

Syntheses of Lamellarins I and K by [3+2] Cycloaddition of a Nitron to an Alkyne

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Abstract: Lamellarins I and K were obtained by a new approach based on the 1,3-dipolar cycloaddition of a nitron to an alkyne. The key cycloaddition yields an isoxazoline which rearranges to afford the central pyrrole ring.

Key words: lamellarins, nitron-alkyne 1,3-dipolar cycloaddition

The lamellarin alkaloids are a group of approximately 35 compounds isolated from the marine prosobranch mollusc *Lamellaria* sp., the marine ascidian *Didemnum* sp. and the sponge *Dendrilla cactos*. Most of them have the basic structure **1**, although about a third have a double bond between carbons 8 and 9.¹ The lamellarins have an interesting range of pharmacological activities, including antitumour and anti HIV-1 properties, reversal of multidrug resistance (MDR), and immunomodulatory activity.^{1,2} For example, it has been claimed that lamellarins I (**1a**) and K (**1b**) not only exhibit potent cytotoxic activity *in vitro* against multidrug-resistant cell lines, but also reverse their MDR at noncytotoxic concentrations,^{2,3} and the results of *in vivo* tests of lamellarin K are consistent with these claims.³

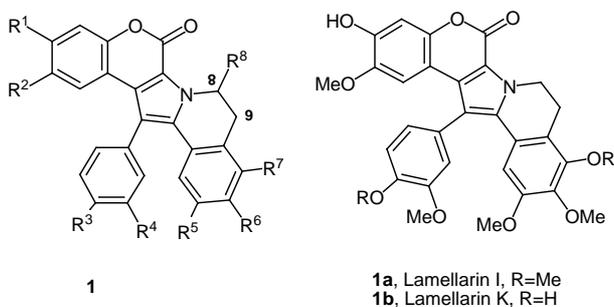
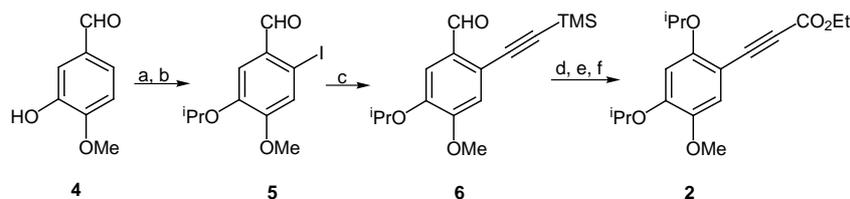
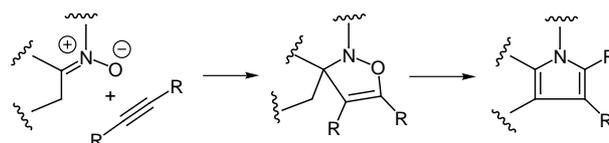


Figure 1



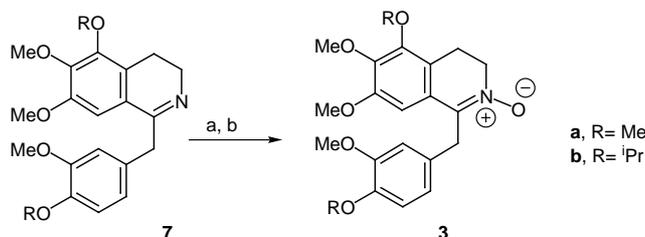
Scheme 2 a) K_2CO_3 , 2-bromopropane/DMF, 100 °C, 24 h, 98%; b) I_2 , $F_3C CO_2 Ag$, $CHCl_3$, rt, 3 h, 80%; c) TMSA, $PdCl_2(PPh_3)_2$, CuI, Et_3N , dioxane, 45 °C, 3 h, 98%; d) MCPBA, $NaHCO_3$, CH_2Cl_2 , rt, 30 min, 73%; e) K_2CO_3 , 2-bromopropane/DMF, 100 °C, 24 h, 45%; f) LDA, THF, -78 °C, 15 min; then $ClCO_2Et$, -78 °C → rt, 94%

The first synthesis of lamellarins was published by Steglich's group in 1997.⁴ Since then, new procedures for the preparation of lamellarins of general structure **1** have been developed by Banwell⁵ and Ishibashi.⁶ Banwell's synthesis of lamellarin K is particularly interesting because it is convergent, versatile and high-yielding.^{5a} The key step in this approach is the formation of the central pyrrole ring by an intramolecular 1,3-dipolar cycloaddition between an azomethine ylide and an alkyne.⁵ Here we present a related approach to lamellarins which relies on two known facts: a) nitrones add to alkynes to form isoxazolines,⁷ and b) isoxazolines with a $R-CH_2-$ substituent at position 3 can be rearranged to pyrroles⁸ (Scheme 1).

