This article was downloaded by: [The University of Manchester Library] On: 23 December 2014, At: 06:50 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

Trimethylsilyl Trifluoromethanesul Fonate (Tmsotf) Catalyzed Amidoalkylation of Silylenolethers. Stereocontrolled Syntheses of (+/-)-Sedamine and (+/-)-Norsedamine

R. A. Pilli^a & L. C. Dias^a

^a Universidade Estadual de Campinas, Instituto de Química , C.P. 615, 13081, Campinas, SP, Brazil Published online: 18 Nov 2011.

To cite this article: R. A. Pilli & L. C. Dias (1991) Trimethylsilyl Trifluoromethanesul Fonate (Tmsotf) Catalyzed Amidoalkylation of Silylenolethers. Stereocontrolled Syntheses of (+/-)-Sedamine and (+/-)-Norsedamine, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 21:21, 2213-2229, DOI: <u>10.1080/00397919108055456</u>

To link to this article: http://dx.doi.org/10.1080/00397919108055456

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

TRIMETHYLSILYL TRIFLUOROMETHANESULFONATE(TMSOTf) CATALYZED AMIDOALKYLATION OF SILYLENOLETHERS. STEREOCONTROLLED SYNTHESES OF (+/-)-SEDAMINE AND (+/-)-NORSEDAMINE.

R.A. Pilli * and L.C. Dias Universidade Estadual de Campinas, Instituto de Química, C.P. 615 13081 Campinas, SP, Brazil

Trimethylsilyl trifluoromethanesulfonate(TMSOTf) catalyzed addition of 1-trimethylsilyloxy-1-phenylethene to N-ethoxycarbonyland N-tert-butoxycarbonyl-2-ethoxypiperidines (3a and 3b, respectively) afforded the corresponding ketocarbamates 4a and 4b, in excellent yields. Stereoselective conversion to (±)-norsedamine (7) and (±)-sedamine (8) is described.

The attachment of an α -amidoalkyl group to nucleophiles is a well known transformation and a value le alternative to the Mannich reaction. Both its intermolecular and intramolecular version have been comprehensively reviewed and many examples of its application to the synthesis of biologically important molecules are known^{1,2}. Among the nucleophiles amenable to α -amidoalkylation, silylenol-

2213

Copyright © 1991 by Marcel Dekker, Inc.

ethers and silylketeneacetals have been explored as regioselective partners. Since the original report of ureidoalkylation of silylenolethers by Danishefsky et. al.³ many novel routes to the preparation of N-acyl iminium ions have been described². Heterolyses of α -heterosubstituted amides and carbamates is the most often employed method for **in situ** generation of N-acyl iminium ions where the heterosubstituent may be an oxygen, halogen, nitrogen, sulfur or phosphorous species².

Particularly relevant to this subject is the partial reduction of imides to cyclic hydroxy or alkoxylactams. Sodium borohydride in the presence of hydrochloric acid is the method of choice⁴ although other reducing agents⁵⁻⁷ and methodologies are also available⁸. Anhydrous titanium(IV) chloride is generally employed in stoichiometric amounts to generate the acyl iminium ion while better yields of amidoalkylation products are usually observed with acyl iminium ions derived from carbamates rather than from amides^{9,10}.

In 1984, Sekiya and coworkers¹¹ reported the **in situ** generation of N,N-bissilyliminium salts from N,N-bis(trimethylsilyl) methoxymethylamines and catalytic amounts of trimethylsilyl trifluoromethanesulfonate (TMSOTf), which reacted with silylketeneacetals to afford N,N-bis(trimethylsilyl)- β -aminoesters in high yields. We have described the TMSOTf promoted addition of silylenolethers to Schiff bases¹² and the diastereoselective reduction of the β -aminoketones¹³.

Here we disclose our results on the addition of 1-trimethylsilyloxy-1-phenylethene to N-ethoxycarbonyl-2-ethoxypiperidine (3a) and N-tert-butoxycarbonyl-2-ethoxypiperidine (3b) promoted by catalytic amounts of TMSOTf (Scheme 1) and its application to the total synthesis of (+/-)-norsedamine (7), (+/-)-sedamine (8) and (+/-)-allosedamine (9).

