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Synthesis and functional studies of THF-gramicidin hybrid ion channels †

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THF–gramicidin hybrids **2–4** with the L-THF amino acid **1** in positions 11 and 12 and compounds **5–8** with the D-THF amino acid **ent-1** in positions 10 and 11 were synthesized and their ion channel properties were studied by single-channel-current analysis. The replacement of positions 11 and 12 by the L-THF amino acid **1** gave a strongly reduced channel performance. In contrast, replacement of positions 10 and 11 by the D-THF amino acid **ent-1** gave rise to new and interesting channel properties. For the permeability ratios, the ion selectivity shifts from Eisenman I towards Eisenman III selectivity and the channels display ms-dynamics. Most remarkable is the asymmetric compound **8**, which inserts selectively into a DPhPC membrane and displays voltage-directed gating dynamics.

Introduction

Biological ion channels transport cations or anions through the phospholipid bilayer of biological membranes.¹ The channel transport is a passive process driven by a concentration gradient or a membrane potential. The control of biochannels (opening/closing) occurs *via* ligand gating, voltage gating, mechanosensitive response or temporary covalent modification. Recent progress in X-ray crystallography has given much insight into the structure and function of biological potassium channels and chloride channels.² Genetically encoded channel malfunction can cause channelopathies leading to cardiovascular, pulmonary or neuronal diseases.³

The design, synthesis and functional analysis of synthetic ion channels contributes to the understanding of channel function and control.⁴ A long term objective for synthetic ion channels is their implantation into biological systems.⁵ Another possible application lies in the area of sensing and nanoelectric devices.⁶

As part of an ongoing project on synthetic ion channels we focus on several objectives:

- the correlation of the channel function with structural data for the channel-active conformation;⁷
 - the use of molecular switches for channel gating;8
 - ion-selectivity;9
 - implantation of synthetic channels into biological systems;⁵
- the control of transversal asymmetry for channel insertion and function. 10

The naturally occuring, ion channel-active gramicidin A (gA) is a suitable lead structure for the design of synthetic channels. ¹¹ gA is a pentadecapeptide with an *N*-formyl and an aminoethanol *C*-terminus.

Its alternative sequence of L- and D-amino acids favours β -helical secondary structures. ¹² In a membrane, two gA monomers form a head-to-head hydrogen-bridged dimer, which

† Electronic supplementary information (ESI) available: HPLC-traces and ¹H-NMR spectra for the THF-gA-hybrids and Table S1: conductances of compounds **5**, **6**, **7** and gramicidin A for monovalent cations. See http://www.rsc.org/suppdata/ob/b3/b303249n/

acts as an ion channel for monovalent cations. The structure of the channel-active conformation was determined as a right-handed single-stranded β -helix with 6.3 residues per turn by solid state-NMR. 12

A covalent linkage of the two *N*-termini *via* a succinyl bridge blocks the dissociation of the channel-active dimer and opens the way for studying unimolecular channels.¹³ When synthetic building blocks are coupled with the gA-motif, hybrid channels result which show new functionality (photoswitching with azobenzene,¹⁴ ion selectivity with cyclohexyl ether amino acids⁹).

The tetrahydrofuran amino acids (THF amino acids) 1 and ent-1 insert an ether oxygen atom into the channel architecture. ¹⁵⁻¹⁷ The gA-channel entrance is formed by the first turn of the β-helix which consists of D-Leu10-Trp11-D-Leu12-Trp13-D-Leu14-Trp15. This is the place where the water molecules of the cation hydration shell are removed until two water molecules per cation remain for the single-file region of the channel interior. ¹⁸ The exchange of two amino acids of the entrance region for a THF amino acid leads to an alternation of the available cation binding sites (amide—ether oxygen). Two possible modifications of the gA-entrance sequence I were chosen for closer examination (Scheme 1):

- replacement of Trp11-D-Leu12 with the L-THF amino acid 1 resulting in the sequence II;
- replacement of D-Leu10-Trp11 with the D-THF amino acid **ent-1** resulting in the sequence **III**.

Both replacements occur with retention of stereochemical information at the α carbon atoms of the amino acids. Here, we report on the synthesis and functional studies of the following series of THF-gA hybrids. The replacement by the L-THF amino acid 1 results in the target compounds 2-4 (Fig. 1).

The replacement by the D-THF amino acid ent-1 leads to the target structures 5–8 (Fig. 2).

Results and discussion

The stereoselective synthesis of the THF amino acids 1 and ent-1 used L- and D-alanine as enantiopure starting materials. ¹⁵ The synthetic route is exemplified in Scheme 2 for the preparation of compound 13, the *N*-Boc-protected form of ent-1.

Reduction of *N*-Boc-D-alanine methyl ester **9** gave the aldehyde **10**, which was allowed to react with buten-1-yl lithium cuprate to produce the alcohol **11** (diastereoselectivity 20 : 1,

chelation control). Epoxidation of the terminal double bond and a subsequent acid-mediated intramolecular epoxide opening gave, after chromatographic separation, the D-cis-THF alcohol (2R,5S)-12 and the corresponding trans-THF alcohol together with the minor stereoisomers from the cuprate addition. A one-step oxidation of (2R,5S)-12 led to the desired N-Boc-protected D-cis-THF amino acid 13 (Scheme 2). The synthesis of the target peptides was accomplished by fragment coupling in solution. Although protocols for the solid-phase synthesis of gA are available, 19 we chose the fragment-coupling strategy 20 for the following reasons: higher purity, larger quantities, modular approach with predictable diversity.

The retrosynthetic analysis of the target peptides is summarized in Scheme 3. All four monomers 2, 3, 5, and 6 share the common tetramer 14. Thus, disconnection between D-Leu4 and Ala5 leads to the decamer 15 containing the L-THF amino acid

Scheme 2 (a) DIBAL-H (2.0 eq.), CH₂Cl₂–n-hexane 30 : 1, -60 °C; (b) (i) lithiation of 1-bromo-3-butene (4.5 eq.): t-BuLi (9.9 eq.), Et₂O–n-hexane 1 : 1, -78 °C; (ii) cuprate-addition: 1-lithium-3-butene (4.5 eq.), CuBr·SMe₂ (2.2 eq.), Et₂O–n-hexane 2 : 1, -40 °C \longrightarrow -80 °C; (c) MCPBA (2.0 eq.), CH₂Cl₂, 0 °C \longrightarrow rt; (d) CSA (0.04 eq.), CH₂Cl₂, rt; (e) iodobenzene diacetate (2.2 eq.), 2,2,6,6-tetramethylpiperidin-1-yloxyl free radical (0.05 eq.), CH₂Cl₂, rt.

Boc-D-THF-OH

1 and the decamer 16 which includes the D-THF amino acid ent-1.

The covalently linked dimers 4, 7, and 8 were assembled from the two building blocks 17 and 18, as well as the decamer 15, the decamer 16 and the undecamer 19. The use of the tetramer clamp 17 results in final Gly–Ala coupling steps, which in our hands proved to be more efficient than the previously reported

Z-A-v-V-v-W-I-L-THF-W-I-W
$$\stackrel{H}{\longrightarrow}$$
 OR

15

14

Z-A-v-V-v-W-D-THF-I-W-I-W $\stackrel{H}{\longrightarrow}$ OR

16

R = TBDPS

2 $\stackrel{\longrightarrow}{\longrightarrow}$ 3 $\stackrel{\longrightarrow}{\longrightarrow}$ 14 + 15

6 $\stackrel{\longrightarrow}{\longrightarrow}$ 5 $\stackrel{\longrightarrow}{\longrightarrow}$ 14 + 16

V-G-OMe

V-G-OMe

V-G-OBn

17

18

19

R = TBDPS

4

17 + 2 x 18 + 2 x 15

7

17 + 2 x 18 + 2 x 16

8

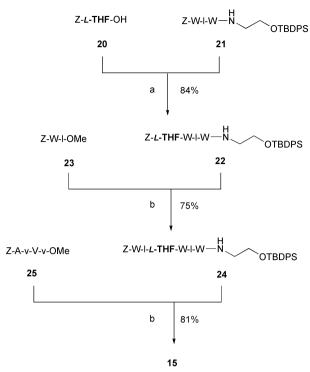
17 + 2 x 18 + 19 + 16

succinyl–Val or tartrate–Val couplings for the synthesis of covalently linked gA dimers. 13,20 The lipophilic TBDPS-group at the aminoethanol terminus increases the solubility in organic solvents and eases chromatographic purification. Z/OMe protections were used throughout the synthesis due to their mild deprotection conditions and the possibility of leaving the Trp side chain unprotected.

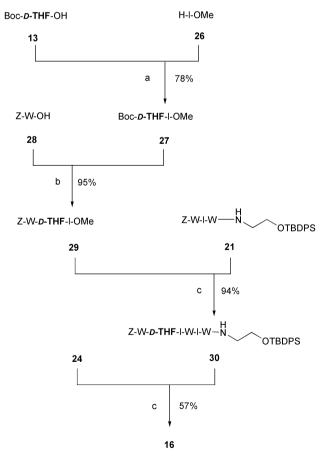
The synthesis of the decamer 15 containing the L-THF 1 utilizes HBTU–HOBt couplings in CH₂Cl₂ or CH₂Cl₂–DMF to assemble the peptide sequence (Scheme 4). Reaction of Z-L-THF-OH 20 with the ethanolamine-tagged tripeptide 21 ²⁰ resulted in the tetramer 22. A detailed discussion of the proper choice of deprotection conditions (H₂, Pd/C and LiOH–THF–MeOH) is given in our earlier work on minigramicidins.²⁰ Coupling of 22 with the dipeptide 23 ²⁰ gave the hexamer 24, which was further elaborated using the tetramer 25 ²⁰ to yield the desired decamer 15.

The decamer 16 containing the D-THF ent-1 was synthesized starting with the coupling of Boc-D-THF-OH 13 and H-D-Leu-OMe 26 to the dimer 27 (Scheme 5). Addition of Z-Trp-OH 28 led to 29. The latter was coupled with the ethanolamine-tagged tripeptide 21 to produce the hexamer 30. The conjugation of 24 and 30 completed the assembly of the decamer 16.

