

# A Novel Route to 4-Aminopyrazoles and Aminopyrazolo[4,3-*b*]pyridines

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3-Oxo-2-arylhydrazononitriles **6** are readily converted into 4-aminopyrazoles **1** *via* reaction with  $\alpha$ -haloketones, chloroacetonitrile and ethyl chloroacetate. The aminopyrazoles are readily converted into aminopyrazolo[4,3-*b*]pyridines upon treatment with malononitrile. Compounds are readily diazotized to yield unstable diazonium salts that readily cyclized into pyrazolo[4,3-*c*]pyridazines.

**Key words:** 2-Arylhyaazononitriles, Aminopyrazoles,  $\alpha$ -Haloketones, Aminopyrazolo[4,3-*b*]pyridines, Pyrazolo[4,3-*c*]pyridazine

## Introduction

The synthesis of 4-aminopyrazole-5-carboxylates **1** is now receiving considerable interest [1–4] as these compounds are important precursors to important pharmaceuticals among which Viagra (**2**) and allopurinol (**3**) are two well-known compounds [5–8]. Moreover, 4-aminopyrazoles can also be considered as interesting precursors to dyes. 4-Derivatives of **1** are generally prepared *via* nitration of pyrazole **4** and reduction of formed nitro derivatives **5** [9]. This approach is both multistageous and environmentally non-friendly.

## Results and Discussion

In conjunction to our interest in synthesis and reactivity of aminopyrazoles we report a simple and efficient route to **1**. Moreover, results of our investigation on the reactivity of **1** is reported. The work enabled developing a novel efficient route to pyrazolo[4,3-*b*]pyridines (Scheme 1).

We have found that **6** reacts readily with chloroacetonitrile in dioxane in the presence of triethylamine (10 equivalents) to yield products of a molecular formula corresponding to **7** or isomeric **1a** with excellent yield. Structure **1a** was readily established based on  $^1\text{H}$  NMR data that revealed the absence of a signal for a methylene function and the appearance of a two-protons  $\text{D}_2\text{O}$ -exchangeable signal at  $\delta = 5.24$  ppm for the amino function. It is assumed that **1a** has resulted from *in situ* cyclization of initially formed **7**. Recently, a similar result has been reported with poor to moderate yield [10].

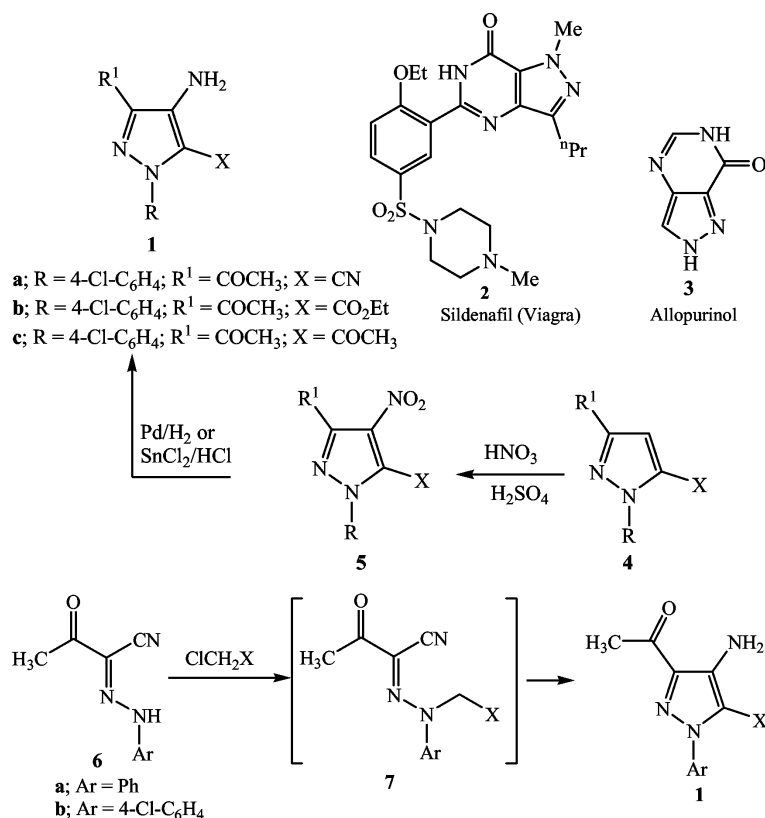
Similar to its behavior toward chloroacetonitrile, compound **6** also reacted with ethyl chloroacetate and with chloroacetone to yield **1b, c**, respectively, in good yields.

