(1–2 × 10^8 MSH u./g.) was obtained. Eight grams was mixed with 150 ml. of 0.1 *N* acetic acid and centrifuged. Sixteen grams oxycellulose was added to the supernate and the mixture shaken for 75 min. Oxycellulose was removed by centrifugation, washed with 0.1 *N* acetic acid, then shaken in 100 ml. of 80% acetic acid for 60 min. The supernate was diluted with equal quantities of water. Lyophilization yielded 0.35–0.5 g. product (1–2 × 10^9 MSH u./g.). One and a half grams of this fraction was distributed through a 12-tube countercurrent system at 5° using *sec*-BuOH and 0.5% aqueous trichloroacetic acid. The contents of tubes 4–6 were combined and lyophilized. Approximately 0.5 g. of solids (3–4 × 10^9 MSH u./g.) was obtained. Forty mg. was subjected to paper electrophoresis at 5°, 18 volts/cm., 8–10 hours, *p*H 8.9 using barbiturate–acetate–hydrochloric acid buffer (u = 0.056). Four components were visualized with 1% brom phenol blue staining. That moving fastest toward the cathode was extracted with 20% acetic acid and lyophilized. The product was dissolved in 1 ml. of 0.2 *N* acetic acid and subjected to paper electrophoresis at *p*H 4.9 using pyridine–acetic acid buffer (u = 0.1), 5°, 18 volts/cm., 10–12 hours. Staining revealed a single component moving toward the cathode. The active area was extracted with 20% acetic acid and lyophilized. The white solid, 2.5 mg., (1.5–2.5 × 10^{10} MSH u./g.) represented about 30% of the total MSH activity placed on the first electrophoretic run at alkaline *p*H. Ninhydrin reaction of hydrolyzed extracts of different parts of the filter paper run at *p*H 4.9 showed MSH activity associated with the predominant color response.

TABLE

Amino acid[9]	Per cent.[12]	Molecular ratio[13]
Aspartic	3.8	3
Glutamic	5.3	3
Serine	3.4	3
Glycine	2.3	3
Tyrosine	4.1	2
Lysine	5.3	3
Arginine	4.0	2
Valine	2.9	2
Phenylalanine	4.7	3
Alanine	1.4	1
Cystine[10]	5.3	2
Proline	3.7	3
Leucine	1.7	1
Threonine	1.3	1
Histidine	0.6	0
Tryptophan[11]	...	2
Total	49.8	34

(9) Semi-quantitative amino acid analyses were done by A. M. Gross and W. F. White of the Research Department, Armour Laboratories using filter paper chromatography and determining the intensity of ninhydrin stained areas with a densitometer; J. F. Rowland and A. M. Gross, *Anal. Chem.*, **26**, 502 (1954).

(10) Cystine and cysteine are not distinguished in the analysis. However, cysteine is probably absent because MSH is not oxidized and reduced readily as would be expected were this amino acid present. Methionine was not tested for.

(11) Tryptophan was determined by ultraviolet absorption after subtracting tyrosine from the total value; A. B. Lerner and C. P. Barnum, *Arch. Biochem.*, **10**, 417 (1946).

(12) Tryptophan, methionine, moisture and ash were not included in the total amino acid per cent. analysis.

(13) Molecular ratios are given in whole numbers and represent only approximate values.

The active fraction moved as a single component (staining with bromophenolblue) on paper electrophoresis at *p*H 1.4, 4.9, 8.9, 11.3 and 12.2. Since at *p*H 11.3 movement towards the anode was slight compared with dextran, the iso-electric *p*H was estimated to be in the region of 10.5–11. On the basis of amino acid composition minimum molecular weight was estimated at 4500. MSH activity of the final product was approximately 500 times that of the original hog posterior pituitary powder with little ACTH activity.[14] This fraction behaved as a single component when distributed in a 97 tube countercurrent apparatus employing the solvents described previously. Although the MSH preparation, assumed to be a polypeptide, was tested by electrophoresis and countercurrent distribution, other criteria for homogeneity remain to be satisfied.

(14) MSH has little if any vasopressin or ACTH activity. Assays set to detect one unit each of ACTH or vasopressin, using 378 and 449 mcg. of MSH, respectively showed no activity.

DIVISION OF DERMATOLOGY AARON B. LERNER
UNIVERSITY OF OREGON MEDICAL SCHOOL TEH H. LEE
PORTLAND 1, OREGON

RECEIVED JANUARY 13, 1955

A NEW METHOD OF FORMING PEPTIDE BONDS

Sir:

We wish to describe a new and very useful method of forming peptide or other amide bonds. The two components, one containing a free carboxyl function and the other a free amino group, couple directly and rapidly in high yield on treatment with N,N′-dicyclohexylcarbodiimide at room temperature.