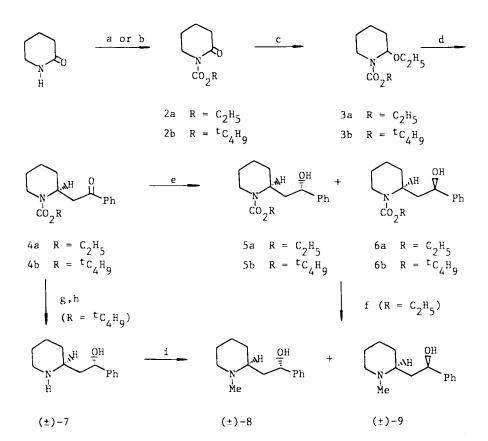
After extensive experimental optimization, the best yields of ketocarbamates 4a,b were achieved when 5-10 mol % of TMSOTf was added to a mixture of ethoxycarbamate 3a or 3b and 1-trimethyl-silyloxy-1-phenylethene, in CH₂Cl₂ at -78^oC. A smooth reaction took place and after quenching with aqueous NaHCO₃, the residue was purified by column chromatography to afford ketocarbamates 4a and 4b in 94% and 97% yield, respectively.

Although the preparation of α -phenacylpiperidine and its derivatives has long been known^{9,14}, the high yields and mild conditions employed in the TMSOTf catalyzed amidoalkylation seems to be appropriate for acid sensitive nucleophiles. As an initial application of this methodology, we prepared the racemic forms of norsedamine (7), sedamine (8) and allosedamine (9), alkaloids found in Sedum acre¹⁵.

Attempts of stereoselective reduction of ketocarbamates 4a and 4b with several reducing agents (e.g., LiAlH_4 , $\text{Zn}(\text{BH}_4)_2$, $\text{LiBH}(\text{C}_2\text{H}_5)_3$ and $\text{KBH}(\text{C}_2\text{H}_5)_3$) led only to modest diastereoselection (3:1 ratio of **5a,b/6a,b** with $\text{LiBH}(\text{C}_2\text{H}_5)_3$ in THF, at -78° C). The relative configuration of **5a** and **6a** was assigned after LiAlH_4 reduction to (+/-)-sedamine (8) and (+/-)-allosedamine (9), respectively.

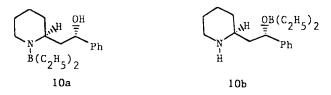
From our previous results in the diastereoselective reduction of β -aminoketones we reasoned that higher levels of stereocontrol





a) i. LDA,THF, -78°C; ii. NCCO₂Et, -78°C, 2h (67%) or i. LDA,THF, -78°C; ii. C1CO₂Et,-78°C (62%); b) ^tBuO-COC-O-^tBu,Et₃N,DMAP,CH₂Cl₂, 25°C, 7h(83%) or i.LDA,THF,-78°C;ii.BOC-ON,THF,-78°C(80%);c)i.NaBH₄, EtOH,-23°C,3h;ii.2N HCl,EtOH,pH 3.0,lh,iii.KOH,EtOH,pH 7.0 (90--95%);d)PhCH(OSiMe₃)CH₂,TMSOTf(5mo1%),CH₂Cl₂,-78°C(95-97%); e)LiBH(C₂H₅)₃,THF,-78°C(100%);f)LiAlH₄,THF,20h(100%);g)i.CF₃CO₂H 25°C,30 min,ii.aq. NaHCO₃(98%);h)Zn(BH₄)₂,THF,-78°C(95%) or i. LiBH(C₂H₅)₃,THF,-78°C, ii.H₂O₂,MeOH,pH 7.0(70%);i)HCHO,CH₃CN,' NaBH₃CN,25°C(88%). could be achieved if the reduction was carried out with α -phenacylpiperidine itself. Trifluoroacetic acid treatment of ketocarbamate 4b afforded the corresponding secondary β -aminoketone which was immediately reduced with $Zn(BH_4)_2$ to a 6:1 mixture of aminoalcohols from which (+/-)-norsedamine (7) (double doublet, δ 4.92 ppm, J = 10.8 and 2.7 Hz) cristallized as the major component (CH₂Cl₂-hexanes), in 75% yield.

Surprisingly, the reduction of α -phenacylpiperidine with LiBH(C₂H₅)₂ (2.0 equiv.) led to a 6:1 mixture of diastereoisomeric products to which structures 10a or 10b was tentatively assigned to the major isomer.



