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A Novel One-Step Synthesis of 2-Methoxycarbonyl-thieno[2,3-b]quinolines and 3-Hydroxy-2-methoxycarbonyl-2,3-dihydrothieno[2,3-b]-quinolines*

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Of the various methods for the synthesis of the thieno[2,3-b]quinoline ring¹⁻⁴ and 2,3-dihydro derivatives^{5,6}, only one method^{1,2} can be applied to the synthesis of 2-substituted thieno[2,3-b]quinolines. This method is a four-step synthesis starting with the ring cleavage of quinoline with thiocarbonyl chloride. Besides the use of the latter reagent, the method has the drawback that the starting quinolines required for the synthesis of 5,6,7- or 6,7,8-trisubstituted 2-methoxycarbonylthieno[2,3-b]quinolines are not readily available. Similarly, the synthesis of 2,3-dihydrothieno[2,3-b]quinolines, reported earlier^{5,6}, involves several steps and it cannot be used to incorporate substituents at position 2 and 3. We now report a novel one-step synthesis of 2-methoxycarbonylthieno[2,3-b]quinolines (4) and 3-hydroxy-2-meth-

oxycarbonyl-2,3-dihydrothieno[2,3-b]quinolines (5); these compounds may be substituted in the 3,5,6,7, and/or 8 positions. The synthesis of compounds 5 involves a new type of intermolecular cyclocondensation. The present investigation includes a new, convenient synthesis of 2-chloro-3-acylquinolines (3).

2-Chloro-3-formylquinolines (1 a-g) are prepared according to Ref. ^{7,8}. The reaction of compounds 1 a-g with methyl mercaptoacetate in dimethylformamide in the presence of anhydrous potassium carbonate affords 2-methoxycarbonylthieno[2,3-b]quinolines (4a-g) in 70-80% yields along with 3-hydroxy-2-methoxycarbonyl-2,3-dihydrothieno[2,3-b]quinolines (5a-g) which can be isolated in 10-15% yields by column chromatography. On the other hand, when the reaction of compounds 1a-g with methyl mercaptoacetate is performed in dimethylformamide/water (9/1) in the presence of potassium carbonate, compounds 5a-g are obtained in 60-80% yields along with compounds 4a-g in 10-12% yields.

The I.R. carbonyl absorption band of 3-hydroxy-2-methoxycarbonyl-2,3-dihydrothieno [2,3-b] quinolines (5a-g)

1-6	X¹	X ²	X ³	X4	R (for 2-6 only)
а	Н	Н	Н	Н	H
b	H	OCH ₃	н	н	н
С	н	OCH3	OCH ₃	н	н
ď	осн₃	OCH ₃	OCH ₃	н	Н
е	н	0C ₂ H ₅	OC ₂ H ₅	н	н
f	Н	OCH ₃	OCH ₃	NO ₂	н
g	Н	OC ₂ H ₅	OC ₂ H ₅	NO ₂	Н
h	н	Н	н	н	CH ₃
i	н	OCH ₃	OCH ₃	н	CH ₃
j	н	н	н	н	C ₆ H ₅
k	н	OCH ₃	OCH₃	н	C ₆ H ₅

appears between 1735 and 1745 cm⁻¹, which indicates that the hydroxy group is not involved in hydrogen bonding. Further, the coupling constant of the methine protons $(J_{2,3})$ ranged between 8.5 and 9.0 Hz. Based on these observations, the stereochemistry of compounds $\mathbf{5a-g}$ may be assigned as trans. Reaction of compounds $\mathbf{5a-g}$ with acetic anhydride/pyridine affords the dehydration products $\mathbf{4a-g}$ in almost quantitative yield.

The reaction of aldehydes 1a and 1c with methyl- and phenylmagnesium halides affords the secondary alcohols 2h-k. Oxidation of these alcohols with pyridinium chlorochromate gives the corresponding 2-chloro-3-acylquinolines (3h-k) in nearly quantitative yields. Of these, only quin-

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Table 1. 2-Chloro-3-(1-hydroxyalkyl)-quinolines (2) and 2-Chloro-3-acylquinolines (3) prepared

