Electrophilic Cyclization of Buta-1,3-diynylarenes: Synthesis of Precursors of (Z)-3-Ene-1,5-diyne Systems Fused to Heterocycles

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Abstract: A simple, convenient, and promising strategy for the synthesis of 2-ethynyl-3-iodo-benzothiophenes, -benzofurans, and -indoles based on electrophilic cyclization of easily available *ortho*-functionalized (buta-1,3-diynyl)arenes was developed. The unique potential of using these compounds as starting materials for the synthesis of enediyne systems, containing thiophene, furan, and pyrrole units is demonstrated.

Key words: electrophilic cyclization, alkynes, heterocycles, enediynes

Electrophilic cyclization of ethynylarenes is well known as an efficient method for the synthesis of fused halogenated heterocycles,¹ including 3-iodobenzothiophenes,^{1d,2} -benzofurans,^{1b,c,i,3} and -indoles.^{1e,4} These compounds are of great interest for the synthesis of a wide variety of heterocyclic compounds with special biological activity⁵ and physical properties.⁶

While the cyclization of *ortho*-substituted ethynylarenes was investigated in details, there are only a few examples of iodocyclization of buta-1,3-diynyl derivatives. Thus, symmetrically^{1b,7} and asymmetrically^{1b} substituted 1,4-diarylbutadiynes underwent a double cyclization to give symmetrical and asymmetrical bisheterocyclic compounds. The formation of 4-iodo-3-ethynylisocoumarine was also detected as an intermediate of double cyclization.^{1b}

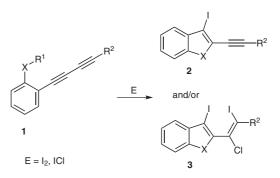
Recently, we demonstrated the Richter-type cyclization of *ortho*-(alka-1,3-diynyl)arenediazonium salts to be a short route to 3-alkynyl-4-halocinnolines.⁸ Aiming at the development of a general and efficient approach towards haloethynylheterocycles, as direct precursors of enediyne systems, fused to a heterocyclic core, we reasoned that such compounds may be accessed via electrophilic cyclization of *ortho*-functionalized buta-1,3-diynyarenes. It should be noted that there is no general method to construct (Z)-3-ene-1,5-diyne system attached to a variety of heterocycles. Reported syntheses of such systems usually include transformation of functional groups to the ethynyl moieties⁹ or double cross-coupling of terminal acetylenes

SYNLETT 2011, No. 4, pp 0517–0520 Advanced online publication: 11.02.2011 DOI: 10.1055/s-0030-1259547; Art ID: G33910ST © Georg Thieme Verlag Stuttgart · New York with dihalosubstituted heteroarenes,¹⁰ which are not widely accessible. Meanwhile, heterocycles possessing an enediyne system are promising objects for the medicinal chemistry as analogues of naturally occurred enediyne antibiotics, which display strong antineoplastic activity.¹¹

Herein we report for the first time the electrophilic cyclization of *ortho*-buta-1,3-diynylthiophenol, -aniline, and -phenol derivatives, which led to the formation of benzannulated 2-ethynyl-3-iodo-*S*,*N*,*O*-five-membered heterocycles. Despite the fact that, Pd-, Cu-, and base-catalyzed cyclizations of buta-1,3-diynylaniline and buta-1,3-diynylphenol derivatives are convenient methods for the synthesis of 2-ethynylindoles and 2-ethynylbenzofurans,¹² no data has been reported about application of the electrophilic cyclization for the producing of this type of heterocycles, having both ethynyl and halogen moieties.

First of all, a series of *ortho*-buta-1,3-diynylthiophenol **1a–d**, -phenol **1g–i**, and *N*,*N*-dimethylaniline derivatives **1e**,**f** were obtained as the starting materials. Compounds **1a–d**,**g–i** were prepared from corresponding *ortho*-iodo-thiophenol or -phenol derivatives and buta-1,3-diynes by Sonogashira coupling^{13a} using procedure developed in our group recently.^{13b} *N*,*N*-Dimethylaniline derivatives **1e**,**f** were synthesized by methylation of corresponding *ortho*-buta-1,3-diynylaniline derivatives^{13b,c} with MeI.

