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Stereospecificity in the Au-catalysed cyclisation of monoallylic diols. Synthesis of (+)-isoalthalactone†

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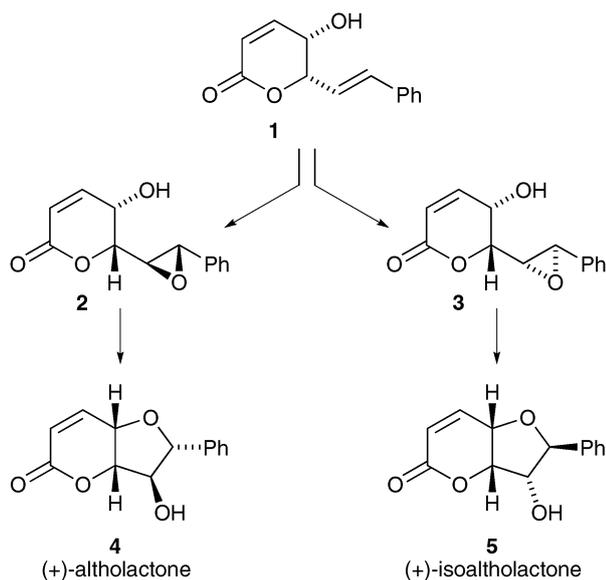
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We describe a concise synthesis of (+)-isoalthalactone via a Au-catalysed cyclisation of a monoallylic diol to form the tetrahydrofuran ring. Analogous cyclisations show that the stereochemical outcome is dictated by the stereochemistry of the diol substrate.

(+)-Althalactone¹ (a.k.a. goniothalenol)² **4**, Scheme 1, was originally isolated from a specimen of an unnamed *Polyalthia* species, its gross structure established by chemical degradation, and its stereochemistry proposed on the basis of NMR and CD spectroscopy, later confirmed by total synthesis of the enantiomers.³ Subsequently, interesting biological activity was found, including an antiproliferative effect on breast cancer cells,⁴ induction of apoptosis in HL-60 cells,⁵ and toxicity

towards P388 cells.² Thirteen years after the first report of (+)-althalactone, the diastereomer (+)-isoalthalactone **5** was described as an isolate from combined extracts of *Goniothalamus malayanus*, *G. montanus*, and *G. tapis*.⁶ Both compounds may be considered⁶ to be biosynthesised by 5-*endo* cyclisation⁷ of the epoxides **2** and **3** which, in turn, could arise from oxidation of either face of 5-hydroxygoniothalamine (**1**). With one exception⁸ⁱ the syntheses⁸ of (+)-isoalthalactone rely on exactly this strategy in order to construct the THF ring although there are variations in the relative timing of the ring-forming steps.

Our interest in this area arose out of a research programme based on the cyclisation of alcohols of the form **6** (Scheme 2) onto alkenes, following electrophilic activation, to provide functionalised tetrahydrofuran cores **7** and **8** stereoselectively.⁹ In this context, we assessed Aponick's Au-catalysed cyclisation of monoallylic diols that appeared in the literature whilst our study was underway.^{10,11} At the time we had in hand ketone **10** (Scheme 3), available from D-ribonolactone acetonide (**9**) in four steps,¹² and conceived a short synthesis of (+)-isoalthalactone from this compound based on cyclisation under Aponick's conditions. The synthesis initiated with cross metathesis of racemic alkene **11**¹³ and, following diastereoselective reduction with L-Selectride,¹² monoallylic diol **13** was obtained as an epimeric mixture at the allylic alcohol centre. Application of Aponick's conditions afforded three diastereomers of the cyclisation products (**14–16**). Such compounds have been shown to cyclise to (iso)althalactone^{8g–i,14} and, initially, separated diastereomer **14** was elaborated to (+)-isoalthalactone by sequential acetonide and ester hydrolysis then lactonisation¹⁵ of the *Z*-enoic acid formed *in situ*. In parallel, the inseparable diastereomers **15** and **16** were taken through an efficient one-pot process consisting of acetonide deprotection then lactonisation following a solvent swap from methanol to benzene; the diol derived from acetonide **15** was converted back into the acetonide *in situ* in order to allow isolation of (+)-isoalthalactone.