The incorporation of the $-B(C_2H_5)_2$ residue was evident from the ¹H-NMR (two triplets at & 0.79 and 0.88 ppm), ¹³C-NMR (two methyls at & 8.31 and 8.83 ppm) and mass spectra (m/z 273). Attempted acetylation of 10 did not allow us to distinguish between structures 10a and 10b, since elimination of $-B(C_2H_5)_3$ ocurred under the conditions employed (Ac₂O/Et₃N/4-DMAP in CH₂Cl₂ at room temperature). Nevertheless, the relative configuration of the major isomer could be unambiguously established after H₂O₂ oxidation which afforded (+/-)-norsedamine (7), in 70% yield. These results were rationalized through chelation controlled hydride addition to α -phenacylpiperidine. (+/-)-Sedamine (8) was uneventfully prepared 17 , in 88% yield, by reductive methylation of (+/-)-norsedamine (7) using the formaldehyde/NaBH₂CN protocol 18 .

Experimental

Unless otherwise noted materials were obtained from commercial suppliers and were used without further purification. Tetrahydrofuran (THF) and ether were distilled from sodium-benzophenone immediately prior to use. Diisopropylamine, triethylamine and dichloromethane were distilled from calcium hydride. All reactions involving organometallic reagents or trimethylsilyl trifluoromethanesulfonate (TMSOTf) were carried out under an argon atmosphere. Melting points were determined on a Kofler apparatus and are uncorrected. ¹H-NMR spectra were determined in CDC1, solution at 300 MHz and 13 C-NMR spectra in CDCl₃ solution at 75.5 MHz unless otherwise noted. Chemical shifts are expressed in ppm downfield from internal tetramethylsilane and ¹H-NMR data are tabulated in order: multiplicity (s, singlet; d, doublet; t, triplet; br, broad; dt, double triplet; q, quartet; dq, double quartet; m, multiplet), number of protons, coupling constants in hertz. Infrared spectra were recorded on a Perkin Elmer 399B spectrophotometer and mass spectra with a Varian MAT 311A spectrometer. Elemental analyses were performed on a Perkin Elmer 2400 CHN Analyser, at Instituto de Química, Unicamp.

N-Ethoxycarbonyl-2-ethoxypiperidine (3a) and N-tert-butoxycarbonyl-2-ethoxypiperidine (3b). To a solution of $2a^{19}$ or $2b^{19}$ (1.0 mmol) in ethanol (10 ml) was added NaBH₄ (0.152 g; 4.0 mmol) in one portion at -23° C. This solution was kept at -23° C for 3 hours and quenched with aqueous HCl (1%) until pH 3.0. The mixture was stirred for an additional 30-45 min. period at -23° C, neutralized with ethanolic KOH (1%), poured into water and extracted with CH₂Cl₂.

The combined organic extracts were dried over anhydrous MgSO₄ and the solvent was removed under reduced pressure. The crude material was purified in silica gel (9:1 hexane-ethyl acetate with 1% of triethylamine) yielding 0.189 g (94% yield) of **3a** or 0.217 g (95% yield) of **3b** as colorless oils.

3a: ¹H-NMR: δ 1.18 (t, 3H, J = 6.9), 1.27 (t, 3H, J = 6.9); 1.4-1.9 (m, 6H), 3.0 (m, br, 1H), 3.42 (m, 2H), 3.95 (m, br, 1H), 4.0-4.3 (m, 2H), 5.6 (m, 1H).

¹³C-NMR: δ 14.55, 14.65, 25.11, 30.47, 39.25, 60.64,
61.26, 66.49, 80.36, 155.36.

3b: ¹H-NMR: δ 1.19 (t, 3H, J = 7.0), 1.46 (s, 9H), 1.30-1.65 (m, 6H), 2.92 (m, br, 1H), 3.42 (m, 2H), 3.8 (m, 1H), 5.44 (m, 1H).

¹³C-RMN: δ 155.23, 80.41, 79.70, 61.75, 39.18, 37.92,
 28.34, 25.13, 18.49 and 15.00 ppm.

IR(film): 2868, 2969, 2934, 1701, 1414, 1261, 1166, 1091, 1076 cm⁻¹.

MS (70 eV): m/z 229 (M⁺, 0.5%), 184 (27%), 144 (17%), 128 (100%), 100 (16%), 84 (87%), 83 (31%), 59 (25%), 57 (98%), 45 (36%), 43 (55%), 41 (45%).