We have investigated the possible utility of compound **1** as precursors to pyrazoloazines. It was found that compound **1a** reacted with malononitrile in refluxing ethanol and in the presence of piperidine to yield the pyrazolo[4,3-*b*]pyridine derivative **9** (Scheme 2). Initial formation of **8** seems most likely; however, attempted isolation of **8** failed. Clearly, the aromaticity of the cyclization product is a driving force for the immediate formation of **9**.

Diazotization of **1** afforded unstable diazonium salts **10** that readily cyclized into pyrazolo[4,3-*c*]pyridazine **11**. Acetylation of compound **1** in acetic anhydride afforded **13** in excellent yield. The IR spectrum of compound **15** shows absorptions of two carbonyl group at 1710 and 1673, a cyano group at 2230 and a NH group at  $3337\text{ cm}^{-1}$ . The  $^1\text{H}$  NMR spectrum reveals two  $\text{CH}_3$  groups at  $\delta = 2.10$  and  $2.58$  ppm which indicates that no further cyclization has occurred.

## Experimental Section

All melting points were measured on a Gallenkamp electrothermal melting point apparatus and are uncorrected. IR spectra were recorded as KBr pellets on a Pye Unicam SP 3-300 Spectrophotometer.  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded in deuterated dimethylsulfoxide [ $\text{D}_6$ ]-DMSO or deuterated chloroform ( $\text{CDCl}_3$ ) at 300 MHz on a Varian Gemini NMR spectrometer using tetramethylsilane (TMS)



Scheme 1.

as an internal reference and results are expressed as  $\delta$  values. Mass spectra were performed on a Shimadzu GCMS-QP 1000 Ex mass spectrometer at 70 eV. Elemental analyses were carried out at the Microanalytical Center of Cairo University. The azo derivatives **6a, b** were prepared as previously reported in the literature [11].

#### General procedure for the preparation of compounds **1a–c**

To a solution of **6b** (2.21 g, 0.01 mol) in dioxane (25 ml) and triethylamine (10.1 g, 0.1 mol) chloroacetonitrile, ethyl chloroacetate or chloroacetone (1 ml, 0.016 mol) was added. The reaction mixture was refluxed for 3 h, then evaporated *in vacuo*, and the solid product was filtered off and crystallized from ethanol.

#### 3-Acetyl-4-amino-1-(4-chlorophenyl)-1H-pyrazole-5-carbonitrile (**1a**)

M. p. 156 °C. Yield: 85%. – IR:  $\nu$  = 3451, 3348 (NH<sub>2</sub>); 2217 (CN), 1671 (CO) cm<sup>-1</sup>. – <sup>1</sup>H NMR (300 MHz, [D<sub>6</sub>]-DMSO):  $\delta$  = 2.61 (s, 3H, CH<sub>3</sub>), 5.24 (s, 2H, NH<sub>2</sub>), 7.50 (d, 2H, *J* = 12 Hz, 2H Ar), 7.69 (d, 2H, *J* = 12 Hz, 2H Ar). – <sup>13</sup>C NMR (75 MHz, [D<sub>6</sub>]-DMSO):  $\delta$  = 197.91 (CO), 139.03 (C-3), 137.62, 135.94 (C-4), 134.62, 128.81, 122.23,

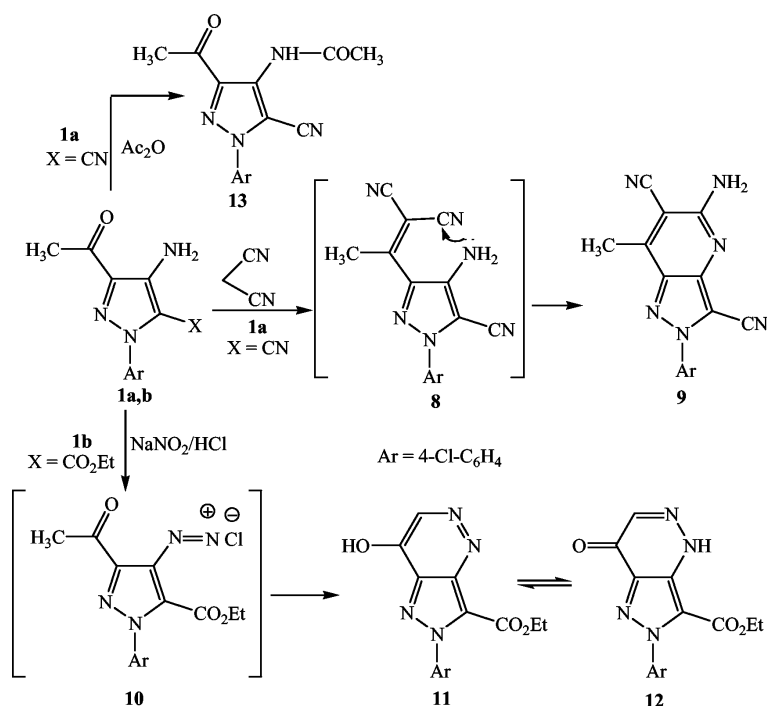
119.75 (C-5), 115.91 (CN), 24.80 (CH<sub>3</sub>). – MS (EI, 70 eV): *m/z* (%) = 260 (20) [M<sup>+</sup>]. – C<sub>12</sub>H<sub>9</sub>ON<sub>4</sub>Cl (260.6): calcd. C 55.29, H 3.48, N 21.49, Cl 13.60; found C 55.10, H 3.60, N 21.30, Cl 13.50.

