

**Note****Synthesis of 1D-1,2-anhydro-*myo*-inositol**

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Glycosidase inhibitors are of major interest for the study of glycosidase activity and in the control of disorders of carbohydrate metabolism<sup>1</sup>. Naturally occurring<sup>2</sup> and mechanism-based inhibitors<sup>3</sup> have been reported. Conduritol B epoxide (**5**, 1,2-anhydro-*myo*-inositol), the first reported site-directed irreversible inhibitor, is still considered to be one of the most suitable glucosidase inhibitors<sup>1a</sup>. Racemic **5** reacts with several  $\beta$ -D-glucosidases by nucleophilic attack of a carboxylate anion of the active site on the protonated epoxide to give an ester of (+)-*chiro*-inositol<sup>1a</sup>, which indicates that only D-**5** had reacted.

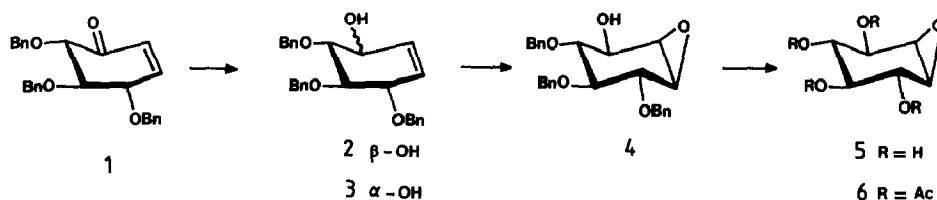
As a part of a project on the synthesis of derivatives of glycosylphosphatidylinositol which might be implicated in a second-messenger mechanism for the signal transduction of insulin<sup>5</sup>, we have found that 2,4/3-tribenzyloxycyclohex-5-enone<sup>6</sup> (**1**) can be reduced stereoselectively and then epoxidised to give 1D-1,2-anhydro-*myo*-inositol (**5**) in good yield.

Racemic **5** has been synthesised<sup>3a,7</sup> from *myo*-inositol and from benzene, and the L form has been prepared<sup>8</sup> from an optically active *chiro*-inositol derivative but, apparently, no synthesis of the D form has been reported.

Reduction of **1** with lithium aluminum hydride or di-isobutylaluminum hydride gave<sup>9</sup> a 6:4 mixture of the epimeric alcohols **2** and **3**. In our hands, the reduction of **1** with lithium aluminum hydride gave a 1:1 mixture and the use of sodium borohydride at –75° gave a 1:3 mixture.

The presence of cerium chloride<sup>10</sup> in borohydride reductions changes the stereoselectivity when applied<sup>11</sup> to cyclohexenones related to **1**. Thus, similar reduction of **1** at –75° afforded a 14:1 mixture of **2** and **3**, from which the desired isomer **2** was isolated in a yield of 90%. The nature of protecting groups affects the stereoselectivity, since our results are in sharp contrast with those<sup>12c</sup> obtained when *tert*-butyldimethylsilyl ethers were used.

Epoxidation of **2** with *m*-chloroperoxybenzoic acid yielded 84% of **4**, debenzylation of which afforded 1D-1,2-anhydro-*myo*-inositol (**5**, 81%),  $[\alpha]_D +66^\circ$  (cf. –70° for the L enantiomer<sup>8</sup>). The  $^1\text{H}$ - and  $^{13}\text{C}$ -n.m.r. spectra and analytical data of the tetra-acetate (**6**) of **5** accorded with the structure proposed.



#### EXPERIMENTAL

**General methods.** — Melting points were determined on a Kofler hot-stage apparatus and are uncorrected. T.l.c. was performed on Silica Gel GF<sub>254</sub> (Merck) with detection by charring with sulfuric acid. Column chromatography was performed on Silica Gel (Merck 70–230). <sup>1</sup>H-N.m.r. spectra were recorded with a Varian XL-300 (300 MHz) or Bruker AM-200 (200 MHz) spectrometer, and <sup>13</sup>C-n.m.r. spectra with a Bruker AM-200 (50 MHz) spectrometer. Optical rotations were determined with a Perkin-Elmer 141 polarimeter.

