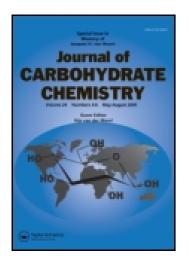
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IMPROVEMENT ON LIPASE CATALYSED REGIOSELECTIVE O-ACYLATION OF LACTOSE:A CONVENIENT RUOTE TO 2'-O- FUCOSYLLACTOSE

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COMMUNICATION

IMPROVEMENT ON LIPASE CATALYSED REGIOSELECTIVE O-ACYLATION OF LACTOSE: A CONVENIENT ROUTE TO 2'-O-FUCOSYLLACTOSE¹

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Although enormous progress has been made in the past ten years, the synthesis of complex carbohydrates still remains a severe problem. Conventional syntheses of these molecules often dictate multiple protections and deprotections of various hydroxyl groups, dramatically increasing the number of steps required. On the other hand, enzymes offer the opportunity to carry out cheap, highly chemo and regioselective transformations, providing new approaches to tackle many synthetic problems encountered in carbohydrate synthesis. In particular, lipases allow highly regioselective *O*-acylations. Therefore, a growing interest in enzymatic manipulation of protecting groups for the synthesis of oligosaccharides has been developed during recent years.²

Recently, we reported on the use of lipase catalysed acylation for designing useful building blocks in oligosaccharide synthesis.^{3,4} Excellent selectivity for 6'-*O*-acylation was observed on various disaccharidic substrates (such as lactose, maltose and cellobiose) using different acylating agents,³ *Candida antarctica* lipase being the best biocatalyst. However, our procedure contained a main drawback: the high polarity of the sugar substrates required the use of *tert*-amyl alcohol as a solvent, whose high boiling point made the work up of the reaction mixture quite laborious, thus rendering the method impractical. An important improvement

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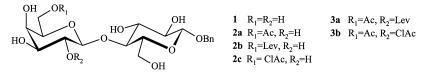
Entry	Substrate	Acylating Agent	Conditions	Product	Yield (%)
1	1	Vinyl acetate	THF, 40°C, 3 days	2a	73
2	1	Vinyl acetate	CH ₃ CN, 40°C, 4 days	2a	64
3	1	Trifluoroethyl levulinate	THF, 40°C, 4 days	2b	83
4	1	Vinyl chloroacetate	THF, 40°C, 2 days	2c	78
5	2a	Trifluoroethyl levulinate	CH ₃ CN, 45°C, 4 days	3 a	65
6	2a	Vinyl chloroacetate	CH_3CN , 45°C, 5 days	3b	62

Table 1. Experimental Conditions and Yields of Enzymatic Acylation Reactions

was achieved when we observed that when using relatively low boiling solvents such as THF or CH_3CN , the enzymatic acylations proceeded with equal efficiency, despite the poor solubility of the substrates. For example, employing THF as a solvent, we could introduce a chloroacetyl group at the 6'-OH of lactose with very high regioselectivity and chemical yield (78%) and with a much easier product recovery.⁴ This finding prompted us to explore in more detail the behaviour of the *Candida antarctica* lipase in different solvents and on various biologically relevant substrates.⁵ In the present communication we show further achievements obtained in the functionalization of lactose and a new straightforward synthesis of 2'-O-fucosyllactose **7**.⁶

In a preliminary screening of different organic solvents, THF and CH₃CN gave the best results. In THF, 6'-O-acylated lactosides were obtained regioselectively from 1^7 at 40 °C in 73–83 % yields from different acylating agents with *Candida antarctica* lipase⁸ (Table 1). The same reaction performed in CH₃CN (Table 1, entry 2) showed less selectivity, giving, besides the 6'-O-monoacylated derivative as the main product, fair amounts (16%) of diacylated disaccharides. However, when compound **2a** (Scheme 1) was submitted to a second acylation with lipase from *Candida antarctica* in CH₃CN at 45 °C, an excellent regioselectivity for the 2'-OH was observed with both the acylating agents employed (Table 1, entries 5 and 6). All the reactions were very clean, without formation of by-products, allowing full recovery of the unreacted substrate. Thus, a double sequential acylation, changing the solvent and using orthogonal protecting groups, provides an easy access to a family of lactose building blocks.