#### Ethyl 3-acetyl-4-amino-1-(4-chlorophenyl)-1H-pyrazole-5-carboxylate (**1b**)

M. p. 143 °C. Yield: 75%. – IR:  $\nu$  = 3492, 3380 (NH<sub>2</sub>), 1711, 1672 (CO) cm<sup>-1</sup>. – <sup>1</sup>H NMR (300 MHz, [D<sub>6</sub>]-DMSO):  $\delta$  = 1.23 (t, 3H, *J* = 7.2 Hz, CH<sub>3</sub>), 2.60 (s, 3H, CH<sub>3</sub>), 4.21 (q, 2H, *J* = 7.2 Hz, CH<sub>2</sub>), 5.78 (s, 2H, NH<sub>2</sub>), 7.29 (d, 2H, *J* = 12 Hz, 2H Ar), 7.35 (d, 2H, *J* = 12 Hz, 2H Ar). – MS (EI, 70 eV): *m/z* (%) = 307 (27) [M<sup>+</sup>]. – C<sub>14</sub>H<sub>14</sub>O<sub>3</sub>N<sub>3</sub>Cl (307.7): calcd. C 54.64, H 4.59, N 13.65, Cl 11.52; found C 54.60, H 4.40, N 13.70, Cl 11.60.

#### 1-[5-Acetyl-4-amino-2-(4-chlorophenyl)-2H-pyrazol-3-yl]-ethanone (**1c**)

M. p. 138 °C. Yield: 72%. – IR:  $\nu$  = 3445, 3350 (NH<sub>2</sub>), 1678, 1669 (CO) cm<sup>-1</sup>. – <sup>1</sup>H NMR (300 MHz, [D<sub>6</sub>]-DMSO):  $\delta$  = 2.55 (s, 3H, CH<sub>3</sub>), 2.61 (s, 3H, CH<sub>3</sub>), 6.31 (s, 2H, NH<sub>2</sub>), 7.39 (d, 2H, *J* = 12 Hz, 2H Ar), 7.52 (d, 2H, *J* = 12 Hz, 2H Ar). – MS (EI, 70 eV): *m/z* (%) = 277 (18)



Scheme 2.

[M<sup>+</sup>]. – C<sub>13</sub>H<sub>12</sub>O<sub>2</sub>N<sub>3</sub>Cl (277.7): calcd. C 56.22, H 4.36, N 15.13, Cl 12.77; found C 56.10, H 4.35, N 15.20, Cl 12.80.

*5-Amino-2-(4-chlorophenyl)-7-methyl-2H-pyrazolo[4,3-*b*]pyridine-3,6-dicarbonitrile (9)*

