

# Stereoselective Vinylation of Aryl *N*-(2-Pyridylsulfonyl) Aldimines with 1-Alkenyl-1,1-heterobimetallic Reagents

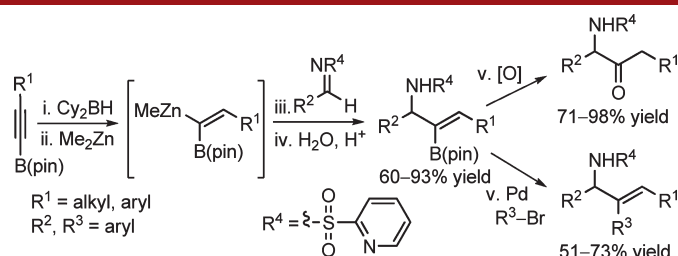
Nusrah Hussain, Mahmud M. Hussain, Muhammed Ziauddin,  
Plengchat Triyawatanyu, and Patrick J. Walsh\*

*P. Roy and Diana T. Vagelos Laboratories, University of Pennsylvania, Department of Chemistry, 231 South 34th Street, Philadelphia, Pennsylvania 19104-6323, United States*

pwalsh@sas.upenn.edu

Received October 14, 2011

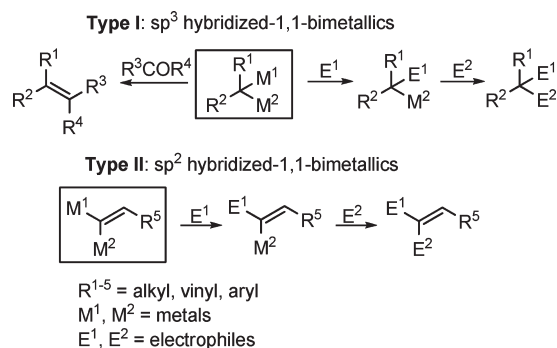
## ABSTRACT



Vinylation of aryl *N*-(2-pyridylsulfonyl) aldimines with versatile 1-alkenyl-1,1-borazinc heterobimetallic reagents is disclosed. In situ hydroboration of air-stable B(pin)-alkynes followed by chemoselective transmetalation with dimethylzinc and addition to aldimines provides B(pin)-substituted allylic amines in 53–93% yield in a one-pot procedure. The addition step can be followed by either B–C bond oxidation to provide  $\alpha$ -amino ketones (71–98% yield) or Suzuki cross-coupling to furnish trisubstituted 2-arylated (*E*)-allylic amines (51–73% yield).

Highly stereoselective construction of C–C double bonds remains a challenge in organic synthesis.<sup>1</sup> In this regard,  $sp^3$  and  $sp^2$  hybridized heterobimetallic reagents of type I and II (Scheme 1) are potentially useful intermediates, because each metal–carbon bond can be chemoselectively exploited in C–C bond forming reactions.<sup>2–4,6</sup> Furthermore, these versatile heterobimetallic reagents can be employed in tandem reactions, minimizing isolation

Scheme 1. 1,1-Heterobimetallics in Organic Synthesis



(1) Corey, E. J.; Cheng, X.-M. *The Logic of Chemical Synthesis*; John Wiley & Sons: New York, 1988.

(2) Marek, I.; Normant, J. -F. *Chem. Rev.* **1996**, *96*, 3241.

(3) (a) Marek, I. *Chem. Rev.* **2000**, *100*, 2887. (b) Marek, I. *Actual. Chim.* **2003**, 15.

(4) Waas, J. R.; Sidduri, A.; Knochel, P. *Tetrahedron Lett.* **1992**, *33*, 3717.

(5) (a) Ho, T. L. *Tandem Organic Reactions*; Wiley and Sons: New York, 1992; p 502. (b) Wender, P. A.; Miller, B. L. *Organic Synthesis: Theory and Applications*; Hudlicky, T., Ed.; JAI Press: Greenwich, CT, 1993; Vol. 2, p 27. (c) Parsons, P. J.; Penkett, C. S.; Shell, A. J. *Chem. Rev.* **1996**, *96*, 195. (d) Schmalz, H.-G.; Geis, O. *Handbook of Organopalladium Chemistry for Organic Synthesis*; Negishi, E., Ed.; Wiley: New York, 2002; Vol. 2, p 2377. (e) Nicolaou, K. C.; Montagnon, T.; Snyder, S. A. *Chem. Commun.* **2003**, 551. (f) Tietze, L. F.; Beifuss, U. *Angew. Chem., Int. Ed.* **1993**, *32*, 131. (g) Walsh, P. J.; Kozlowski, M. C. *Fundamentals of Asymmetric Catalysis*; University Science Books: Sausalito, CA, 2008.