It was found that *ortho*-buta-1,3-diynylthioanisole derivatives $1\mathbf{a}-\mathbf{c}$ can be efficiently converted into 2-ethynyl-3iodobenzothiophenes $2\mathbf{a}-\mathbf{c}$ by using of iodine in acetonitrile at room temperature (Scheme 1, Table 1, entries 1– 3).



Scheme 1 Electrophilic cyclization of *ortho*-buta-1,3-diynylthiophenol (X = S), -aniline (X = NMe), and -phenol (X = O) derivatives. For R^1 , R^2 , see Table 1.

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It should be noted that cyclization of trimethylsilyl derivative 1d under the same conditions gave the mixture of silylated and desilylated products 2d and 2d' in a molar ratio of 4.5:1 (entry 4).¹⁴ Unfortunately, complete separation of this mixture by column chromatography could not be achieved due to the close retention times of compounds 2d and 2d'. However, it was possible to isolate the silvlated compound 2d in 30% yield. When the reaction was carried out in dry dichloromethane (r.t., 3 h) without treatment of the reaction mixture with aqueous Na₂S₂O₃ solution at the end, a mixture of 2d and 2d' with the molar ratio 23:1¹⁴ was obtained (entry 5). Treatment of this mixture with K₂CO₃ in MeOH (r.t., 5 h) gave 2-ethynyl-3-iodobenzothiophene 2d' in 50% yield. It should be noted, that this compound is extremely interesting as a substrate for the synthesis of cyclododeca-1,5,9-triene-3,7,11triyne derivatives.¹⁵

Cyclization of aniline derivatives **1e**,**f** did not proceed in acetonitrile at room temperature and required a higher temperature due to the lower nucleophilicity of dimethylamino group compared to methylthio group. Thus, cyclization of *N*,*N*-dimethylamino-*ortho*-buta-1,3-diynyl-anilines **1e**,**f** proceeded smoothly at 40 °C (entries 6 and 7) to give 2-ethynyl-3-iodoindole derivatives **2e**,**f** in good yields.

With regard to the synthesis of 2-ethynyl-3-iodobenzofuran derivatives, we showed that iodocyclization of *ortho*-(4-phenylbuta-1,3-diynyl)anisole (**1g**) with 1 equivalent of iodine in acetonitrile at 40 °C gave the desired product **2g** in a poor yield (entry 8). Treatment of anisole **1g** with a more active electrophilic agent (iodine monochloride, 2 equiv) in dichloromethane led to the formation of byproduct **3g** with an iodochlorinated triple bond (entry 9). In the case of an equimolar ratio of reagents a mixture of **2g** and **3g** was detected by TLC and GC-MS monitoring. O-Benzylated *ortho*-dodeca-1,3-diynylphenole **1h** with iodine did not undergo cyclization even at 70 °C in acetonitrile at all, which is in accordance with the known leaving-group activities in electrophilic cyclization-type reactions.^{1h} Nevertheless, we found that the use of iodine monochloride in acetonitrile is favorable for selective cyclization of O-benzylated derivative **1h**. In this case 2-ethynylbenzofurane **2g** was isolated in 49% yield (entry 10).

However, when we attempted to use iodine monochloride in acetonitrile for the cyclization of *ortho*-dodeca-1,3-diynylanisole (**1i**), the iodochlorination of the triple bond in the 3-iodo-2-decynylbenzofurane proceeded smoothly to yield the benzofurane **3i** with iodochlorinated triple bond (entry 11).¹⁶

Thus we demonstrated that benzannulated five-membered-*S*,*N*,*O*-heterocycles, possessing both ethynyl moiety and iodine atom, can be directly accessed via electrophilic cyclization of the corresponding functionalized *ortho*-buta-1,3-diynylarenes. The ability of *ortho*buta-1,3-diynylthiophenol, -aniline, and phenol derivatives to undergo electrophilic cyclization is in accordance with the relative nucleophilicity of the functional groups and decreases among S > N > O.

