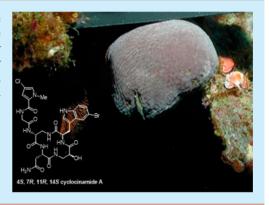
Application of the DP4 Probability Method to Flexible Cyclic Peptides with Multiple Independent Stereocenters: The True Structure of Cyclocinamide A

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Supporting Information

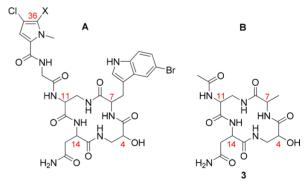
ABSTRACT: A DP4 protocol has been successfully utilized to establish the true structure of the natural product cyclocinamide A, a flexible cyclic peptide with four isolated stereocenters. Benchmarking the necessary level of theory required to successfully predict the NMR spectra of three previously synthesized isomers of cyclocinamide A led to the prediction of the natural stereochemistry as 4S, 7R, 11R, 14S, which has been confirmed by total synthesis.



In 1997, Crews and co-workers described the isolation, structure, and biological activity (solid tumor activity in a disk diffusion assay) of the cyclic peptide cyclocinamide A (CC-A). This initial work, together with a second publication, proposed the all-S stereoisomer (1a, Figure 1A) for this natural product constructed of α - and β -amino acids. Initial synthetic work, which did not benefit from the stereochemical claims made in the second CC-A publication, produced the 4R, 7S, 11R, 14S (1b) and 4R, 7S, 11S, 14S (1c) isomers. However, neither was found to be the natural product. Ireland and co-workers isolated a similar compound that they dubbed cyclocinamide B (CC-B, 2), which was claimed to differ from its predecessor in two ways: the addition of a second chlorine atom at C36 and defined stereochemistry of 4S, 7R, 11S, 14R.

One of us (UCSC) has prepared 1a,⁶ the 4S, 7S, 11R, 14S isomer 1d,⁶ 2,⁷ and *ent*-1b⁷ (i.e., the compound with the core of 2 and the glycine-pyrrole side chain of compounds 1). None of these compounds corresponded to either natural product as determined by comparison of the ¹H and ¹³C NMR data. However, it was determined that CC-A and CC-B were either enantiomers or identical with respect to their stereochemistry by comparison of their respective NMR data.⁷

Herein, we report the assignment of the stereochemistry of cyclocinamide A as 1e (4S, 7R, 11R, 14S) through the extension of the DP4 analysis to this 14-membered,



1a (4S, 7S, 11S, 14S), X = H), "all-S", UCSC synthesis 1b (4R, 7S, 11R, 14S, X = H), "4R, 11R", Montana State synthesis 1c (4R, 7S, 11S, 14S, X = H), "4R", Wayne State synthesis 1d (4S, 7S, 11R, 14S, X = H), "11R", UCSC synthesis ent-1b (4S, 7R, 11R, 14R, X = H), "7R, 14R", UCSC synthesis 1e (4S, 7R, 11R, 14S, X = H), "7R, 11R", present synthesis 2 (4S, 7R, 11S, 14R, X = CI), UCSC synthesis

Figure 1. (A) Cyclocinamide structures 1 and 2, stereochemistry and shorthand designation; (B) truncated structure of CC-A 3 for computational studies.

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conformationally flexible, cyclic peptide with four unrelated stereocenters. In addition, we present the total synthesis of this isomer, thus verifying the DP4 prediction.