We took advantage from our chemo-enzymatic approach to design a new convenient route to the synthesis of 2'-O-fucosyllactose 7, which, after lactose, was found to be the most abundant component of the human milk oligosaccharides



Scheme 1.

762



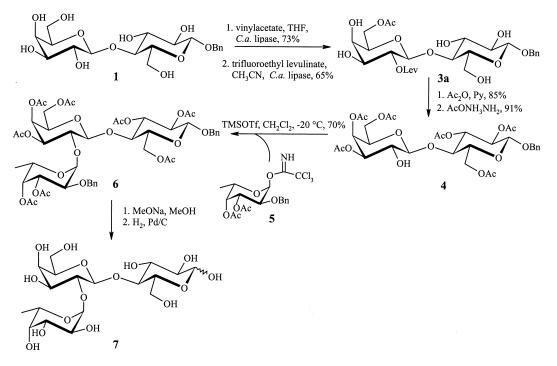


CONVENIENT ROUTE TO 2'-O-FUCOSYLLACTOSE

in 77 % of Caucasian women classified as secretors.⁹ It seems reasonable that 2'-*O*-fucosyllactose may play an important role in determining the antiadhesive properties of human milk oligosaccharides, which have been demonstrated to represent a protection against infections for breast-fed infants during the lactation period.¹⁰

To the best of our knowledge, only three chemical syntheses of 2'-O-fucosyllactose have been reported so far, all of them relying on a tri-O-isopropylidene lactose derivative as a key intermediate.¹¹ However, our procedure offers some advantages, such as better overall yield and easier removal of the protective groups. Our synthesis of the 2'-O-fucosyllactose was accomplished as follows. An acetyl and a levulinoyl group were sequentially introduced on the benzyl lactoside 1^7 to give compound **3a** in 47% overall yield (95% conversion). After conventional acetylation of the remaining hydroxyl groups and selective removal of levulinoyl ester, acceptor **4** was glycosylated with the α -trichloroacetimidate of the 3,4 di-Oacetyl-2-O-benzyl-L-fucopyranose 5^{12} to give the protected 2'-O-fucosyllactose 6^{13} in 70 % yield (Scheme 2).

The deprotection to 2'-O-fucosyllactose was very cleanly and easily accomplished by standard Zemplèn deacetylation and hydrogenolysis (Pd/C, H₂, MeOH/H₂O, 24 h) of the remaining benzyls leading to 2'-O- fucosyllactose 7 in quantitative yield (Scheme 2). The optical rotation ($[\alpha]_D$ – 52.3, *c* 0.6, H₂O, after three days) as well as ¹H and ¹³C NMR data from compound 7 are in agreement with those reported in the literature;^{6,14} ¹H NMR (300 MHz, D₂O): δ 5.51, H-1"





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 (α/β) , J = 2.7 Hz, 5.43, H-1 α , J = 3.7 Hz, 4.72, H-1' (α/β) , J = 7.7 Hz, (the signal of H-1 β is overlapped with the solvent peak at δ 4.80–4.85); ¹³C NMR (75.44 MHz, D₂O): δ 103.11 (C-1'), 102.15 (C-1"), 98.72 (C-1 β), 94.64 (C-1 α).

In conclusion, we described a new and easy chemo-enzymatic synthesis of the biologically relevant 2'-O-fucosyllactose exploiting a versatile lactose building block, obtained by a double sequential acylation at 6'-OH and 2'-OH with two orthogonal protecting groups catalysed by lipase from *Candida antarctica*.

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REFERENCES

- 1. Dedicated to Prof. Joachim Thiem on the occasion of his 60th anniversary.
- 2. a) Khan, R.; Gropen, L.; Konowicz, P.A.; Matulovà, M.; Paoletti, S. Enzymic regioselective hydrolysis of peracetylated reducing disaccharides, specifically at the anomeric center: intermediates for the synthesis of oligosaccharides. Tetrahedron Lett. **1993**, *34*, 7767–7770.

b) Palmer, D. C.; Terradas, F. Regioselective enzymatic deacetylation of sucrose octaacetate in organic solvents. Tetrahedron Lett. **1994**, *35*, 1673–1676.

c) Cai, S.; Hakomori, S.; Toyokuni, T. Application of protease-catalyzed regioselective esterification in synthesis of 6'-deoxy-6'-fluoro- and 6-deoxy-6-fluorolactosides. J. Org. Chem. **1992**, *57*, 3431–3437.

