

Cite this: *Chem. Commun.*, 2011, **47**, 10488–10490

www.rsc.org/chemcomm

COMMUNICATION

Rhodium-catalyzed asymmetric addition of arylboroxines to β -alkoxyacrylate esters†

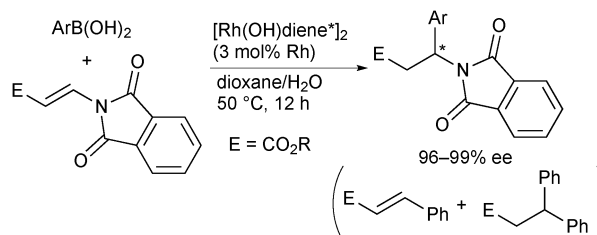
Takahiro Nishimura,* Atsuyuki Kasai, Makoto Nagaosa and Tamio Hayashi*

Received 11th July 2011, Accepted 29th July 2011

DOI: 10.1039/c1cc14150c

Asymmetric addition of arylboroxines to β -alkoxyacrylate esters proceeded in the presence of a rhodium complex coordinated with a chiral diene ligand to give high yields of β -alkoxy- β -arylcarboxylic acid esters with very high enantioselectivity.

Rhodium-catalyzed asymmetric addition of organoboron reagents to α,β -unsaturated carbonyl compounds is a powerful tool to construct a stereogenic carbon center,¹ because a wide variety of aryl- and alkenyl groups can be introduced into the β -position in high yields and high enantioselectivity.² Most studies so far have focused on the addition to α,β -unsaturated carbonyl compounds substituted with alkyl or aryl groups at the β -position, while the addition to those substituted with heteroatoms has been less developed.^{3–5} The rhodium-catalyzed addition of arylboronic acids to α,β -unsaturated carbonyl compounds proceeds *via* an oxa- π -allylrhodium intermediate, which then undergoes hydrolysis to give a hydroarylation product **I** (Scheme 1).⁶ On the other hand, in the addition to α,β -unsaturated carbonyl compounds bearing a strongly electronegative atom such as nitrogen or oxygen at the β -position, β -elimination from the oxa- π -allylrhodium intermediate giving a substitution product **II** becomes a problem as a competitive reaction,⁷ and the preferential hydrolysis of the oxa- π -allylrhodium intermediate is important for the selective formation

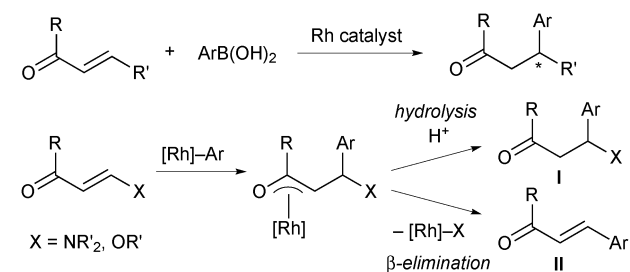


Scheme 2 Asymmetric addition to β -phthaliminoacrylate esters.

of the addition product. Recently, we reported rhodium-catalyzed asymmetric addition of arylboronic acids to β -phthaliminoacrylate esters (Scheme 2). The reaction was successfully carried out by use of a hydroxorhodium/chiral diene catalyst giving β -aryl- β -*N*-phthaloylamino acid esters in high yields with high enantioselectivity, while elimination of phthalimide was observed in the reaction with a bisphosphine ligand (binap) or KOH as a base.⁸

Chiral β -hydroxy- and β -alkoxy carboxylic acid derivatives are important structural components in natural products and pharmaceuticals, and a number of methods to access β -alkoxy carbonyl compounds in a stereoselective manner have been reported in the aldol reaction⁹ and the oxa-Michael reaction.^{10,11} Our straightforward approach to synthesize chiral β -alkoxy carboxylic acid derivatives is focusing on the rhodium-catalyzed asymmetric conjugate arylation of β -alkoxyacrylate esters. Here we report the asymmetric addition of arylboroxines to β -alkoxyacrylate esters, which are readily available from propiolic acid esters and alcohols.¹² The reaction giving chiral β -alkoxy- β -arylcarboxylic acid esters with extremely high enantioselectivity is realized by use of a rhodium/chiral diene catalyst.