In contrast to other schemes for carboxyl activation involving mixed anhydride formation, the reaction is not sensitive to moisture; indeed, it may be carried out in aqueous solution. The remarkable selectivity of the reagent is attested by the successful use of carbobenzoxyserine as an acylating moiety without protection of the hydroxyl group. No racemization was detected employing as the acylating agent a dipeptide derivative in which an optically active amino acid furnished the free carboxyl function (carbobenzoxyglycyl-L-phenylalanine), an observation of considerable importance in the synthesis of larger peptides by joining units containing two, three or more amino acids. The co-product, N,N′-dicyclohexylurea, has a very low solubility in most organic or aqueous solvents, and, in all cases tried, is easily separated.

$$RCO_2H + NH_2R' + C_6H_{11}N{=}C{=}NC_6H_{11} \longrightarrow RCONHR' + C_6H_{11}NHCONHC_6H_{11}$$

The simplicity, convenience and efficiency of this technique may be illustrated by the synthesis of a tripeptide derivative. After a 4-hour period at room temperature, a solution in tetrahydrofuran of carbobenzoxyglycyl-L-phenylalanine containing a slight excess of crystalline N,N′-dicyclohexylcarbodiimide[1] and ethyl glycinate was treated with a small amount of acetic acid (to decompose the

(1) Readily prepared by the method of R. Herbeck and M. Pezzati, *Ber.*, **71**, 1933 (1938).

excess reagent). The insoluble urea was removed, the solvent was replaced by ethyl acetate, and the solution was washed with dilute acid and aqueous potassium bicarbonate. The addition of petroleum ether afforded 87% of crystalline carbobenzoxyglycyl-L-phenylalanylglycine ethyl ester; m.p. 118–119°, $[\alpha]^{27}D$ −13.5° [ethanol] (reported[2]: m.p. 116–118°, $[\alpha]^{25}D$ −12°). In a similar fashion we have prepared a variety of dipeptide derivatives, including the following examples.

In methylene chloride, phthaloyl-L-phenylalanylglycine ethyl ester was produced in 92% yield; m.p. 161–162°, $[\alpha]^{26.6}D$ −146°, (reported[3]: m.p. 161–162°, $[\alpha]^{29.5}D$ −146°). In aqueous tetrahydrofuran, a product of the same quality was obtained in 72% yield. Phthaloyl-L-alanyl-L-proline benzyl ester (74%) was isolated with m.p. 101–102°, $[\alpha]^{26.5}D$ −135° [ethanol]. *Anal.* Calcd. for $C_{23}H_{22}N_2O_5$: C, 67.98; H, 5.42; N, 6.90. Found: C, 68.07; H, 5.52; N, 6.77. Carbobenzoxy-L-serine and ethyl glycinate coupled to give carbobenzoxy-L-serylglycine ethyl ester (59%) in tetrahydrofuran: m.p. 106–107°, [ethanol], reported,[4] m.p. 105–107°. Phthaloyl-L-phenylalanyl-L-leucine ethyl ester (91% yield) had a m.p. of 109–110°, $[\alpha]^{25.4}D$ −115° [ethanol]. *Anal.* Calcd. for $C_{25}H_{28}N_2O_5$: C, 68.78; H, 6.47; N, 6.42. Found: C, 68.50; H, 6.59; N, 6.48.

DEPARTMENT OF CHEMISTRY JOHN C. SHEEHAN
MASSACHUSETTS INSTITUTE OF TECHNOLOGY
CAMBRIDGE 39, MASSACHUSETTS GEORGE P. HESS[5]

RECEIVED JANUARY 11, 1955

(2) G. Anderson and R. Young, THIS JOURNAL, **74**, 5307 (1952).
(3) J. Sheehan, D. Chapman and R. Roth, *ibid.*, **74**, 3822 (1952).
(4) J. S. Fruton, *J. Biol. Chem.*, **146**, 463 (1942).
(5) Aided by a fellowship from the National Foundation for Infantile Paralysis.

9α-HALO-11β-HYDROXY AND 11-KETO DERIVATIVES OF PROGESTERONE, DESOXYCORTICOSTERONE AND 17α-HYDROXYPROGESTERONE

Sir:

In previous communications[1,2] there have been described the synthesis of 9α-halogenated derivatives of cortisone and hydrocortisone and shown that the glucocorticoid activity of these substances increased with decreasing atomic weight of the halogen atom. The most active member of that series, 9α-fluorohydrocortisone acetate possessed about 11 times the activity of cortisone acetate in the rat liver glycogen assay. Soon thereafter it was found that in addition to being potent glucocorticoids these compounds were highly effective in controlling electrolyte balance and in maintaining life in the rat[3], dog[4,5] and in man.[4,6]