Prod- uct	Me- thod	Yield [%]		Molecular Formula ^a	M.S. (M ⁺) <i>m/e</i>	I.R. (KBr) ν [cm ⁻¹]	¹ H-N.M.R. (CDCl ₃ /TMS _{int}) δ [ppm]
2h	A	90	72–73°	C ₁₁ H ₁₀ ClNO (207.7)	207	3350 (OH)	1.40-1.50 (d, 3 H, $-\text{CH}_3$, $J = 9.0 \text{ Hz}$); 3.30 (br.s, 1 H, OH); 5.05-5.40 (q, 1 H, $-\text{CH}-\text{OH}$); 7.1-8.0 (m, 4 H _{aron}); 8.30 (s, 1 H, 4-H)
2i	A	96	172–173°	C ₁₃ H ₁₄ ClNO ₃ (267.7)	267	3500 (OH)	1.98-2.05 (d, 3H, CH—C \underline{H}_3 , $J = 6.3$ Hz); 2.60 (br. s, 1H, O \underline{H}); 3.90, 3.95 (2s, 6H, 2OC \underline{H}_3); 5.15-5.35 (q, 1H, C \underline{H} —OH); 6.88 (s, 1H, 8- \underline{H}); 7.28 (s, 1H, 5-H); 8.08 (s, 1H, 4- \underline{H})
2ј	Α	90	8990°	C ₁₆ H ₁₂ ClNO (269.7)	269	3400 (OH)	3.32 (br. s, 1 H, OH); 6.10 (s, 1 H, СН); 6.9–7.9 (m, 9 H _{atom}); 8.30 (s, 1 H, 4-Н)
2k	A	91	152–153°	C ₁₈ H ₁₆ CINO ₃ (329.8)	329	3300 (OH)	3.52 (br. s, 1 H, OḤ); 3.88 (s, 6 H, 2 ОСḤ ₃); 6.18 (s, 1 H, СḤ—OH); 6.88 (s, 1 H, 8-Ḥ); 7.0-7.45 (m, 6 H, 5-Ḥ and 5 H _{arom}); 8.11 (s, 1 H, 4-Ḥ)
3h	В	97	7576°	C ₁₁ H ₈ CINO (205.6)	205	1705 (COCH ₃)	2.70 (s, 3 H, CO—CH ₃); 7.00–8.05 (m, 4H _{arom}); 8.25 (s, 1 H, 4-H)
3i	В	96	172°	C ₁₃ H ₁₂ CINO ₃ (265.7)	265	1665 (COCH ₃)	2.72 (s, 3H, CO—CH ₃); 3.94, 3.99 (2s, 6H, 2OCH ₃); 6.99 (s, 1H, 8-H); 7.26 (s, 1H, 5-H); 8.19 (s, 1H, 4-H)
3ј	В	92	82-83°	C ₁₆ H ₁₀ ClNO (267.7)	267	1685 (COCH ₃)	$6.9-8.0$ (m, $9 H_{arom}$); 8.12 (s, $1 H, 4-H$)
3k	В	97	142-144°	C ₁₈ H ₁₄ ClNO ₃ (327.8)	327	1670 (COCH ₃)	3.92, 3.98 (2s, 6H, 2OC \underline{H}_3); 7.01 (s, 1H, 8- \underline{H}); 7.28 (m, 4H, 5- \underline{H} , 3'- \underline{H} , 4'- \underline{H} , 5'- \underline{H}); 7.72–7.84 (dd, 2H, 2'- \underline{H} , 6'- \underline{H} , $J_o = 8.10$ Hz, $J_m = 2.7$ Hz); 7.99 (s, 1H, 4- \underline{H})

 $[^]a$ The microanalyses showed the following maximum deviations from the calculated values: C $\pm 0.39;$ H $\pm 0.36;$ N $\pm 0.36.$