N-Ethoxycarbonyl-2-phenacylpiperidine (4a) and N-tert--butoxycarbonyl-2-phenacylpiperidine (4b).

To a solution of **3a** or **3b** (1.0 mmol) and 1-trimethylsilyloxy--1-phenylethene (0.222 g, 1.15 mmol) in CH_2Cl_2 (2.0 ml) at $-78^{\circ}C$ and under an atmosphere of nitrogen was added trimethylsilyl trifluoromethanesulfonate (0.010 ml, 0.011 g, 0.05 mmol).

The resulting mixture was stirred for 1 hour at $-78^{\circ}C$ and quenched by the addition of saturated aqueous NaHCO₃. The resulting mixture was let to warm up to room temperature and it was extracted with CH_2Cl_2 (3 x 10 ml) and the combined organic extracts were dried over MgSO₄. The organci solvent was removed under reduced pressure and the crude material was purified by silica gel column chromatography (10:1 hexane-ether) to afford 0.258 g (0.94 mmol, 94% yield) of 4a, as a colorless oil, and 0.294 g (0.97 mmol, 97% yield) of 4b as a white solid (m.p. $81-82^{\circ}C$).

4a: ¹H-NMR: 6 1.19 (t. 3H, J = 6.6), 1.43 (m, 1H), 1.63
(m, 5H), 2.88 (t, 1H, J = 13.0), 3.13 (m, 2H), 4.03 (m, 3H), 4.77
(m, 1H), 7.42 (m, 2H), 7.51 (m, 1H), 7.98 (d, 2H, J = 7.8).

¹³C-NMR: δ 14.57, 18.88, 25.33, 28.04, 39.36, 39.88, 48.35, 61.51, 128.74, 129.13, 133.68, 137.26, 156.16, 199.12. IR(film): 2980, 2940, 2860, 1690, 1420 and 1260 cm⁻¹.

Elemental analysis calcd. for C₁₆H₂₁NO₃: C-69.81, H-7.64, N-5.09. Found: C-69.44, H-7.66, N-5.04.

4b: 1 H-NMR: δ 1.37 (s, 9H), 1.64 (m, 6H), 2.89 (ddd, 1H, J = 13.2, 12.6 and 2.7), 3.16 (dd, 1H, J = 14.4 and 6.1), 3.22

(1H, dd, J = 14.4 and 8.4), 4.05 (d, br, 1H, J = 12.6), 4.84 (m, 1H), 7.48 (m, 2H), 7.58 (m, 1H), 8.0 (d, 2H, J = 7.2).

¹³C-NMR: δ 18.90, 25.34, 28.36, 28.55, 39.29, 39.49,
 48.34, 79.84, 128.73, 129.08, 133.57, 137.29, 155.24, 199.13.

IR(KBr): 2975, 2945, 2875, 1770, 1715, 1295, 1250, 1160 and 1145 $\rm cm^{-1}$.

MS (70 eV): 303 (M⁺, 9%), 247 (11%), 229 (9%), 191 (62%), 128 (32%), 105 (39%), 84 (100%), 77 (16%), 57 (87%).

(1SR,2SR)- and (1RS,2SR) N-Ethoxycarbonyl-2-(β-hydroxy-β--phenylethyl)-piperidine (5a and 6a).

To a stirred solution of **4a** (0.137 g, 0.5 mmol) in THF (3.0 ml), at -78° C and under nitrogen atmosphere, was added dropwise a 1.0 M solution of LiBH $(C_2H_5)_3$ in THF (0.7 ml, 0.7 mmol). After 2 hours at -78° C, the reaction was quenched by addition of water (2.0 ml) and allowed to warm up to room temperature.

After extraction with ether (3 x 2.0 ml), the combined organic phases were washed with brine (2 x 2.0 ml), dried over $MgSO_4$ and evaporated under reduced pressure. Column chromatography on silica gel (85:15 hexane-ether) afforded 0.102 g (0.37 mmol, 74% yield) of 5a and 0.037 g (0.12 mmol, 24% yield) of 6a.

5a: ¹H-NMR: δ 1.21 (t, 3H, J = 7.2), 1.38 (m, 2H), 1.59 (m, br, 5H), 1.89 (m, 1H), 2.16 (m, 1H), 2.85 (m, 1H), 3.89 (m, br, 1H), 4.09 (q, 2H, J = 7.2), 4.42 (m, br, 1H), 4.73 (m, br, 1H), 7.2-7.4 (m, 5H).

