CEPHALOSPORIN NITROGEN MUSTARD CARBAMATE PRODRUGS FOR "ADEPT".

Rikki P. Alexander ^a*, Nigel R. A. Beeley ^a, Maraid O'Driscoll ^a, Faye P. O'Neill ^a, T. Andrew Millican ^a, Andrew J. Pratt ^b and Frances W. Willenbrock ^a.

a) CELLTECH, 216, Bath Road, Slough, SL1 4EN, Berkshire, UK.
b) The Dyson Perrins Laboratory, South Parks Road, Oxford OX1 3QY, Oxfordshire, UK.
* Author to whom correspondance should be addressed.

Key words: ADEPT, β-lactamase, Moncional Antibody, Nitrogen Mustard, Prodrug.

Abstract: A series of cephalosporin carbamates, (6a-d) and (7), functionalised with masked nitrogen mustards has been synthesised. These are potential prodrugs for interaction with a monoclonal antibody- β -lactamase tumour targeting system. The β -lactamase catalysed hydrolysis of (6a-d) and (7) to generate free nitrogen mustards has been characterised.

Conventional anti-cancer chemotherapy is frequently associated with serious side effects. Indeed, a number of promising cytotoxic agents have failed to progress to clinical evaluation due to their lack of selectivity over normal cells. Earlier attempts to capitalise on using enzymes endogenous to tumour tissue to deliver a cytotoxic agent via cleavage of a prodrug did result in tumour regression in animal models ¹. However, many such enzymes are so similar to those of normal tissues that no real benefit emerged. The ready availability of monoclonal antibodies which target specifically to tumour tissue allows the concept of Antibody Directed Enzyme Prodrug Therapy (ADEPT) to be realised ². This consists of i) localisation of an antibody-enzyme conjugate or chimeric construct to a tumour; ii) administration of a relatively non-toxic substrate for the enzyme (prodrug) which can liberate on cleavage a cytotoxic agent in the vicinity of the tumour.

In this context, β -lactamases are of particular interest for the following reasons: i) they are small, soluble monomeric enzymes without a mammalian counterpart; ii) they not only hydrolyse penicillins but also cephalosporins, via a mechanism involving expulsion of a 3' leaving group and iii) the enzyme is particularly tolerant to a wide variety of substituents at this position ³. Accordingly, we have devised a series of cephalosporin 3' carbamate derivatives which allow the release of a cytotoxic agent on exposure to β -lactamase as shown below. In the case of the nitrogen mustards, which owe their cytotoxicity to alkylation of DNA via



 $HN(CH_2CHMeBr)_2, H_2NC_6H_4N(CH_2CH_2CI)_2; R = \frown$

an aziridinium intermediate, such carbamates should be relatively innocuous. The chemical realisation of such species was not trivial, involving careful choice of protecting groups to prevent lactonisation and reagents which minimised the isomeristion of $\Delta 3$ to $\Delta 2$ cephalosporins ⁴. Having examined both pnitrobenzyl and t-butyl esters to protect the 4-carboxylate we eventually settled upon the diphenyl methyl ester.



Selective hydrolysis of the 3' acetate of cephalothin (1) ⁵ followed by esterification with diphenyldiazomethane gave the corresponding 4-ester (2) ⁶. Treatment of (2) with carbonyldiimidazole ⁷, phosgene, diphosgene ⁸ or triphosgene ⁹ followed by diethanolamine failed to provide the desired products. However reaction of (2) with p-nitrophenyl chloroformate (1eq., 1eq. pyridine in THF) ¹⁰ gave the isolable intermediate carbonate (3a) as a white solid. Upon exposure of (3a) to diethanolamine (1eq. in pyridine at 0°C, 2hrs.) a mixture of $\Delta 2$ and $\Delta 3$ isomers was formed in favour of the undesired $\Delta 2$ isomer (3:1 by 300 MHz.NMR). In order to obtain satisfactory formation of the desired $\Delta 3$ carbamate it was necessary to isolate and purify the pentafluorophenyl carbonate (3b). Chloroformate formation in situ (C₆F₅OH/NaH/COCl₂ in THF) followed by addition of (2) in the presence of pyridine (1eq.) gave (3b) (70%). Exposure to diethanolamine (1eq., -26°C in DMF) yielded the desired $\Delta 3$ isomer (4a) as the major product. The ratios of $\Delta 2$ to $\Delta 3$ isomers for a variety of amines (best conditions) are summarised in table 1.

Amine	Product	Solvent	Δ3:Δ2	Yield
HN(CH2CH2OH)2	(4a)	DMF	34:1	95%
HN(CH2CHMeOH)2	(4b)	DMF	26:1	95%
H2NC6H4N(CH2CH2OH)2	(5)	DMSO/C ₅ H ₅ N (1:1)	16:1	83%
HN(CH2CH2CI)2 .HCI	(4c)	C₅H₅N	28:1	77%

Table 1: Acylation of carbonate (3b) with various secondary amines.