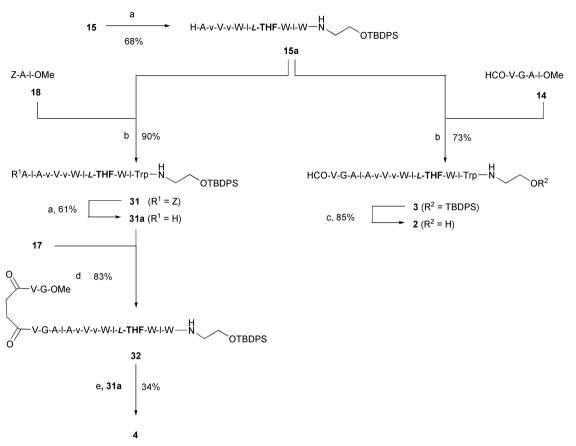
To proceed towards the L-THF-gA hybrids 2, 3, and 4 (Scheme 6), the decamer 15 was N-deprotected by hydrogenolysis and the resulting amine 15a was coupled with the N-formyl tetramer 14 to yield compound 3 in 73% yield. Cleavage of the TBDPS-group in 3 with 5% HF in CH₃CN gave 2. Treatment of the methyl ester 18 with LiOH resulted in the corresponding carboxylic acid, which was coupled with the amine 15a to give the dodecamer 31. After hydrogenolytic clevage of the Z-group the amine 31a was obtained and subsequently conjugated with the tetrameric clamp 17 to produce compound 32. Hydrolysis of the methyl ester in 32 and



Scheme 4 (a) *Z*-group cleavage: Pd/C, H_2 (1 bar), MeOH, rt; coupling: EDC (1.5 eq.), HOBt (3.0 eq.), (*i*-Pr)₂EtN (3.0 eq.), CH₂Cl₂, 0 °C \rightarrow rt; (b) methyl ester deprotection: 2.5 eq LiOH, THF \rightarrow H₂O 3 : 1, 0 °C; *Z*-group cleavage: Pd/C, H_2 (1 bar), MeOH, 40 °C; coupling: HBTU (1.5 eq.), HOBt (3.0 eq.), (*i*-Pr)₂EtN (3.0 eq.), CH₂Cl₂ \rightarrow DMF 5 : 1, \rightarrow 15 °C.



Scheme 5 (a) EDC (1.5 eq.), HOBt (3.0 eq.), $(i\text{-Pr})_2\text{EtN}$ (3.0 eq.), CH₂Cl₂, 0 °C—rt; (b) *Boc*-group cleavage: TFA, CH₂Cl₂, rt; coupling: EDC (1.5 eq.), HOBt (3.0 eq.), $(i\text{-Pr})_2\text{EtN}$ (3.0 eq.), CH₂Cl₂, 0 °C—rt; (c) methyl ester deprotection: 2.5 eq LiOH, THF–H₂O 3: 1, 0 °C; *Z*-group cleavage: Pd/C, H₂ (1 bar), MeOH, 40 °C; coupling: HBTU (1.5 eq.), HOBt (3.0 eq.), $(i\text{-Pr})_2\text{EtN}$ (3.0 eq.), CH₂Cl₂–DMF 5: 1, -15 °C.



Scheme 6 (a) Pd/C, H_2 (1 bar), EtOH, 50 °C; (b) methyl ester deprotection: 2.5 eq LiOH, THF- H_2 O 3 : 1, 0 °C; coupling: HBTU (1.5 eq.), HOBt (3.0 eq.), (*i*-Pr)₂EtN (3.0 eq.), CH₂Cl₂-DMF 5 : 1, -15 °C; (c) HF (5% in CH₃CN), CH₂Cl₂-DMF 10 : 1, rt; (d) benzyl ester deprotection: Pd/C, H_2 Cl bar), MeOH, rt; coupling: HBTU (1.5 eq.), HOBt (3.0 eq.), (*i*-Pr)₂EtN (3.0 eq.), CH₂Cl₂-DMF 5 : 1, -15 °C; (e) methyl ester deprotection: 2.5 eq LiOH, THF- H_2 O 3 : 1, 0 °C; coupling: HATU (1.5 eq.), HOAt (3.0 eq.), (*i*-Pr)₂EtN (3.0 eq.), CH₂Cl₂-DMF 5 : 1, -15 °C.

coupling of the resulting carboxylic acid with the amine 31a led to the symmetrical bis-L-THF channel hybrid 4.

The decamer 16 was converted into the free amine 16a and coupled with the carboxylic acid prepared from the tetrameric methyl ester 14 (Scheme 7). The desired monomeric D-THF-gA hybrid 6 could be obtained in 87% yield. Deprotection of the silyl ether led to compound 5. In order to get to the linked dimers containing the D-THF amino acid ent-1, the methyl ester 18 was transformed into the corresponding carboxylic acid and coupled with the amine 16a to yield the dodecamer 33 and its Z-deprotected form 33a, respectively. The benzyl ester 17 could be converted into the free acid and coupled with the amine 33a to produce compound 34. Hydrolysis of the methyl ester in 34 and subsequently coupling with a second molecule of 33a gave the symmetrically bis-D-THF channel hybrid 7. After deprotection of the methyl ester in 35,20 the coupling of the resulting carboxylic acid with the amine 33a produced the unsymmetrical mono-D-THF channel hybrid 8.

All gA-hybrid channels **2–8** were purified by RP-HPLC and characterized by ¹H-NMR and MS (see experimental section and electronic supplementary information (ESI)†). The characterisation of ion channels by single-channel currents is a single-molecule technique and has to fulfil high-purity criteria in order to avoid misinterpretation of channel-active impurities. The advanced synthetic intermediates of Schemes 6 and 7 were all tested routinely for ion channel activity. HPLC-separation was additionally monitored by single-channel current analysis.

For the functional analysis of the ion channels, planar phospholipid bilayers (black-lipid membranes, Fig. 3)¹ were used. If not stated otherwise, soybean lecithin was the phospholipid of choice. The channel formers were added from a methanol stock solution to the *cis* and/or the *trans* side of the measurement chamber. Details are given in the experimental section.

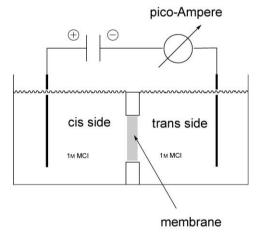
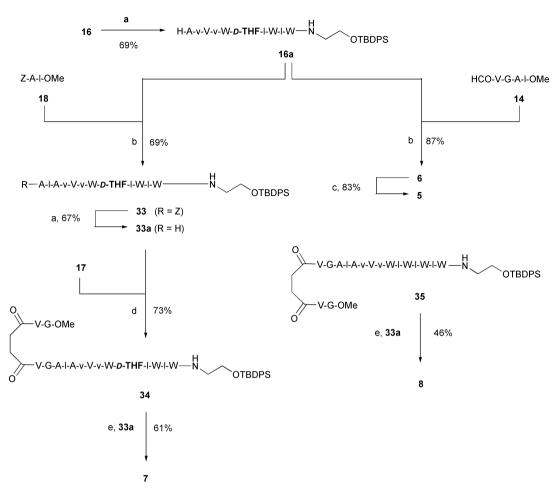


Fig. 3 Experimental setup for single-channel measurements in planar phospholipid bilayers.

The replacement of Trp11-D-Leu12 of gA with the L-THF amino acid 1 resulted in a considerable change in the ion channel properties. Selected traces are shown in the upper part of Fig. 4. For the monomeric hybrids 2 and 3 it was necessary to add the compounds in high concentrations from the *cis* and the *trans* side to see any channels at all. Conductivities were reduced to 10% compared with gA. The observation of single channel currents was restricted to Cs⁺. For compounds 2 and 3 multiple noisy events were found. The covalently linked dimer 4 showed better resolved channel signals at lower concentrations (Fig. 4c). Here, it was sufficient to add the compound from the *cis* side only. The channel conductivity of 4 was still low and allowed the detection of Cs⁺ currents only. Thus, the use of the L-THF-amino acid 1 in positions 11 and 12 of gA reduces the transport efficiency of the gA channel strongly.



Scheme 7 (a) Pd/C, H_2 (1 bar), EtOH, 50 °C; (b) methyl ester deprotection: 2.5 eq LiOH, THF– H_2O 3 : 1, 0 °C; coupling: HBTU (1.5 eq.), HOBt (3.0 eq.), (*i*-Pr)₂EtN (3.0 eq.), CH₂Cl₂–DMF 5 : 1, -15 °C; (c) HF (5% in CH₃CN), CH₂Cl₂–DMF 10 : 1, rt; (d) Pd/C, H_2 (1 bar), EtOH, 50 °C; coupling: HBTU (1.5 eq.), HOBt (3.0 eq.), (*i*-Pr)₂EtN (3.0 eq.), CH₂Cl₂–DMF 5 : 1, -15 °C; (e) methyl ester deprotection: 2.5 eq LiOH, THF– H_2O 3 : 1, 0 °C; coupling: HATU (1.5 eq.), HOAt (3.0 eq.), (*i*-Pr)₂EtN (3.0 eq.), CH₂Cl₂–DMF 5 : 1, -15 °C.

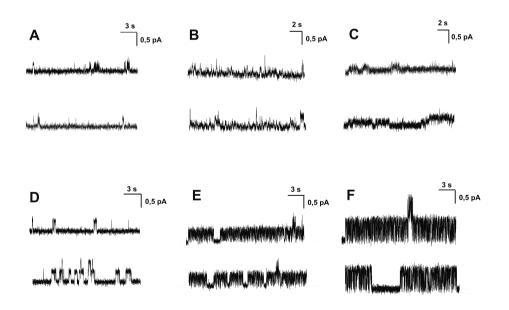


Fig. 4 Channel traces of THF-gA-hybrids. The upper part shows the results from the replacement of positions 11 and 12 by the L-THF amino acid 1(A: 2, B: 3, C: 4); the lower part shows the results from the replacement of positions 10 and 11 by the D-THF amino acid ent-1 (D: 5, E: 6, F: 7) in asolectin (1 M CsCl, +170mV).

The effect of the replacement of D-Leu10-Trp11 with the D-THF amino acid **ent-1** was examined next. The monomeric D-THF hybrids **5** and **6** as well as the symmetric, dimeric D-THF-gA hybrid **7** show single channel behavior for Cs⁺, Rb⁺, K⁺ and Na⁺ ions (Table 1). The conductivities were much lower

for 5, 6, and 7 than for gA but the channels were uniform and well defined.

Representative traces are shown in the lower part of Fig. 4. The conductivity series $Cs^+ > Rb^+ > K^+ > Na^+$ is similiar to that for gA (see electronic supplementary information (ESI)†).