¹³C-NMR: § 14.58, 19.01, 25.39, 29.06, 39.50, 39.98, 48.51, 61.55, 72.48, 126.09, 127.65, 128.69, 145.04, 156.65. IR(film): 3424, 3100, 3077, 3063, 3031, 2986, 2942, 2880, 1650, 1430, 1265 and 1170 cm⁻¹.

MS (70 eV) m/z 277 (M⁺, 20%), 202 (5%), 156 (100%), 128 (17%), 105 (29%), 84 (26%), 45 (22%).

6a: ¹H-NMR: δ 1.29 (t, 3H, J = 7.2), 1.4-1.7 (m, 6H), 1.7-1.85 (m, 1H), 2.22 (dt, br, 1H, J = 13.2 and 2.4), 2.84 (dt, br, 1H, J = 13.2 and 2.4), 4.09 (d, br, 1H, J = 13.2), 4.18 (q, 2H, J = 7.2), 4.6-4.8 (m, 3H), 7.2-7.4 (m, 5H).

¹³C-NMR: δ 14.61, 19.14, 25.46, 29.33, 39.33, 40.32,
 47.49, 61.99, 70.01, 125.97, 127.48, 128.69, 144.48, 157.79.

(1SR, 2SR) - and (1RS, 2SR)-N-tert-Butoxycarbonyl-2-(β -Hydroxy-- β -phenylethyl)-piperidine (5b and 6b).

To a stirred solution of 4b (0.151 g, 0.5 mmol) in THF (3.0 ml), at -78° C and under nitrogen atmosphere, was added dropwise a 1.0 M solution of LiBH $(C_2H_5)_3$ in THF (0.7 ml, 0.7 mmol). After 2 hours at -78° C, the reaction was quenched by addition of water (2.0 ml) and allowed to warm up to room temperature.

After extraction with ether (3 x 2.0 ml), the combined organic extracts were washed with brine (2 x 2.0 ml), dried over $MgSO_4$ and evaporated under reduced pressure. Column chromatography on silica gel (hexane-ether 95:5) afforded 5b (0.113 g, 0.37 mmol) as a colorless oil and **6b** (0.037 g, 0.12 mmol) as a colorless solid (m.p. 74.4-75.7°C).

5b: ¹H-NMR: & 1.44 (s, 9H), 1.5-1.6 (m, br, 6H), 1.8-1.92 (m, 1H), 2.0-2.17 (m, 1H), 2.7-2.9 (m, 1H), 3.8-4.0 (m, 2H), 4.39 (m, br, 1H), 4.73 (m, br, 1H), 7.2-7.4 (m, 5H). ¹³C-NMR: δ 19.09, 25.44, 28.48, 29.18, 39.55, 40.37,
 48.56, 72.62, 79.77, 125.75, 127.25, 128.34, 144.74.

IR(film): 3400, 3085, 3064, 3029, 2980, 2940, 2925, 1690, 1665, 1420, 1364, 1274 and 1160 cm⁻¹.

MS (70 eV) m/z 305 (M^+ , 6%), 249 (6%), 205 (7%), 184 (10%), 128 (94%), 84 (86%), 57 (100%).

6b: ¹H-NMR: δ 1.49 (s, 9H), 1.5-1.8 (m, br, 7H), 2.21 (m, 1H),
2.80 (m, 1H), 4.05 (m, br, 1H), 4.44 (m, br, 1H), 4.58 (m, br, 1H),
4.75 (s, br, 1H), 7.2-7.5 (m, 5H).

¹³C-NMR: δ 19.20, 25.52, 28.44, 29.29, 39.54, 40.36,
 46.80, 69.11, 80.38, 125.57, 127.04, 128.29, 128,65, 144.11.

IR(KBr): 3435, 3070, 3020, 2970, 2930, 2870, 2850, 1652, 1645, 1415, 1360, 1280, 1160 cm⁻¹.

MS (70 eV) m/z 305 (M⁺, 12%), 249 (15%), 205 (16%), 184 (20%), 128 (100%), 104 (32%), 84 (98%), 77 (25%), 57 (97%), 43 (24%), 41 (54%).

