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A Convergent Enantioselective Synthesis of the Tetrahydroisoquinoline Unit in the Spiro Ring of Ecteinascidin 743

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Summary: An efficient enantioselective synthesis of the spiro tetrahydroisoquinoline unit in ecteinascidin 743 is described, employing (+)-tetrahydrocarvone as a readily available and recoverable chiral auxiliary. The synthetic sequence involves a triply-convergent stereoselective bisannulation to construct the tetrahydroisoquinoline framework within 2. Copyright © 1996 Elsevier Science Ltd

The ecteinascidins make up a family of rare marine-derived alkaloids that exhibit unusually potent *in vivo* antitumor activity and are currently in preclinical development.¹ Structurally the members of this family, exemplified by ecteinascidin 743 (1), incorporate a piperazine-bridged bis(tetrahydroisoquinoline) framework similar to that of the saframycin class of antitumor antibiotics.² Unique to the ecteinascidins is the presence of a bridged ten-membered lactone that incorporates a benzylic sulfide linkage and, in the case of 1, a third spiro tetrahydroisoquinoline unit. In connection with our efforts directed toward the enantioselective synthesis of the ecteinascidins, the tetrahydroisoquinoline subunit 2, the bridged lactone intermediate 3 and the *N*-protected α -amino aldehyde 4 were envisioned to function as the key intermediates in the construction of the tris(tetrahydroisoquinoline) structure within 1. Herein is described a short, convergent and enantioselective synthesis of 2 employing tetrahydrocarvone as a readily available and recoverable chiral auxiliary.



The preparation of tetrahydroisoquinoline 2 commenced with 3-benzyloxy-4-methoxybenzaldehyde 5, which was subjected to nitroaldol condensation (CH₃NO₂, piperidine, AcOH) to afford nitrostyrene 6 (77%). Reduction of 5 with 3 equiv of LiAlH₄ in THF at 23 °C for 7 h yielded 2-(3-benzyloxy-4-methoxy)phenethylamine which was immediately condensed with 2 equiv of (-)-tetrahydrocarvone 7³ (C₇H₈, 55 °C, 3Å mol sieves, 2 h) to afford the Schiff base 8. The crude imine was treated with 2 equiv of methyl 3-mercaptopyruvate (9)⁴ in the presence of 2 equiv of acetic acid at 23 °C to promote diastereoselective axial thiol



addition to form the N,S-ketal 10. This adduct was directly subjected to cyclization using 2 equiv of methanesulfonic acid under dehydrating conditions (3Å mol sieves, CH₂Cl₂, 23 °C, 16 h) to effect initial iminium formation (*i.e.*, 11) and subsequent electrophilic substitution on the aromatic ring to yield tetrahydroisoquinoline 12 (54% from 6; 6.5:1 mixture of diastereomers, epimeric at C1; ratio invariant with time). It had been expected that the cyclization would generate 2 (with the desired C1-R configuration) by preferred approach of the aromatic moiety to the less hindered face of the iminium group in 11 (*i.e.*, trans to the adjacent equatorial methyl substituent). However, NOE data⁵ and single crystal X-ray analysis (Figure 1)⁶ of the cyclization product revealed that the major diastereomeric component was the C1-S diastereomer 12 in which the methyl ester is trans to the equatorial methyl group in the chiral auxiliary. Thus, it appears that the second ring closure in the transformation of $10 \rightarrow 12$ is an example of a Pictet-Spengler type cyclization that proceeds through a late (product resembling) transition state in which the iminium carbon in 11 is significantly pyramidalized.



Figure 1. X-ray Crystal Structure of 12.