Table 2. 2-Methoxycarbonylthieno[2,3-b]quinolines (4) prepared

Prod- uct	Me- thod	Yield [%]	m.p. [°C]	Molecular formula ^a	M.S. (M ⁺) <i>m/e</i>	I.R. (KBr) v [cm ⁻¹]	¹ H-N.M.R. (CDCl ₃ /TMS _{int} , if not stated otherwise) δ [ppm]
4 a	C E	85 97	155°	C ₁₃ H ₉ NO ₂ S (243.4)	243	1715 (COOCH ₃)	3.95 (s, 3H, COOCH ₃); 7.45-8.12 (m, 5H _{arom}); 8.50 (s, 1H, 3-H)
4 b	C E	87 95	183–184°	C ₁₄ H ₁₁ NO ₃ S (273.3)	273	1720 (COOCH ₃)	(TFA/TMS _{int.}): 3.60, 3.70 (2s, 6H, OCH ₃ , COOCH ₃); 7.10 (br. s, 1H, 5-H); 7.50–7.65 (d, 2H, 7-H, 8-H, $J_o = 9$ Hz); 8.00 (s, 1H, 4-H); 8.93 (s, 1H, 3-H)
4c	C E	80 98	192	C ₁₅ H ₁₃ NO ₄ S (303.3)	303	1720 (COOCH ₃)	(CDCl ₃ /DMSO- <i>d</i> ₆ /TMS _{in}): 3.70, 3.86, 4.00 (3s, 9H, 2OCH ₃ , COOCH ₃); 7.20 (s, 1H, 8-H); 7.27 (s, 1H, 5-Ḥ); 8.01 (s, 1H, 4-H); 8.54 (s, 1H, 3-Ḥ)
4 d	C E	85 96	177178°	C ₁₆ H ₁₅ NO ₅ S (333.4)	333	1715 (COOCH ₃)	3.92, 3.97, 4.00 (3s, 9H, 3OCH ₃); 4.10 (s, 3H, COOCH ₃); 7.20 (s, 1H, 8-H); 8.00 (s, 1H, 4-H); 8.76 (s, 1H, 3-H)
4 e	C E	78 96	199	C ₁₇ H ₁₇ NO ₄ S (331.4)	331	1720 (COOCH ₃)	1.42–1.60 (t, 6H, 2O—CH ₂ —СḤ ₃); 3.90 (s, 3 H, COOCḤ ₃); 4.12–4.34 (q, 4H, 2O—CḤ ₂ —CH ₃); 7.05 (s, 1 H, 8-Ḥ); 7.30 (s, 1 H, 5-Ḥ); 7.91 (s, 1 H, 4-Ḥ); 8.30 (s, 1 H, 3-Ḥ)
4f	C E	79 98.5	195–196°	$C_{15}H_{12}N_2O_6S$ (348.3)	348	1720 (COOCH ₃); 1520, 1340 (NO ₂)	b
4g	C E	76 98	154155°	$C_{17}H_{16}N_2O_6S$ (376.4)	376	1715 (COOCH ₃); 1520, 1340 (NO ₂)	1.45-1.63 (m, 6H, 2O—CH ₂ —CH ₃); 3.90 (s, 3 H, COOCH ₃); 4.05-4.43 (m, 4H, 2O—CH ₂ —CH ₃); 7.46 (s, 1 H, 5-H); 7.95 (s, 1 H, 4-H); 8.34 (s, 1 H, 3-H)
4h	H G	85 92	159	$C_{14}H_{11}NO_2S$ (257.3)	257	1720 (COOCH ₃)	2.72 (s, 3H, CH ₃); 3.88 (s, 3H, COOCH ₃); 7.32-8.19 (m. 4H ₃₅₀₀); 8.39 (s, 1H, 4-H)
4i	H G	90 88	241-242°	$C_{16}H_{15}NO_4S$ (317.4)	317	1725 (COOCH ₃)	2.74 (s, 3H, CH ₃); 3.96, 3.98, 4.00 (3s, 9H, 3OCH ₃); 7.02 (s, 1H, 8-H); 8.20 (s, 1H, 4-H)
4j	C E	87 95	201-202°	$C_{19}H_{13}NO_2S$ (319.4)	319	1730 (COOCH ₃)	3.78 (s, 3H, COOCH ₃); 7.30–8.15 (m, 9H _{arom}); 8.23 (s, 1H, 4-H)
4k	C E	84 93	204°	C ₂₁ H ₁₇ NO ₄ S (379.4)	379	1730 (COOCH ₃)	3.76 (s, 3H, COOCH ₃); 3.90 4.00 (2s, 6H, 2OCH ₃); 7.00 (s, 1H, 8-H); 7.38-7.48 (m, 6H, 5-H) + 4H _{arom})

 $[^]a$ The microanalyses showed the following maximum deviations from the calculated values: C $\pm\,0.37;$ H $\pm\,0.33,$ N $\pm\,0.38.$

^b The ¹H-N.M.R. spectrum could not be recorded because the poor solubility of compound 4f in the usual N.M.R. solvents.