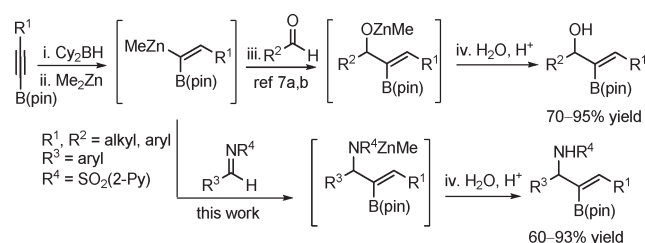
(6) Hussain, M. M.; Walsh, P. J. *Acc. Chem. Res.* **2008**, *41*, 883.

and purification of intermediates.<sup>5</sup> These attributes allow for rapid development of molecular complexity from simple building blocks.

As part of our program in developing stereoselective C–C bond forming reactions,<sup>6</sup> we have reported the generation of 1-alkenyl-1,1-heterobimetallic reagents

based on boron and zinc from readily available, air-stable B(pin)-substituted alkynes (Scheme 2).<sup>7a</sup> Thus, regioselective hydroboration of B(pin)-alkynes generates the 1,1-bis(boro) intermediates.<sup>7a,8</sup> Chemoselective transmetalation of the more reactive vinyl-BCy<sub>2</sub> bond generates 1-alkenyl-1,1-heterobimetallic reagents. The difference in reactivity between Zn–C vs B–C bonds allows for selective reaction at the Zn–C bond with aldehydes to yield B(pin)-substituted allylic zinc alkoxide intermediates. The alkoxide intermediates were then employed in various tandem reactions to form an array of compounds such as B(pin)-substituted allylic alcohols,<sup>7a–c</sup> α-hydroxy ketones,<sup>7a</sup> tri-substituted (*E*)-allylic alcohols,<sup>7a</sup> B(pin)-substituted cyclopropyl alcohols,<sup>7b</sup> and B(pin)-substituted allylic acetates.<sup>7d</sup>

**Scheme 2.** Generation of 1-Alkenyl-1,1-heterobimetallics of Boron/Zinc and Additions to Electrophiles

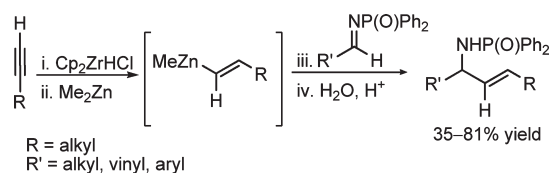


Herein, we report the addition of alkenyl-1,1-heterobimetallic reagents to *N*-(2-pyridylsulfonyl) aldimines to furnish B(pin)-substituted allylic amines (Scheme 2, lower part). The addition can be followed by oxidation of the B–C bond to provide α-aminoketones or by Suzuki cross-coupling to provide densely functionalized trisubstituted (*E*)-allylic amines.

Allylic amines<sup>9</sup> are important pharmacophores that can exhibit significant biological properties. Examples include Acrivastine (Semprex),<sup>10</sup> Flunarizine,<sup>11</sup> and several GABA uptake inhibitors.<sup>12</sup> As a result, additions to imines have attracted considerable attention. For example, Wipf and co-workers reported the addition of vinylzinc reagents

to aldimines activated with a diphenylphosphinoyl moiety (Scheme 3).<sup>13</sup> Carretero<sup>14</sup> and co-workers demonstrated that the reactivity of *N*-sulfonyl imines could be increased in the presence of an appropriately positioned heteroaryl group. Using this strategy, they developed the alkylation of aryl *N*-(2-pyridylsulfonyl) aldimines with organozinc halides.<sup>14b</sup> The Carretero and Toru groups both have utilized the *N*-pyridylsulfonyl as a novel stereocontrol element in enantioselective Mannich-type reactions with silyl enol ethers in the presence of chiral copper catalysts.<sup>15</sup> Various related nucleophilic reagents, such as dialkyl zinc,<sup>5,16,17</sup> alkynylzinc,<sup>5,18</sup> diethylaluminum cyanide,<sup>19</sup> and Danishefsky's diene,<sup>20</sup> have also been investigated in imine addition reactions to yield the desired amines.