Table 1 Single-channel conductances of compounds **7**, **8** and succ(gA-TBDPS)₂ for different cations

	Conductance/pS		
Compound	1 м CsCl	1 м KCl	1 м NaCl
8	5.7	2.2	0.6
7	7.0	1.8	0.6
Succ(gA-TBDPS) ₂	19.1	14.0	7.7

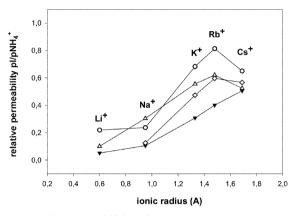


Fig. 5 Relative permeabilities of compounds 5, 6, 7 and gA for monovalent cations in relation to the ionic radius of the cation; $-\bigcirc$ 6; $-\Delta$ 7; $-\Phi$ 5; $-\Psi$ gA.

The permeability ratios relative to NH_4^+ reveal a tendency from an Eisenman I towards an Eisenman III selectivity with $Rb^+ > Cs^+$ (see Fig. 5). This behavior is most pronounced for compound 6 and indicates the presence of the donor ether oxygen as a new binding site in the channel.

The single channel events for **6** and **7** display dynamic behavior in the millisecond range (openings and closings: 10–20 ms). This occurs with the monomeric compound **6** and the dimeric compound **7**. Therefore, in the case of **6**, the dimer formation and dissociation cannot be the reason for this type of channel opening and closing. Its cause is the presence of the THF amino **ent-1**, which leads to a type of conformational gating.

The asymmetric compound 8 in its extended β-helical conformation can be schematically described as a cylinder-type structure with a gramicidin (grami) end and a THF end. In Figs. 6-8 the THF end is depicted in red and the grami end in blue. If compound **8** is added from the *cis* side to the membrane there are two modes of insertion possible: a THF-end-first mode and a grami-end-first mode. If the insertion occurs randomly, two types of channel, A and B, should be possible. Let us assume that both channels are different in functionality, so that both are distinguishable by their current traces. In channel type A the cations enter the channel through the grami end and leave it through the THF end. In channel B the cations enter the channel through the THF end and leave it through the grami end. The entrance and the exit site are determined by the membrane potential. Independent of the membrane potential one expects the occurence of both channel types, A and B, if the insertion occurs randomly (Fig. 6). If the insertion occurs selectively, one type of channel either A or B is possible (Fig. 7). A positive potential at the cis side should lead to the observation of channel type A (Fig. 7A). After change of the potential the channel type B should be observed (Fig. 7B).

The addition of compound **8** from the *cis* side to a diphytanoyl phosphatidyl choline (DPhPC) membrane with 1 m CsCl and a membrane potential of +160 mV at the *cis* side gave rise to only one type of channel with a very long dwell time (type 1, Fig. 8A). Change of the potential to -160 mV gave rise to a second type of channel (type 2, Fig. 8B) with a long overall dwell time but with the same short closings and openings that

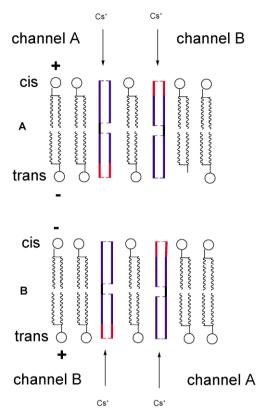


Fig. 6 Two channel types A and B resulting from random insertion of 8 into a membrane.

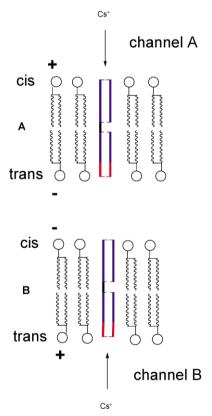


Fig. 7 One channel type either A or B resulting from selective insertion of 8 into a membrane.

were observed for compounds 6 and 7. Switching the potential back to +160 mV gave rise to the first channel type. This behavior indicates a selective insertion of compound 8 into the membrane. A control experiment to support this hypothesis was done by the addition of compound 8 from the *trans* side. This time, at a membrane potential of +160 mV, the type 2

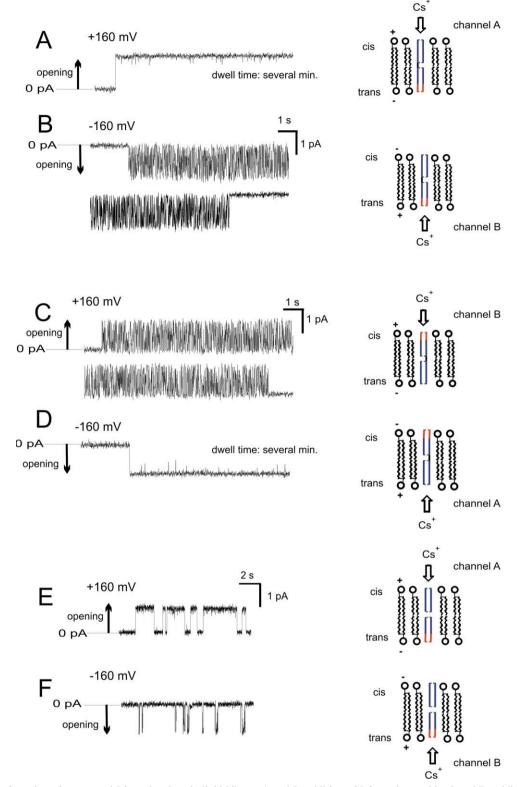


Fig. 8 Selective insertion of compound 8 into the phospholipid bilayer. A and B: addition of 8 from the *cis* side; C and D: addition of 8 from the *trans* side, E and F: addition of gA from the *cis* side and 5 from the *trans* side. Current traces were recorded with 1 M CsCl and DPhPC membranes.

channel (Fig. 8C) and at -160 mV the type 1 channel (Fig. 8D) were observed.

Having proven that a selective insertion had occurred, the question of a *THF-first* or a *grami-first* insertion was addressed next. Towards this end, the monomer compound 5 was added to the *trans* side and gA to the *cis* side. By this token, a predictable *trans* orientation of the THF end and a *cis* orientation of the grami end were accomplished. Mixed 5–gA channels were formed that showed single channel behavior as depicted in Figs. 7 E and F. Because of the lack of covalent succinyl linker,

the dwell times were shorter, but the characteristics in the channel dynamics observed in Figs. 7A–D were found again. If the cations enter the channel from the grami end, stable grami-type channels are formed (Fig. 8E). If the cations enter from the THF end, channels with ms-dynamics are formed (Fig. 8F). It has to be noted, that after 30 min of experiments no transversal mixing was noticeable. The TBDPS-deprotected compounds 5 and gA gave reproducible mixed channels, while their protected counterparts 6 and gA-TBDPS caused trans-bilayer motion (flip/flop) problems due to their higher

lipophilicity resulting in the gA-TBDPS dimer as the major detectable species.

The asymmetric orientation leads to a potential-dependent function, which can be compared with the voltage gating of biological channels. The ms-dynamics (short closings and openings) occur if a positive potential is applied at the THF end. A type of potential-driven conformational switching (36 \rightarrow 37) around the THF building block in the gA sequence may cause this behavior. As depicted in Fig. 9 the positive potential next to the THF end may lead a switch into a partial β -turn, 15 channel-inactive form of 37.

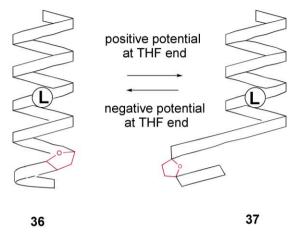


Fig. 9 Schematic representation of a potential-driven conformational switch from an active conformation **36** into an inactive conformation **37**.

The voltage dependence of the gating dynamics for compound 8 was studied over a broader potential range between +200 mV and -200 mV (Fig. 10). Due to the low conductivities in the range between +50 mV and -50 mV no data could be collected in this region. The data show that, for negative potentials, the long lived channel of type A dominates. In the positive range of channel type B a decrease in the dwell time with an increase of the potential is observed.

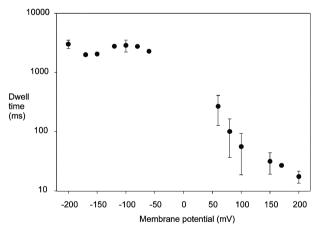


Fig. 10 Voltage dependence of the dwell time for 8 in DPhPC and $1\ \text{M}$ CsCl.

No significant difference between positive and negative potential was observed for the conductivity of **8**. There is a rectifying effect on the channel-gating dynamics, but not on the channel conductivities. Single-channel currents for Cs⁺, K⁺ and Na⁺ are summarized in Table 1. Succ(gA-TBDPS)₂²⁰ was used as a reference compound. The conductivities of **7** and **8** are of comparable size and smaller than for succ(gA-TBDPS)₂. with an increased selectivity for Cs⁺ *versus* K⁺ and Na⁺.

For a detailed understanding of the potential-dependent function of **8** it will be necessary to perform a detailed

structural NMR-analysis of compound **8** in a membrane-like environment with different electrical potentials. This is a challenging problem which hopefully can be solved. Data from other examples of potential-dependent channel tranport by Matile ^{4f} and Kobuke ^{4h} will be useful for comparison. The structural reasons for the selective insertion of **8** into the membrane also remain to be clarified.

Conclusion

In summary, it was found that THF amino acids are valuable building blocks to modulate the ion channel activity of gA. The choice of the THF amino acid and the positions of replacement are crucial. The replacement of positions 11 and 12 by the L-THF amino acid 1 leads to compounds 2, 3, and 4 with a drastically reduced channel performance. In contrast, replacement of positions 10 and 11 by the p-THF amino acid ent-1 gave rise to componds 5–8 with new and interesting channel properties. The ion selectivity shifts from Eisenman I towards Eisenman III selectivity and the channels display ms-dynamics. Most remarkable is the asymmetric compound 8, which inserts selectively into a DPhPC membrane and displays voltage-dependent gating dynamics.

Experimental

Analytical techniques

Single channel measurement. Planar lipid membranes were prepared by painting a solution of soybean lecithin (40%, Avanti Polar Lipids, Alabaster, AL) or diphytanoylphosphatidylcholin (DPhPC, Avanti Polar Lipids, Alabaster, AL) in *n*-decane (25 mg ml $^{-1}$) over the aperture of a polystyrene cuvette with an inner diameter of 0.15 mm. All experiments were performed at ambient temperature. The used electrolyte solutions at a concentration of 1 m each were unbuffered. Probes of monomolecular channels, dissolved in methanol, were added to the trans or cis side (containing the measuring electrode) of the cuvette. Probes of bimolecular channels, dissolved in methanol, were added to both trans and cis sides of the cuvette. Current detection and recording was performed with a patch-clamp amplifier Axopatch 200B, a Digidata A/D converter and pClamp6 software (Axon Instruments, Foster City, MA). The acquisition frequency was 5 kHz. The data were filtered with a digital filter at 50 Hz for further analysis.

