

α -LNA, locked nucleic acid with α -D-configuration

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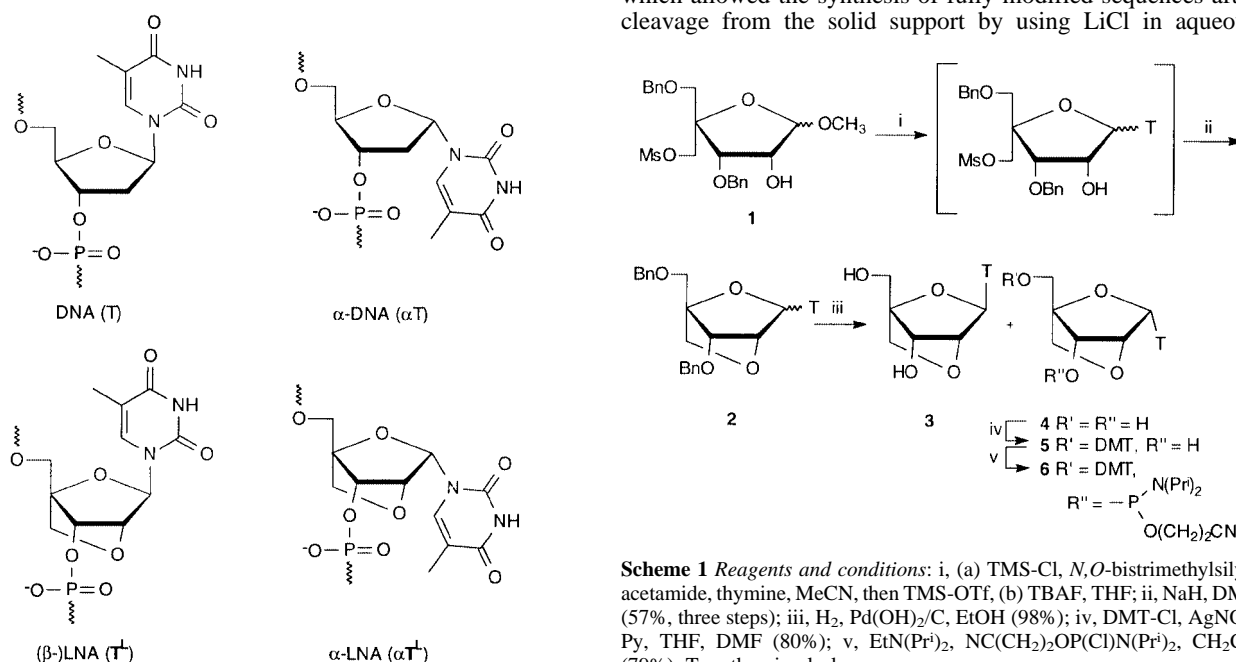
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The bicyclic thymine monomer of α -LNA (α T^L) was efficiently synthesised and used in the synthesis of α -LNA sequences: incorporation of single α T^L-monomers in α -configured oligothymidylates destabilises the affinity towards both complementary DNA and RNA, whereas a fully modified α -LNA sequence displays a very efficient recognition of complementary RNA.

Conformationally restricted oligonucleotide analogues have been intensively investigated for their abilities in high affinity nucleic acid recognition.¹ As a prime example, LNA (locked nucleic acid)[†] has recently been introduced as a nucleic acid analogue displaying unprecedented affinity towards DNA and RNA.² The anomeric inverted analogue of DNA (α -DNA) has been demonstrated to hybridise efficiently with complementary DNA and RNA with a parallel strand orientation, and to be highly resistant towards degradation by nucleases.³ Chemically modified analogues of α -DNA have also been investigated,⁴ including the introduction of α -configured bicyclic nucleoside monomers.⁵

The α -2'-deoxynucleoside monomers in α -DNA as well as the β -2'-deoxynucleoside monomers in DNA exist in an equilibrium between the two low-energy N- and S-type conformational ranges.^{6,7} In none of the α -DNA analogues, investigated so far,^{4,5} have the monomers been efficiently restricted towards N-type conformations. Hoping to obtain an unprecedented parallel nucleic acid recognition and, thereby, a new tool in the development of diagnostic probes and antisense therapeutics, we therefore decided to examine the incorporation into oligonucleotides of an α -nucleoside analogue which is conformationally locked in an N-type conformation, i.e. α -LNA.[‡]