Generation of the enantiomer of 12 thus became straightforward, and was accomplished using (+)-tetrahydrocarvone 13³ as the chiral auxiliary. Condensation of 13 with the intermediates 6 and 9 using the same reaction sequence outlined above⁷ gave the tetrahydroisoquinoline product 14 as the principal cyclized product in comparable yield and diastereoselectivity. Selective cleavage of the methyl ester function in 14 was accomplished by reaction with 4 equiv of sodium *n*-propylmercaptide in DMF at 45 °C for 3 h to yield the corresponding carboxylic acid. It is worthy to note that although 14 was formed as a chromatographically inseparable mixture of

C1-epimers, the desired C1-R diastereomer underwent ester deprotection faster than its C1-S counterpart, presumably a result of steric shielding of the methyl ester group by the equatorial methyl substituent on the tetrahydrocarvone ring system in C1-epi-14. Thus, quenching of the ester cleavage reaction at *ca.* 85%



conversion afforded the free carboxylic acid exclusively as the single desired diastereomer in 84% yield. The final steps in the sequence included acid hydrolysis (1 N HCl in MeOH, reflux, 3.5 h) followed by hexane extraction of the reaction mixture to isolate the chiral auxiliary **13** with quantitative recovery. The resulting tetrahydroisoquinoline carboxylic acid was directly treated with 4 equiv of di-*tert*-butylpyrocarbonate and DMAP (0.8 equiv) in pyridine resulting in N- and S- acylation to afford, after silica gel chromatography, acid **15** (98% yield), suitable for esterification with appropriate synthetic pentacyclic intermediates in the synthesis of **1**.

In summary, an efficient enantioselective synthesis of the spiro tetrahydroisoquinoline unit in ecteinascidin 743 is described, which employs (+)-tetrahydrocarvone as a chiral controller and which follows an unexpected and unusual stereochemical course. The key feature of the synthetic sequence involved a triply-convergent stereoselective bisannulation employing the intermediates 6, 9 and 13 to give, after deprotection, the selectively protected tetrahydroisoquinoline product 15. The strategy outlined in the synthesis of 15 may facilitate synthetic access not only to members of the ecteinascidin antitumor agents, but also to a host of related structures of potential biological utility. An enantioselective total synthesis of ecteinascidin 743 (1) has recently been completed.^{8,9}

References and Notes

- (a) Rinehart, K. L.; Shield, L. S. in *Topics in Pharmaceutical Sciences*, eds. Breimer, D. D.; Crommelin, D. J. A.; Midha, K. K. (Amsterdam Medical Press, Noordwijk, The Netherlands), 1989, pp. 613.
 (b) Rinehart, K. L.; Holt, T. G.; Fregeau, N. L.; Keifer, P. A.; Wilson, G. R.; Perun, T. J., Jr.; Sakai, R.; Thompson, A. G.; Stroh, J. G.; Shield, L. S.; Seigler, D. S.; Li, L. H.; Martin, D. G.; Grimmelikhuijzen, C. J. P.; Gäde, G. J. Nat. Prod. 1990, 53, 771. (c) Rinehart, K. L.; Sakai, R; Holt, T. G.; Fregeau, N. L.; Perun, T. J., Jr.; Seigler, D. S.; Wilson, G. R.; Shield, L. S. Pure Appl. Chem. 1990, 62, 1277. (d) Rinehart, K. L.; Holt, T. G.; Fregeau, N. L.; Stroh, J. G.; Keifer, P. A.; Sun, F.; Li, L. H.; Martin, D. G. J. Org. Chem. 1990, 55, 4512. (e) Wright, A. E.; Forleo, D. A.; Gunawardana, G. P.; Gunasekera, S. P.; Koehn, F. E.; McConnell, O. J. J. Org. Chem. 1990, 55, 4508. (f) Sakai, R.; Rinehart, K. L.; Guan, Y.; Wang, H.-J. Proc. Natl. Acad. Sci. USA 1992, 89, 11456.
- For previous work on the synthesis of the saframycins see: (a) Fukuyama, T.; Sachleben, R. A. J. Am. Chem. Soc. 1982, 104, 4957. (b) Fukuyama, T.; Yang, L.; Ajeck, K. L.; Sachleben, R. A. J. Am.

Chem. Soc. 1990, 112, 3712. (c) Saito, N.; Yamauchi, R.; Nishioka, H.; Ida, S.; Kubo, A. J. Org. Chem. 1989, 54, 5391.