The failure to find correspondence between the claimed and true structures of natural products by synthetic means has been the experience of many other researchers⁸ and has led us to a new strategy. The prediction of ¹H and ¹³C chemical shifts by purely computational means has advanced greatly in recent years. In addition, the classic problem of natural product identification, that of having one set of experimental NMR data that could be assigned to one of several stereochemically possible structures, has been explored by Smith and Goodman¹⁰ with the development of the DP4 application. Use of this procedure results in the assignment of a probability of identity between a given experimental set of ¹H and/or ¹³C NMR spectra and each of the corresponding computationally derived data sets for the stereoisomers.¹¹

Implementation of this methodology to the present problem, however, was not without its challenges. The stereogenic centers in CC-A are isolated from one another; no useful NOE data beyond the nearest neighbor relationships were obtained, and the four amide bonds effectively isolate each spin system. Furthermore, there was no detectable interaction between the glycine-pyrrole fragment at C11, the tryptophan residue at C7, and the asparagine side chain at C14. In addition, the flexibility of CC-A would require the incorporation of a number of low-energy conformations that contribute to the final calculated spectra.

Structure 3 was employed in the calculations. Such an approach was deemed reasonable due to the lack of interactions between side chains on the macrocycle (*vide supra*). In addition, we had previously argued that evaluation of the chemical shift data rested most reasonably on the ring sp³ carbons and hydrogens, where all the stereochemistry resides, and not on the more conformationally mobile side chains.

The work initially focused on computing the ¹H and ¹³C chemical shifts of the three isomers most recently prepared: **1a** (all-*S*), **1d** (11*R*), and *ent*-**1b** (7*R*, 14*R*). The computational protocols would be continually refined until the DP4 application could assign a high probability of identity upon comparison of a given experimental set of ¹H and ¹³C chemical shifts with the corresponding calculated spectra for that isomer when challenged with the calculated spectra of all three isomers. Once this benchmarking task was complete, the same methodology would be applied to all possible stereoisomers of cyclocinamide A for comparison with the experimental shifts of the natural product. ¹²

Table 1 gives an overview of the initial conformational search results for each isomer using molecular mechanics calculations (Spartan 10¹³). However, the initial levels of theory and basis sets employed within Gaussian 09¹⁴ used to refine the energy levels and provide the calculated spectral data did not produce adequate results. Specifically, we were unable to obtain correspondence of the experimental vs calculated values for the 7R, 14R isomer with the DP4 method from our initial calculations. Successful implementation was finally achieved by an additional geometry optimization from the conformations with energies <4 kcal mol⁻¹ from the minimum using B3LYP/6-311+G(d,2p), which also included a frequency calculation on the final structures to obtain the Gibbs free energies. As shown in

Table 1. Computational Data and DP4 Predictions

		minimum energy conformations		
isomer	stereochemistry	molecular mechanics ^a	Gaussian ^b	DP4 protocol predictions c (%)
1a	all-S	105	13	0.0
ent-1b	7R, 14R	39	8	26.3
1c	4 <i>R</i>	64	16	0.0
1d	11 <i>R</i>	102	19	0.3
1e	7R, 11R	121	15	73.2
1f	7R	71	8	0.2
1g	11R, 14R	61	9	0.0
1h	14R	50	7	0.0

"Number of low energy conformations from molecular mechanics calculations. "Number of low energy (<4 kcal/mol⁻¹) conformations from single-point energy calculations. "Probabilities using DP4 applet from comparison of cyclocinamide A (CC-A) experimental NMR data with computational NMR data from all isomers.

Table 1, the number of low energy conformations ranged from 7 for the 14R structure to 19 for 1d (11R). These structures were then used to calculate the ¹H and ¹³C NMR shielding constants with the mPW1PW91/6-311+G(3df,2p) level of theory. To transform the shielding constants to chemical shift data, N-methylacetamide in DMSO was employed as the reference, calculated at the same level of theory. The Gibbs free energy of each conformer was used to assess its contribution to the Boltzmann distribution of structures contributing to the final NMR shielding constants. From these results, 92–99% identity was obtained from the DP4 analysis for each of the three structures; only the core sp³ centers (¹H and ¹³C spectra) were used in the comparison (Supporting Information, Tables S1 and S2). ¹⁷

With a confirmed computational method in place, the final DP4 analysis was performed on all eight possible diastereomers of CC-A. Table 1 gives the final results, which identified the 7R, 11R isomer 1e as the most probable stereochemical match to the natural product. The only other isomer with an appreciable, but distinctly lower, probability of identity with CC-A was *ent-*1b, which the previous synthetic work had shown was not the desired product.