Entry	Substrate				Conditions ^a	Product, yield (%) ^b		
	1	Х	\mathbb{R}^1	\mathbb{R}^2		\mathbb{R}^2	2 ¹⁹	3
1	1a	S	Me	Ph	А	Ph	2a 83	_
2	1b	S	Me	C ₈ H ₁₇	А	$C_{8}H_{17}$	2b 84	-
3	1c	S	Me	(CH ₂) ₂ OH	А	(CH ₂) ₂ OH	$2c^{20} 67$	-
4	1d	S	Me	TMS	А	TMS H	2d 61 2d' 13	_
5	1d	S	Me	TMS	В	TMS H	2d 66 2d' 4	-
6	1e	NMe	Me	Ph	С	Ph	2e 75	-
7	1f	NMe	Me	C ₈ H ₁₇	С	C ₈ H ₁₇	2f 88	_
8	1g	0	Me	Ph	D	Ph	2g 16	_
9	1g	0	Me	Ph	Е	Ph	2g traces	3g 59°
10	1h	0	Bn	Ph	F	Ph	2g 49	3g traces
11	1i	0	Me	C ₈ H ₁₇	F	C ₈ H ₁₇	2i traces	3i 38°

Table 1Synthesis of Compounds 2 and 3

^a Conditions A: I₂ (1 equiv), MeCN, r.t., 3 h. Conditions B: I₂ (1 equiv), dry CH_2Cl_2 , r.t., 3 h. Conditions C: I₂ (1 equiv), MeCN, 40 °C, 5 h. Conditions D: I₂ (1 equiv), MeCN, 40 °C, overnight. Conditions E: ICl (1 equiv), CH_2Cl_2 , r.t., overnight, then ICl (1 equiv), r.t., 1 h. F: ICl (1 equiv), MeCN, r.t., overnight, then ICl (0.5 equiv), r.t., 3 h.

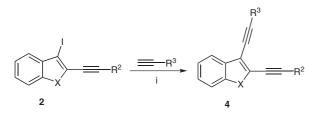
^b All yields are given for individual compounds after purification by column chromatography; yields of **2d** and **2d'** (entries 4, 5) were calculated in accordance to the molar ratio of **2d** and **2d'** (see ref. 14), which was determined from ¹H NMR spectra of mixtures.

^c Position of iodine and chlorine atoms was determined according to HMBC experiment.

Encouraged by the successful development of iodocyclization toward 3-ethynyl-2-iodoheterocycles we turned our attention to the synthesis of enediyne systems. There are a few reported examples of enediynes fused to thiophene,^{1b,9,17} pyrrole,¹⁸ and furan rings,⁹ and only few known compounds with different substituents attached to the triple bonds^{18a,b} because of the lack of methods for their synthesis.

So, the approach towards benzannulated 2-ethynyl-3iodo-S,N,O-five-membered heterocycles described above should open a simple way for the synthesis of a variety of enediynes systems containing heterocyclic rings and different substituents attached to the triple bonds.

To explore this opportunity we synthesized several enediyne systems using the Sonogashira coupling protocol in the presence of $Pd(PPh_3)_4$ as a catalyst and diisopropanolamine (DIPA) as a base. All enediynes **4a**–i were isolated in moderate to good yields without optimization of the reaction conditions (Scheme 2, Table 2).



Scheme 2 Synthesis of benzothiophene-, indole-, and benzofurancontaining enediyne systems. *Reagents and conditions*: (i) $Pd(PPh_3)_4$, CuI, Ph₃P, DMF, DIPA, 50 °C, overnight.

Table 2Synthesis of Enediynes 4

Entry	Subs	trate		Acetylene	Product	Yield (%)
	2	Х	\mathbb{R}^2	R ³	4 ²¹	
1	2a	S	Ph	C ₆ H ₁₃	4a	71
2	2a	S	Ph	(CH ₂) ₃ OH	4b	72
3	2a	S	Ph	TMS	4c	83
4	2b	S	C_8H_{17}	Ph	4d	65
5	2b	S	C_8H_{17}	4-MeOC ₆ H ₄	4e	59
6	2c	S	(CH ₂) ₂ OH	TMS	$4f^{22}$	63
7	2f	NMe	$C_8 H_{17}$	Ph	4 g	76 ^a
8	2g	0	Ph	C ₆ H ₁₃	4h	47
9	2g	0	Ph	TMS	4 i	69

^a Reaction was carried out at 40 °C.

In summary, for the first time the electrophilic cyclization of *ortho*-buta-1,3-diynylthiophenol, -aniline, and -phenol derivatives was accomplished to give benzannulated 3iodo-2-ethynylheterocycles. This is a promising and direct approach towards heterocyclic compounds possessing iodine and ethynyl moiety at neighboring atoms, which are precursors of various enediyne systems fused to heterocyclic core.