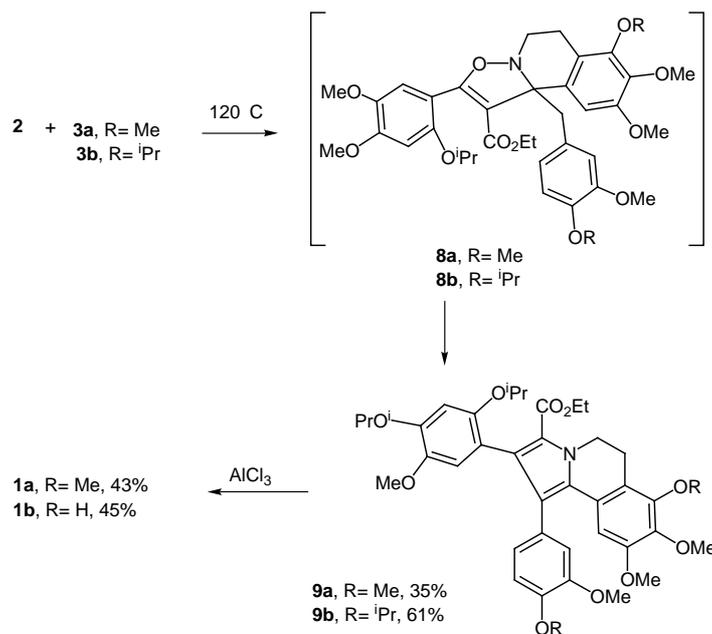


Scheme 1

As synthetic targets we selected lamellarins I and K. Application of our procedure to their synthesis requires preparation of alkyne **2** and nitrones **3**. Alkyne **2** was synthesized from isovanillin (**4**) as shown in Scheme 2. Protection of the hydroxyl group of **4** as an isopropyl ether by treatment with K_2CO_3 and iPrBr , followed by treatment with iodine and silver trifluoroacetate, afforded iodide **5**, which upon palladium-catalyzed coupling with trimethylsilylacetylene (TMSA) yielded **6**. Treatment with MCPBA transformed the formyl group of **6** into a hydroxyl group, which was then protected with K_2CO_3 and iPrBr as above, with accompanying loss of the TMS group. Finally, deprotonation of the resulting alkyne with LDA



Scheme 3 a) NaBH_4 , MeOH, rt, 1 h, 96-97%; b) H_2O_2 , Na_2WO_4 , MeOH, rt, 3 h, 45-49%



Scheme 4

and carboxylation with ethyl chloroformate afforded **2** in good yield.

N-Oxides **3** were obtained from dihydroisoquinolines **7**, which were prepared using standard isoquinoline alkaloid procedures.⁹ Reduction of the imine double bond of **7** with sodium borohydride, followed by a non-optimized oxidation with sodium tungstate¹⁰ (45-49% yield), afforded *N*-oxides **3**.

The key cycloaddition step was carried out by heating a mixture of nitron **3** and alkyne **2** under argon for 18 h at 120 °C in a sealed tube. Under these conditions compounds **9** were obtained in moderate yields.¹¹ The cycloaddition appears to take place with the expected regiochemistry to form the isoxazoline **8**, which upon heating undergoes rearrangement to the corresponding pyrrole **9**.

Finally, lamellarins I and K were obtained from **9a** and **9b**, respectively, by removal of the isopropyl protecting groups¹² with concomitant acid-catalyzed lactonization.¹³

Acknowledgement

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References and Notes

- Urban, S.; Hickford, S. J. H.; Blunt, J. W.; Munro, M. H. G. *Current Organic Chemistry* **2000**, *4*, 765-807. Lamellarins O, P, Q and R have a different structure characterized by a non-fused polysubstituted pyrrole nucleus.
- Andersen, R. J.; Faulkner, D. J.; Cun-heng, H.; Van Duyne, G. D.; Clardy, J. *J. Am. Chem. Soc.* **1985**, *107*, 5492. Quesada, A. R.; Gravalos, M. D. G.; Puentes, J. L. F. *Br. J. Cancer* **1996**, *74*, 677. Boger, D.; Soenen, D. R.; Boyce, C. W.; Hedrick, C. W.; Jin, Q. *J. Org. Chem.* **2000**, *65*, 2479. Reddy, M. V. R.; Rao, M. R.; Rhodes, D.; Hansen, M. S. T.; Rubins, K.; Bushman, F. D.; Venkateswarlu, Y.; Faulkner, D. J. *J. Med. Chem.* **1999**, *42*, 1901.
- Puentes, J. L. F.; García, D.; Rodríguez, A. US Patent. 5,852,033, 1995.
- a) Heim, A.; Terpin, A.; Steglich, W. *Angew. Chem. Int. Ed. Engl.* **1997**, *36*, 155. b) Peschko, C.; Winklhofer, C.; Steglich, W. *Chem.-Eur. J.* **2000**, *6*, 1147.