In order to synthesise the α -LNA thymine monomer (α T^L, **4**, Scheme 1) we first investigated the coupling of thymine to appropriate bicyclic carbohydrate precursors.⁹ However, this approach has never been optimised to give the target compound in a satisfactory yield, and we hereby introduce an alternative synthetic strategy. The starting methyl furanoside **1** was obtained, as described previously,⁹ and used in a modified Vorbrüggen nucleobase coupling reaction (Scheme 1). After varying the reaction conditions, the best result in terms of yield and ratio of products was obtained by using *in situ* TMS-protection of both the 2'- hydroxy functionality in **1** and the thymine, followed by coupling by using TMS-triflate as the Lewis acid in refluxing acetonitrile for seven days. After desilylation, the mixture of nucleosides was reacted with sodium hydride to give the anomeric mixture **2** (α : β \approx 1.3:1 according to ¹H NMR) in 57% overall yield. After removal of the benzyl groups by hydrogenation, the β - and α -LNA monomers (**3**² and **4**,⁹ respectively) were obtained in 98% yield and separated. The configuration of **4** was confirmed by comparison of NMR data with its exact enantiomer^{8d} and by NOE-difference spectra as mutual contacts were observed between H-1' and H-3' and between H-5'' and H-6, respectively. The α T^L monomer **4** was prepared for incorporation into oligonucleotide sequences by protection, by using the dimethoxytrityl (DMT) group to give **5**, followed by phosphitylation to give the phosphoramidite synthon **6**.[§] This compound was used in automated solid phase synthesis of oligonucleotides by using the phosphoramidite approach.¹⁰ In connection with the α -thymidine (α T) phosphoramidite, the α -LNA sequences were obtained (Table 1) by using tetrazole activation and 10 min coupling times giving >98% stepwise coupling yields. The modified oligomers **8–10** and **12**, **13** were synthesised by using the DMT-ON mode on universal CPG-support (Biogenex), which allowed the synthesis of fully modified sequences after cleavage from the solid support by using LiCl in aqueous



Scheme 1 Reagents and conditions: i, (a) TMS-Cl, *N,O*-bistrimethylsilylacetamide, thymine, MeCN, then TMS-OTf, (b) TBAF, THF; ii, NaH, DMF (57%, three steps); iii, H₂, Pd(OH)₂/C, EtOH (98%); iv, DMT-Cl, AgNO₃, Py, THF, DMF (80%); v, EtN(Pr)₂, NC(CH₂)₂OP(Pr)₂, CH₂Cl₂ (79%). T = thymine-1-yl.

Table 1 Hybridisation data for α -LNA sequences and reference strands

Sequence	dA ₁₄ complement		rA ₁₄ complement		rA ₆ CA ₇ complement <i>T_m</i> /°C ^a
	<i>T_m</i> /°C ^a	ΔT_m /°C ^b	<i>T_m</i> /°C ^a	ΔT_m /°C ^b	
7 5'-T ₁₄	33.0		30.0		n.d. ^c
8 5'- α T ₁₄	32.0		43.0		n.d. ^c
9 5'- α T ₇ T ₄ T ₆	25.5	-6.5	35.0	-8.0	n.d. ^c
10 5'- α T ₅ T ₄ T ₅	26.0	-1.5	24.5	-4.6	n.d. ^c
11 5'-T ₁₀	22.0		20.0		n.d. ^c
12 5'- α T ₁₀	18.0		33.5		22.0
13 5'- α T ₁₀	no <i>T_m</i> ^d		45.0	+1.2 ^e ; +2.5 ^f	37.0

^a Melting temperatures (*T_m*) obtained from the maxima of the first derivatives of the melting curve (*A*₂₆₀ vs. temperature) recorded in a buffer containing 10 mM Na₂HPO₄, 100 mM NaCl, 0.1 mM EDTA, pH 7.0 using 1.5 μ M concentrations of the two complementary sequences, assuming identical extinction coefficients for all thymine nucleotides. ^b The change in *T_m* value per modification compared with the reference strand **8**. ^c Not determined. ^d No clear cooperative transition was seen. ^e Compared with **12**. ^f Compared with **11**.

