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# Direct aroylation of $\beta$ -cyclodextrin on the secondary hydroxyl face: a convenient equilibrium-dependent preparation of mono-3-aroyl- $\beta$ -cyclodextrin via C2-*O*-aroyl to C3-*O*-aroyl group migration

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**Abstract** A new and convenient method is presented for the preparation of mono-3-*O*-benzoyl- $\beta$ -cyclodextrin by direct regioselective benzoylation of  $\beta$ -cyclodextrin using *N*-benzoylimidazole in carbonate buffer solution. This process involved equilibrium-dependent C2-*O*-aroyl to C3-*O*-aroyl group migration to improve the yield.

Keywords Cyclodextrin  $\cdot$  Benzoylation  $\cdot$  Isomerization  $\cdot$  Equilibrium

# Introduction

Cyclodextrins (CDs) are well-known macrocyclic oligosaccharides which have hollow truncated cone structures capable of forming inclusion complexes with a variety of organic molecules in aqueous solutions. They have attracted widespread interest as hosts for building supramolecular structures, and their derivatives have evolved into a versatile class of host molecules with applications in enzyme mimics, molecular recognition, drug delivery, and chiral discrimination, to mention a few [1-4].

In recent years, selective benzoylation of the C-6 hydroxyl groups has been extensively studied, and the obtained mono-6-*O*-benzoyl- $\beta$ -CDs have been used as novel supramolecular photosensitizing hosts in photochirogenesis

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[5–11]. Since the more open secondary hydroxyl face of CDs is stated to be catalytically very important [12-15], benzoylations of this face are believed to produce valuable derivatives for asymmetry catalysis. However only very few reactions are known to introduce a single benzoyl group at the secondary hydroxyl face of CDs, especially at the 3 position, presumably because hydroxyl groups at the 3 position are the most inaccessible and difficult to be modified [16]. Hao et al. [17] benzoylated one of the C-3 hydroxyl groups with benzoyl chloride in alkaline acetonitrile solution, but the isolated yield was relatively low (16 %). Herein, we describe a new and convenient method for the preparation of mono-3-O-benzoyl- $\beta$ -CD by direct benzoylation of  $\beta$ -CD using N-benzoylimidazole in a carbonate buffer solution, an equilibrium process which involved C2-O-aroyl to C3-Oaroyl group migration to improve the yield (Scheme 1).

# **Results and discussion**

In previous work [18], we developed a simple and efficient method for direct regioselective benzoylation of a single C-2 hydroxyl group of  $\beta$ -CD, and found that its hydrolysis occurred via benzoyl group migration, i.e., the C-3 hydroxyl group of  $\beta$ -CD attacked the carbonyl carbon of the C2-*O*-benzoyl group. Therefore, mono-3-*O*-benzoyl- $\beta$ -CD can be prepared through isomerization of the corresponding mono-C2-*O*-benzoyl isomer.

A mixture of  $\beta$ -CD, one molar equivalent of *N*-(*p*-methylbenzoyl)imidazole, and carbonate buffer in DMF was stirred at 60 °C for 2 h. The reaction regioselectively generated mono-3-*O*-(*p*-methylbenzoyl)- $\beta$ -CD (1), mono-2-*O*-(*p*-methylbenzoyl)- $\beta$ -CD, and multi-benzoates. The mixture of mono-C2-*O*-benzoyl and mono-C3-*O*-benzoyl isomers (1), isolated by a simple open reversed-phase





column chromatography, was purified by preparative HPLC using 20 % aqueous MeOH as the eluent. The fractions containing the mono-C2-*O*-benzoyl isomer were concentrated under reduced pressure, dissolved in 20 % aqueous MeOH, heated at 50 °C for 1 h until the equilibrium of isomerization was established, and then purified by preparative HPLC. Through several isomerization and purification cycles, the mono-C3-*O*-benzoyl isomer (1) was obtained in good yield (39 %).

The structure of **1** was confirmed using ESI–MS and NMR spectra. Its ESI–MS spectrum exhibited the molecular ion  $[M + Na]^+$  at m/z = 1,275. <sup>13</sup>C NMR spectroscopy is an effective technique for the analysis of cyclic oligosaccharides. Breslow [19] reported that aroylation of a hydroxyl group of CDs usually leads to a downfield chemical shift of the carbon carrying the hydroxyl ( $\alpha$  carbon), but a small upfield chemical shift of the  $\beta$  carbon and a still smaller shift of the  $\gamma$  carbon. In the <sup>13</sup>C NMR spectra of **1**, the peak at  $\delta = 78.5$  ppm (C-4') clearly indicates that the substituent is at the 3 position of  $\beta$ -CD.

Using N-(p-methoxybenzoyl)imidazole as an aroylating reagent afforded **2** in 35 % yield.

In conclusion, mono-3-*O*-benzoyl- $\beta$ -CD was prepared by direct benzoylation of  $\beta$ -CD on the secondary hydroxyl face using *N*-benzoylimidazole in carbonate buffer solution in good yield. The process involved an isomerization equilibrium, i.e., mono-2-*O*-benzoyl- $\beta$ -CD was isomerized to the corresponding mono-3-*O*-benzoyl- $\beta$ -CD. This benzoylation method can be highly useful for the preparation of other benzoates of  $\beta$ -CD as macrocyclic host molecules.