Based on this analysis, we prepared isomer 1e using an analogous route to that employed in the production of 2 and ent-1b (Scheme 1). Thus, methyl R-5-bromotryptophan 4 was coupled with 5 (prepared as previously reported⁶) to afford dipeptide 6, which itself was subjected to TFA deprotection and coupled with commercially available (S)-Fmoc-Asn(Tr)-OH 7 to provide the 4S, 7S, 11R-tripeptide 8 (CC-A numbering). Additional deprotection and coupling, this time with (R)-10, provided tetrapeptide 11 in a moderate 49% yield from 8, but with 40% of recovered 9, which could be recycled. Carrying out this reaction for an extended time did not lead to improved yields. The conversion of 11 to seco-acid 13 required deprotection of the Fmoc amine and carboxylic acid saponification. The latter reaction proved sensitive, with the best results being obtained from carefully monitored treatment with a base at 0 °C for 3 h; extended exposure of 13 to the base at room temperature led to extensive decomposition. Cyclization was accomplished by treating 13 with DEPBT, affording compound 14 in 71% yield. Reductive deprotection of 14 removed both the Boc and the terminal amide trityl group, setting up amine coupling with carboxylic acid 156 to give hexapeptide 16 in

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Scheme 1. Synthesis of 7R, 11R-Cyclocinamide A

66% yield (2 steps). Finally, TBDPS removal by treatment with TBAF gave the target alkaloid 1e in 44% yield.

The ¹H and ¹³C NMR spectra of **1e** were compared to the values originally reported by Crews¹ (Table S3). Figures S1 and S2 provide a graphical comparison of the ring sp³ carbons and hydrogens of all four of the recently prepared CC-A isomers (1a, ent-1b, 1d, and 1e) to the natural material. Table S4 provides the corresponding ¹³C data in tabular form color-coded for the magnitude of chemical shift variation from that of the natural material; the same comparison of natural CC-B with synthetic 2 is also provided in this table. The close correspondence of the 7R, 11R isomer spectral data with corresponding natural product data strongly support the assignment of compound 1e as cyclocinamide A. The most notable difference in the synthetic sample is the doublet appearing at δ 6.00 ppm, which we assign to the hydroxyl OH group (a likely candidate for exchange in the natural sample). 18 Otherwise, the largest disparity between our synthetic sample and the originally isolated compound is the magnitude of the dextrorotatory specific rotation: $[\alpha]^{21}_{D}$ +102.5 for the synthetic material vs $[\alpha]^{21}_{D}$ +29 for the original isolated sample (c 0.1, MeOH for both).

In conclusion, we have succeeded in establishing the absolute stereochemistry and structure of the marine natural product cyclocinamide A. In addition, we have demonstrated

that the DP4 protocol can be effectively utilized to guide synthetic target identification on flexible molecules, provided the necessary level of theory for the problem can be ascertained.¹⁹

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.8b01756.

Details and tables of computational data, comparison of NMR data between final synthetic compound and natural product, full experimental, and NMR spectra (PDF)

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Notes

The authors declare no competing financial interest.

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- (12) With four stereogenic centers, 16 possible stereoisomers (8 pairs of enantiomers) are possible. However, since the analysis was by NMR, which is transparent to enantiomers, the determination would be among the 8 unique diastereomers. See Supporting Information for full details.
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- (16) See Supporting Information for all calculation tables.
- (17) It should be noted that the synthesis of the all-S isomer 1a was established through the use of Marfey's method and subsequent HPLC analysis, establishing the stereochemical purity of the product. The convergence of experimental and calculated NMR data not only validated this computational approach but also verified that the final cyclizations leading to 1d and ent-1b occurred without racemization, thus preserving stereochemical integrity.
- (18) The free hydroxyl group in cyclocinamide B is observed at δ 6.0; see ref 5.
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