*1D-1,2,3-Tri-O-benzyl-(1,3/2,4)-5-cyclohexene-1,2,3,4-tetrol* (**2**, *1,2,3-tri-O-benzylconduritol B*). — A stirred mixture of **1** (ref. 6) (565 mg, 1.36 mmol), methanol (70 mL), and CeCl<sub>3</sub>·7H<sub>2</sub>O (490 mg, 0.97 equiv.) at -75° was treated with sodium borohydride (76 mg, 5.88 equiv.) for 5 h. Water (40 mL) was then added, volatile solvents were evaporated under vacuum, the aqueous phase was extracted with dichloromethane (2 × 50 mL), and the combined extracts were dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated. Column chromatography (1:3 ethyl acetate–hexane) of the residue afforded **2** (515 mg, 91%), m.p. 116–119°, [α]<sub>D</sub> + 117° (*c* 0.9, chloroform); lit.<sup>11</sup> [α]<sub>D</sub> + 114.6°. N.m.r. data: <sup>1</sup>H (200 MHz, C<sub>6</sub>D<sub>6</sub>), δ 7.40–7.20 (m, 15 H, 3 Ph), 5.23 (m, 2 H, H-1,2), 4.67 (d, 1 H, PhCH), 4.58 (s, 2 H, PhCH<sub>2</sub>), 4.38 (d, 1 H, PhCH), 4.19 (s, 2 H, PhCH<sub>2</sub>), 3.92 (m, 1 H, H-6), 3.83 (m, 1 H, H-3), 3.47 (dd, 1 H, J<sub>3,4</sub> 7.5, J<sub>4,5</sub> 10.3 Hz, H-4), 3.17 (dd, 1 H, J<sub>5,6</sub> 7.7, J<sub>4,5</sub> 10.3 Hz, H-5), 1.44 (d, 1 H, J 4.4 Hz, HO-6); <sup>13</sup>C (CDCl<sub>3</sub>), 138.6 (C-ipso), 129.4 [C-1 (or 2)], 128.6, 128.4, 127.9, 127.8, 127.7, 127.6 (aromatic), 127.0 [C-2 (or 1)], 84.3, 83.3, 80.5, 75.3, 72.2, 71.9, and 71.8 p.p.m.

Anal. Calc. for  $C_{22}H_{28}O_4$ : C, 77.87; H, 6.78. Found: C, 77.60; H, 6.70.

*1D-1,2-Anhydro-4,5,6-tri-O-benzyl-myo-inositol* (**4**). — A solution of **2** (100 mg, 0.24 mmol) in dichloromethane (5 mL) was treated with 55% *m*-chloroperoxybenzoic acid (121 mg, 1.60 equiv.) at room temperature for 2 days. Aqueous 10% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (5 mL) and dichloromethane (5 mL) were then added, and the organic phase was dried (Na<sub>2</sub>SO<sub>4</sub>) and concentrated. Column chromatography (1:2 ethyl acetate–hexane) of the residue gave **4** (87 mg, 84%), m.p. 147–150°, [α]<sub>D</sub> + 78° (*c* 0.4, chloroform). N.m.r. data (CDCl<sub>3</sub>): <sup>1</sup>H (300 MHz), δ 7.40–7.30 (m, 15 H, 3 Ph), 4.94 (d, 1 H, PhCH), 4.82 (s, 2 H, PhCH<sub>2</sub>), 4.78 (d, 1 H, PhCH), 4.71 (d, 1 H, PhCH), 4.63 (d, 1 H, PhCH), 4.02 (dd, 1 H, J<sub>1,6</sub> 1.7, J<sub>5,6</sub> 8.1 Hz, H-6), 3.93 (d, 1 H, J<sub>3,4</sub> 7.4 Hz, H-3), 3.48 (m, 2 H, H-4,5), 3.41 (m, 1 H, H-1), 3.23 (d, 1 H, J<sub>1,2</sub> 3.8 Hz, H-2); <sup>13</sup>C, 138.3, 138.2, and 137.4 (C-ipso), 128.6, 128.5, 128.4, 128.3, 128.0, 127.9, 127.8, 127.8, 127.7, 127.6 (aromatic), 83.3, 79.5, 79.3, 75.6, 75.2, 73.2, 71.9, 56.2 and 53.6 p.p.m. (oxirane).

Anal. Calc. for  $C_{27}H_{28}O_5$ : C, 74.98; H, 6.52. Found: C, 75.10; H, 6.86.

**1D-1,2-Anhydro-myo-inositol (5).** — A solution of **4** (427 mg, 0.99 mmol) in methanol (50 mL) and ethyl acetate (10 mL) was treated with hydrogen in the presence of 10% Pd-C at room temperature overnight, then filtered, and concentrated. Column chromatography (1:3 methanol-chloroform) of the residue afforded **5** (129 mg, 81%), m.p. 158–160°, lit.<sup>8,9</sup> 160°;  $[\alpha]_D +66^\circ$  (*c* 0.6, D<sub>2</sub>O); <sup>13</sup>C-N.m.r. data (D<sub>2</sub>O):  $\delta$  76.0, 72.6, 71.9, 71.7, 58.6 and 57.9 p.p.m. (oxirane).

*Anal.* Calc. for C<sub>6</sub>H<sub>10</sub>O<sub>5</sub>: C, 44.45; H, 6.22. Found: C, 44.29; H, 6.35.