- (5) a) Banwell, M.; Flynn, B.; Hockless, D. *Chem. Commun.* **1997**, 2259. b) Banwell, M.; Flynn, B.; Hockless, D.; Longmore, R. W.; Rae, D. *Aust. J. Chem.* **1999**, *52*, 755.
- (6) Ishibashi, F.; Miyazaki, Y.; Ywao, M. *Tetrahedron* **1997**, *53*, 5951.
- (7) a) Tufariello, J. J. in *1,3-Dipolar Cycloaddition Chemistry*, Padwa, A., Ed., John Wiley and Sons, New York, 1988, vol. 2, pp. 83-68. b) Breuer, R.; Aurich, H. G.; Nielsen, A. T. in *Nitrones, Nitronates and Nitroxides*; Patai, S., Ed., John Wiley and Sons, New York, 1989, pp. 139-312. c) Little, R. D. in *Comprehensive Organic Synthesis*, Trost, B. M.; Fleming, I.; Paquette, L.A., Eds., Pergamon Press, Oxford, 1991, pp. 239-270.
- (8) a) Zhao, B-X.; Yu, Y.; Eguchi, S. *Tetrahedron* **1996**, *52*, 12049. b) Schmidt, G.; Stracke, H. U.; Winterfeldt, E. *Chem. Ber.* **1970**, *103*, 3196.
- (9) a) Shamma, M. *The Isoquinoline Alkaloids*, Academic Press, New York, 1972. b) Shamma, M.; Moniot, J. M. *Isoquinoline Alkaloid Research 1972-1977*, Plenum Press, New York, 1978.
- (10) Murahashi, S.; Mitsui, H.; Shiota, T.; Tsuda, T.; Watanabe, S. *J. Org. Chem.* **1990**, *55*, 1736.
- (11) Experimental procedure for the cycloaddition. A solution of **3b** (37 mg, 0.08 mmol) and **2** (38 mg, 0.12 mmol) in toluene (1 mL) was heated under argon in a sealed tube for 18 h at 120 °C. Then the solvent was evaporated in vacuo and the residue was chromatographed (silicagel, hexane/ether) to yield **9b** (38 mg, 61% yield) as an oil. Spectroscopic data for **9b**: UV (EtOH), λ_{max} : 304 nm. IR (NaCl), ν : 1694 cm^{-1} . ^1H NMR (CDCl_3), δ : 6.73 (m, 2H), 6.61 (s, 1H), 6.50 (s, 1H), 6.40 (m, 2H), 4.50-4.32 (m, 5H), 4.05-3.98 (m, 3H), 3.74 (s, 3H), 3.52 (s, 3H), 3.46 (s, 3H), 3.25 (s, 3H), 3.10-2.90 (m, 2H), 1.26-0.78 (m, 27H) ppm. ^{13}C NMR (CDCl_3), 162.2 (C), 151.4 (C), 150.1 (C), 149.5 (C), 148.2 (C), 145.8 (C), 145.7 (C), 144.5 (C), 141.3 (C), 130.0 (C), 129.0 (C), 127.8 (C), 124.0 (C), 123.2 (CH), 122.0 (C), 120.1 (C), 119.8 (C), 119.6 (C), 115.9 (CH), 114.8 (CH), 107.4 (CH), 104.7 (CH), 75.5 (CH), 71.7 (CH), 71.6 (CH), 71.4 (CH), 60.5 (CH_3), 59.7 (CH_2), 56.1 (CH_3), 55.7 (CH_3), 55.1 (CH_3), 42.6 (CH_2), 23.0 (CH_2), 22.7 (CH_3), 22.2 (CH_3), 21.9 (CH_3), 15.2 (CH_3), 13.7 (CH_3) ppm. LRMS, m/z (%): 745 (M^+ , 34). HRMS, m/z calcd. for $\text{C}_{43}\text{H}_{55}\text{NO}_{10}$ 745.382598, found 745.385468.
- (12) Banwell, M.; Flynn, B.; Stewart, S. G. *J. Org. Chem.* **1998**, *63*, 9139.
- (13) Spectroscopic data for these synthetic compounds are identical to those published for lamellarins I and K. Carroll, A. R.; Bowden, B. F.; Coll, J. C. *Aust. J. Chem.* **1993**, *46*, 489.

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