ammonia. The oligomers were purified by using disposable reverse phase chromatography cartridges (Cruachem), which yielded products with >90% purity, as judged from capillary gel electrophoresis. The compositions of α -LNA sequences were verified from MALDI-MS spectra.[¶]

The α -LNA sequences **9**, **10** and **13**, as well as their α -DNA counterparts **8** and **12**, were mixed with their DNA and RNA complements and the resulting hybridisation data are shown in Table 1. Compared to the unmodified sequence **7**, the affinity of the unmodified α -sequence **8** towards dA₁₄ is, as expected,³ similar. However, the introduction of one or four α -LNA monomers (**9** or **10**, respectively) results in strongly decreased affinities towards dA₁₄. Towards complementary RNA, rA₁₄, **8** has a higher affinity than **7** and the destabilising effect of one or four α -LNA monomers is even more pronounced. However, in both cases, the introduction of a block of α -LNA monomers diminishes the combined destabilising effect (comparing ΔT_m for **9** and **10**). This suggests that the N-type conformation α -LNA monomers do not have the ability to alter neighbouring nucleosides and, thereby, change the overall single strand conformation towards a form which is more preferable for duplex formation. As judged from NMR studies, this is an important feature of the original (β -)LNA.¹¹

The fully modified α -LNA sequence **13** displays strong recognition of the complementary RNA-strand ($\Delta T_m = +1.2$ °C per monomer compared to α -DNA **12** and +2.5 °C compared to the unmodified oligodeoxynucleotide sequence **11**). On the other hand, no clear cooperative transition was seen when **13** was mixed with dA₁₄. This indicates either that α -LNA is unable to recognise DNA and is thereby extraordinarily RNA-selective or, alternatively, that the two sequences might form a secondary structure not detectable by UV-spectroscopy at 260 nm. Thus, a broad non-cooperative transition at 40–45 °C is seen in the mixture, but a similar transition is also observed for **13** alone. Even though the possibility of self-melting has been described earlier for longer α -oligothymidylate sequences,¹² the broad transition and low hyperchromicity observed for **13** alone does not indicate the melting process of a duplex structure, but rather a transition between secondary forms of single strands. The presence of a duplex between **13** and rA₁₄ was confirmed by the fact that a clear melting transition of a duplex between **13** and a mis-matching complementary sequence was observed with *T_m* decreased by 8 °C (Table 1). Furthermore, thermal stabilities measured at higher ionic concentrations (data not shown) were increased, as expected, for all the duplexes involving **11**–**13** and RNA-complements as well as **11** and **12** with dA₁₄, whereas no clear cooperative transitions at increased temperature were observed with **13**:dA₁₄ or with **13** alone.

In conclusion, α -LNA is able to form a duplex with complementary RNA with a high thermal stability comparable to other stereoisomers of LNA,⁸ even though the original (β)LNA still displays the highest affinity towards complementary nucleic acids. Nevertheless, α -LNA in the present oligothymidylate sequence displays the highest affinity towards RNA of any α -D-configured oligonucleotide analogue. However, from the sequences presented here we are not able to determine whether α -LNA prefers a parallel strand orientation upon hybridisation. This subject is under investigation in our laboratories via the synthesis of α -LNA sequences with mixed nucleobase compositions, in addition to further examination of the properties and applications of α -LNA.

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Notes and references

† LNA is defined as an oligonucleotide containing one or more LNA monomers which are bicyclic nucleosides preorganized in N-type conformations. α -LNA is therefore defined as an oligonucleotide containing one or more monomeric α -D-LNA nucleosides in connection to unmodified α -D-nucleosides.

‡ Three other stereoisomers of LNA have been recently introduced⁸ and their affinities towards both complementary RNA and the enantiomeric L-RNA have been investigated.^{8c} In that sense, an α -LNA sequence has been examined in form of the duplex between its enantiomer α -L-LNA and L-RNA, but only as an (almost) fully modified sequence and only against RNA.^{8c}

§ Selected data for **6**: δ_p (CDCl₃, 121.5 MHz with 85% H₃PO₄ as external standard) 150.9, 151.1.