- Lay, L.; Panza, L.; Riva, S.; Khitri, M.; Tirendi, S. Regioselective acylation of disaccharides by enzymatic transesterification. Carbohydr. Res. 1996, 291, 197–204.
- La Ferla, B.; Lay, L.; Poletti, L.; Russo, G.; Panza, L. Easy chemo-enzymatic synthesis of human milk trisaccharides from a common selectively protected lactose building block. J. Carbohydr. Chem. 2000, 19, 331–343.
- La Ferla, B.; Lay, L.; Russo, G.; Panza, L. Regioselective lipase acylation as a useful tool for separation and selective protection of β-D-Gal(1–4)-D-GlcNAc and β-D-Gal(1–3)-D-GlcNAc disaccharides. Tetrahedron: Asymmetry 2000, 11, 3647–3651.
- a) Kuhn, R.; Baer, H.H.; Gauhe, A. Fucosido-lactose, das Trisaccharid der Frauenmilch. Chem. Ber. 1955, 88, 1135–1146.
 b) Kuhn, R.; Baer, H.H.; Gauhe, A. Kristallisierte Fucosido-lactose. Chem. Ber. 1956, 89, 2513.
- Jung, K-H.; Hoch, M.; Schmidt, R.R. Selectively protected lactose and 2-azido lactose, building blocks for glycolipid synthesis. Liebigs Ann. Chem. 1989, 1099–1106.
- 8. General procedure for enzymatic acylation reaction: The substrate (1 or 2a) (100 mg) suspended in the solvent (10 mL) and the activated ester (2 mL) were shaken in the presence of *Candida antarctica* lipase (300 mg), immobilized on macroporus acrylic resin (CHIRAZYME[®] L-2, c.-f. C2, lyo; purchased from Roche Diagnostics), at the temperature reported in Table 1. The enzyme was filtered off and the solvent removed at reduced pressure. The acylated product and the unreacted substrate were isolated by chromatography on silica gel with ethyl acetate/methanol 9/1.
- Kunz, C.; Rudloff, S. Biological functions of oligosaccharides in human milk. Acta Pediatr. 1993, 82, 903–912.



ORDER		REPRINTS
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CONVENIENT ROUTE TO 2'-O-FUCOSYLLACTOSE

 a) Coppa, G.V.; Gabrielli, O.; Giorgi, P.; Catassi, C.; Montanari, M. P.; Varaldo, P. E.; Nichols, B. L. Preliminary study of breastfeeding and bacterial adhesion to uroepithelial cells. The Lancet **1990**, *335*, 569–571.

b) Cravioto, A.; Tello, A.; Villafan, H.; Ruiz, J.; Del Vedovo, S.; Neeser, J-R. Inhibition of localized adhesion of enteropathogenic Escherichia coli to HEp-2 cells by immunoglobulin and oligosaccharide fractions of human colostrum and breast milk. J. Infect. Dis. **1991**, *163*, 1247–1255.

c) Laegreid, A.; Otnaess, A. B. K.; Fuglesang, J. Human and bovine milk: comparison of ganglioside composition and enterotoxin-inhibitory activity. Pediatr. Res. **1986**, *20*, 416–421.

d) Newburg, D. S.; Pickering, L. K; McCluer, R. H.; Cleary, T. G. Fucosylated oligosaccharides of human milk protect suckling mice from heat-stable enterotoxin of Escherichia coli. J. Infect. Dis. **1990**, *162*, 1075–1080.

 a) Abbas, S.A.; Barlow, J.J.; Matta, K. L. Synthetic studies in carboydrates. Part XIV. Synthesis of *O*-α-L-fucopyranosyl-(1–2)-*O*-β-D-galactopyranosyl-(1–4)-D-glucopyranose (2'-*O*-α-L-fucopyranosyllactose). Carbohydr. Res. **1981**, *88*, 51–60.
 b) Jain, R. K.; Locke, R.D.; Matta, K. L. Synthetic studies in carboydrates. Part

b) Jam, K. K., Locke, K.D., Matta, K. L. Symmetric studies in carboydrates. Fart LXXVII. A convenient synthesis of O- α -L-fucopyranosyl-(1–2)-O- β -D-galactopyranosyl-(1–4)-D-glucopyranose (2'-O- α -L-fucopyranosyllactose). Carbohydr. Res. **1991**, 212, c1-c3.