It appeared of great interest to ascertain what influence variations in the side-chain might have upon the adrenocorticoid activity of such halogenated derivatives. For this purpose we have prepared the 9α-halo derivatives (halogen = Br, Cl, F) of 11β-hydroxyprogesterone, 11β,17α-dihydroxyprogesterone and corticosterone acetate and of the corresponding 11-ketones by a synthetic route paralleling that described in our earlier publications.[1,2] This synthesis proceeds from the 11-mesylates of the requisite 11α-hydroxy derivatives[7] (11α-hydroxyprogesterone mesylate, m.p. 165–167°; $[\alpha]^{23}D$ +135° (*c*, 0.77 in $CHCl_3$); λ_{max}^{alc} 238 mμ (ε = 17,200); *Anal.* C, 64.81; H, 7.63; S, 7.48. Epicorticosterone 11α-mesylate 21-acetate, m.p. 156–157°; $[\alpha]^{23}D$ +144° (*c*, 0.92 in $CHCl_3$); λ_{max}^{alc} 238 mμ (ε = 16,600); *Anal.* C, 61.52; H, 7.07. 11α,17α-Dihydroxyprogesterone 11α-mesylate, m.p. 150–152°; $[\alpha]^{23}D$ +64° (*c*, 0.49 in $CHCl_3$); λ_{max}^{alc} 238 mμ (ε = 18,200); *Anal.* C, 62.11; H, 7.71; S, 7.11), via the 9,11-unsaturated steroids (9(11)-dehydro-17α-hydroxyprogesterone, m.p. 214–216°; $[\alpha]^{23}D$ +67° (*c*, 0.82 in $CHCl_3$); λ_{max}^{alc} 239 mμ (ε = 18,450); *Anal.* C, 76.52; H, 8.46), to the 9α,11β-bromohydrins (see table). The latter on treatment with base yielded the 9β,11β-epoxides (9β,11β-oxidoprogesterone, amorphous, $[\alpha]^{23}D$ +61° (*c*, 1.55 in $CHCl_3$); λ_{max}^{alc} 243 mμ (ε 13,600). 9β,11β-Oxidodesoxycorticosterone acetate, m.p. 137–138°; $[\alpha]^{23}D$ +61° (*c*, 0.66 in $CHCl_3$); λ_{max}^{alc} 243 mμ (ε = 15,100); *Anal.* C, 71.81; H, 8.10. 9β,11β-Oxido-17α-hydroxyprogesterone, m.p. 183–184°; $[\alpha]^{23}D$ −32° (*c*, 1.02 in $CHCl_3$); λ_{max}^{alc} 243 mμ (ε = 16,600); *Anal.* C, 72.99; H, 8.11), which upon reaction with the requisite hydrogen halides formed the 9α-chloro- and 9α-fluoro-11β-hydroxy derivatives. Oxidation with chromic acid furnished the corresponding 11-ketones. Alternatively, the 9α-chloro-derivatives could be prepared by allowing the 9(11)-unsaturated steroids to react with N,N′-dichlorodimethylhydantoin in the presence of perchloric acid.[8]

The physical properties of the halogenated steroids and the activities of representative compounds in the liver glycogen and sodium retention assays in the adrenalectomized rat are listed in the accompanying table. As had been observed previously in the 9α-halohydrocortisone series both gluco- and mineralocorticoid activities were found to increase with decreasing atomic weight of the halogen atom. No significant differences were noted between the activities of the 11β-hydroxy and 11-keto derivatives. Outstanding among the compounds tested were 9α-fluoro-11β-hydroxy and 11-ketoprogesterone, which although lacking both the 17- and 21-hydroxyl groups approximately equalled cortisone acetate in glucocorticoid activity. The most potent mineralocorticoids of this series were 9α-fluorocorticosterone acetate and 9α-

(1) J. Fried and E. F. Sabo, THIS JOURNAL, **75**, 2273 (1953).
(2) J. Fried and E. F. Sabo, *ibid.*, **76**, 1455 (1954).
(3) A. Borman, F. M. Singer and P. Numerof, *Proc. Soc. Exp. Biol. Med.*, **86**, 570 (1954).
(4) G. W. Liddle, M. M. Pechet and F. C. Bartter, *Science*, **120**, 496 (1954).
(5) W. W. Swingle, C. Baker, M. Eisler, S. J. Le Brie and L. J. Brannick, *Proc. Soc. Exp. Biol. Med.*, In press.
(6) A. Goldfien, G. W. Thorn, P. M. Beigleman and J. C. Laidlaw, *J. Clin. Endocrinology*, **14**, 782 (1954).
(7) J. Fried, R. W. Thoma, J. R. Gerke, J. E. Herz, M. N. Donin and D. Perlman, THIS JOURNAL, **74**, 3962 (1952).
(8) The course of this reaction was dependent on the nature of the side chain. Thus, 9(11)-dehydro-17α-hydroxyprogesterone afforded the desired chlorohydrin in about 50% yield. On the other hand, treatment of 9(11)-dehydroprogesterone with N,N′-dichlorohydantoin resulted in a mixture containing more than one atom equivalent of chlorine from which 9α-chloro-11β-hydroxyprogesterone could be isolated only after reduction with chromous chloride. It appears likely that the extra chlorine atom reducible by chromous chloride is located in the 17-position.