**Scheme 3.** Wipf's Vinylation of Aryl Diphenylphosphinoyl Imines via Vinylzinc Reagents



Our first task in the addition of bimetallics to imines was to find a suitable imine activating group. The bimetallic reagent was generated and allowed to react with activated imines at  $-18\text{ }^{\circ}\text{C}$  (Table 1). *N*-Tosylimines gave a trace addition product with our alkenyl heterobimetallic reagents (entry 1). Rather, a significant amount of reduction product was isolated. The *N*-Boc imine behaved similarly, failing to furnish the desired amine (entry 2). When the activating group was changed to diphenylphosphinoyl, less than 30% of the allylic amine was isolated. Gratifyingly, the bimetallic addition to *N*-pyridyl sulfonyl imine occurred smoothly in 73% yield in toluene at  $-18\text{ }^{\circ}\text{C}$  to furnish the desired product (entry 4). The addition was then optimized with the *N*-pyridyl sulfonyl imines. Switching the solvent from toluene to dichloromethane improved the yields slightly (entry 4 vs 7), while, in THF, almost no product was formed (entry 5). Dimethylzinc performed better than diethylzinc (entry 7 vs 9). Increasing the reaction temperature from  $-18$  to  $-10\text{ }^{\circ}\text{C}$  led to a diminished yield (entry 6 vs 7). With the optimized conditions in entry 7, the scope of the reaction was examined.

Aryl aldimines with electron-donating or -withdrawing groups were good substrates, providing the B(pin)

(7) (a) Li, H.; Carroll, P. J.; Walsh, P. J. *J. Am. Chem. Soc.* **2008**, *130*, 3521. (b) Hussain, M. M.; Li, H.; Hussain, N.; Ureña, M.; Carroll, P. J.; Walsh, P. J. *J. Am. Chem. Soc.* **2009**, *131*, 6516. (c) Hussain, M. M.; Hernández-Toribio, J.; Carroll, P. J.; Walsh, P. J. *Angew. Chem., Int. Ed.* **2011**, *50*, 6337. (d) Hussain, M. M.; Walsh, P. J. *Angew. Chem., Int. Ed.* **2010**, *49*, 1834.

(8) Srebnik, M. *Tetrahedron Lett.* **1991**, *32*, 2449.

(9) (a) Kovacic, P. *Med. Hypotheses* **2007**, *69*, 1105. (b) Kovacic, P. *Med. Hypotheses* **2006**, *67*, 151.

(10) (a) Martín, V. S.; Woodward, S. S.; Katsuki, T.; Yamada, Y.; Ikeda, M.; Sharpless, K. B. *J. Am. Chem. Soc.* **1981**, *103*, 6237. (b) Slater, J. W.; Zechnich, A. D.; Haxby, D. G. *Drugs* **1999**, *57*, 31.

(11) (a) Straub, H.; Koehling, R.; Speckmann, E. *J. Brain Res.* **1994**, *658*, 119. (b) Ashton, D.; Reid, K.; Willems, R.; Marrannes, R.; Wauguier, A. *Drug Develop. Res.* **1986**, *8*, 396.

(12) Gibson, J. R.; Manna, V. K.; Salisbury, J. J. *Int. Med. Res.* **1989**, *17*, 28.

(13) Wipf, P.; Kendall, C.; Stephenson, C. R. J. *J. Am. Chem. Soc.* **2003**, *125*, 761.

(14) (a) Esquivias, J.; Arrayás, R. G.; Carretero, J. C. *J. Org. Chem.* **2005**, *70*, 7451. (b) Esquivias, J.; Arrayás, R. G.; Carretero, J. C. *Angew. Chem., Int. Ed.* **2006**, *45*, 629.

(15) (a) González, A. S.; Arrayás, R. G.; Carretero, J. C. *Org. Lett.* **2006**, *8*, 2977. (b) Nakamura, S.; Sano, H.; Nakashima, H.; Kubo, K.; Shibata, N.; Toru, T. *Tetrahedron: Asymmetry* **2007**, *48*, 5565.

(16) Charette, A. B.; Boezio, A. A.; Côté, A.; Moreau, E.; Pytkowicz, J.; Desrosiers, J. -N.; Legault, C. *Pure Appl. Chem.* **2005**, *77*, 1259.

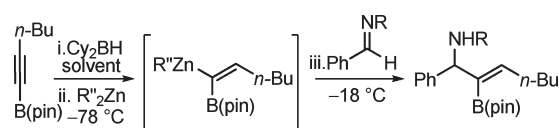
(17) Porter, J. R.; Traverse, J. F.; Hoveyda, A. H.; Snapper, M. L. *J. Am. Chem. Soc.* **2001**, *123*, 984.