(+/-)-Sedamine (8)

A solution of **5a** (0.138 g, 0.5 mmol) in THF (0.5 ml) was added dropwise to a stirred suspension of LiAlH_4 (0.030 g, 0.79 mmol) in ether (0.6 ml). The reaction mixture was refluxed for 20 hours and quenched by the addition of water (0.1 ml), 15% aqueous NaOH (0.1 ml) and water (0.3 ml), respectively. The resulting mixture was extracted with ether (3 x 1.0 ml), dried over MgSO₄ and the organic solvent was evaporated under reduced pressure. The crude material was recrystallized from CH_2Cl_2 -hexane to afford 0.106 g (0.48 mmol, 96% yield) of a colorless solid (m.p. 89.6- -90.1° C, lit. 14d: 90°C).

¹H-NMR (CDC1₃/D₂O) δ 1.2-1.8 (m, 7H), 2.12 (m, 1H), 2.50 (s, 3H), 2.4-2.7 (m, 1H), 2.82 (m, 1H), 3.04 (m, 1H), 4.90 (dd, 1H, J = 10.6 and 2.8), 7.2-7.4 (m, 5H).

¹³C-NMR: δ 20.54, 22.45, 25.84, 39.90, 40.09, 51.44, 61.14, 75.01, 125.99, 127.46, 128.72, 146.15.

IR(KBr): 3396, 3166, 3076, 3016, 2985, 2945, 2925, 2915, 2915, 2846, 2814, 2786, 1456, 1376, 1196, 1075 and 1058 cm⁻¹.

MS (70 eV) m/z 219 (M⁺, 11%), 112 (3%), 99 (13%), 98 (100%), 84 (2%), 79 (5%), 77 (5%), 70 (11%).

(+/-)-Allosedamine (9)

The same procedure described above afforded 0.017 g (0.08 mmol, 81% yield) of (+/-)-allosedamine (9) (m.p. $68.2-70.0^{\circ}$ C, lit. 14d: $68-69^{\circ}$ C) from 0.030 g (0.1 mmol) of **6a**.

¹H-NMR: & 1.25-1.31 (m, 2H), 1.6-1.8 (m, 4H), 1.8-2.0 (m, 2H), 2.17 (m, IH), 2.30 (m, 1H), 2.54 (s, 3H), 3.09 (m, 1H), 3.6-4.4 (s, br, 1H), 5.13 (dd, 1H, J = 10.5 and 3.3), 7.2-7.4 (m, 5H).

¹³C-NMR: δ 24.28, 25.42, 29.22, 39.43, 43.85, 56.90, 62.60, 71.88, 125.58, 126.85, 128.20, 145.48.

IR(KBr): 3263, 3084, 3062, 3025, 2990, 2941, 2859, 2800, 1601, 1491, 1463, 1378 and 1061 cm^{-1} .

MS (70 eV) m/z 219 (M^+ , 12%), 113 (5%), 99 (11%), 98 (100%), 70 (9%).

a-Phenacylpiperidine

A solution of 3b (0.210 g, 0.69 mmol) in trifluoroacetic acid (1.5 ml) was stirred for 30 min. at room temperature. The mixture was then diluted with CH_2Cl_2 (5.0 ml), neutralized with saturated aqueous NaHCO₃ and washed with water. The organic phase was dried over MgSO₄ and the organic solvent was removed under reduced pressure to give 0.199 g (0.67 mmol, 98% yield, of α -phenacylpiperidine as a pale yellow oil²¹.

¹H-NMR: $(CDC1_3/D_2O) \delta 1.2-1.5 \text{ (m, 4H)}, 1.5-1.63 \text{ (m, 2H)},$ 1.65-1.85 (m, 1H), 2.5-2.8 (m, 1H), 3.0-3.2 (m, 3H), 7.5-7.8 (m, 3H), 7.95 (d, 2H, J = 7.0).

¹³C-NMR (CDC1₃/D₂O) δ 24.76, 25.96, 32.66, 45.53, 46.87, 52.96, 128.14, 128.69, 133.28, 137.14, 199.55.

IR(film): 3326, 3056, 3026, 2926, 2848, 1686, 1448, 1206, 1176 and 1136 cm⁻¹.

MS (70 eV) m/z 203 (M⁺, 41%), 173 (23%), 105 (67%), 98 (29%), 91 (41%), 84 (100%), 77 (74%), 59 (96%), 43 (76%).