The tetra-acetate (**6**) of **5** had m.p. 108–111°,  $[\alpha]_D +80^\circ$  (*c* 0.5, chloroform). N.m.r. data (C<sub>6</sub>D<sub>6</sub>): <sup>1</sup>H (300 MHz),  $\delta$  5.53 (dd, 1 H, J<sub>5,6</sub> 9.0, J<sub>4,5</sub> 10.6 Hz, H-5), 5.33 (dd, 1 H, J<sub>1,6</sub> 1.7, J<sub>5,6</sub> 9.0 Hz, H-6), 5.30 (dd, 1 H, J<sub>3,4</sub> 8.0, J<sub>4,5</sub> 10.6 Hz, H-4), 5.20 (d, 1 H, J<sub>3,4</sub> 8.0 Hz, H-3), 3.11 (dd, 1 H, J<sub>1,6</sub> 1.7, J<sub>1,2</sub> 3.7 Hz, H-1), 2.76 (d, 1 H, J<sub>1,2</sub> 3.7 Hz, H-2), 1.71, 1.67, 1.67, and 1.66 (4 s, 3 H each, 4 Ac); <sup>13</sup>C, 169.9, 169.5, 169.3, 169.0, 71.6, 71.4, 70.9, 68.1, 54.5 and 53.9 (oxirane), 20.2, 20.1, and 20.0 p.p.m. (2 C).

*Anal.* Calc. for C<sub>14</sub>H<sub>18</sub>O<sub>9</sub>: C, 50.91; H, 5.49. Found: C, 51.15; H, 5.67.

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#### REFERENCES

- (a) P. Lalagerie, G. Legler, and J. M. Yon, *Biochemie*, 64 (1982) 977–1000; (b) E. Truscheit, W. Frommer, B. Junge, L. Müller, D. Schmidt, and W. Wingender, *Angew. Chem. Int. Ed. Engl.*, 20 (1981) 744–761.
- D. D. Schmidt, W. Frommer, B. Junge, L. Müller, W. Wingender, E. Truscheit, and D. Schafer, *Naturwissenschaften*, 64 (1977) 535–536; D. D. Schmidt, W. Frommer, T. Müller, and E. Truscheit, *ibid.*, 66 (1979) 584–585.
- (a) G. Legler, *Methods Enzymol.*, 46 (1977) 368–381; (b) P. J. Marshall, M. L. Sinnott, P. J. Smith, and D. Widdows, *J. Chem. Soc., Perkin Trans. I*, (1981) 366–376; M. K. Tong and B. Ganem, *J. Am. Chem. Soc.*, 110 (1988) 312–313; S. G. Withers, I. P. Street, P. Bird, and D. H. Dolphin, *ibid.*, 109 (1987) 7530–7531; S. G. Withers, K. Rupitz, and I. P. Street, *J. Biol. Chem.*, 263 (1988) 7929–7932; S. Halazy, C. Danzin, A. Ehrhard, and F. Gerhart, *J. Am. Chem. Soc.*, 111 (1989) 3484–3485; G. Caron and S. G. Withers, *Biochem. Biophys. Res. Commun.*, 163 (1989) 495–499.
- (a) G. Legler, *Hoppe-Seyler's Z. Physiol. Chem.*, 349 (1968) 767–774; (b) G. Legler and S. N. Hasnain, *ibid.*, 351 (1970) 25–31; A. Quaroni, E. Gershon, and G. Semenza, *J. Biol. Chem.*, 249 (1974) 6424–6433.
- A. R. Saltiel and P. Cuatrecasas, *Proc. Natl. Acad. Sci. U.S.A.*, 83 (1986) 5793–5797; J. M. Mato, K. L. Kelly, A. Abler, and L. Jarett, *J. Biol. Chem.*, 262 (1987) 2131–2137; A. R. Saltiel and P. Cuatrecasas, *Am. J. Physiol.*, 255 (1988) c1.
- D. Semeria, M. Philippe, J.-M. Delaumeny, A.-M. Sepulchre, and S. D. Gero, *Synthesis*, (1983) 710–713.
- G. Legler, *Hoppe-Seyler's Z. Physiol. Chem.*, 345 (1966) 197–214; T. L. Nagabushan, *Can. J. Chem.*, 48 (1970) 383–384; K. J. Lee, S. A. Boyd, and N. S. Radin, *Carbohydr. Res.*, 144 (1985) 148–154; S. J. Angyal, V. Bender, and J. H. Curtin, *J. Chem. Soc., C*, (1966) 798–800; N. S. Radin and R. R. Vunnam, *Methods Enzymol.*, 72 (1981) 673–684; M. Nakajima, I. Tomida, N. Kurikara, and S. Takei, *Chem. Ber.*, 92 (1959) 173–178.
- S. J. Angyal and J. S. Murdoch, *Aust. J. Chem.*, 22 (1969) 2417–2428; D. Mercier, A. Olesker, S. D. Gero, and J. E. G. Barnett, *Carbohydr. Res.*, 18 (1971) 227–231.
- G. Vass, P. Krausz, B. Quiclet-Sire, J.-M. Delaumeny, J. Cleophax, and S. D. Gero, *C. R. Acad. Sci. Ser. II*, 301 (1985) 1345–1346.
- A. L. Gemal and J. L. Luche, *J. Am. Chem. Soc.*, 103 (1981) 5454–5459.
- N. Chida, M. Ohtsuka, K. Nakazawa, and S. Ogawa, *J. Chem. Soc., Chem. Commun.*, (1989) 436–438; C. Le Drian, E. Vieira, and P. Vogel, *Helv. Chim. Acta*, 72 (1989) 338–347; C. Le Drian, J.-P. Vionnet, and P. Vogel, *ibid.*, 73 (1990) 161–168.