(18) Zani, L.; Alesi, S.; Cozzi, P. G.; Bolm, C. *J. Org. Chem.* **2006**, *71*, 1558.

(19) Nakamaru, S.; Sato, N.; Sugimoto, M.; Toru, T. *Tetrahedron: Asymmetry* **2004**, *15*, 1513.

(20) Mancheño, O. G.; González, A. S.; Arrayás, R. G.; Carretero, J. C. *J. Am. Chem. Soc.* **2004**, *126*, 456.

**Table 1.** Optimization of the Addition of Alkenyl-1,1-heterobimetallics to *N*-Pyridyl Sulfonyl Imines



entry	R <sub>2</sub> 'Zn	solvent	R	yield (%) <sup>a</sup>
1	Me <sub>2</sub> Zn	toluene	SO <sub>2</sub> Tol	trace
2	Me <sub>2</sub> Zn	toluene	Boc	trace
3	Me <sub>2</sub> Zn	toluene	P(O)Ph <sub>2</sub>	<30
4	Me <sub>2</sub> Zn	toluene	SO <sub>2</sub> (2-Py)	73
5	Me <sub>2</sub> Zn	THF <sup>b</sup>	SO <sub>2</sub> (2-Py)	trace
6	Me <sub>2</sub> Zn	CH <sub>2</sub> Cl <sub>2</sub>	SO <sub>2</sub> (2-Py)	68 <sup>b</sup>
7	Me <sub>2</sub> Zn	CH <sub>2</sub> Cl <sub>2</sub>	SO <sub>2</sub> (2-Py)	80
8	Et <sub>2</sub> Zn	toluene	SO <sub>2</sub> (2-Py)	64
9	Et <sub>2</sub> Zn	CH <sub>2</sub> Cl <sub>2</sub>	SO <sub>2</sub> (2-Py)	65

<sup>a</sup> Isolated yields. <sup>b</sup> Reaction performed at –10 °C.

**Table 2.** Addition of Alkenyl-1,1-heterobimetallics to *N*-Pyridyl Sulfonyl Imines

entry	borane	imine	allylic amines	yield (%) <sup>a</sup>
1				80
2				68
3				87
4				93
5				60
6				70
7				53

<sup>a</sup> Isolated yields.

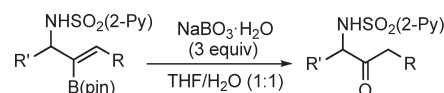
substituted allylic amines in 53–93% yield (Table 2). The air-stable B(pin)-substituted alkynes can contain

aromatic or aliphatic substituents (R = aryl, alkyl). Even the bulky *tert*-butyl-substituted B(pin) alkyne underwent addition to generate the corresponding allylic amine in 60% yield (entry 5). Substitution at the *ortho* position of the aldimine resulted in a slightly lower yield (entry 7 vs 3–5).

Having established vinylation of aldimines with our heterobimetallics, we sought to examine tandem reactions involving the B–C bond. Two such reactions are B–C bond oxidation and Suzuki cross-coupling.

We envisioned that oxidation of the 2-B(pin)-substituted allylic amines would provide access to valuable α-amino ketones, which have important biological activity.<sup>21</sup> In the presence of NaBO<sub>3</sub>·H<sub>2</sub>O<sup>22</sup> in THF/H<sub>2</sub>O (1:1) at rt, B(pin)-substituted allylic amines were smoothly oxidized to the corresponding α-amino ketones in 71–98% yield (Table 3).

**Table 3.** Oxidation of Allylic Amines to α-Amino Ketones



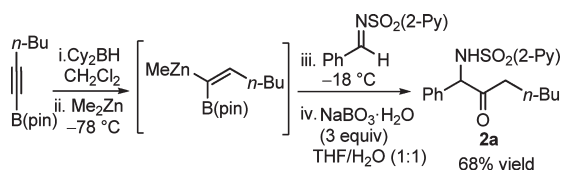
entry	allylic amines	amino ketones	yield (%) <sup>a</sup>
1	<b>1a</b>		80
2	<b>1b</b>		75
3	<b>1c</b>		96
4	<b>1d</b>		98
5	<b>1e</b>		87
6	<b>1f</b>		71
7	<b>1g</b>		87

<sup>a</sup> Isolated yields.