(+/-)-Norsedamine (7)

a) Zn(BH₄)₂ reduction

To a stirred solution of α -phenacylpiperidine (0.099 g, 0.49 mmol) in THF (2.0 ml), at -78°C and under a nitrogen atmosphere, was added dropwise a 0.16 M ethereal solution of $\text{Zn}(\text{BH}_4)_2$ (0.12 ml, 0.98 mmol). After 2 hours at -78°C the reaction was quenched with saturated aqueous NH₄Cl (2.0 ml). After extraction with ether (3 x 2.0 ml), the combined organic phases were washed with brine (2 x 2.0 ml), dried over MgSO₄ and evaporated under reduced pressure. Fractional crystallization of the residue (CH₂Cl₂-hexane) afforded 0.075 g (0.37 mmol, 75% yield) of (+/-)-nor-sedamine (7), m.p. 91-93°C (lit. 20: 92-93°C).

b) LiBH(C₂H₅)₃ reduction

To a stirred solution of α -phenacylpiperidine (0.099 g, 0.49 mmol) in THF (2.0 ml), at -78°C and under nitrogen atmosphere, was added dropwise a 1.0 M solution of LiBH $(C_2H_5)_3$ in THF (1.0 ml, 1.0 mmol). After 2 hours at -78°C, the reaction was quenched by addition of a saturated NH₄Cl solution (2.5 ml) and extracted with CH₂Cl₂ (3 x 3.0 ml). The combined organic extracts were dried over MgSO₄ and the solvent was removed under reduced pressure to afford 0.104 g (77% yield) of 10, as a colorless solid (m.p. 93.5-96.0°C, after recrystallization from CH₂Cl₂-hexane.

¹H-NMR: δ 0.21 (m, 2H), 0.57 (m, 1H), 0.69 (m, 1H), 0.79 (t, 3H, J = 7.8). 0.88 (t, 3H, J = 7.8), 1.25 (m, 2H), 1.45 (m, 2H), 1.84 (m, 4H), 2.37 (m, 2H), 3.12 (m, 2H), 4.73 (d, 1H, J = 11.1), 7.18 (m, 1H), 7.29 (t, 2H, J = 7.2), 7.41 (d, 2H, J = 6.6).

¹³C-NMR: δ 8.31, 8.83, 23.36, 26.10, 33.74, 44.07, 45.24,
55.82, 69.80, 125.75, 126.54, 128.10, 146.51.

IR(KBr): 3226, 3086, 3064, 3024, 2946, 2926, 2896, 2846, 2826, 1446, 1126 and 1111 ${\rm cm}^{-1}.$

MS (70 eV) m/z 273 (M^+ , 0.8%), 245 (16%), 244 (87%), 243 (23%), 104 (11%), 91 (7%), 84 (81%), 72 (22%), 59 (34%), 45 (100%).

0.104 g (0.38 mmol) were taken up in methanol (8.0 ml) and to this solution were added a solution containing 30% of H_2O_2 (2.5 ml) and phosphate buffer (5.0 ml). The mixture was stirred 30 min. at room temperature and the volatiles were removed under reduced pressure. The residue was extracted with CH_2Cl_2 (3 x 5.0 ml) and the combined organic phases were dried over MgSO₄ and the solvent was removed underreduced pressure to afford, after recrystallization from CH_2Cl_2 -hexane, 0.070 g (0.34 mmol, 70% overall yield) of (+/-)-norsedamine (7), m.p. 91-93°C (lit. 20: 92-93°C).

¹H-NMR: δ 1.10 (m, 1H), 1.31 (m, 2H), 1.46-1.71 (m, 5H), 1.82 (m, 1H), 2.63 (m, 1H), 2.88 (m, 1H), 3.06 (m, 1H), 4.92 (dd, 1H, J = 10.8 and 2.7), 7.22 (m, 1H), 7.34 (m, 4H).

¹³C-NMR: & 24.48, 27.32, 34.26, 45.19, 46.00, 58.25, 75.45, 125.54, 126.94, 128.15, 145.27.

IR(KBr): 3400, 3301, 3140, 3064, 3023, 1450, 1382, 1300, 1212, 1176, 1157, 742 cm⁻¹.

MS (70 eV m/z 205 (M^+ , 11%), 149 (6%), 98 (6%), 85 (7%), 84 (100%), 77 (4%), 56 (10%).