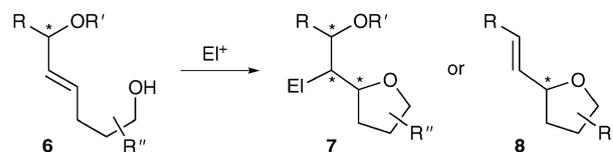
Scheme 1 Key steps in the proposed biogenesis of styryl lactones althalactone and isoalthalactone.

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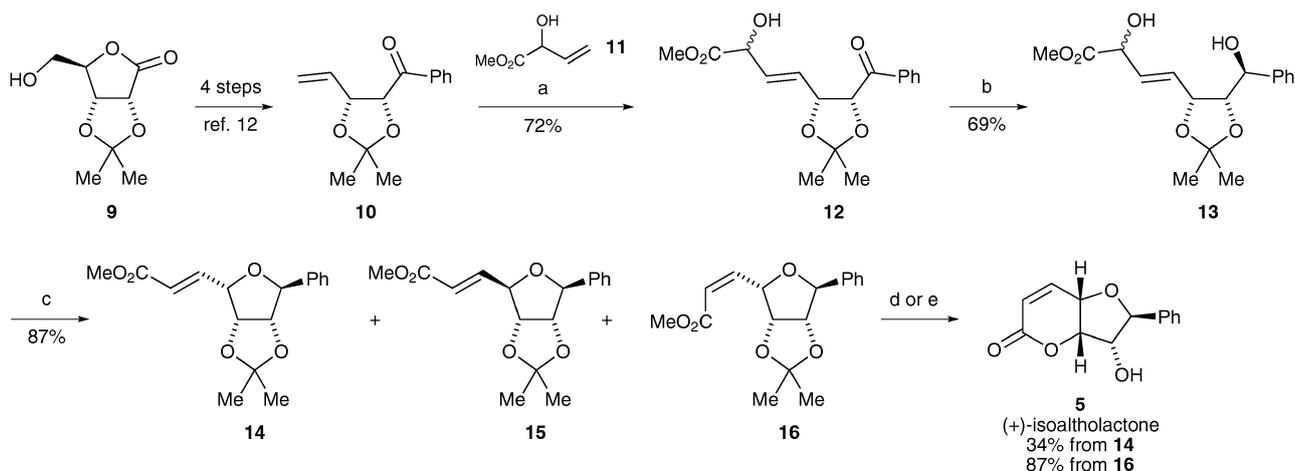
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† Electronic supplementary information (ESI) available: Experimental procedures, spectroscopic data, and support for the stereochemistry of cyclisation products. See DOI: 10.1039/c1cc11805f



Scheme 2 Allylic stereoselection accompanying electrophilic cyclization.



Scheme 3 Reagents and conditions. (a) (\pm)-**11**, cat. Grubbs II, CH_2Cl_2 , reflux, 2 h; (b) L-Selectride, THF, -78°C , 0.5 h; (c) cat. Ph_3PAuCl , cat. AgOTf , 4 Å MS, CH_2Cl_2 , 20°C , 2 h ($dr \approx 1:1:1$); (d) from **14**: 1. cat. TsOH, MeOH, reflux, 18 h; 2. aq. NaOH, *i*-PrOH, 20°C , 15 min (85% from **14**); 3. 2,4,6-trichlorobenzoyl chloride, pyridine, 20°C , 18 h (40%); (e) from **15** & **16**: cat. TsOH, MeOH, 50°C , 18 h; replace solvent with C_6H_6 , 20°C , 2 h; replace solvent with $(\text{MeO})_2\text{CH}_2$, acetone, MeOH, 20°C , 10 min [one-pot].