Diethanolamine cephems (4a,b) were converted to their corresponding halo derivatives (2.2 eq. PPh₃, NCS or NBS in THF)¹¹ with complete retention of double bond integrity by 300 MHz. NMR. The arylaminodiol (5) was chlorinated (2.2 eq. PPh₃.Cl₂ complex in dry pyridine, 0°C)¹² again retaining the Δ 3 double bond. Deprotection (TFA, CH₂Cl₂, Anisole) gave (6a), (6b), (6c), (6c) or (7) after purification by reverse phase HPLC.



Examination of the prodrug principle was achieved by exposure of the cephalosporin mustards to the class C β -lactamase from *Enterobacter clocae* P99 ¹³. The kinetic parameters for β -lactamase catalysed hydrolysis of each prodrug are summarised in Table 2. In the case of prodrug (6a), enzyme catalysed hydrolysis was also monitored by ¹H NMR at 300 MHz. Liberation of free *bis*-chloroethylamine was concomitant with β -lactam hydrolysis. The toxicity ratios (mouse lymphoma cells) of these nitrogen mustard prodrugs in the presence and absence of β -lactamase along with other biological data will be reported elswhere.

Prodrug	К _М (µМ)	k _{cat} (s ⁻¹)	K _{cat} /K _M (µM ⁻¹ s ⁻¹)
Ceph-CO-N(CH ₂ CH ₂ Cl) ₂ (6a)	108.5	510	4.7
Ceph-CO-N(CH ₂ CHMeCl) ₂ (6c)	182	130	0.7
Ceph-CO-N(CH ₂ CH ₂ Br) ₂ (6b)	71.5	150	2.1
Ceph-CO-N(CH ₂ CHMeBr) ₂ (6d)	40	120	3.0
Ceph-CO-HNC ₆ H ₄ N(CH ₂ CH ₂ Cl) ₂ (7)	142.5	181	1.3

Table 2:	Michaelis-Menton parameters ¹⁴ for hydrolysis of cephem carbamate
	prodrugs by Enterobacter clocae P99 β-lactamase.

References¹⁵.

- M. E. Whisson and T. A. Connors, Nature, 206, 689 (1965); T. A. Connors 1. and M. E. Whisson, Nature, 210, 866 (1966)
- 2. K. D. Bagshawe, Br. J. Cancer, 56, 531 (1987)
- 3. H. A. Albrecht, G. Bestid, K-K Chan, J. G. Christenson, R. Cleeland, K. H. Deitcher, N. H. Georgopapdakou, D. D. Keith, D. L. Pruess, J. Sepinwall, A. C. Specian Jr., R. L. Then, M. Weigele, K. F. West and R. Yang, J. Med. Chem., 33, 77 (1990)
- R. B. Woodward, K. Heusler, J. Gostelli, P. Naegali, W. Oppolzer, R. Ramage, 4. S. Ranganathan and H. Vorbruggen, J. Amer. Chem. Soc., 88, 852 (1966); G. V. Kaiser, R. D. G. Cooper, R. E. Koehler, C. F. Murphy, J. A. Webber, I. G. Wright and E. M. Van Heyningen, J. Org. Chem., 35, 2430 (1970)
- 5.
- S. Mobashery and M. Johnson, *J. Biol. Chem.*, <u>17</u>, 7879 (1986) H. Yamanaka, T. Chiba, K. Kawabata, H. Takasugi, T Masugi and T. Takaya, 6. J. Antibiot., 38, 1738 (1985)
- H. A. Staab and A. Mannschreck, Chem. Ber., 95, 1294 (1962) 7.
- L. N. Pridgen, J. Prol Jr., B. Alexander and L. Gillyard, J. Org. Chem., 54, 8. 3231 (1989)
- 9. H. Eckert and B. Forster, Angew. Chem. Int. Edit., 26, 894 (1987)
- 10. R. L. Letsinger and K. K. Ogilvie, J. Org. Chem., 32, 296 (1967)
- 11. A. K. Bose and B. Lai, Tetrahedron Letts., 3937 (1973)
- G. A. Wiley, R. L. Hershkowitz, B. M. Reinz and B. C. Chung, J. Amer. Chem. 12. Soc., 86, 964 (1964)
- 13. P. C. Fleming, M. Goldner and D. G. Glass, Lancet, j, 1399 (1963)
- Kinetic measurements were made in 0.1 M phosphate buffer, pH 7.0, 37 °C. 14. Hydrolysis was monitored by absorbance change at 265 nM. Extinction coefficients were determined from absorbance changes after complete hydrolysis. Kinetic parameters were obtained from either complete progress curves using half-time analysis of the integrated Michaelis-Menton equation (C. W. Wharton and R. J. Szawelski, Biochem. J., 203, 351 (1982)) or initial rates over a range of substrate concentrations using the program "Enz Fitter" of Dr. Robin J. Leatherbarrow.
- 15. For an alternative cephalosporin based approach to "ADEPT" see: T. A. Shepherd, L. N. Jungheim, D. L. Meyer and J. Starling, BioMed. Chem. Lett., 1, 21 (1991)

(Received in UK 5 April 1991)