A mixture of **1a** (2.6 g, 0.01 mol), malononitrile (0.66 g, 0.01 mol) and triethylamine (1.01 g, 0.01 mol) in ethanol (25 ml) was refluxed for 4 h. The solvent was evaporated under vacuo, and the crude product was collected and crystallized from ethanol. – M.p. 202 °C. Yield: 77%. – IR:  $\nu$  = 3460, 3350 (NH<sub>2</sub>), 2220, 2215 (CN) cm<sup>-1</sup>. – <sup>1</sup>H NMR (300 MHz, [D<sub>6</sub>]-DMSO):  $\delta$  = 2.48 (s, 3H, CH<sub>3</sub>), 5.68 (s, 2H, NH<sub>2</sub>), 7.29 (d, 2H, *J* = 12 Hz, 2H Ar), 7.45 (d, 2H, *J* = 12 Hz, 2H Ar). – <sup>13</sup>C NMR (75 MHz, [D<sub>6</sub>]-DMSO):  $\delta$  = 157.05, 148.73, 143.58, 137.91, 134.67, 129.84, 125.30, 124.35, 114.83 (CN), 111.14 (CN), 110.88, 97.95, 15.16 (CH<sub>3</sub>). – MS (EI, 70 eV): *m/z* (%) = 308 (22) [M<sup>+</sup>]. – C<sub>15</sub>H<sub>9</sub>N<sub>6</sub>Cl (308.7): calcd. C 58.36, H 2.94, N 27.22, Cl 11.48; found C 58.50, H 3.11, N 27.35, Cl 11.50.

*Ethyl 2-(4-chlorophenyl)-7-hydroxy-2H-pyrazolo[4,3-*c*]pyridazine-3-carboxylate (11)*

A mixture of **1b** (3.07 g, 0.01 mol), sodium nitrite (0.69 g, 0.01 mol) and concentrated hydrochloric acid (2 ml) in acetic acid (15 ml) was stirred at r. t. for 2 h. The reaction mixture was poured into ice-water, the product so formed was collected by filtration and crystallized from ethanol to give **11**. – M.p. 210 °C. Yield: 72%. – IR:  $\nu$  = 3421 (OH), 1728 (CO)

cm<sup>-1</sup>. – <sup>1</sup>H NMR (300 MHz, [D<sub>6</sub>]-DMSO):  $\delta$  = 1.15 (t, 3H, *J* = 7.5 Hz, CH<sub>3</sub>), 4.32 (q, 2H, CH<sub>2</sub>), 7.40 (d, 2H, *J* = 12 Hz, 2H Ar), 7.66 (d, 2H, *J* = 12 Hz, 2H Ar), 7.76 (s, 1H), 13.30 (s, 1H, OH). – <sup>13</sup>C NMR (75 MHz, [D<sub>6</sub>]-DMSO):  $\delta$  = 173.40 (CO), 166.52 (C-7), 140.12, 138.38, 136.22 (C-3), 128.21, 127.24, 120.23, 116.41, 112.60, 108.83, 61.63 (CH<sub>2</sub>), 12.55 (CH<sub>3</sub>). – MS (EI, 70 eV): *m/z* (%) = 318 (31) [M<sup>+</sup>]. – C<sub>14</sub>H<sub>11</sub>O<sub>3</sub>N<sub>4</sub>Cl (318.7): calcd. C 52.76, H 3.48, N 17.58, Cl 11.12; found C 52.70, H 3.55, N 17.60, Cl 11.10.

*N-[3-Acetyl-1-(4-chlorophenyl)-5-cyano-1H-pyrazol-4-yl]-acetamide (13)*

A solution of **1a** (2.6 g, 0.01 mol) in acetic anhydride (25 ml) was refluxed for 3 h. The reaction mixture was cooled and the solid that separated was collected and crystallized from ethanol to give **13**. – M.p. 195 °C; yield: 75%. – IR:  $\nu$  = 3337 (NH), 2230 (CN), 1707, 1673 (CO) cm<sup>-1</sup>. – <sup>1</sup>H NMR (300 MHz, [D<sub>6</sub>]-DMSO):  $\delta$  = 2.10 (s, 3H, CH<sub>3</sub>), 2.58 (s, 3H, CH<sub>3</sub>), 4.45 (s, 1H, NH), 7.22 (d, 2H, *J* = 10 Hz, 2H Ar), 7.45 (d, 2H, *J* = 10 Hz, 2H Ar). – <sup>13</sup>C NMR (75 MHz, [D<sub>6</sub>]-DMSO):  $\delta$  = 193.60 (CO), 168.23 (CO), 140.84, 136.59, 134.62, 129.79, 128.22, 127.18, 126.03, 110.94 (CN), 27.21 (CH<sub>3</sub>), 22.84 (CH<sub>3</sub>). – MS (EI, 70 eV): *m/z* (%) = 302 (24) [M<sup>+</sup>]. – C<sub>14</sub>H<sub>11</sub>O<sub>2</sub>N<sub>4</sub>Cl (302.7): calcd. C 55.55, H 3.66, N 18.51, Cl 11.71; found C 55.50, H 3.55, N 18.40, Cl 11.80.

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