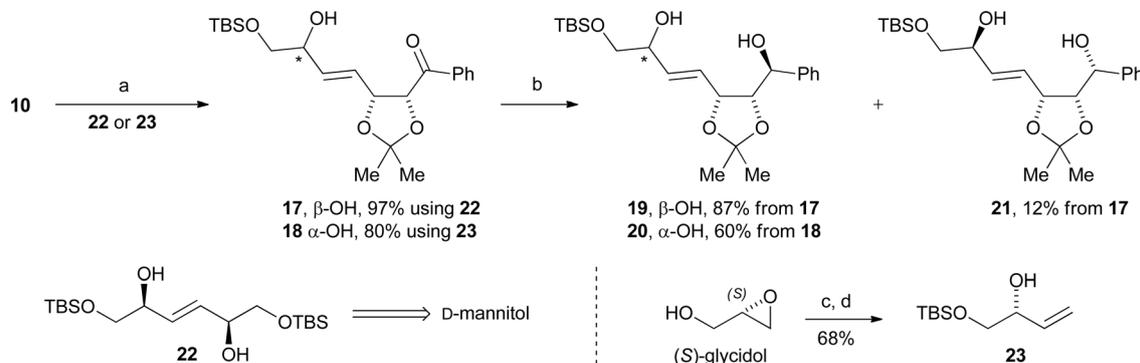
Our synthesis, at nine steps overall from D-ribonolactone, compares favourably with the shortest of those so far published. Interestingly, during the key step of this synthesis, the 2:1 diastereomeric ratio of alcohols **13** was carried through into the products; that is, the ratio of *E*:*Z*- α,β -unsaturated esters (**14**/**15**:**16**) was *ca.* 2:1, as was the ratio of products in which the enoate side chain is located *endo*- or *exo*- on the trioxabicyclo[3.3.0]octane ring (**14**/**16**:**15** respectively).

On this basis we wondered if these cyclisations are stereospecific to the extent that correct choice of monoallylic diol configuration would yield a single product diastereomer. This point had not arisen in Aponick's original publications nor in reported applications of the methodology.¹⁶ Very recently, Aponick showed that in the synthesis of tetrahydropyrans the configuration at the newly-formed allylic centre is determined by both the alkene and the allylic alcohol configurations; however, no results were reported concerning the formation of tetrahydrofurans.¹⁷ This information was not available to us at the time,⁹ and the allylic alcohol stereochemistry in isomers **13** was not assigned; therefore, we could not draw any conclusions on the stereospecificity based on the results presented in Scheme 3. Alkene **11** had not been reported as a single enantiomer so, in order to probe this further, we prepared the stereodefined epimeric ketones **17** and **18**

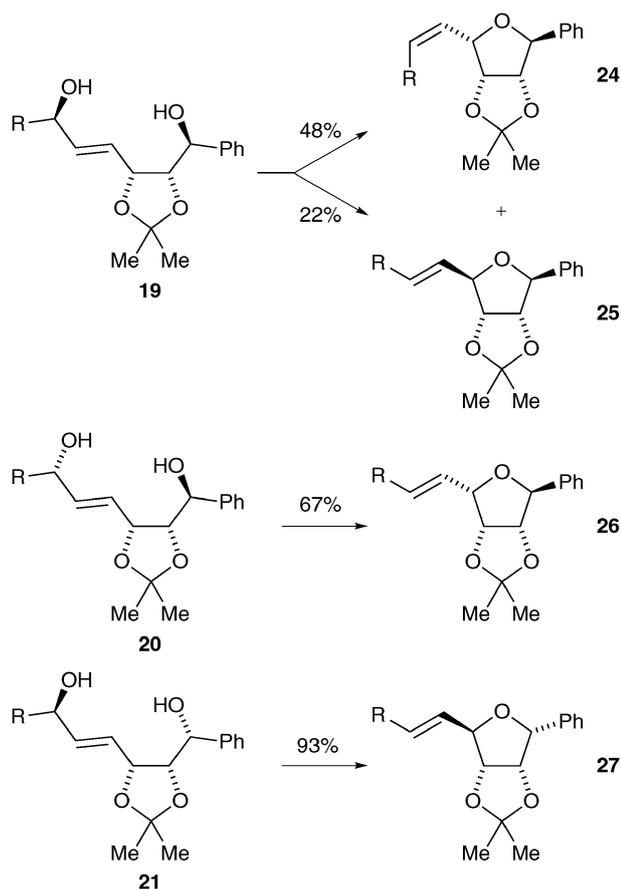
(Scheme 4) by cross metathesis.^{18,19} The L-Selectride reduction of ketone **17** gave a 7.2:1 mixture of diols **19** and **21**, epimeric at the benzylic alcohol; the corresponding reduction of ketone **18** gave a single diastereomer (**20**) but in lower yield.