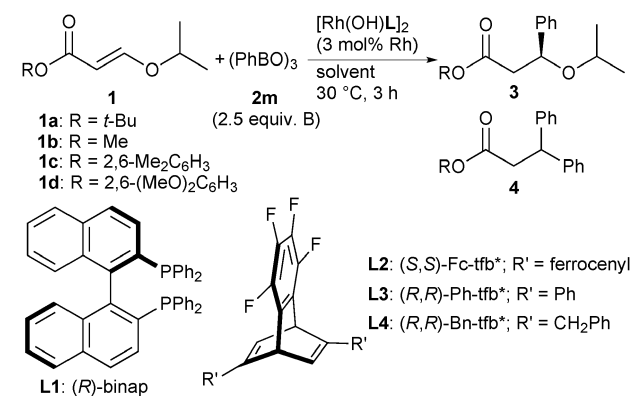
We found that the catalytic activity of a rhodium complex for the reaction of β -alkoxyacrylate esters is higher with a diene ligand than with a bisphosphine ligand (Table 1). Thus, treatment of *tert*-butyl 3-isopropoxypropenoate (**1a**) with phenylboroxine (**2m**) (2.5 equiv. of **B**) in the presence of $[\text{Rh}(\text{OH})((R)\text{-binap})]_2$ (3 mol% of Rh) in 1,4-dioxane/ H_2O (9:1) at 30 °C for 3 h, which is one of the best catalytic conditions in the asymmetric addition of arylboronic acids to α,β -unsaturated carbonyl compounds,⁶ gave the addition product **3am** in 6% yield, where most of the phenylboroxine (**2m**) was consumed to give benzene by protodeborylation (entry 1). The yield of the addition product **3am** was low (5%) even in the reaction in the presence of KOH (40 mol%)



Scheme 1 Rhodium-catalyzed addition to α,β -unsaturated carbonyl compounds.

Department of Chemistry, Graduate School of Science, Kyoto University, Sakyo, Kyoto 606-8502, Japan.
E-mail: tnishi@kuchem.kyoto-u.ac.jp, thayashi@kuchem.kyoto-u.ac.jp;
Fax: +81 75 753 3988; Tel: +81 75 753 3983

† Electronic supplementary information (ESI) available: Experimental procedures and compound characterization data. See DOI: 10.1039/c1cc14150c

Table 1 Rhodium-catalyzed addition of phenylboroxine (**2m**) to **1**^a

Entry	1	L and conditions	Conv. ^b (%)	3 ^b (%)	4 ^b (%)	ee ^c (%)
1	1a	L1 A	8	6 (3am)	<1	— ^d
2 ^e	1a	L1 A	19	5 (3am)	2 (6f)	— ^d
3	1a	cod A	45	27 (3am)	12	—
4	1a	cod B	25	18 (3am)	3	—
5	1a	cod C	38	30 (3am)	2	—
6	1a	L2 C	92	82 (3am)	10	98
7	1b	L2 C	89	86 (3bm)	2	>99.5
8	1c	L2 C	100	90 (3cm)	10	>99.5
9	1d	L2 C	100	97 (3dm)	3	>99.5
10	1d	L3 C	91	78 (3dm)	11	99.2
11	1d	L4 C	87	77 (3dm)	9	94
12	1d	L1 C	<1	<1 (3dm)	0	— ^d
13 ^g	1d	L2 C	100	96 (3dm)	3	>99.5

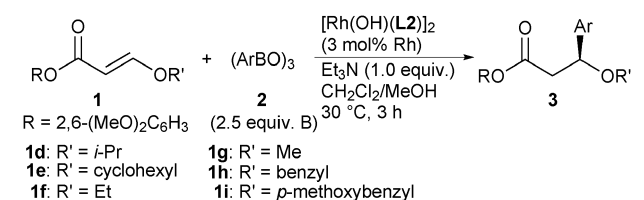
^a Reaction conditions: **1** (0.200 mmol), (PhBO)₃ (**2m**) (0.167 mmol), [Rh(OH)L]₂ (3 mol% of Rh) at 30 °C for 3 h. Conditions A: the reaction in dioxane/H₂O (9:1; 0.8 mL). Conditions B: the reaction in CH₂Cl₂/MeOH (1:1; 0.8 mL). Conditions C: the reaction in CH₂Cl₂/MeOH (1:1; 0.8 mL) in the presence of Et₃N (0.20 mmol). ^b A conversion of **1** and yields of **3** and **4** were determined by ¹H NMR. ^c The ee was determined by HPLC analysis with chiral stationary phase columns. ^d Not determined. ^e Performed in the presence of KOH (40 mol%) at 60 °C. ^f The value in parentheses is an yield of *t*-butyl cinnamate. ^g PhB(OH)₂ (0.50 mmol) was used instead of (PhBO)₃; cod: cycloocta-1,5-diene.