- 3. The (-)- and (+)- enantiomers of tetrahydrocarvone (12 and 13) were prepared from (+)- and (-)-carvone, respectively, by a two-step reduction procedure followed by epimerization of the α-Me group (3 atm H₂, cat. Rh(PPh₃)₃Cl, C₆H₆, 23 °C, 23 h; 1 atm H₂, MeOH, cat. 10% Pd-C, 23 °C, 2.5 h; NaOMe, MeOH, 23 °C, 18 h) to minimize double bond transposition and concomitant racemization. Rotations: [α]_D²³ for 12 = -15.5 and for 13 = +15.5 (c=3, EtOH).
- 4. The method of preparation of 9 was analogous to that described for ethyl 3-mercaptopyruvate: Taylor, E. C.; Reiter, L. A. J. Am. Chem. Soc. 1989, 111, 285.
- 5. NOE experiments were performed on the LiAlH₄ reduction product of 12:



- 6. Crystal structure solved by Dr. Axel Fischer.
- 7. Preparation of 14: To a solution of 6 (173 mg, 0.608 mmol, 1 equiv) in THF (6 mL), at -78 °C was added solid LiAlH4 (69 mg, 1.82 mmol, 3.0 equiv), and the resulting suspension was stirred at 23 °C for 7 h before the excess reducing agent was quenched with the sequential dropwise addition of water (69 µL), 3 N aqueous sodium hydroxide solution (69 µL), and water (210 µL) at 0 °C. The suspension was diluted with EtOAc (20 mL), and the solids were removed by filtration. The filtrate was concentrated, and a suspension of the crude amine, 13 (188 mg, 1.22 mmol, 2.0 equiv), and crushed activated 4 Å molecular sieves (~550 mg) in toluene (2.5 mL) was stirred at 55 °C for 2 h. The reaction mixture was allowed to cool to 23 °C prior to the sequential addition of solid 9 (163 mg, 1.22 mmol, 2.0 equiv) and acetic acid (35 µL, 0.608 mmol, 1.0 equiv). The mixture was stirred at 23 °C for 2 h, then was diluted with EtOAc (50 mL). The mixture was washed sequentially with saturated aqueous NaHCO3 solution (50 mL) and saturated aqueous NaCl solution (50 mL), and then was dried (Na2SO4) and concentrated. The residue was dissolved in CH₂Cl₂ (30 mL), and to this solution was added crushed activated 3 Å molecular sieves (~1 g) and CH₃SO₃H (79 µL, 1.22 mmol, 2.0 equiv). The mixture was stirred at 23 °C for 16 h and then filtered. The filtrate was washed with saturated aqueous NaHCO3 solution (30 mL) and saturated aqueous NaCl solution (30 mL), and then was dried (Na₂SO₄) and concentrated. The residue was purified by filtration through a plug of silica gel (10% EtOAc in CH2Cl2 eluent) followed by flash column chromatography (10% EtOAc in hexanes) to give 14 (167 mg, 54% 4 steps) as a 6.5:1 mixture of diastereomers at C1. Rf 0.18 (10% ethyl acetate in hexanes); $[\alpha]_D^{23} = +116$ (c = 0.30, CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 7.46 (m, 2H), 7.40 (m, 2H), 7.33 (m, 1H), 6.90 (s, 1H), 6.66 (s, 1H), 5.12 (s, 2H), 3.86 (s, 3H), 3.77 (d, 1H, J = 9.7 Hz), 3.77 (m, 1H), 3.63 (s, 3H), 3.10 (dd, 1H, J = 6.8, 12.1 Hz), 2.87 (m, 1H), 2.64 (d, 1H, J = 9.7 Hz), 2.61 (m, 1H), 1.93 (m, 1H), 1.72 (m, 2H), 1.61 (m, 2H), 1.51 (m, 1H), 1.43 (m, 1H), 1.24 (m, 2H), 0.85 (m, 9H); FTIR (neat film) 2955 (s), 2930 (s), 1724 (s), 1511 (s), 1463 (m), 1446 (m), 1368 (m), 1259 (s), 1241 (s), 1214 (s), 1164 (s), 1018 (m), 1001 (m) (cm⁻¹); HRMS (CI⁺) m/z: Calcd for C₃₀H₄₀NO₄S (MH⁺) 510.2678, found 510.2700.
- 8. Corey, E. J.; Gin, D. Y.; Kania, R. S. J. Am. Chem. Soc. submitted.
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