The three-step methodology described herein enables the construction of asymmetrically substituted enediyne systems with absolute regiocontrol. We also demonstrated that these substituents may contain such functional groups as OH and TMS, which is very important for further enediyne modifications.

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References and Notes

- (1) (a) Larock, R. C. In Acetylene Chemistry; Diederich, F.; Stang, P. J.; Tykwinski, R. R., Eds.; Wiley-VCH: Weinheim, 2005, 51–99. (b) Mehta, S.; Larock, R. C. J. Org. Chem. 2010, 75, 1652; and references cited therein.
 (c) Manarin, F.; Roehrs, J. A.; Brandão, R.; Nogueira, C. W.; Zeni, G. Synthesis 2009, 4001. (d) Cho, C.-H.; Neuenswander, B.; Larock, R. C. J. Comb. Chem. 2010, 12, 278. (e) Worlikar, S. A.; Neuenswander, B.; Lushington, G. H.; Larock, R. C. J. Comb. Chem. 2019, 11, 875.
 (f) Cho, C.-H.; Neuenswander, B.; Lushington, G. H.; Larock, R. C. J. Comb. Chem. 2009, 11, 875.
 (f) Cho, C.-H.; Neuenswander, B.; Lushington, G. H.; Larock, R. C. J. Comb. Chem. 2009, 11, 900. (g) Manarin, F.; Roehrs, J. A.; Gay, R. M.; Brandao, R.; Menezes, P. H.; Nogueira, C. W.; Zeni, G. J. Org. Chem. 2009, 74, 2153. (h) Mehta, S.; Waldo, J. P.; Larock, R. C. L. Org. Chem. 2009, 74
 - S.; Waldo, J. P.; Larock, R. C. J. Org. Chem. **2009**, 74, 1141. (i) Cho, C.-H.; Neuenswander, B.; Lushington, G. H.; Larock, R. C. J. Comb. Chem. **2008**, 10, 941.
- (2) (a) Yue, D.; Larock, R. C. J. Org. Chem. 2002, 67, 1905.
 (b) Flynn, B. L.; Verdier-Pinard, P.; Hamel, E. Org. Lett. 2001, 3, 651. (c) Yue, D.; Larock, R. C. Tetrahedron Lett. 2001, 42, 6011.
- (3) (a) Arcadi, A.; Cacchi, S.; Fabrizi, G.; Marinelli, F.; Moro, L. Synlett **1999**, 1432. (b) Yue, D.; Yao, T.; Larock, R. C. J. Org. Chem. **2005**, 70, 10292. (c) Okitsu, T.; Nakazawa, D.; Taniguchi, R.; Wada, A. Org. Lett. **2008**, 10, 4967.
- (4) (a) Barluenga, J.; Trincado, M.; Rubio, E.; Gonzalez, J. M. *Angew. Chem. Int. Ed.* 2003, *42*, 2406. (b) Amjad, M.; Knight, D. W. *Tetrahedron Lett.* 2004, *45*, 539. (c) Yue, D.; Larock, R. C. *Org. Lett.* 2004, *6*, 1037. (d) Yue, D.; Yao, T.; Larock, R. C. *J. Org. Chem.* 2006, *71*, 62.
- (5) (a) Wei, S.; Coleman, R. S.; Lowary, T. L. Org. Biomol. Chem. 2009, 7, 3709. (b) Santin, E. P.; Khanwalkar, H.; Voegel, J.; Collette, P.; Mauvais, P.; Gronemeye, H.; de Lera, A. R. ChemMedChem 2009, 4, 780. (c) Yang, L.-Y.; Chang, C.-F.; Huang, Y.-C.; Lee, Y.-J.; Hu, C.-C.; Tseng, T.-H. Synthesis 2009, 1175. (d) Bang, H. B.; Han, S. Y.; Choi, D. H.; Yang, D. M.; Hwang, J. W.; Lee, H. S.; Jun, J.-G. Synth. Commun. 2009, 39, 506.
- (6) (a) Zhou, Y.; Liu, W.-J.; Ma, Y.; Wang, H.; Qi, L.; Cao, Y.; Wang, J.; Pei, J. J. Am. Chem. Soc. 2007, 129, 12386.
 (b) Kobayashi, T.; Hasegawa, Y. US 2005258398, 2005.
- (7) Barluenga, J.; Trincado, M.; Rubio, E.; Gonzalez, J. M. *Angew. Chem. Int. Ed.* **2003**, *42*, 2406.
- (8) (a) Vinogradova, O. V.; Sorokoumov, V. N.; Vasylevsky, S. F.; Balova, I. A. *Tetrahedron Lett.* **2007**, *48*, 4907.
 (b) Vinogradova, O. V.; Sorokoumov, V. N.; Balova, I. A. *Tetrahedron Lett.* **2009**, *50*, 6358.