at 60 °C (entry 2). The β-alkoxyacrylate ester was prone to eliminate the alkoxy group during the present rhodium-catalyzed reaction to give *tert*-butyl cinnamate (6%) and *tert*-butyl 3,3-diphenylpropanoate (**4am**: 2%), which is the phenylation product of *tert*-butyl cinnamate formed by elimination of the isopropoxy group. On the other hand, the use of [Rh(OH)(cod)]₂ gave 27% yield of **3am** and 12% of **4am** (entry 3). The selectivity giving the addition product **3am** was higher in the solvent system of CH₂Cl₂/MeOH (1:1) (**3am**/**4am** = 18%/3%) (entry 4), and the addition of triethylamine (1.0 equiv.) increased the chemoselectivity (**3am**/**4am** = 30%/2%) (entry 5). Chiral diene ligands^{13,14} based on tetrafluorobenzobarrelenes (tfb)¹⁵ displayed a high catalytic activity and enantioselectivity. Thus, the reaction of **1a** in the presence of [Rh(OH)((*S,S*)-Fc-tfb*)₂ (**L2**)]₂ (Fc: ferrocenyl)^{15a} (3 mol% of Rh) gave the 1,4-addition product **3am** in 82% yield, whose ee was 98% (entry 6). The formation of 10% yield of the diphenylation product **4am** was also observed. The ester group of **1** had a significant influence on both the reactivity and selectivity (entries 7–9). The addition to methyl ester **1b** gave the addition product **3bm** in 86% yield

with over 99.5% ee and a small amount (2%) of methyl 3,3-diphenylpropanoate (**4bm**) (entry 7). The reactions of 2,6-dimethylphenyl ester **1c** and 2,6-dimethoxyphenyl ester **1d** proceeded with complete conversion to give the corresponding addition products **3cm** and **3dm** in 90 and 97% yield, respectively (entries 8 and 9), where the chemoselectivity was high in the reaction of **1d** (**3dm**/**4dm** = 97%/3%) (entry 9). The enantioselectivities observed for **3cm** and **3dm** were extremely high (>99.5% ee). The reaction was also catalyzed by rhodium complexes coordinated with Ph-tfb* (**L3**) or Bn-tfb* (**L4**), but their catalytic activities and stereoselectivities were lower than those observed with the ferrocene-substituted diene **L2** (entries 10 and 11). The reaction was not catalyzed at all by [Rh(OH)(binap (**L1**))]₂ under the same reaction conditions (entry 12). Phenylboronic acid can be used as well as phenylboroxine (**2m**) to give **3dm** with the same high chemo- and enantioselectivity (entry 13). The absolute configuration of **3dm** obtained with (*S,S*)-**L2** was assigned to be *S* by analogy with (*S*)-**3im** (*vide infra*).

The present catalytic system can be applied to the asymmetric addition of several arylboroxines to β-alkoxyacrylates with all extremely high enantioselectivity (Table 2). Aryl groups (**2m–2t**) having a variety of substituents were successfully introduced at the β position of the β-isopropoxyacrylate **1d** giving the corresponding addition products (**3dm–3dt**) in high yields with over 99.5% ee (entries 1–8). The addition of phenylboroxine (**2m**) to β-alkoxyacrylates where the alkoxy groups are cyclohexyloxy (**1e**), ethoxy (**1f**), methoxy (**1g**), benzyloxy (**1h**), and *p*-methoxybenzyloxy (**1i**) gave the corresponding addition products **3em–3im** with over 99.5% ee (entries 9–13).¹⁶

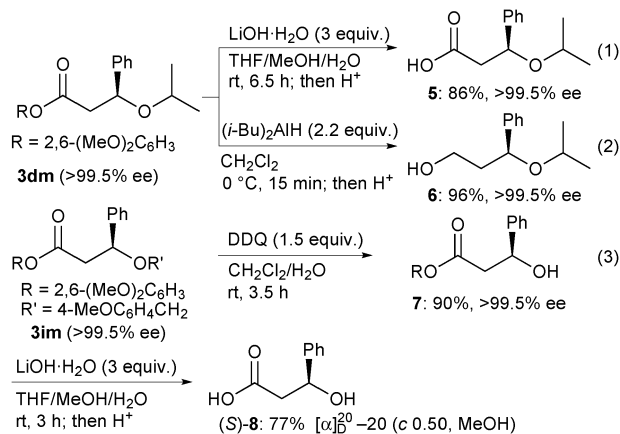
The β-alkoxy-β-arylcarboxylic acid esters obtained here with almost perfect enantioselectivity can be converted into