# Experimental

Analytical and preparative HPLC column chromatographies were done using a Perkin-Elmer Series 200 HPLC system with a UV–Vis detector. A Kromasil 100-10-C18 column (4.6 mm  $\times$  250 mm) was used for the analytical HPLC, and a Zorbax SB-C18 column (10 mm  $\times$  250 mm) was used for preparative HPLC. NMR spectra were recorded on Bruker AM-600 spectrometer (<sup>1</sup>H 600 MHz and <sup>13</sup>C 150 MHz) in DMSO- $d_6$  solutions with tetramethylsilane as

standard. The ESI–MS experiments were performed using a ThermoQuest Finnigan LCQ<sup>DECA</sup> system equipped with an ESI source (ThermoQuest LC/MS Division, San Jose, CA, USA). Carbonate buffer (0.2 M, pH 9.9) was prepared by mixing equal volumes of 0.2 M sodium carbonate and 0.2 M sodium bicarbonate. *N*-Benzoylimidazole was prepared according to a literature procedure [20]. All other chemicals were of commercial grade and used without further purification.

#### General experimental procedure

To a stirred solution of 2 g  $\beta$ -CD (1.76 mmol) with one molar equiv. of N-benzoylimidazole in 60 cm<sup>3</sup> DMF, 12 cm<sup>3</sup> carbonate buffer (0.2 M, pH 9.9) was added. The reaction mixture was heated at 60 °C for 2 h. Then the mixture was neutralized with 1 NHCl, evaporated in vacuo to a volume of ca. 5 cm<sup>3</sup>, and 300 cm<sup>3</sup> of acetone was added to precipitate cyclodextrin derivatives. The collected solid was isolated on an open RP-18 column eluted with H<sub>2</sub>O/MeOH (10-30 %) to give a mixture of mono-C2-O-benzoyl and mono-C3-Obenzoyl isomers. The mixture was dissolved in 70 cm<sup>3</sup> 20 % aqueous MeOH, heated at 50 °C for 1 h, and then purified by preparative HPLC using 20 % aqueous MeOH as the eluent. The fractions containing the mono-C2-O-benzoyl isomer were evaporated in vacuo, dissolved in 30 cm<sup>3</sup> 20 % aqueous MeOH, heated at 50 °C for 1 h, and purified by preparative HPLC. The processes were carried out twice, and all the fractions containing the mono-C3-O-benzoyl isomer were combined and lyophilized to give the pure mono-C3-Obenzoyl isomer.

# Mono-3-O-(p-methylbenzoyl)- $\beta$ -CD (1, C<sub>50</sub>H<sub>76</sub>O<sub>36</sub>)

Yield 0.87 g (39 %); ESI–MS: m/z = 1,275 ([M + Na]<sup>+</sup>); <sup>1</sup>H NMR (600 MHz, DMSO- $d_6$ ):  $\delta = 2.35$  (s, 3H), 3.20–3.74 (m, 39H), 3.77 (t, 1H), 3.86 (br, 1H), 4.33–4.58 (m, 7H), 4.70–4.87 (m, 6H), 4.92 (d, 1H), 5.35 (t, 2H), 5.50–5.90 (m, 12H), 7.26 (d, 2H), 7.84 (d, 2H) ppm; <sup>13</sup>C NMR (150 MHz, DMSO- $d_6$ ):  $\delta = 21.5$ , 60.4, 71.8, 72.2, 72.8, 73.1, 73.5, 75.1, 78.5 (C-4'), 81.7, 81.8, 82.0, 102.0, 102.2, 102.4, 128.9, 129.1, 129.9, 142.8, 166.1 ppm. *Mono-3-O-(p-methoxybenzoyl)-β-CD* (**2**,  $C_{50}H_{76}O_{37}$ ) Yield 0.79 g (35 %); ESI–MS: m/z = 1,291 ([M + Na]<sup>+</sup>); <sup>1</sup>H NMR (600 MHz, DMSO- $d_6$ ):  $\delta = 3.20-3.75$  (m, 39H), 3.76 (t, 1H), 3.82 (s, 3H), 3.89 (br, 1H), 4.38–4.61 (m, 7H), 4.74–4.87 (m, 6H), 4.94 (d, 1H), 5.34 (t, 1H), 5.39 (br, 1H), 5.48–6.08 (m, 12H), 7.00 (d, 2H), 7.91 (d, 2H) ppm; <sup>13</sup>C NMR (150 MHz, DMSO- $d_6$ ):  $\delta = 55.8$ , 60.4, 71.5, 71.9, 72.2, 72.5, 72.8, 73.1, 73.4, 75.0, 78.6 (C-4'), 81.8, 82.0, 82.1, 102.1, 102.3, 102.4, 113.9, 122.5, 131.9, 162.8, 165.8 ppm.

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