(+/-)-Sedamine (8)

To a stirred solution of 7 (0.205 g, 1.0 mmol) and 37% aqueous formaldehyde (0.4 ml, 5.0 mmol) in acetonitrile (3.0 mmol) was added sodium cyanoborohydride (0.156 g, 2.5 mmol). The reaction mixture was stirred for 45 min. at room temperature and then glacial acetic acid was added dropwise until pH 5.0. Stirring was continued for an additional 20 min. when 1% aqueous KOH was added until pH 7.0.

The resulting mixture was extracted with ether $(3 \times 5 \text{ ml})$, the combined organic extracts were washed with brine, dried over MgSO₄ and evaporated under reduced pressure to give an oil which was purified by column chromatography on silica gel (92:8 CHCl₃-MeOH) to give 0.193 g (0.88 mmol, 88% yield) of (+/-)-sedamine (8), m.p. 89.6-90.1°C (lit. 14d, 90°C).

Acknowledgements

The authors acknowledge financial support from International Foundation for Science (IFS, Sweden), Fundação de Amparo a Pesquisa do Estado de São Paulo (Brazil) and Conselho Nacional de Desenvolvimento Científico e Tecnológico (Brazil).

References and notes

- 1. a. Zaugg, H.E., Synthesis, 85(1984); b. ibid, 181(1984)
- 2. Speckamp, W.N. and Hiemstra, H., Tetrahedron 41,4367(1985)
- Danishefsky, S., Guingant, A. and Prisbylla, M., Tetrahedron Lett. 2033(1980)
- Hubert, J.C., Wijnberg, J.B.P.A., and Speckamp, W.N., Tetrahedron 31,1437(1975)
- Tagmann, E., Sury, E. and Hoffmann, K., Helv. Chim. Acta 37, 185(1954)
- 6. Winterfeldt, E., Synthesis, 617(1975)
- 7. Fisher, M.J., and Overman, L.E., J. Org. Chem. 55,1447(1990)
- 8. Shono, T., Tetrahedron 40,811(1984)
- 9. Shono, T., Matsumura, Y. and Tsubata, K., J. Am. Chem. Soc. 103,1172(1981)
- Ikeda, K., Terao, Y. and Sekiya, M. Chem. Pharm. Bull.
 29,1156(1981)
- 11. Okano, K., Morimoto, T. and Sekiya, M., J. Chem. Soc. Chem. Commun. 883(1984)
- Pilli, R.A. and Russowsky, D., J. Chem. Soc. Chem. Commun. 1053(1987)

- Pilli, R.A., Russowsky, D. and Dias, L.C., J. Chem. Soc.
 Perkin Trans. I 1213(1990)
- 14. a. Marion, L., Lavigne, R. and Lemay, L., Can. J. Chem.
 29,347(1951); b. Beyerman, H.C. and Enthoven, P.H., Rec. Trav.
 Chim. Pays Bas 75,82(1956); c. Schopf, C., Braun, F.,
 Burkhardt, K., Dummer, G. and Muller, J. Justus Liebigs Ann.
 der Chemie 626,123(1959); d. Beyerman, H.C., Eveleens, W. and
 Muller, Y.M.F., Rec. Trav. Chim. Pays-Bas 75,63(1956)
- 15. Kolesnikov, D.G. and Schwarzmann, A.G., Zh. Obshch. Khim. (USSR) 9,2156(1939); C.A. 34,4072(1940)
- 16. Hootele, C., Halin, F. and Thomas, S. Tetrahedron 41,5563 (1985)
- 17. (+/-)-Sedamine (8) and (+/-)-norsedamine (7) were characterized by comparison of their melting points and spectroscopic properties (¹H- and ¹³C-NMR) with those described in ref. 14d, 16 and 20
- 18. a. Wanner, K.T., and Kartner, A., Heterocycles 26,921(1987);
 b. Borch, R.F. and Hassid, A.I., J. Org. Chem. 37,1673(1972)
- 19. Ketocarbamates 2a (67% yield) and 2b (83% yield) were prepared according to the procedures described by Speckamp et. alli [(Tetrahedron Lett. 4799(1986)] and Grieco et. alli [(J. Org. Chem. 48,2424(1983)], respectively
- 20. Tirel, P., Vaultier, M. and Carrie, R., Tetrahedron Lett. 1947(1989)
- 21. Trigo, G.G., Galvez, E. and Sollhuber, M.M., Heterocycles 16,1625(1979)

(Received in USA 21 June, 1991)