Table 2 Asymmetric addition of arylboroxine **2** to **1**^a

Entry	1	Ar (2)	Yield ^b (%)	ee ^c (%)
1	1d	Ph (2m)	95 (3dm)	>99.5
2	1d	2-MeC ₆ H ₄ (2n)	91 (3dn)	>99.5
3	1d	3-MeC ₆ H ₄ (2o)	96 (3do)	>99.5
4	1d	4-MeC ₆ H ₄ (2p)	94 (3dp)	>99.5
5 ^d	1d	4-MeOC ₆ H ₄ (2q)	91 (3dq)	>99.5
6 ^e	1d	4-ClC ₆ H ₄ (2r)	90 (3dr)	>99.5
7 ^f	1d	4-CF ₃ C ₆ H ₄ (2s)	68 (3ds)	>99.5
8	1d	2-Naphthyl (2t)	90 (3dt)	>99.5
9	1e	Ph (2m)	92 (3em)	>99.5
10	1f	Ph (2m)	87 (3fm)	>99.5
11	1g	Ph (2m)	75 (3gm)	>99.5
12	1h	Ph (2m)	88 (3hm)	>99.5
13	1i	Ph (2m)	88 (3im)	>99.5

^a Reaction conditions: **1** (0.200 mmol), (PhBO)₃ (**2**) (0.167 mmol), [Rh(OH)(**L2**)]₂ (3 mol% of Rh), Et₃N (0.20 mmol), MeOH (0.40 mL), CH₂Cl₂ (0.40 mL) at 30 °C for 3 h. ^b Isolated yield of **3**. ^c Determined by HPLC analysis with chiral stationary phase columns. ^d Performed with (4-MeOC₆H₄BO)₃ (0.200 mmol). ^e For 12 h. ^f For 24 h.

some functionalized compounds without loss of their enantiomeric purity (eqn (1)–(3)). Thus, basic hydrolysis of **3dm** gave the corresponding β -isopropoxycarboxylic acid **5** in 86% yield (eqn (1)), and the reduction of **3dm** by treatment with (*i*-Bu)₂AlH gave alcohol **6** in 96% yield (eqn (2)). Treatment of β -*p*-methoxybenzyloxycarboxylic acid ester **3im** with 2,3-dichloro-5,6-dicyano-*p*-benzoquinone (DDQ), followed by basic hydrolysis gave β -hydroxycarboxylic acid (*S*)-**8**, whose absolute configuration was determined by comparison of its specific rotation with the value reported previously (eqn (3)).¹⁷



In summary, we have developed a rhodium-catalyzed asymmetric addition of arylboroxines to β -alkoxyacrylate esters giving β -alkoxy- β -arylcarboxylic acid esters in high yields with very high enantioselectivity, which was realized by use of a rhodium/chiral diene catalyst.