NMR-spectroscopy. NMR spectroscopy was carried out on Bruker AMX300, DPX600, AMX500 and AX400 instruments. All resonances are referenced to residual solvent signals (CDCl₃: 7.25 ppm, DMSO-*d*₆: 2.49 ppm).

HPLC. HPLC was carried out on a Rainin Dynamax HPLC-System, DA-Detector. Column: Jupiter C5 250 \times 4 mm. Method 1 (M1): 65%—85% B in 25 min, A: H₂O; B: acetonitrile–isopropanol 2:1; 0.7 ml min⁻¹, 30 °C. Method 2 (M2): 60%—100% B in 20 min, A: H₂O + 0.1% TFA; B: acetonitrile–isopropanol 2:1 + 0.1% TFA; 0.7 ml min⁻¹, 30 °C. Method 3 (M3) 55%—65% B in 20 min, A: H₂O; B: acetonitrile–isopropanol 2:1; 0.7 ml min⁻¹, 30 °C. Method 4 (M4): 70%—100% B in 25 min, A: H₂O + 0.1% TFA; B: acetonitrile–isopropanol 2:1 + 0.1% TFA; 0.7 ml min⁻¹, 30 °C.

General procedures

The term "aqueous work-up" refers to an extraction of the organic layer with 1 M NaHSO₄, sat. NaHCO₃ and brine consecutively (a third of the volume of the organic layer each), followed by drying over MgSO₄ and evaporation. Thin layer chromatography (TLC) and flash column chromatography (FCC) were performed on silica gel (Merck Si60, 40–60 μm). Columns for FCC were slurry-packed. TLC-spots were visual-

ised by UV-light and by staining with 2% molybdophosphoric acid in EtOH or 2% p-methoxy-benzaldehyde in EtOH followed by 20% $\rm H_2SO_4$ and heating. Optical rotations are given in 10^{-1} deg cm² g⁻¹

Methyl ester deprotection (GP 1). The methyl ester was dissolved in THF (0.05 M) and cooled to 0 °C. 2.5 eq. LiOH-H₂O was added as an aqueous solution (1 : 3 v/v H₂O-THF) and stirred until complete conversion. The mixture was then neutralised with a 2 M HCl solution, THF was evaporated and the residue was partitioned between EtOAc (0.01 M) and 1 M NaHSO₄. The aqueous layer was extracted with EtOAc, the collected organic fractions were washed with brine and dried over MgSO₄. After evaporation the residue was dried to constant weight (0.01 Torr). The crude carboxylic acid proved to be sufficiently pure (TLC, NMR, HPLC).

Z-Group cleavage and benzyl ester deprotection (GP 2 A+B). The Z-protected compound (or benzyl ester) was dissolved in the indicated solvent (0.02 M) and 10 mol% (2 mol% for benzyl ester deprotection) Pd from Pd/C (5%, Degussa E101 NO/W) was added. The reaction mixture was then degassed, purged with H₂ and hydrogenated under 1 bar at the given temperature. (A): the mixture was vigorously stirred until complete conversion (TLC). The catalyst was then filtered off on Celite® 512 (2 cm). After evaporation and coevaporation with MeOH-toluene the residue was dried to constant weight (0.01 Torr). The crude amine obtained was sufficiently pure. (B): After stirring for 1.5 h the reaction was stopped and the catalyst filtered off on Celite® 512 (2 cm). The fractions were collected, the mixture was reduced to the initial volume and hydrogenated a second time under the same conditions for 2.5 h. After filtration on Celite[®], evaporation and coevaporation with MeOH-toluene the residue was dried to constant weight (0.01 Torr). In several cases the crude amines had to be further purified by FCC.

EDC–HOBt-mediated coupling (GP 3). The materials were dissolved in CH_2Cl_2 (indicated concentration) and cooled to 0 °C. EDC, HOBt, and diisopropylethylamine (1.5 eq, 3.0 eq and 3.0 eq respectively, relative to the carboxylic acid) were then added consecutively. After 30 min at 0 °C the reaction mixture was stirred at room temperature until conversion was complete.

HBTU–HOBt-mediated coupling (GP 4). The materials were dissolved in CH_2Cl_2 –DMF 5 : 1 (indicated concentration) and cooled to -15 °C. HBTU, HOBt, and disopropylethylamine (1.5 eq, 3.0 eq and 3.0 eq respectively, relative to the carboxylic acid) were then added consecutively. After 2 h at -15 °C the reaction mixture was stirred at room temperature until conversion was complete.

HATU–HOAt-mediated coupling (GP 5). The materials were dissolved in CH_2Cl_2 –DMF 5 : 1 (indicated concentration) and cooled to -15 °C. HATU, HOAt, and diisopropylethylamine (1.5 eq, 3.0 eq and 3.0 eq respectively, relative to the carboxylic acid) were then added consecutively. After 2 h at -15 °C the reaction mixture was stirred at room temperature for 22 h.

Work-up of preparative-HPLC acidic fractions (GP 6). The collected acidic HPLC-fractions were neutralised with NaHCO₃ (pH paper) and reduced (50%) by evaporation. The obtained mixture was then diluted with CHCl₃ (same volume) and after phase separation, the aqueous layer was extracted four times with CHCl₃–i-PrOH 5 : 1. The collected organic fractions were dried over Na₂SO₄ and after evaporation the obtained pure compound was dried to constant weight.

Experimental procedures and compound characterisation

(2R)-2-N-t-Butoxycarbonyl-aminopropanal 10. 29.5 ml of DIBAL-H (1 m in n-hexane, 29.5 mmol) was added dropwise at

-60 °C to a solution of N-Boc-D-alanine methyl ester 9 in CH₂Cl₂ (0.1 m, 150 ml). After 1.5 h the reaction was quenched with i-PrOH (15 ml), warmed to room temperature and the resulting mixture was stirred for 3 h with K-Na-tartrate (30 ml, 1 M). The layers were then separated and the aqueous phase was extracted with CH₂Cl₂ (3 × 20 ml). The collected organic layers were washed with brine and dried over MgSO₄. After evaporation and FCC (220 g, CH₂Cl₂-acetone 25:1) 2.0 g (11.6 mmol, 78%) of pure aldehyde 10 was obtained. $R_f = 0.68 \text{ (CH}_2\text{Cl}_2\text{-}$ acetone 20:1); v_{max}(CHCl₃)/cm⁻¹ 3687, 3620, 3441, 3013, 2979, 1734, 1707, 1501, 1451, 1393, 1369, 1236, 1166, 1048, 878, 668; $\delta_{\rm H}$ (75 MHz; CDCl₃; Me₄Si) 1.27 (3 H, d, J 7.5, Ala 3-H₃), 1.39 $(9H, s, 3 \times t\text{-Bu-H}_3), 4.09-4.24 (1H, m, 2-H), 4.94-5.11 (1H, m, 2-H)$ NH), 9.50 (1H, s, CHO); $\delta_{\rm C}$ (300 MHz; CDCl₃; Me₄Si) 14.8 $(3-CH_3)$, 28.3 (t-Bu-CH₃), 65.4 (2-CH), 199.7 (1-CHO); $[a]_D =$ -33.1, $[a]_{365} = -174.0$, $(c = 1.06, CHCl_3, 25 °C)$.

(2R)-2-N-t-Butoxycarbonylamino-6-hepten-3-ol 11. 22 ml of a 1.7 M solution of t-BuLi in n-hexane were added at -30 °C to 30 ml dry Et_2O . The mixture was then cooled to -78 °C and 1-bromo-3-butene (1.7 ml, 17.1 mmol) was added dropwise under continuous stirring. In a second flask CuBr·SMe₂ (1.77 g, 8.6 mmol) was dried under vacuum (0.01 Torr) and successively suspended in 50 ml dry Et₂O. After cooling to -78 °C, the metallation product 1-lithium-3-butene (52 ml Et₂O-n-hexane 1: 1 mixture) was transferred into the suspension via cannula within 1 min. The mixture was warmed to -40 °C and after 5 min a colour change from yellow to dark green indicated the formation of the cuprate. After cooling to -80 °C a solution of the aldehyde 10 (654 mg, 3.8 mmol) in Et₂O (20 ml) was added dropwise and the resulting mixture was stirred for 1.5 h. The reaction was quenched at room temperature by addition of 50 ml of half sat. NH₄Cl. After phase separation the organic layer was extracted with aqueous NH₃ (3 × 50 ml), washed with sat. NH_4Cl (2 × 30 ml), brine and dried over anhydrous MgSO₄. Evaporation followed by FCC (100 g, CH₂Cl₂-acetone 20:1) gave 614 mg (2.7 mmol, 71%) of the chromatographically inseparable diastereomeric mixture (11 syn/anti 20 : 1, estimated by ¹H NMR upon integration of the C-1 CH₃ signals) as a colourless oil. $R_f = 0.32$ (CH₂Cl₂-acetone 20 : 1); v_{max} (CHCl₃)/ cm⁻¹ 3442, 3016, 2980, 2935, 1702, 1502, 1453, 1393, 1368, 1236, 1166, 917; m/z (EI) 229.1678 (M+. C₁₂H₂₃NO₃ requires 229.1678).

