH<sub>3</sub>OH), was obtained in 72% yield as amorphous solid. The structure of 1 was verified on

the basis of FAB mass spectrum of m/z 899 (M + Na)<sup>4</sup> and 915 (M + K)<sup>2</sup> as well as <sup>3</sup>H and <sup>13</sup>C NMR spectra in  $D_2O^{20}$ . Studies including the detailed characterization of **I** and its possibility as a host compound of inclusion chemistry are in progress.<sup>21</sup>

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- 17 Thermal coupling of 5 and 7 at 120°C for 2 h with or without  $\alpha$ -MS sometimes afforded much higher yield, however those conditions were not effective for larger scales (2  $\sim$  3 mmol)
- 18. NMR chemical shifts along with coupling constants in parenthesis (Hz) of anomeric protons and carbons in CDCl, are as follows: 5, 8 4 86 (1 64) and 8 97 0 (167 3), 7, 8 6 09 (br s) and 8 91 3 (181.7); 8, 8 4 83 (1.65), 5.31 (1 65), and 8 96 8 (168 4), 99.8 (172.1); 9, 8 4.84 (1 1), 5 23 (br s), and 8 97.0 (165 8), 99 8 (170 2); 10, 8 4.85 (1 65), 5 28 (1 65), 5 30 (1 65), and 8 99.08 (168.8), 99 59 (176 1), 99.84 (171.7); 11, 8 97 0 (170 2), 99 6 (170 2), 99.7 (170.2), 12, 8 5.29 (2H. bt s), 6 06 (1 64), and 8 91 4 (183 1), 99 6 (172 8), 99 7 (172 8), 13, 8 4.86 (br s), 5 27-5 31 (5H. complex), and 8 97.0, 99.2, 99.3, 99.4, 99 7, 15, 8 5 26 (2H. br s), 5.29 (1H. br s), 5.31 (2H. br s), 6 07 (1 47), and 8 91 5, 99 3, 99.4, 99 8, 99.8; 16, 8 84 8 (137 8), 85 7 (149.6), 99 4 (176.1), 99.4 (176.1), 99 6 (173.1), 99.8 (176 1), 17, 8 84.8 (135.5), 85 7 (149.6), 99 4 (170.0), 99.4 (170.0), 99 7 (172.0), 99 7 (172.0).
- 19  $^{1}$ H (400 MHz) and  $^{13}$ C (100 MHz) NMR spectrum of 2 m CDCl<sub>3</sub>  $^{3}$  8 1.44 (18H, d, J = 6 1 Hz), 3 56 (6H, t, J = 8 8 Hz), 3 76 (6H, dd, J = 8 8, 2.4 Hz), 3.81 (12H, m), 4.25 (6H, d, J = 12 1 Hz), 4 31 (6H, d, J = 12.1 Hz), 4 48 (6H, d, J = 12 1 Hz), 4.53 (6H, d, J = 12.1 Hz), 4.92 (6H, s), 7.15-7.26 (60H, m), and 818.4q, 68.5d, 71 5t, 72 5t, 76 7d, 78 4d, 100 9d ( $J_{CP} = 163.8$  Hz), 127 0d, 127 4d, 127 5d, 127 6d, 127.6d, 128.0d, 128.1d, 128.3d, 138.3s, 138.4s
- 20.  $^{1}$ H (600 MHz) and  $^{13}$ C (150 MHz) NMR spectrum of 1 in D<sub>2</sub>O (*tert*-butyl alcohol as internal standard at  $\delta$  1.25 and  $\delta$  32.25, respectively).  $\delta$  1 39 (18H, d, J = 6.4 Hz), 3 48 (6H, t, J = 9.0 Hz), 3 88 (6H, m), 3.89 (6H dd, J = 9.0, 2 6 Hz), 4 01 (6H t, J = 2.6 Hz), 4.91 (6H, d, J = 2.6 Hz), and  $\delta$  20 0q, 70 9d, 72 5d, 73 0d, 104.8d
- 21. We propose the common name " $\alpha$ -cycloawaodorin" to 1, the first cyclooligosaccharide of L-series, in connection with the area where the compound has been prepared