(21) (a) Hanada, M.; Sugawara, K.; Koko, K.; Toda, S.; Nishiyama, Y.; Tomita, K.; Yamamoto, H.; Konishi, M.; Oki, T. *J. Antibiot.* **1992**, 45, 1746. (b) Soares, C. O.; Alves, M. J. M.; Becharaa, E. J. H. *Free Radic. Med.* **2011**, 50, 1760.

(22) Kabalka, G. W.; Shoup, T. M.; Goudgaon, N. M. *Tetrahedron Lett.* **1989**, 30, 1483.

**Scheme 4.** Tandem Addition/B–C Bond Oxidation To Yield  $\alpha$ -Amino Ketone **2a**



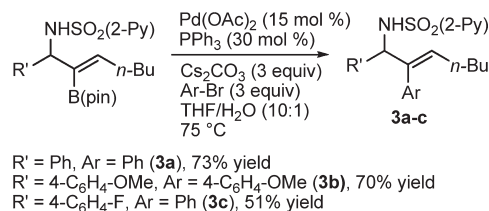
The addition/oxidation reaction can also be executed in a tandem fashion. Thus, after completion of the bimetallic addition to the aldimine, the reaction mixture was subjected to  $\text{NaBO}_3 \cdot \text{H}_2\text{O}$  to provide the  $\alpha$ -amino ketone in 68% yield in one pot (Scheme 4).

We next utilized the B–C bond in Suzuki cross-coupling reactions. In the presence of  $\text{Pd}(\text{OAc})_2$  (15 mol %),  $\text{PPh}_3$  (30 mol %),  $\text{Cs}_2\text{CO}_3$  (3 equiv), and aryl bromide (3 equiv) in  $\text{THF}/\text{H}_2\text{O}$  (10:1) at 75 °C, the B(pin)-substituted allylic amines smoothly underwent cross-coupling to furnish the 2-arylated trisubstituted (*E*)-allylic amines in 51–73% yield (Scheme 5). No (*Z*)-double bond isomers were observed in these reactions.

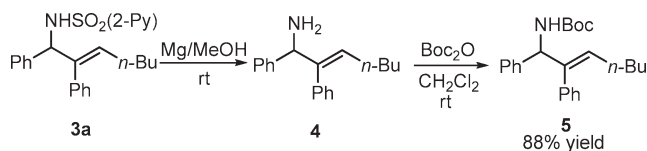
Although the 2-pyridyl sulfonyl group is essential for the addition step, its removal is important for applications of the products. The 2-pyridyl sulfonyl group was readily cleaved on treatment of **1a** with magnesium in MeOH to liberate the free amine **4** (Scheme 6).<sup>23,24</sup> The free amine **4** was then transformed into its Boc-derivative **5** on treatment with  $\text{Boc}_2\text{O}$  at rt in 88% overall yield (Scheme 6).

In summary, the nucleophilic addition of 1-alkenyl-1,1-borozinc heterobimetallic reagents to aryl *N*-(2-pyridyl-sulfonyl) aldimines has been developed. This protocol provides a variety of B(pin)-substituted allylic amines in good yields. The addition step can be followed by a tandem oxidative cleavage of the B–C bond to furnish valuable  $\alpha$ -amino ketones or by Suzuki cross-coupling to form

**Scheme 5.** Suzuki Cross-Coupling of Allylic Amines



**Scheme 6.** Removal of the 2-Pyridyl Sulfonyl Group followed by Boc-Protection



2-arylated trisubstituted (*E*)-allylic amines. It is noteworthy that 2-arylated trisubstituted (*E*)-allylic amines are not currently accessible via the Tsuji–Trost reaction, because 2-arylated allylic acetates are not good substrates for the allylic substitution reaction.<sup>7d</sup> Given that amino ketones and allylic amines are important pharmacophores,<sup>10–12,21</sup> we anticipate that the methods described herein will be useful to the synthetic community.

**Acknowledgment.** We acknowledge the NIH (NIGMS GM58101) and the NSF (CHE-0848467). Funds for instrumentation were provided by the NIH for an MS (1S10RR023444). M.M.H. thanks the University of Pennsylvania SAS for a Dissertation Completion Fellowship.

**Supporting Information Available.** Procedures and full characterization of new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

(23) Pak, C. S.; Lim, D. S. *Synth. Commun.* **2001**, *31*, 2209.  
(24) Arrayás, R. G.; Cabrera, S.; Carretero, J. C. *Org. Lett.* **2005**, *7*, 219.