Notes and references

- For reviews, see (a) M. P. Sibi and S. Manyem, *Tetrahedron*, 2000, **56**, 8033; (b) N. Krause and A. Hoffmann-Röder, *Synthesis*, 2001, 171; (c) K. Fagnou and M. Lautens, *Chem. Rev.*, 2003, **103**, 169; (d) T. Hayashi and K. Yamazaki, *Chem. Rev.*, 2003, **103**, 2829; (e) S. Darses and J.-P. Genet, *Eur. J. Org. Chem.*, 2003, 4313; (f) J. Christoffers, G. Koripelly, A. Rosiak and M. Rössle, *Synthesis*, 2007, 1279; (g) G. Berthon and T. Hayashi, in *Catalytic Asymmetric Conjugate Reactions*, ed. A. Cordova, Wiley-VCH, Weinheim, 2010, p. 1.
- (a) Y. Takaya, M. Ogasawara, T. Hayashi, M. Sakai and N. Miyaara, *J. Am. Chem. Soc.*, 1998, **120**, 5579; (b) M. Sakai, H. Hayashi and N. Miyaara, *Organometallics*, 1997, **16**, 4229.
- For examples of metal-catalyzed asymmetric 1,4-addition of organometallic reagents to β -silyl α,β -unsaturated carbonyl compounds, see: (a) R. Shintani, K. Okamoto and T. Hayashi, *Org. Lett.*, 2005, **7**, 4757; (b) M. A. Kacprzynski, S. A. Kazane, T. L. May and A. H. Hoveyda, *Org. Lett.*, 2007, **9**, 3187.
- For examples of rhodium-catalyzed asymmetric 1,4-addition of organometallic reagents to dihydropyridones and quinolones, see: (a) R. Shintani, N. Tokunaga, H. Doi and T. Hayashi, *J. Am. Chem. Soc.*, 2004, **126**, 6240; (b) R. Shintani, T. Yamagami, T. Kimura and T. Hayashi, *Org. Lett.*, 2005, **7**, 5317; (c) X. Zhang, J. Chen, F. Han, L. Cun and J. Liao, *Eur. J. Org. Chem.*, 2011, 1443.
- For examples of rhodium-catalyzed 1,4-addition of organoboron reagents to dihydropyranones and chromenones, see: (a) J. Ramnauth, O. Poulin, S. S. Bratovanov, S. Rakhit and S. P. Maddaford, *Org. Lett.*, 2001, **3**, 2571; (b) J. Chen, J. Chen, F. Lang, X. Zhang, L. Cun, J. Zhu, J. Deng and J. Liao, *J. Am. Chem. Soc.*, 2010, **132**, 4552.
- T. Hayashi, M. Takahashi, Y. Takaya and M. Ogasawara, *J. Am. Chem. Soc.*, 2002, **124**, 5052.
- In most examples of addition of carbon nucleophiles to β -alkoxy- α,β -unsaturated carbonyl compounds, the formation of substitution products or double addition products, which derived from the substitution products, is reported, see: (a) G. A. Molander and H. C. Brown, *J. Org. Chem.*, 1977, **42**, 3106; (b) M. G. Gorbunova, I. I. Germ and V. P. Kukhar, *J. Fluorine Chem.*, 1999, **65**, 25; (c) M. Bergdahl, M. Eriksson, M. Nilsson and T. Olsson, *J. Org. Chem.*, 1993, **58**, 7238; (d) N. Ito and T. Etoh, *J. Chem. Soc., Perkin Trans. 1*, 1996, 2397; (e) G. Bartoli, M. Bartolacci, M. Bosco, G. Foglia, A. Giuliani, E. Marcantoni, L. Sambri and E. Torregiani, *J. Org. Chem.*, 2003, **68**, 4594; (f) W. Wang and T. Ikemoto, *Tetrahedron Lett.*, 2005, **46**, 3875; (g) Y. Liu, K. Bakshi, P. Zavalij and M. P. Doyle, *Org. Lett.*, 2010, **12**, 4304.
- T. Nishimura, J. Wang, M. Nagaosa, K. Okamoto, R. Shintani, F. Kwong, W. Yu, A. S. C. Chan and T. Hayashi, *J. Am. Chem. Soc.*, 2010, **132**, 464.
- For recent reviews of the aldol reaction, see: (a) N. Kumagai, *Chem. Pharm. Bull.*, 2011, **59**, 1; (b) J. Li and D. Menche, *Synthesis*, 2009, 2293; (c) T. Brodmann, M. Lorenz, R. Schäckel, S. Simsek and M. Kalesse, *Synlett*, 2009, 174; (d) L. M. Geary and P. G. Hultin, *Tetrahedron: Asymmetry*, 2009, **20**, 131; (e) G. Guillena, C. Najera and D. J. Ramón, *Tetrahedron: Asymmetry*, 2007, **18**, 2249; (f) B. Schetter and R. Mahrwald, *Angew. Chem., Int. Ed.*, 2006, **45**, 7506; (g) C. Palomo, M. Oiarbide and J. M. García, *Chem. Soc. Rev.*, 2004, **33**, 65.