(2S,5R,1'R)-2-(1'-t-Butoxycarbonylamino-ethyl)-5-hydroxymethyl-tetrahydrofuran (2S,5R)-12. The diastereomeric alcohol mixture 11 614 mg (2.70 mmol) was dissolved in 40 ml CH₂Cl₂ and cooled to 0 °C. A solution of MCPBA (m-chloroperbenzoic acid) in CH₂Cl₂ (20 ml, 0.26 M) was then added dropwise and the reaction was stirred overnight at room temperature. After dilution with CH₂Cl₂ (20 ml) the mixture was washed with sat. Na₂SO₃ (2×20 ml), brine (2×15 ml) and dried over anhydrous Na₂SO₄. The clear solution obtained was reduced to 30 ml and stirred for 3 h with 24 mg (0.10 mmol) of CSA (camphorsulfonic acid). CH₂Cl₂ (30 ml) was then added and the mixture was washed with sat. NaHCO₃, brine and dried over MgSO₄. After evaporation and FCC (90 g MTBE-PE 3:1) 250 mg of the cis-THF-alcohol 2S,5R-12 (1.02 mmol, 40% over two steps) were obtained as a colourless oil as well as 275 mg of the inseparable diastereomeric mixture (2S,5S)-12, (2R,5R)-12 and (2R,2S)-12 (20:1:1 respectively). $R_f = 0.38$ (MTBE-PE 3 : 1); v_{max} (CHCl₃)/cm⁻¹ 3439, 3009, 2980, 2877, 1699, 1510, 1455, 1368, 1336, 1238, 1165, 1100, 1076, 884; δ_{H} (300 MHz; CDCl₃; Me₄Si) 1.08 (3H, d, J 6.7, 2'-H₃), 1.42 (9H, s, $3 \times t$ -Bu-H₃), 1.67–1.73 (1H, m, 3-*H*H), 1.82–2.01 (3 H, m, 4-H₂ and 3-HH), 2.15 (1H, s, OH), 3.45 (1H, dd, J 10.6 and 7.0, CHHOH), 3.65-3.73 (2H, m, CHHOH and 2-H), 3.99-4.03 (1H, m, 5-H), 4.62 (1H, d, J 8.4, NH); $\delta_{\rm C}$ (75 MHz; CDCl₃; Me₄Si) 18.1 (2'-CH₃), 26.3 (3-CH₂), 28.3 (t-Bu-CH₃), 28.8 (4-CH₂), 50.3 (1'-CH), 64.0 (CH₂OH), 79.4 (t-Bu-C(CH₃)₃),

80.3 (5-CH), 83.4 (2-CH), 156.6 (C=O, Boc); $[a]_D = +$ 3.0, $[a]_{436} = +$ 3.9 (c = 1.10, CHCl₃, 25 °C); m/z (FAB) 246.1711 (M + H⁺. C₁₂H₂₄NO₄ requires 246.1705).

(5S,2R,1'R)-5-(1'-t-Butoxycarbonylamino-ethyl)-tetrahydrofuran-2-carboxylic acid (Boc-D-THF-OH, 13). 280 mg (1.13 mmol) of the alcohol 2S, 5R-12 were dissolved in 15 ml CH₂Cl₂. Two drops of H₂O, 800 mg (2.48 mmol) of iodobenzene diacetate (IBDA) and 9.0 mg (0.06 mmol) of TEMPO were added and the mixture was vigorously stirred for 3 h at room temperature. After dilution with 40 ml CH₂Cl₂ and phase separation, the organic layer was washed with sat. Na₂S₂O₃ (4 × 15 ml) until reduction of the residual IBDA was complete (KI/ starch paper). The organic layer was then extracted with sat. NaHCO₃ (4×10 ml) and after phase separation the collected aqueous fractions were acidified with 2 m HCl to pH = 3 (pH paper) and extracted with CH_2Cl_2 (4 × 30 ml). The organic layer was washed with brine and dried with MgSO₄. After evaporation and drying to constant weight (0.01 Torr) 278 mg (1.07 mmol, 95%) of the pure carboxylic acid 13 was obtained. $R_f =$ 0.31 (CHCl₃–MeOH–HCO₂H 200 : 10 : 1); v_{max} (CHCl₃)/cm⁻¹ 3442, 3100, 2975, 2932, 2877, 1748, 1691, 1600, 1510, 1430, 1395, 1380, 1343, 1244, 1164, 1086, 840, 770; $\delta_{\rm H}$ (300 MHz; CDCl₃; Me₄Si) 1.18 (3H, d, J 6.8, 2'-H₃), 1.44 (9H, s, 3 × t-Bu-H₃), 1.66 (1H, dddd, J 12.3, 10.4, 9.1 and 8.1, 3-HH), 1.93 (1H, dddd, J 12.5, 7.8, 6.2 and 3.3, 3-HH), 2.09 (1H, dddd, J 12.7, 8.1, 4.0 and 3.3, 4-HH) 2.30 (1H, dddd, J 12.7, 10.4, 9.0 and 7.8, 4-HH), 3.65 (1H, dq, J 6.8 and 3.2, 1'-H), 3.98 (1 H, ddd, J 9.1, 6.2 and 3.3, 5-H), 4.42 (1H, dd, J 9.0 and 4.0, 2-H); $\delta_{\rm C}$ (75 MHz; CDCl₃; Me₄Si) 19.5 (2'-CH), 25.5 (4-CH₂), 28.6 (t-Bu-CH₃), 85.8 (5-CH), 158.9 (C=O, Boc). $[a]_D = + 14.0$, $[a]_{365} = +43.8$ $(c = 1.80 \text{ in EtOH}, 25 ^{\circ}\text{C}); m/z \text{ (FAB) } 260.1511 \text{ (M} + \text{H}^{+}\text{.}$ $C_{12}H_{22}NO_5$ requires 260.1498).

Z-L-THF-Trp-Leu-Trp-2-(t-butyldiphenylsilyloxy)-ethyl-

amide 22. 1.70 g (1.85 mmol) tripeptide 21 were N-deprotected according to GP 2 (MeOH, rt, 2 h; $R_f = 0.38$ in CHCl₃-MeOH- $NH_3(aq.)$ 100 : 10 : 1). The crude product and 390 mg (1.33) mmol) of carboxylic acid 20 were then combined and coupled according to GP 3 (0.03 m, 3 h). After aqueous work-up, FCC (120 g, 3% MeOH in CHCl₃) gave 1.19 g (1.12 mmol, 84%) of ethylamide 22 as a colourless foam. $R_f = 0.49$ (CHCl₃-MeOH- $HCO_2H 100 : 10 : 5$); $\delta_H (300 \text{ MHz}; CDCl_3; Me_4Si) 0.60 (3H, d, d)$ J 6.0, Leu δ-H₃), 0.64 (3H, d, J 6.4, Leu δ-H₃), 0.97 (12H, s, 2"-H₃, and $3 \times t$ -Bu-H₃), 1.08–1.19 (2H, m, Leu β -H₂), 1.20–1.40 (2H, m, Leu γ-H and THF 3- or 4-CHH), 1.58-1.77 (1H, m, THF 3- or 4-CHH), 1.83-2.00 (2H, m, THF 3- or 4-CHH), 2.90 (1H, dd, J 13.9 and 10.2, Trp β-HH), 3.05–3.21 (3H, m, 1 × Trp β-HH and 2 × Trp β-HH), 3.28 (2H, t, J 7.4, 1-H₂), 3.60 (2H, t, J 6.4, 2-H₂), 3.62-3.75 (2H, m, THF 1'-CH and THF 5-H), 4.09-4.27 (2H, m, THF 2-H and Leu α-H), 4.37-4.57 $(2H, m, 2 \times Trp \alpha - H), 4.94 (1H, d, J 12.4, Z - HH), 5.05 (1H,$ J 12.4, Z-HH), 6.87–6.95 (2H, m, 2 × ar. CH), 6.98–7.08 (3H, m, $3 \times \text{ar. CH}$), 7.12 (1H, s, ar. CH), 7.22–7.34 (7H, m, $7 \times \text{ar.}$ CH), 7.36-7.45 (6H, m, $6 \times$ ar. CH), 7.48-7.69 (7H, m, THF NH and $6 \times \text{ar. CH}$), 7.93–8.05 (2H, m, Leu NH and AE NH), 8.15 (1H, d, J 8.7, Trp NH), 8.35 (1H, d, J 7.2, Trp NH), 10.78 (1H, s, indole NH), 10.80 (1H, s, indole NH); $[a]_D = +38.3$, $[a]_{365} = + 134.2 (c = 1.2, CHCl_3, 20 °C); HPLC: 12.8 min (M1);$ m/z (FAB) 1060.5379 (M + H⁺. $C_{61}H_{74}N_7O_8Si$ requires 1060.5368).

Z-Trp-D-Leu-L-THF-Trp-D-Leu-Trp-2-(*t***-butyldiphenyl-silyloxy)-ethylamide 24.** 627 mg (1.35 mmol) of dipeptide **23** were *C*-deprotected according to GP 1 and 1.06 g (1.00 mmol) of peptide **22** were *N*-deprotected according to GP 2 (MeOH, 40 °C, 3 h, $R_{\rm f} = 0.26$ in CHCl₃–MeOH–NH₃ (aq.) 100 : 10 : 1). The crude products were combined and coupled according to GP 4 (0.02 M, 1 h). After aqueous work-up, FCC (123 g 3%

MeOH in CHCl₃-5% MeOH in CHCl₃ + 0.5% HCO₂H) gave 1.02 g (0.75 mmol, 75%) of peptide **24** as a colourless foam. R_f = 0.49 (CHCl₃-MeOH-NH₃ (aq.) 100 : 10 : 1); $\delta_{\rm H}$ (400 MHz; DMSO- d_6 ; Me₄Si) (ppm): 0.63 (3H, d, J 6.3, Leu δ -H₃), 0.67 (d, J 6.6, 3H, Leu δ -H₃), 0.80 (d, J 5.9, 3H, Leu δ -H₃), 0.83 (3H, d, J 5.9, Leu δ-H₃), 0.94–1.03 (12H, m, THF C-2'-H₃ and 3 \times t-Bu-H₃), 1.03–1.12 (1H, m, Leu γ-H), 1.13–1.22 (2H, m, Leu β-H₂), 1.24–1.36 (1H, m, THF 4-HH), 1.39–1.54 (3H, m, Leu β -H₂ and Leu γ-H), 1.64–1.74 (1H, m, THF 3-HH), 1.85–1.94 (2H, m, THF 4-HH and THF 3-HH), 2.65 (1H, dd, J 8.8 and 14.0, Trp β -HH), 2.91 (1H, dd, J 10.2 and 14.0, Trp β -HH), 3.05–3.21 (4H, m, Trp β -HH and 3 × Trp β -HH), 3.21–3.28 (2H, m, AE 2-H₂), 3.61 (2H, t, J 6.6, AE 1-H₂), 3.67–3.75 (1H, m, THF 5-H), 3.87 (1H, dt, J 2.3 and 7.0, THF 1'-H), 4.12 (1H, t, J 5.5, THF 2-H), 4.19-4.27 (1H, m, Leu α -H), 4.37-4.53 (4H, m, Leu α-H and 3 × Trp α-H), 4.94 (1H, d, J 12.4, Z-HH), 5.05 $(1H, d, J 12.4, Z-HH), 6.89-6.97 (3H, m, 3 \times ar. CH), 6.99-7.06$ (3H, m, 3 × ar. CH), 7.08 (1H, s, ar. CH), 7.14 (2H, d, J 15.6, 2 × ar. CH), 7.26–7.33 (8H, m, 8 × ar. CH), 7.38–7.48 (7H, m, Trp NH and $6 \times \text{ar. CH}$), 7.53–7.65 (7H, m, $7 \times \text{ar. CH}$), 7.98 (1H, t, J 5.8, AE NH), 8.06 (1H, d, J 7.9, Leu NH), 8.09-8.15 (2H, m, Leu NH and Trp NH), 8.18 (1H, d, J 8.6, THF NH), 8.38 (1H, d, J7.3, Trp NH), 10.73 (1H, s, indole NH), 10.79 (2H, s, indole NH); $[a]_D = +19.5$, $[a]_{365} = +90.8$ (c = 1.3, CHCl₃, T = 20 °C); HPLC: 14.4 min (M1); m/z (ESI) 1381.8 (M + Na⁺. C₇₈H₉₄- $N_{10}NaO_{10}Si$ requires 1381.7).