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- (9) (a) Sahu, B.; Namboothiri, I. N. N.; Persky, R. *Tetrahedron Lett.* 2005, *46*, 2593. (b) Sahu, B.; Muruganantham, R.; Namboothiri, I. N. N. *Eur. J. Org. Chem.* 2007, 2477.
- (10) (a) Kim, C.-S.; Russell, K. C. J. Org. Chem. 1998, 63, 8229.
 (b) Kumarasinghe, E. S.; Peterson, M. A.; Robins, M. J. *Tetrahedron Lett.* 2000, 41, 8741.
- (11) For recent reviews, see: (a) Jones, G. B.; Fouad, F. S. *Curr. Pharm. Des.* **2002**, *8*, 2415. (b) Maretina, I. A.; Trofimov, B. A. *Russ. Chem. Rev.* **2006**, *75*, 825. (c) Shao, R.-G. *Curr. Mol. Pharm.* **2008**, *1*, 50.
- (12) (a) Fiandanese, V.; Bottalico, D.; Marchese, G.; Punzi, A. *Tetrahedron* 2008, 64, 53. (b) Fiandanese, V.; Bottalico, D.; Marchese, G.; Punzi, A. *Tetrahedron* 2008, 64, 7301.
- (13) (a) Sonogashira, K.; Tohda, Y.; Hagihara, N. *Tetrahedron Lett.* **1975**, *16*, 4467. (b) Balova, I. A.; Sorokoumov, V. N.; Morozkina, S. N.; Vinogradova, O. V.; Knight, D. W.; Vasilevsky, S. F. *Eur. J. Org. Chem.* **2005**, 882. (c) Liang, Y.; Tao, L.-M.; Zhang, Y.-H.; Li, J.-H. *Synthesis* **2008**, 3988.
- (14) Relative molar ratio of 2d and 2d' was determined by ¹H NMR analysis of the integral intensity of the signals of acetylene hydrogen atom ($\delta = 3.74$ ppm) of 2d', TMS group ($\delta = 0.32$ ppm) of 2d and multiplet signals of aromatic hydrogen atoms ($\delta = 7.38-7.50$ and 7.67–7.76 ppm) of 2d and 2d' (CDCl₃, 400 MHz, TMS as reference).
- (15) (a) Sooloki, D.; Kennedy, V. O.; Tessier, C. A.; Youngs, W. J. *Synlett* **1990**, 427. (b) Sooloki, D.; Bradshaw, J. D.; Tessier, C. A.; Youngs, W. J. *Organometallics* **1994**, *13*, 451.
- (16) It was determined by TLC that O-benzyl-2-(dodeca-1,3diynyl)phenol reacts with ICl in MeCN in similar way giving benzofuran **3i** with iodochlorinated triple bond.
- (17) Ebata, H.; Miyazaki, E.; Yamamoto, T.; Takimiya, K. Org. Lett. 2007, 9, 4499.
- (18) (a) Barluenga, J.; Fananas-Mastral, M.; Andina, F.; Aznar, F.; Valdes, C. *Organometallics* 2008, *27*, 3593.
 (b) Kitagaki, S.; Katoh, K.; Ohdachi, K.; Takahashi, Y.; Shibata, D.; Mukai, C. *J. Org. Chem.* 2006, *71*, 6908.
 (c) Shvartsberg, M. S.; Vasilevskii, S. F.; Prihod'ko, T. A. *Izv. Acad. Nauk SSSR, Ser. Khim.* 1982, *11*, 2524.
- (19) General Procedure for Electrophilic Cyclization of ortho-Buta-1,3-diynylthiophenol, -Aniline, and -Phenol Derivatives Using I₂