Table 3. 3-Hydroxy-2-methoxycarbonyl-2,3-dihydrothieno[2,3-b]quinolines (5) prepared

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Prod- uct	Me- thod	Yield [%]	m.p. [°C]	Molecular Formula*	M.S. (M ⁺) m/e	I.R. (KBr) v [cm ⁻¹ H]	1 H-N.M.R. (solvent/TMS $_{ m int}$) δ [ppm]
5a	D	75	147°	C ₁₃ H ₁₁ NO ₃ S (261.3)	261	3300 (OH); 1735 (COOCH ₃)	(CDCl ₃ /DMSO- d_6): 3.60 (br.s, 1H, CH—OH); 3.76 (s, 3H, COOCH ₃); 4.38–4.48 (d, 1H, CH—OH, $J = 9$ Hz); 5.14–5.24 (d, 1H, CH—COOCH ₃ , $J = 9$ Hz); 7.3–8.0 (m, 5H _{arom})
5b	D	75	189°	C ₁₄ H ₁₃ NO ₄ S (291.3)	291	3350 (OH); 1740 (COOCH ₃)	(CDCl ₃ /DMSO-d ₆): 3.60 (br. s, 1 H, CH—OH); 3.70 (s, 3 H, COOCH ₃); 3.78 (s, 3 H, OCH ₃); 4.32— 4.42 (d, 1 H, CH—OH, $J = 9$ Hz); 5.26—5.36 (d, 1 H, CH—COOCH ₃ , $J = 9$ Hz); 7.25 (br. s, 1 H, 5-H); 7.46 (br. s, 1 H, 7-H); 7.56 (br. s, 1 H, 8-H); 8.04 (s, 1 H, 4-H)
5c	D	80	195°	$C_{15}H_{15}NO_5S$ (321.4)	321	3300 (OH); 1740 (COOCH ₃)	b
5d	D	73	165166°	$C_{16}H_{17}^{'}NO_{6}S$ (351.4)	351	3400 (OH); 1735 (COOCH ₃)	(CDCl ₃): 3.72 (br.s, 4H, CH—OḤ, COOCḤ ₃); 3.82, 3.90, 3.98 (3s, 9H, 3OCḤ ₃); 4.39—4.48 (d, 1H, CḤ—OH, <i>J</i> = 8.5 Hz); 5.69–5.78 (d, 1H, CḤ—COOCH ₃ , <i>J</i> = 8.5 Hz); 7.06 (s, 1H, 8-Ḥ); 8.15 (s, 1H, 4-Ḥ)
5e	D	65	120-121°	C ₁₇ H ₁₉ NO ₅ S (349.4)	349	3350 (OH); 1735 (COOCH ₃)	(CDCl ₃): 1.44–1.62 (t, 6H, 2O—CH ₂ —CH ₃); 3.67 (br.s, 1 H, CH—OH); 3.75 (s, 3 H, COOCH ₃); 4.25–4.46 (m, 5 H, 2O—CH ₂ —CH ₃ , CH—OH); 5.51–5.61 (d, 1 H, CH—COOCH ₃ . $J = 9$ Hz); 7.18 (s, 1 H, 8-H); 7.35 (s, 1 H, 5-H); 8.00 (s, 1 H, 4-H)
5f	D	74	207–208°	$C_{15}H_{14}N_2O_7S$ (366.35)	366	3400 (OH); 1740 (COOCH ₃); 1520, 1335 (NO ₂)	(3, 111, 0.11), 7.33 (3, 111, 3-11), 6.00 (3, 111, 4-11)
5g	D	76	178179°	C ₁₇ H ₁₈ N ₂ O ₇ S (394.4)	394	3380 (OH); 1740 (COOCH ₃); 1510, 1340 (NO ₂)	(CDCl ₃ /DMSO- <i>d</i> ₆): 1.2–1.6 (m, 6H, 2O—CH ₂ —CH ₃); 3.65 (br.s, 1H, CH—OH); 3.73 (s, 3H, COOCH ₃); 4.00–4.43 (m, 5H, 2O—CH ₂ —CH ₃ , CH—OH); 5.54–5.68 (d, 1H, CH—COOCH ₃ , <i>J</i> = 8.5 Hz); 7.40 (s, 1H, 5-H); 7.95 (s, 1H, 4-H)
5j	D	80	194°	C ₁₉ H ₁₅ NO ₃ S (337.4)	337	3250 (OH); 1745 (COOCH ₃)	(CDCl ₃): 3.69 (s, 3H, COOCH ₃); 3.83 (s, 1H, OH); 4.60 (s, 1H, СН—СООСН ₃); 7.25–7.93 (m, 10H ₄₀₀₀)
5k	D	81	181	C ₂₁ H ₁₉ NO ₅ S (397.45)	397	3300 (OH); 1740 (COOCH ₃)	(CDCl ₃ /DMSO- <i>d</i> ₆): 3.63 (s, 3 H, COOCH ₃); 3.82 (br. s, 4 H, OCH ₃ , OH); 3.90 (s, 3 H, OCH ₃); 4.70 (s, 1 H, CH—COOCH ₃); 6.90 (s, 1 H, 8-H); 7.0—7.6 (m, 7 H, 5-H, 4-H, 5 H _{arom})