Z-Ala-D-Val-Val-D-Val-Trp-D-Leu-L-THF-Trp-D-Leu-Trp-2-(t-butyldiphenylsilyloxy)-ethylamide 15. 480 mg (0.9 mmol) of tetrapeptide 25 were C-deprotected according to GP 2 and 800 mg (0.60 mmol) of peptide 24 were N-deprotected according to GP 1 (MeOH, 40 °C, 3 h; $R_f = 0.12$ in CHCl₃-MeOH-HCO₂H 100:10:1). The crude products were combined and coupled according to GP 4 (0.01 M, 2 h). After aqueous work-up, FCC $(120 \text{ g}, 3 \rightarrow 4\% \text{ MeOH in CHCl}_3)$ gave 828 mg (0.48 mmol 81%)of product 15 as a colourless foam. $R_f = 0.55$ (CHCl₃-MeOH- HCO_2H 100 : 10 : 1); δ_H (500 MHz; DMSO- d_6 ; Me₄Si) 0.56– $0.63~(9H, m, 2 \times Leu~\delta-H_3~and~Val~\gamma-H_3), 0.64-0.67~(6H, m, Leu~\Delta-H_3)$ δ -H₃ and Val γ-H₃), 0.69 (3H, d, J 6.2, Leu δ -H₃), 0.77–0.87 $(12H, m, 4 \times Val \gamma - H_3), 0.99 (9H, s, 3 \times t - Bu - H_3), 0.95 - 1.05$ (4H, m, THF 2'-H₃ and Leu γ-H), 1.20 (3H, d, J 6.9, Ala β-H₃),1.08-1.23 (3H, m, Leu β -H₂ and Leu γ -H), 1.28-1.39 (2H, m, THF 4-HH and THF 3-HH), 1.40-1.49 (2H, m, Leu β -H₂), 1.66–1.74 (1H, m, THF 4-HH), 1.84–1.94 (2H, m, Val β-H and THF 3-HH), 1.96–2.06 (2H, m, $2 \times \text{Val }\beta$ -H), 2.89–3.01 (2H, m, $2 \times \text{Trp }\beta\text{-}H\text{H}$), 3.22-3.08 (4H, m, Trp $\beta\text{-}H\text{H}$ and $3 \times \text{Trp}$ β -HH), 3.23–3.31 (2H, m, AE 2-H₂), 3.61 (2H, t, J 6.4, AE 1-H₂), 3.68-3.75 (1H, m, THF 5-H), 3.85-3.93 (1H, m, THF 1'-H), 4.09-4.19 (2H, m, THF 2-H and Ala α -H), 4.20-4.28(3H, m, Val α -H and 2 × Leu α -H), 4.29–4.36 (2H, m, 2 × Val α -H), 4.41–4.47 (1H, m, Trp α -H), 4.51 (1H, dd, J 7.3 and 14.2, Trp α -H), 4.57 (1H, dd, J 7.7 and 14.1, Trp α -H), 4.98 (1H, d, J 13.0, Z-HH), 5.01 (1H, d, J 13.0, Z-HH), 6.89–6.97 (3H, m, 3 × ar. CH), 6.99–7.06 (3H, m, 3 × ar. CH), 7.08 (1H, s, ar. CH), 7.15 (2H, d, J 7.1, 2 × ar. CH), 7.26–7.37 (8H, m, 8 × ar. CH), 7.38–7.48 (7H, m, 6 × ar. CH and Ala NH), 7.53–7.60 (3H, m, $3 \times \text{ar. CH}$), 7.62 (4H, d, J 6.6, $4 \times \text{ar. CH}$), 7.83 (1H, d, J 9.9, NH), 7.86 (1H, d, J 9.2, NH), 7.90 (1H, d, J 8.7, THF NH), 7.92 (1H, d, J 8.7, NH), 7.99 (1H, t, J 4.8, AE NH), 8.02 (1H, d, J 8.3, NH), 8.10–8.17 (2H, m, 2 × Trp NH), 8.23 (1H, d, J 8.3, NH), 8.46 (1H, d, J 7.1, Trp NH), 10.74 (1H, s, indole NH), 10.79 (2H, s, indole NH); HPLC: 16.6 min (M1); m/z (ESI) $1728.6 \, (M + H^+. C_{96} H_{127} N_{14} O_{14} Si \text{ requires } 1727.9).$

H-Ala-D-Val-Val-D-Val-Trp-D-Leu-L-THF-Trp-D-Leu-Trp-2-(*t*-butyldiphenylsilyloxy)-ethylamide 15a. 400 mg (231 μmol) of peptide 15 were *N*-deprotected according to GP 2A (EtOH, 50 °C). After filtration, evaporation, FCC (20 g, CHCl₃–MeOH–NH₃ 100: 10: 1) of the crude product gave 250 mg (157 μmol, 68%) of amine 15a. *R*_f = 0.16 (CHCl₃–MeOH–HCO₃H

100 : 10 : 1). HPLC: 11.8 min (M2). m/z (ESI) 1615.6 (M + Na $^+$. $C_{88}H_{120}N_{14}NaO_{12}Si$ requires 1615.9).

HCO-Val-Gly-Ala-D-Leu-Ala-D-Val-Val-D-Val-Trp-D-Leu-L-THF-Trp-D-Leu-Trp-2-(*t*-butyldiphenylsilyloxy)-ethylamide 3. 55.0 mg (137 µmol) of tetrapeptide 14 were C-deprotected according to GP 1 ($R_f = 0.26$ in CHCl₃-MeOH-HCO₂H 100 : 10:1) and then coupled with 100 mg (62.7 μmol) of amine 15a according to GP 4 (6.0 mm, 3 h). After aqueous work-up and FCC (12 g, CHCl₃-MeOH-HCO₂H 200 : 10 : 1) 90.0 mg (46.0 μmol, 73%) of peptide 3 were obtained as a colourless foam. Prior to single-channel measurement, purification of 10.0 mg of compound 3 by preparative HPLC (250 mm Rainin C 8, 60→99% B in 20 min, A: H₂O; B: acetonitrile-isopropanol 2:1; 10 ml min⁻¹) afforded 8.0 mg (80%) of purest product. $R_f = 0.52 \text{ (CHCl}_3\text{-MeOH-HCO}_2\text{H } 100 : 10 : 1); \delta_H \text{ (400 MHz)}$ DMSO- d_6 ; Me₄Si) 0.55–0.63 (9H, m, 3 × Leu δ -H₃), 0.64–0.71 (9H, m, 3 × Leu δ-H₃), 0.77–0.88 (24H, m, 8 × Val γ-H₃), 0.95– 1.05 (13H, m, THF 2'-H₃, $3 \times t$ -Bu-H₃ and Leu γ -H), 1.19 (3H, d, J 4.3, Ala β-H₃), 1.21 (3H, d, J 4.3, Ala β-H₃), 1.06–1.25 (4H, m, Leu γ-H, Leu β-H₂ and THF 4-HH), 1.27–1.37 (2H, m, THF 3-HH and Leu β -HH), 1.39–1.49 (3H, m, Leu β -HH and Leu β-H₂), 1.50–1.60 (1H, m, Val β-H), 1.65–1.75 (1H, m, THF 4-HH), 1.83–2.07 (5H, m, THF 3-HH, Leu γ -H and 3 \times Val β -H), 2.86–3.02 (2H, m, Trp β -H₂), 3.07–3.20 (4H, m, 2 × Trp β -H₂), 3.21–3.31 (2H, m, AE 1-H₂), 3.60 (2H, t, J 6.4, AE 1-H₂), 3.67-3.76 (3H, m, Gly α -H₂ and THF 5-H), 3.89 (1H, dd, J 14.7 and 7.4, THF 1'-H), 4.12 (1H, t, J 5.8, Val α-H), 4.17–4.34 (9H, m, $3 \times \text{Val }\alpha\text{-H}$, $3 \times \text{Leu }\alpha\text{-H}$, $2 \times \text{Ala }\alpha\text{-H}$ and THF 2-H), 4.40– 4.58 (3H, m, $3 \times \text{Trp }\alpha\text{-H}$), 6.88–6.96 (3H, m, $3 \times \text{ar. CH}$), 6.98– 7.05 (3H, m, $3 \times \text{ar. CH}$), 7.06 (1H, d, J 1.9, ar. CH), 7.12 (1H, d, J 1.9, ar. CH), 7.14 (1H, d, J 1.9, ar. CH), 7.26–7.32 (3H, m, $3 \times \text{ar. CH}$), 7.36–7.47 (6H, m, $6 \times \text{ar. CH}$), 7.51–7.64 (7H, m, $7 \times \text{ar. CH}$), 7.85 (1H, d, J 8.6, NH), 7.88–8.02 (7H, m, $7 \times \text{NH}$), 8.04 (1H, s, CHO), 8.09-8.26 (5H, m, 5 × NH), 8.31 (1H, t, J 5.6, AE NH), 8.46 (1H, d, J 7.0, Trp NH), 10.71 (1H, s, indole NH), 10.77 (2H, s, indole NH); HPLC: 14.3 min (M1); m/z (ESI) 1962.7 (M + H⁺. $C_{105}H_{149}N_{18}O_{17}Si$ requires 1962.1).