¶ MALDI-MS: *m/z* ([M – H][–] (found/calc.): **9** (4227.1/4223.8); **10** (4309.2/4307.8); **13** (3261.6/3260.1).

- 1 P. Herdewijn, *Liebigs Ann. Chem.*, 1996, 1337; P. Herdewijn, *Biochim. Biophys. Acta*, 1999, **1489**, 167.
- 2 S. K. Singh, P. Nielsen, A. A. Koshkin and J. Wengel, *Chem. Commun.*, 1998, 455; A. A. Koshkin, S. K. Singh, P. Nielsen, V. K. Rajwanshi, R. Kumar, M. Meldgaard, C. E. Olsen and J. Wengel, *Tetrahedron*, 1998, **54**, 3607; S. Obika, D. Nanbu, Y. Hari, J. Andoh, K. Morio, T. Doi and T. Imanishi, *Tetrahedron Lett.*, 1998, **39**, 5401; A. A. Koshkin, P. Nielsen, M. Meldgaard, V. K. Rajwanshi, S. K. Singh and J. Wengel, *J. Am. Chem. Soc.*, 1998, **120**, 13 252; J. Wengel, *Acc. Chem. Res.*, 1999, **32**, 301.
- 3 F. Morvan, B. Rayner, J.-L. Imbach, S. Thenet, J.-R. Bertrand, J. Paoletti, C. Malvy and C. Paoletti, *Nucleic Acids Res.*, 1987, **15**, 3421; M. Durand, J. C. Maurizot, N. T. Thuong and C. Helene, *Nucleic Acids Res.*, 1988, **16**, 5039.
- 4 C. H. Gottfredsen, J. P. Jacobsen and J. Wengel, *Bioorg. Med. Chem.*, 1996, **4**, 1217; K. Pongracz and S. M. Gryaznov, *Nucleic Acids Res.*, 1998, **26**, 1099; F. Morvan, J. Zeidler and B. Rayner, *Tetrahedron*, 1998, **54**, 71; A. Laurant, M. Laval, F. Debart, J.-J. Vasseur and B. Rayner, *Nucleic Acids Res.*, 1999, **27**, 4151.
- 5 M. Bolli, P. Lubini and C. Leumann, *Helv. Chim. Acta*, 1995, **78**, 2077; R. Zou and M. D. Matteucci, *Bioorg. Med. Chem. Lett.*, 1998, **8**, 3049.
- 6 W. Saenger, *Principles of Nucleic Acid Structure*, Springer, New York, 1984.
- 7 C. Thibaudeau and J. Chattopadhyaya, *Nucleosides Nucleotides*, 1997, **16**, 523.
- 8 (a) V. K. Rajwanshi, A. E. Håkansson, B. M. Dahl and J. Wengel, *Chem. Commun.*, 1999, 1395; (b) V. K. Rajwanshi, A. E. Håkansson, R. Kumar and J. Wengel, *Chem. Commun.*, 1999, 2073; (c) V. K. Rajwanshi, A. E. Håkansson, M. D. Sørensen, S. Pitsch, S. K. Singh, R. Kumar, P. Nielsen and J. Wengel, *Angew. Chem., Int. Ed.*, 2000, **39**, 1656; (d) A. E. Håkansson and J. Wengel, personal communication.
- 9 P. Nielsen and J. Wengel, *Chem. Commun.*, 1998, 2645.
- 10 M. H. Caruthers, *Acc. Chem. Res.*, 1991, **24**, 278.
- 11 M. Petersen, C. B. Nielsen, K. E. Nielsen, G. A. Jensen, K. Bondensgaard, S. K. Singh, V. K. Rajwanshi, A. A. Koshkin, B. M. Dahl, J. Wengel and J. P. Jacobsen, *J. Mol. Recognit.*, 2000, **13**, 44.
- 12 U. Neidlein and C. Leumann, *Tetrahedron Lett.*, 1992, **33**, 8057.