c) Izumi, M.; Tsuruta, O.; Harayama, S.;. Hashimoto, H. Synthesis of 5-thio-L-fucose-containing disaccharides, as sequence-specific inhibitors, and 2'-fucosyllactose, as substrate of α -L-fucosidases. J. Org. Chem. **1997**, *62*, 992–998.

d) Matta, K. L.; Jain, R. K.; Locke, R.D. US Patent 5,438,124; 1 Aug 1995; (CA, **124**, 30256j, 1996).

- 12. Manzoni, L.; Lay, L.; Schmidt, R.R. Synthesis of Lewis a and Lewis x pentasaccharides based on *N*-trichloroethoxycarbonyl protection. J. Carbohydr. Chem. **1998**, *17*, 739–758 and references therein.
- Selected analytical data for trisaccharide 6: $[\alpha]_D 47.7^\circ$ (c 1.05, CHCl₃), ¹H NMR 13. $(300 \text{ MHz}, \text{CDCl}_3)$: δ 7.39–7.21 (m, 10H, H_{Ar}), 5.32 (d, 1H, J_{3',4'} = 3.3 Hz, H-4'), $5.26 (d, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.25 (t, 1H, J_{3'',4''} = J_{4'',5''} = 4.6 \text{ Hz}, \text{H-4''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 1H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 2H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H-1''}), 5.15 (dd, 3H, J_{1'',2''} = 3.9 \text{ Hz}, \text{H$ 1H, $J_{2'',3''} = 10.6$ Hz, H-3"), 5.10 (t, 1H, $J_{2,3} = J_{3,4} = 9.4$ Hz, H-3), 5.01 (dd, 1H, $J_{2',3'}$ = 10.5 Hz, H-3'), 5.00 (t, 1H, H-2), 4.86 (d, 1H, J = 12.4 Hz, CHPh), 4.59 (d, 1H, J = 12.4 Hz, CHPh), 4.59 (unresolved, 1H, H-6a), 4.58 (d, 1H, CHPh), 4.53 (d, 1H, CHPh), 4.51 (d, 1H, $J_{1,2} = 7.6$ Hz, H-1), 4.43 (d, 1H, $J_{1',2'} = 7.6$ Hz, H-1'), 4.37 (qd, 1H, H-5"), 4.30 (dd, 1H, $J_{6a,5} = 4.7$ Hz, $J_{6a,6b} = 12.1$ Hz, H-6b), 4.15–4.02 (m, 2H, H-6'a, H-6'b), 3.88 (t, 1H, $J_{3,4} = J_{4,5} = 9.0$ Hz, H-4), 3.85–3.79 (m, 3H, H-2', H-5', H-2"), 3.62 (ddd, H-5), 2.12–1.86 (6s, 24 H, CH₃CO), 1.13 (d, 3H, J_{5",6"} = 6.6 Hz, H-6"); ¹³C-NMR (75.44 MHz, CDCl₃): δ 170-169 (8 CO), 136-127 (C_{Ar}), 100.71, 99.41, 97.14 (C-1, C-1', C-1"), 73.99, 73.70, 73.19, 73.18, 72.89, 72.30, 72.15, 71.63, 71.29, 70.69, 69.72 (C-2, C-3, C-4, C-5, C-2', C-3', C-4', C-5', C-2", C-3", C-4", C-5"), 73.18, 70.69 (2 CH₂Ph), 20.74-20.46 (CH₃CO), 15.65 (C-6") Anal. Calcd for C₄₈H₆₀O₂₃ (1004.80): C, 57.37; H, 6.02. Found: C, 57.42; H, 6.05.
- 14. Ishizuka, Y.; Nemoto, T.; Fujiwara, M.; Fujita, K.; Nakanishi, H. Three-dimensional structure of fucosyllactoses in an aqueous solution. J. Carbohydr. Chem. **1999**, *18*, 523–533.

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