^{11,12}[L-THF]-gA 2. 20.0 mg (10.2 μmol) of peptide 3 were dissolved in 20 ml CH₂Cl₂-DMF 10: 1. Under stirring at rt 1.8 ml of a 5% solution of HF in CH₃CN were then added dropwise. After 45 min the reaction mixture was diluted with 20 ml CH_2Cl_2 , carefully washed with sat. NaHCO₃ (2 × 10 ml) and brine (1 × 10 ml) and dried over Na₂SO₄. After evaporation, 20.0 mg of the crude peptide 2 were isolated and purified by preparative HPLC (250 mm Rainin C 8, 60%→99% B in 20 min, A: H₂O; B: acetonitrile-isopropanol 2: 1; 10 ml min⁻¹). Pure product 2 (15.0 mg 8.70 µmol, 85%) was isolated as a colourless solid. $R_f = 0.40 \text{ (CHCl}_3\text{-MeOH-HCO}_2\text{H}; 100 : 10 :$ 1); $\delta_{\rm H}$ (400 MHz; DMSO- d_6 ; Me₄Si) 0.56–0.64 (9H, m, 3 × Leu δ -H₃), 0.65–0.73 (9H, m, 3 × Leu δ -H₃), 0.77–0.88 (24H, m, $8 \times \text{Val } \gamma$ -H₃), 0.99 (3H, d, J 6.6, THF 2'-H₃), 0.95–1.08 (2H, m, 2 × Leu γ-H), 1.11–1.28 (5H, m, Leu γ-H and 2 × Leu β-H₂), 1.19 (3H, d, J 7.6, Ala β-H₃), 1.21 (3H, d, J 7.6, Ala β-H₃), 1.28– 1.41 (3H, m, Leu β -H₂ and THF 4-HH), 1.43–1.51 (1H, m, THF 3-HH), 1.52–1.61 (1H, m, Val β-H), 1.69–1.79 (1H, m, THF 4-HH), 1.86–2.08 (5H, m, THF 3-HH, Leu γ -H and 3 \times Val β-H), 2.89 (1H, dd, J 14.5 and 9.9, Trp β-HH), 2.98 (1H, dd, J 14.0 and 8.4, Trp β-HH), 3.06–3.24 (6H, m, Trp β-HH, 3 × Trp β -HH and AE 2-H₂), 3.35–3.43 (1H, m, AE 1-HH), 3.66– 3.80 (3H, m, Gly α-H₂ and THF 5-H), 3.90 (1H, dd, J 15.4 and 7.7, THF 1'H), 4.13–4.33 (10H, m, $4 \times \text{Val } \alpha$ -H, $3 \times \text{Leu } \alpha$ -H, $2 \times \text{Ala } \alpha\text{-H}$ and THF 2-H), 4.43 (1H, dt, J 4.9 and 9.1, Trp α -H), 4.51 (2H, dd, J 9.1 and 4.9, 2 × Trp α -H), 4.67–4.73 (1H, m, AE 1-HH), 6.88–6.97 (3H, m, 3 × ar. CH), 6.98–7.05 (3H, m, $3 \times$ ar. CH), 7.08 (1H, d, J 1.9, ar. CH), 7.09 (1H, d, J 1.9, ar. CH), 7.16 (1H, d, J 1.9, ar. CH), 7.25–7.34 (3H, m, 3 × ar. CH), 7.52–7.62 (3H, m, 3 × ar. CH), 7.89 (1H, t, J 5.6, AE NH), 7.93–8.12 (8H, m, $7 \times NH$ and CHO), 8.17 (1H, d, J 8.6, Trp

NH), 8.22–8.31 (3H, m, 3 × NH), 8.35 (1H, t, J 5.8, Gly NH), 8.47 (1H, d, J 7.6, Trp NH), 10.73 (1H, s, indole NH), 10.79 (2H, s, indole NH); HPLC: 10.9 min (M3); m/z (ESI) 1745.9 (M + Na $^+$. $C_{89}H_{130}N_{18}NaO_{17}$ requires 1746.0).

Z-Ala-D-Leu-Ala-D-Val-Val-D-Val-Trp-D-Leu-L-THF-Trp-D-Leu-Trp-2-(t-butyldiphenylsilyloxy)-ethylamide 31. 43.0 mg (123 µmol) of dipeptide 18 were C-deprotected according to GP 1 and coupled with 60.0 mg (37.7 µmol) of amine 15a according to GP 4 (6.0 mm, 2.5 h). After aqueous work-up and FCC (12 g, 3% MeOH in CHCl₃→4% MeOH in CHCl₃ + 0.4% HCO₂H) 65.0 mg (34.0 µmol, 90%) of peptide 31 were obtained as a colourless foam. $R_f = 0.52$ (CHCl₃-MeOH-HCO₂H 100 : 10 : 1); $\delta_{\rm H}$ (500 MHz; DMSO- d_6 ; Me₄Si) 0.57–0.63 (9H, m, 3 × Leu δ-H₃), 0.65 (6H, d, J 6.4, 2 × Leu δ-H₃), 0.70 (3H, d, J 6.6, Leu δ -H₃), 0.76–0.86 (18H, m, 6 × Val γ -H₃), 0.95–1.01 (13H, m, $3 \times t$ -Bu-H₃, THF 2'-H₃ and Leu γ -H), 1.08–1.16 (3H, m, Leu γ -H and Leu β -H₂), 1.17–1.23 (7H, m, 2 \times Ala β -H₃ and THF 3-HH), 1.28–1.35 (2H, m, THF 4-HH and Leu γ -H), 1.42-1.49 (4H, m, Leu β -H₂), 1.51-1.59 (1H, m, Val β -H), 1.66-1.75 (1H, m, THF 4-HH), 1.84–1.94 (2H, m, Leu γ -H, and THF 3-HH), 1.95–2.05 (2H, m, Val β-H), 2.88–3.00 (2H, m, $2 \times \text{Trp }\beta$ -HH), 3.07–3.21 (4H, m, Trp β -HH and $3 \times \text{Trp }\beta$ -HH), 3.23-3.33 (2H, m, AE 2-H₂), 3.61 (2H, t, J 6.2, AE 1-H₂), 3.71 (1H, dd, J 15.1 and 7.1, THF 5-H), 3.89 (1H, dd, J 14.4 and 7.5, THF 1'-H), 4.07 (1H, t, J 7.2, Ala α -H), 4.14 (1H, t, J 5.6, THF 2-H), 4.19–4.28 (4H, m, Val α -H and 3 × Leu α -H), 4.29–4.35 (3H, m, Ala α -H and 2 × Val α -H), 4.41–4.47 (1H, m, Trp α -H), 4.45 (1H, dd, J 14.7 and 7.1, Trp α-H), 4.57 (1H, dd, J 14.3 and 7.0, Trp α -H), 4.99 (1H, d, J 12.4, Z-HH), 5.02 (1H, d, J 12.1, Z-HH), 6.88–6.96 (3H, m, 3 × ar. CH), 6.98–7.05 (3H, m, 3 × ar. CH), 7.07 (1H, d, J 1.8, ar. CH), 7.13 (1H, d, J 1.7, ar. CH), 7.16 (1H, d, J 1.8, ar. CH), 7.26–7.36 (8H, m, 8 × ar. CH), 7.39– 7.47 (7H, m, $6 \times ar$. CH and Ala NH), 7.53–7.60 (3H, m, $3 \times ar$. CH), 7.61-7.65 (4H, m, $4 \times$ ar. CH), 7.77 (1H, d, J 8.7, NH), 7.86 (1H, d, J 9.9, NH), 7.88-7.93 (2H, m, NH and THF NH), 7.96 (1H, d, J7.1, NH), 7.99 (1H, t, J6.0, AE NH), 8.02 (1H, d, J 8.5, NH), 8.09–8.17 (3H, m, NH and 2 × Trp NH), 8.23 (1H, d, J 8.3, NH), 8.46 (1H, d, J 7.1, Trp NH), 10.73 (1H, s, indole NH), 10.79 (2H, s, indole NH); HPLC: 10.7 min (M4); m/z (ESI) 1934.3 (M + Na⁺. $C_{105}H_{142}N_{16}NaO_{16}Si$ requires 1934.0).

H-Ala-D-Leu-Ala-D-Val-D-Val-Trp-D-Leu-L-THF-Trp-D-Leu-Trp-2-(t-butyldiphenylsilyloxy)-ethylamide 31a. 124 mg (64.9 μmol) of peptide 31 were N-deprotected according to GP 2A (EtOH, 50 °C). After filtration and evaporation FCC (15 g, CHCl₃–MeOH–NH₃(conc.) 100 : 10 : 1) gave 70.0 mg (39.4 μmol, 61%) of amine 31a. $R_{\rm f}=0.28$ (CHCl₃–MeOH–NH₃(conc.) 100 : 10 : 1); HPLC: 7.2 min (M4). m/z (ESI) 1799.4 (M + Na⁺. C₉₇H₁₃₆N₁₆NaO₁₄Si requires 1800.0).