^a The microanalyses showed the following maximum deviations from the calculated values: $C \pm 0.29$; $H \pm 0.32$; $N \pm 0.38$.

olines 3j and 3k undergo intermolecular cyclocondensation with methyl mercaptoacetate to give compounds 4j, k and 5j, k whereas under the same conditions quinolines 3h and 3i undergo a substitution reaction to give the 2-methoxycarbonylmethylthio-3-acylquinolines 6h and 6i, respectively. Addition of piperidine to the methanolic solution of compounds 6h, i leads to intramolecular cyclocondensation to give the 2-methoxycarbonylthieno[2,3-b]quinolines 4h, i. Compounds 4h, i can also be obtained from quinolines 3h, i in a one-pot reaction with methyl mercaptoacetate in the presence of piperidine.

Melting points were measured in a sulfuric acid bath and are uncorrected. Mass spectra were recorded on a JEOL-JMS-D 300 mass spectrometer. ¹H-N.M.R. spectra were recorded on a Perkin-Elmer R-32 instrument except for compounds **2b**, **2g**, and **3g** the spectra of which were recorded on a Varian EM-360 L spectrometer. Typical working procedures are given in the following.

2-Chloro-3-(1-hydroxyethyl)-6,7-dimethoxyquinoline (2i):

Method A: A solution of methylmagnesium iodide is prepared from magnesium turnings (0.9 g. 0.04 mol) and methyl iodide

(5.6 g, 0.04 mol) in ether (50 ml). To this solution is added, dropwise and with stirring, a solution of 2-chloro-3-formyl-6,7-dimethoxy-quinoline (1c; 5.0 g, 0.02 mol) in tetrahydrofuran (50 ml) and the mixture is stirred at 30 °C for 10 min. Aqueous 10 % ammonium chloride (100 ml) is then added and the organic material is extracted with ethyl acetate (2 \times 100 ml). The extract is dried with sodium sulphate and evaporated. Addition of petroleum ether to the residue leads to formation of a solid. This product is isolated by suction and recrystallized from chloroform/petroleum ether (30/70).

3-Acetyl-2-chloro-6,7-dimethoxyquinoline (3i):

Method B: To a well stirred solution of compound 2i (5.3 g, 0.02 mol) in dry dichloromethane (50 ml) is added pyridinium chlorochromate (6.6 g, 0.03 mol) and the mixture is stirred at room temperature (31°C) for 45 min. The suspension is extracted with ether (3 × 100 ml), the extract dried with sodium sulphate and evaporated. The residue is passed through a short band of Florisil® using chloroform/ethyl acetate (9/1) as eluent. Removal of the solvent gives a solid which is recrystallized from methanol.

2-Methoxycarbonylthieno[2,3-b]quinoline (4a):

Method C: A mixture of 2-chloro-3-formylquinoline (1a; 15 g, 0.07 mol), methyl mercaptoacetate (8 g, 0.07 mol), anhydrous potassium carbonate (11.5 g, 0.08 mol), and dry tetrahydrofuran (75 ml)

The ¹H-N.M.R. spectrum could not be recorded because of the poor solubility of the compound in the usual N.M.R. solvents.

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is stirred at room temperature (30–35 °C) for 3 h. Addition of water (250 ml) then leads to the formation of a solid which is isolated by suction and washed with cold water (250 ml). The solid product is purified by column chromatography on silica gel using chloroform as eluent. The eluate is evaporated and the residual product 4a recrystallized from petroleum ether/chloroform (80/20).