The results from the cyclisation reactions are summarised in Scheme 5. Comparing first the benzylic (*S*)-alcohols **19** and **20**: the allylic (*S*)-alcohol (**19**) led to the *endo-Z*- product (**24**) and the *exo-E*- product (**25**) in *ca.* 2:1 ratio respectively whereas the allylic (*R*)-alcohol (**20**) gave the *endo-E*- product (**26**) exclusively. Diastereomer **27** was the only product that could be traced back to the minor benzylic (*R*)-alcohol **21**. With these results we can infer that in the (+)-isoalcoholone synthesis, tetrahydrofuran derivative **14** arose from the diol with the (*R*)-configured allylic alcohol ('**13R**') and diastereomers **15** and **16** arose from cyclisation of diol **13S** (which is therefore the major component in **13**).

In general, these results fit Aponick's recent stereochemical model¹⁷ but some key differences are evident that augment the emerging stereochemical picture. Specifically, (i) our results include the first examples of the formation of *Z*-olefins in this system; (ii) the stereochemical outcome does not depend exclusively on the alkene+allylic alcohol configuration; and (iii) the configuration at the incoming hydroxyl centre is important.



Scheme 4 Reagents and conditions. (a) cat. Hoveyda–Grubbs II, CH_2Cl_2 , 50°C , 1.5–3 h; (b) L-Selectride, THF, $-78^\circ\text{C} \rightarrow 0^\circ\text{C}$, 1 h; (c) TBSCl, imidazole, DMF, $0^\circ\text{C} \rightarrow 20^\circ\text{C}$, 2 h; (d) $\text{Me}_3\text{S}^+\text{I}^-$, BuLi, THF, $-10^\circ\text{C} \rightarrow 20^\circ\text{C}$, 2.5 h.



Scheme 5 Results for the Au-catalysed cyclisation of diols **19**–**21** (cat. Ph₃PAuCl, cat. AgOTf, 4 Å MS, CH₂Cl₂, 20 °C, 2 h). [R = CH₂OTBS].

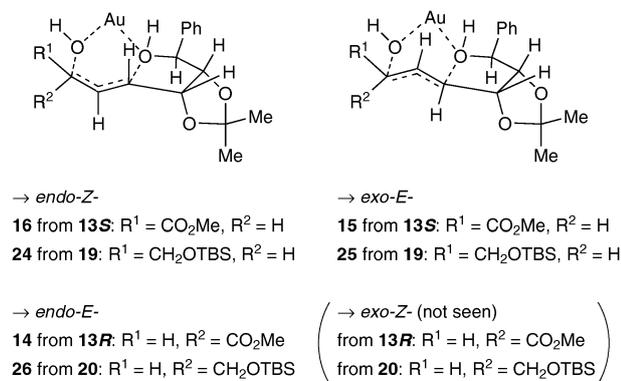


Fig. 1 A *syn*-S_N2' hydroxyl displacement model to relate substrate and product diastereomers.

Our model (Fig. 1) places the Au(I)-centre loosely co-ordinated to both hydroxyls, mediating an overall *syn*-S_N2' displacement. We stress that this is merely a rule-of-thumb mnemonic and a stereochemically equivalent *anti*-alkoxyauration/*anti*-elimination process has been advocated.^{17,20}

In conclusion, application of the Au(I)-mediated cycloetherification of monoallylic diols to the synthesis of (+)-isoaltholactone, and a follow-up study, have revealed further elements of stereocontrol in this reaction. In particular, secondary structural elements temper predictions based upon strict control by the allylic alcohol configuration.

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