To an Ar flushed solution of corresponding o-(buta-1,3diynyl)arene **1** (0.2 mmol) in MeCN (3 mL), a solution of iodine (0.2 mmol, 0.051 g,) in MeCN (2 mL) was added dropwise. The reaction mixture was stirred at corresponding temperature up to disappearance of starting material according to TLC monitoring (see Table 1). Then, the reaction mixture was diluted with 5% aq solution of Na₂S₂O₃ and extracted with CH_2Cl_2 (3 × 7 mL). The combined organic layers were washed with H_2O , dried over anhyd Na_2SO_4 , and concentrated under reduced pressure to yield the crude product, which was purified by column chromatography on silica gel using pentane (**for 2a,b,d,e,g**) or cyclohexane–EtOAc (**2c,f**) as the eluent.

(20) Selected Data for 2c

Mp 73–75 °C. IR (neat): v = 3124 (OH), 2927 (CH), 2222 (C=C), 1449, 1427, 1370, 1326, 1294, 1243, 1192, 1158, 1062, 1031, 1016, 983, 938, 914, 849, 831, 747, 721, 707 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 2.01 (t, *J* = 6.5 Hz, 1 H), 2.83 (t, *J* = 6.1 Hz, 2 H), 3.88–3.93 (m, 2 H), 7.38–7.46 (m, 2 H), 7.67–7.72 (m, 2 H). ¹³C NMR (100 MHz, CDCl₃): δ = 24.4, 60.8, 87.6, 96.9, 122.1, 125.0, 125.7, 126.1, 126.4, 138.6, 140.3 (two signals are overlapping with each other). MS (EI, 70 eV): *m/z* (%) = 328.0(100) [M]⁺, 296.9 (66) [M – CH₂OH]⁺, 171.0(31), 139.1 (5) 126.1 (14). HRMS: *m/z* calcd for C₁₂H₉OSI: 327.9419. Found: 327.9418. Anal. Calcd for C₁₂H₉OSI: C, 43.92; H, 2.76; S, 9.77. Found: C, 43.85; H, 2.69; S, 9.61.

(21) General Procedure for the Synthesis of Enediyne Systems 4

To a stirred solution of 2-ethynyl-3-iodo heterocyle **2** (0.1 mmol) in DMF (2 mL), the alkyne (0.2 mmol), Pd(PPh₃)₄ (5 mol%), Ph₃P (10 mol%), and DIPA (0.4 mmol) were added. The reaction vial was evacuated and flushed with Ar several times. After that 15 mol% of CuI was added, the reaction vial was then sealed and flushed with Ar. The reaction mixture was allowed to stir at 40–50 °C (see Table 2) overnight. After cooling, the reaction mixture was poured into the sat. aq solution of NH₄Cl and extracted with CH₂Cl₂ (3 × 10 mL). The combined organic layers were washed two times with H₂O, dried over anhyd Na₂SO₄, and concentrated under reduced pressure to yield the crude product, which was purified by column chromatography on silica gel using pentane (for **4a,c,d,h,i**) or cyclohexane–EtOAc (for **4b,e–g**) as the eluent.

(22) Selected Data for 4f

IR (film on KBr): v = 3358 (OH), 3061 (CH), 2958 (CH), 2897 (CH), 2223 (C=C), 2149 (C=C), 1458, 1433, 1349, 1317, 1249, 1216, 1160, 1081, 1046, 976, 938, 896, 843, 729, 687, 642 cm⁻¹. ¹H NMR (400 MHz, CDCl₃): δ = 0.32 (s, 9 H), 2.02 (t, *J* = 6.7 Hz, 1 H), 2.83 (t, *J* = 6.1 Hz, 2 H), 3.85–3.90 (m, 2 H), 7.38–7.45 (m, 2 H), 7.70–7.72 (m, 1 H), 7.83–7.85 (m, 1 H). ¹³C NMR (100 MHz, CDCl₃): δ = 0.1, 24.5, 60.8, 75.8, 97.5, 97.7, 101.5, 122.0, 122.8, 123.3, 125.1, 126.1, 127.3, 138.0, 138.5. MS–FAB: *m*/*z* (%) = 299.1 (61) [M⁺ + H], 298.1 (100) [M⁺]. HRMS: *m*/*z* calcd for C₁₇H₁₈OSSi: 298.0848; found: 298.0845.

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