- For a review, see: C. F. Nising and S. Bräse, *Chem. Soc. Rev.*, 2008, **37**, 1218.
- For selected examples of straightforward asymmetric synthesis of β -alkoxy carbonyl compounds, see: (a) T. Kano, Y. Tanaka and K. Maruoka, *Tetrahedron*, 2007, **63**, 8658; (b) B. Checa, E. Gálvez, R. Parelló, M. Sau, P. Romea, F. Urpí, M. Font-Bardía and X. Solans, *Org. Lett.*, 2009, **11**, 2193; (c) A. Cosp, P. Romea, F. Urpí and J. Vilarrasa, *Tetrahedron Lett.*, 2001, **42**, 4629.
- M.-J. Fan, G.-Q. Li and Y.-M. Liang, *Tetrahedron*, 2006, **62**, 6782.
- For reviews of chiral diene ligands, see: (a) R. Shintani and T. Hayashi, *Aldrichimica Acta*, 2009, **42**, 31; (b) C. Defieber, H. Grützmaier and E. M. Carreira, *Angew. Chem., Int. Ed.*, 2008, **47**, 4482.
- For selected examples, see: (a) T. Hayashi, K. Ueyama, N. Tokunaga and K. Yoshida, *J. Am. Chem. Soc.*, 2003, **125**, 11508; (b) Y. Otomaru, K. Okamoto, R. Shintani and T. Hayashi, *J. Org. Chem.*, 2005, **70**, 2503; (c) Y. Otomaru, A. Kina, R. Shintani and T. Hayashi, *Tetrahedron: Asymmetry*, 2005, **16**, 1673; (d) K. Okamoto, T. Hayashi and V. H. Rawal, *Chem. Commun.*, 2009, 4815; (e) J.-F. Paquin, C. Defieber, C. R. J. Stephenson and E. M. Carreira, *J. Am. Chem. Soc.*, 2005, **127**, 10850; (f) F. Läng, F. Breher, D. Stein and H. Grützmaier, *Organometallics*, 2005, **24**, 2997; (g) S. Helbig, S. Sauer, N. Cramer, S. Laschat, A. Baro and W. Frey, *Adv. Synth. Catal.*, 2007, **349**, 2331; (h) Z.-Q. Wang, C.-G. Feng, M.-H. Xu and G.-Q. Lin, *J. Am. Chem. Soc.*, 2007, **129**, 5336; (i) T. Noël, K. Vandyck and J. Van der Eycken, *Tetrahedron*, 2007, **63**, 12961; (j) T. Gendrineau, O. Chuzel, H. Eijsberg, J.-P. Genet and S. Darses, *Angew. Chem., Int. Ed.*, 2008, **47**, 7669; (k) X. Hu, M. Zhuang, Z. Cao and H. Du, *Org. Lett.*, 2009, **11**, 4744; (l) M. K. Brown and E. J. Corey, *Org. Lett.*, 2010, **12**, 172; (m) Y. Luo and A. J. Carnell, *Angew. Chem., Int. Ed.*, 2010, **49**, 2750; (n) G. Pattison, G. Piraux and H. W. Lam, *J. Am. Chem. Soc.*, 2010, **132**, 14373.
- (a) T. Nishimura, H. Kumamoto, M. Nagaosa and T. Hayashi, *Chem. Commun.*, 2009, 5713; (b) T. Nishimura, Y. Ichikawa, T. Hayashi, N. Onishi, M. Shiotsuki and T. Masuda, *Organometallics*, 2009, **28**, 4890; (c) T. Nishimura, T. Kawamoto, M. Nagaosa, H. Kumamoto and T. Hayashi, *Angew. Chem., Int. Ed.*, 2010, **49**, 1638; (d) R. Shintani, M. Takeda, T. Nishimura and T. Hayashi, *Angew. Chem., Int. Ed.*, 2010, **49**, 3969; (e) T. Nishimura, Y. Maeda and T. Hayashi, *Angew. Chem., Int. Ed.*, 2010, **49**, 7324; (f) T. Nishimura, Y. Yasuhara, T. Sawano and T. Hayashi, *J. Am. Chem. Soc.*, 2010, **132**, 7872; (g) T. Nishimura, H. Makino, M. Nagaosa and T. Hayashi, *J. Am. Chem. Soc.*, 2010, **132**, 12865.
- The reaction of a β -aryloxyacrylate, 2,6-dimethoxyphenyl 3-(2-methylphenyloxy)propenoate, with phenylboroxine (**2m**) gave the cinnamate ester (14%) and diphenylpropanoate (85%) under the same reaction conditions.
- O. Pàmies and J.-E. Bäckvall, *Adv. Synth. Catal.*, 2002, **344**, 947.