3-Hydroxy-2-methoxycarbonyl-2,3-dihydrothieno[2,3-b]quinoline (5a):

Method D: A mixture of 2-chloro-3-formylquinoline (1a; 16 g, 0.08 mol), methyl mercaptoacetate (9.6 g, 0.09 mol), potassium carbonate (12.4 g, 0.09 mol), dimethylformamide (72 ml), and water (8 ml) is stirred at room temperature (30–35 °C) for 4–5 h. Dilution of the mixture with water (300 ml) then leads to precipitation of a solid which is isolated by suction, dried, and purified by column chromatography on silica gel using chloroform and chloroform/methanol (70/30) as eluents. Elution with chloroform gives product 4a; product 5a is obtained by elution with chloroform/methanol. Compound 4a is recrystallized from petroleum ether/chloroform (80/20) and compound 5a from methanol. The similarly prepared compounds 5b, f, g are recrystallized from dimethylformamide and compounds 5c, d, e, j, k from methanol.

2-Methoxycarbonylthieno[2,3-b]quinoline (4a):

Method E: A mixture of compound 5a (12 g, 0.04 mol), acetic anhydride (10 ml), and pyridine (0.5 ml) is stirred at $50\,^{\circ}$ C for 30 min. Addition of water (100 ml) then leads to precipitation of a solid which is isolated by suction, washed with water (500 ml), dried, and cyrstallized from petroleum ether/chloroform (80/20) to afford 4a in quantitative yield.

3-Acetyl-2-(methoxycarbonylmethylthio)-quinoline (6h):

Method F: A mixture of 3-acetyl-2-chloroquinoline (3h; 2.05 g, 0.01 mol), methyl mercaptoacetate (1.06 g, 0.01 mol), potassium carbonate (1.38 g, 0.01 mol), and dry dimethylformamide (20 ml) is stirred at room temperature (30 °C) for 2.5 h. Water (75 ml) is then added, the resultant solid product isolated by suction, and recrystallized from chloroform/methanol (20/80); yield of 6h: 2.5 g (91 %); m.p. 140–141 °C.

C₁₄H₁₃NO₃S calc. C 61.09 H 4.72 N 5.09 (275.3) found 61.05 5.06 5.00

M. S.: $m/e = 275 \text{ (M}^+)$.

I. R. (KBr): v = 1740 (COOCH₃); 1680 cm^{-1} (CO—CH₃).

¹H-N.M.R. (CDCl₃/TMS_{int}): $\delta = 2.62$ (s, 3 H, CO—CH₃); 3.65 (s, 3 H, COOCH₃); 3.88 (s, 2 H, S—CH₂—COOCH₃); 7.30-7.77 (m, 4 H_{arom}); 8.43 ppm (s, 1 H, 4—H).

3-Acetyl-6,7-dimethoxy-2-(methoxycarbonylmethylthio)-quinoline (6i):

Method F is used as above; yield of 6i: 95%; m.p. 170°C.

C₁₆H₁₇NO₅S calc. C 57.31 H 5.07 N 4.17 (335.2) found 57.27 4.96 4.41

 $M.S.: m/e = 335 (M^+).$

I. R. (KBr): v = 1745 (COOCH₃); 1675 cm⁻¹ (CO—CH₃).

¹H-N.M.R. (CDCl₃/TMS_{in1}): δ = 2.61 (s, 3 H, CO—CH₃); 3.70 (s, 3 H, S—CH₂—COOCH₃); 3.92 (s, 2 H, S—CH₂—COOCH₃); 3.97, 4.00 (2s, 6 H, 2OCH₃); 7.00 (s, 1 H, 8-H); 7.10 (s, 1 H, 5-H); 8.30 ppm (s, 1 H, 4-H).

2-Methoxycarbonyl-3-methylthieno[2,3-b]quinoline (4h):

Method G: A solution of compound **6h** (2.75 g, 0.01 mol) and piperidine (0.85 g, 0.01 mol) in methanol (25 ml) is stirred at room temperature (30–33 °C) for 3.5 h. Removal of methanol under reduced pressure followed by addition of water (75 ml) to the residue affords product **4h** which is recrystallized from petroleum ether/chloroform (80/20); yield: 2.36 g (92 %).

2-Methoxycarbonyl-3-methylthieno[2,3-b]quinoline (4h):

Method H: A mixture of 2-chloro-3-acetylquinoline (3h; 4.1 g, 0.02 mol), piperidine (1.6 g, 0.02 mol), methyl mercaptoacetate (2.1 g, 0.02 mol), and dry dimethylformamide (30 ml) is stirred at room temperature (30-32 °C) for 2.5 h. Water (75 ml) is then added

and the resultant solid product isolated by suction, dried, and recrystallized from petroleum ether/chloroform (80/20); yield of 4h: 2.18 g (85%).

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