Succ-1-(1'-t-butyldiphenylsilyloxy-11,12[L-THF]-gramicidin **A)-4-(Val-Gly-OMe)-diamide 32.** 30.0 mg (56.2 μmol) of benzyl ester 17 were C-deprotected according to GP 2A. The crude product was coupled with amine 31a (30.0 mg, 17.0 µmol) according to GP 4 (3.0 mм, 3 h). After aqueous work-up FCC (5 g, 3% MeOH in CHCl₃→4% MeOH in CHCl₃ + 0.4% HCO_2H) gave 31.0 mg (14.1 µmol, 83%) of pure diamide 32 as a colourless solid. $R_f = 0.42$ (CHCl₃–MeOH–HCO₂H 100 : 10 : 1); $\delta_{\rm H}$ (600 MHz; DMSO- d_6 ; Me₄Si) 0.52–0.71 (18H, m, 6 × Leu δ -H₃), 0.74–0.89 (30H, m, 10 × Val γ-H₃), 0.91–1.02 (13H, m, $3 \times t$ -Bu-H₃, Leu γ -H and THF 2'-H₃), 1.03–1.13 (3H, m, Leu γ -H and Leu β-H₂), 1.18 (3H, d, J 6.8, Ala β-H₃), 1.19 (3H, d, J 6.8, Ala β-H₃), 1.16–1.23 (1H, m, THF 3-HH), 1.25–1.35 (2H, m, THF 4-HH and Leu γ-H), 1.41–1.51 (4H, m, $2 \times \text{Leu } \beta$ -H₂), 1.52-1.59 (1H, m, Val β -H), 1.64-1.74 (1H, m, THF 4-HH), 1.82-2.07 (5H, m, 4 × Val β-H and THF 3-HH), 2.36-2.45 (4H, m, succ 2-H₂ and 3-H₂), 2.85–3.03 (2H, m, $2 \times \text{Trp } \beta$ -HH), 3.04–3.21 (4H, m, Trp β -HH and 3 × Trp β -HH), 3.22–3.28

(2H, m, AE 2-H₂), 3.60 (3H, s, OC H_3) 3.55–3.63 (3H, m, THF 5-H and AE 1-H₂), 3.63–3.74 (4H, m, 2 × Gly α-H₂), 3.85–3.93 (1H, m, THF 1'-H), 4.03–4.33 (11H, m, 3 × Leu α-H, 5 × Val α-H, THF 2-H and 2 × Ala α-H), 4.37–4.60 (3H, m, 3 × Trp α-H), 6.87–6.95 (3H, m, ar. 3 × CH), 6.97–7.03 (3H, m, 3 × ar. CH), 7.05 (1H, d, J 1.9, ar. CH), 7.11 (1H, d, J 1.9, ar. CH), 7.14 (1H, d, J 1.5, ar. CH), 7.24–7.32 (3H, m, 3 × ar. CH), 7.35–7.45 (6H, m, 6 × ar. CH), 7.49–7.57 (3H, m, 3 × ar. CH), 7.58–7.64 (4H, m, 4 × ar. CH), 7.78 (1H, d, J 8.7, NH), 7.83–7.93 (6H, m, 6 × NH), 7.94–8.05 (4H, m, 4 × NH), 8.06–8.15 (2H, m, 2 × NH), 8.17–8.28 (2H, m, 2 × NH), 8.37 (1H, t, J 5.8, AE NH), 8.45 (1H, d, J 5.3, Trp NH), 10.71 (1H, s, indole NH), 10.76 (2H, s, indole NH); HPLC: 14.7 min (M4); m/z (ESI) 2226.0 (M + Na⁺. C₁₁₆H₁₆₆N₂₀NaO₂₁Si requires 2226.2).

Succ-(11,12[L-THF]-gA-TBDPS)₂ 4. 15.0 mg (6.8 μmol) of methyl ester 32 were C-deprotected according to GP 1 (R_f = 0.20 in CHCl₃-MeOH-HCO₂H 100 : 10 : 1; HPLC: 7.2 min M4). 12.0 mg of the crude carboxylic acid were combined with 12.0 mg (6.8 μmol) of amine **31a** and coupled according to GP 5 (1.0 mm, 12 h). After aqueous work-up 24.0 mg of crude diamide 4 were obtained and purified by preparative HPLC (250 mm Rainin C 4, $60\% \rightarrow 100\%$ B in 25 min, A: $H_2O + 0.1\%$ TFA; B: acetonitrile-isopropanol 2 : 1 + 0.1% TFA; 14 ml min⁻¹). Treatment of the collected fractions according to GP 6 gave 9.0 mg (2.3 µmol, 34%) of pure diamide 4 as a colourless solid. $\delta_{\rm H}$ (500 MHz; DMSO- d_6 ; Me₄Si) 0.55–0.73 (30H, m, 10 × Leu δ-H₃), 0.75–0.91 (54H, m, $16 \times \text{Val } \gamma$ -H₃ and $2 \times \text{Leu } \delta$ -H₃), 0.92–1.05 (26H, m, 6 \times t-Bu-H₃, 2 \times THF 2'-H₃ and 2 \times Leu γ -H), 1.05–1.39 (20 H, m, 4 × Ala β-H₃ and 4 × Leu β-H₂), 1.40–1.59 (6H, m, 2 × Leu γ-H, 2 × Val β-H and 2 × THF 3- or 4-HH), 1.61–1.78 (4H, m, 2 × Leu β -H₂), 1.85–2.08 (12H, m, $2 \times \text{THF } 3$ - or 4-HH, $2 \times \text{THF } 3$ - or 4-HH, $6 \times \text{Val } \beta$ -H and $2 \times \alpha$ Leu γ -H), 2.26–2.32 (2H, m, 2 × THF 3- or 4-HH), 2.36–2.45 (4H, m, succ 2-H₂ and 3-H₂), 2.88–3.05 (6H, m, $6 \times \text{Trp }\beta\text{-}H\text{H}$), 3.08–3.29 (6H, m, 6 × Trp β -HH), 3.42 (4H, t, J 5.3, 2 × AE 2-H₂), 3.56–3.63 (6H, m, $2 \times AE 1$ -H₂ and $2 \times Gly \alpha CHH$), 3.66-3.76 (2H, m, 2 × Leu α -H), 3.86-3.95 (4H, m, 2 × THF 1'-H and $2 \times \text{Val }\alpha\text{-H}$), 4.05–4.15 (6H, m, $2 \times \text{Val }\alpha\text{-H}$, $2 \times \text{Gly}$ α -HH and 2 × Leu α -H), 4.18–4.34 (12H, m, 4 × Val α -H, 4 × Ala α -H, 2 × Leu α -H and 2 × THF 5-H), 4.39–4.56 (6H, m, 6 × Trp α -H), 4.56–4.62 (2H, m, 2 × THF 2-H), 6.89–6.97 (6H, m, $6 \times \text{ar. CH}$), 6.99–7.05 (6H, m, $6 \times \text{ar. CH}$), 7.07 (2H, s, $2 \times \text{ar.}$ CH), 7.15 (4H, d, J 12.8, 4 × ar. CH), 7.30 (6H, t, J 7.7, 6 × ar. CH), 7.38–7.48 (12H, m, 12 × ar. CH), 7.52–7.66 (14H, m, 14 × ar. CH), 7.90 (16H, m, 16 × NH), 8.15-8.21 (6H, m, 6 × NH), 8.24–8.36 (6H, m, 6 × NH), 8.49–8.56 (2H, m, 2 × NH), 10.72 (2H, s, indole NH), 10.80 (4H, s, indole NH); HPLC: 18.6 min (M4); m/z (ESI) 1999.9 (M + $2Na^{+}$. $C_{212}H_{298}N_{36}Na_{2}O_{34}Si_{2}$ requires 1997.1).

Boc-D-THF-D-Leu-OMe 27. 289 mg (1.15 mmol) of carboxylic acid 5 were combined with 616 mg (3.40 mmol) of D-Leucine hydrochloride and coupled according to GP 3 (0.04 m 2 h). After aqueous work-up FCC (35g, 1% MeOH in CHCl₃) gave 346 mg (0.90 mmol, 78%) of pure methyl ester 27 as a colourless foam. $R_f = 0.57$ (CHCl₃-MeOH 20 : 1); δ_H (300 MHz; DMSO-d₆; Me₄Si) 0.84 (3H, d, J 6.4, Leu δ-H₃), 0.91 (3H, d, J 6.4, Leu δ-H₃), 0.99 (3H, d, J 5.5, THF 2'-H₃), 1.29-1.39 (1H, m, Leu β -HH), 1.40 (9H, s, 3 × t-Bu-H₃), 1.54 (1H, ddd, J 13.2, 9.9 and 3.6, THF 4-HH), 1.61-1.71 (1H, m, Leu γ -H), 1.83–1.92 (2H, m, Leu β -HH and THF 3-HH), 2.00–2.09 (1H, m, THF 3-HH), 2.11-2.18 (1H, m, THF 4-HH), 3.64 $(3H, s, OCH_3), 3.65-3.71$ (2H, m, THF 1'-H and Leu α -H), 4.34 (1H, d, J 8.2, THF 5-H), 4.39 (1H, ddd, J 11.5, 8.4 and 3.5 THF 2-H), 7.06 (1H, d, J 8.0, THF NH), 8.68 (1H, d, J 8.6, Leu NH); $[a]_D = -9.3$, $[a]_{365} = -42.1$ (c = 1.0, CHCl₃, 20 °C); HPLC: 7.9 min (M1); m/z (FAB) 387.2491 (M + H⁺. $C_{19}H_{35}N_2O_6$ requires 387.2495).

Z-Trp-D-THF-D-Leu-OMe 29. 216 mg (0.56 mmol) of methyl ester 27 were dissolved in a solution of CF₃CO₂H in CH₂Cl₂ (20 ml, 20%) and stirred 20 min. at room temperature. The reaction mixture was then diluted with CH₂Cl₂ (30 ml), washed carefully with sat. NaHCO₃ (10 ml) and brine (20 ml) and dried over MgSO₄. After evaporation, 160 mg of crude amine ($R_f =$ 0.28 in CHCl₃-MeOH-HCO₂H 100:10:1) were obtained and directly coupled with 379 mg (1.12 mmol) Z-Trp-OH according to GP 3 (0.02 m, 2 h). After aqueous work-up and FCC (7 g, 3% MeOH in CHCl₃) 322 mg (0.53 mmol, 95%) of pure peptide 29 were obtained as a colourless foam. $R_f = 0.50$ (CHCl₃-MeOH- $HCO_2H 100: 10: 1); \delta_H (500 \text{ MHz}; DMSO-d_6; Me_4Si) 0.80 (3H,$ d, J 6.4, Leu δ-H₃), 0.87 (3H, d, J 6.4, Leu δ-H₃), 0.98 (3H, d, J 6.9, THF 2'-H₃), 1.38–1.48 (1H, m, THF 4-HH), 1.52 (1H, ddd, J 13.3, 9.6 and 3.9, THF 3-HH), 1.57-1.67 (1H, m, Leu γ -H), 1.86–1.95 (2H, m, THF 3-HH and Leu β -HH), 2.04–2.13 (2H, m, THF 4-HH and Leu β -HH), 2.96 (1H, dd, J 13.9 and 8.8, Trp β -HH), 3.31 (1H, dd, J 14.4 and 4.8, Trp β -HH), 3.60 $(3H, s, OCH_3)$, 3.76–3.83 (1H, m, Leu α -H), 3.91–3.99 (1H, m, THF C-1'-H), 4.25-4.32 (1H, m, THF 5-H), 4.32-4.41 (2H, m, Trp α-H and THF 2-H), 4.97 (2H, s, Z-H₂), 6.97 (1H, t, J 7.5, ar. CH), 7.01-7.09 (2H, m, $2 \times \text{ar.CH}$), 7.10-7.21 (1H, m, ar. CH), 7.22-7.27 (2H, m, 2 × ar.CH), 7.27-7.37 (4H, m, 4 × ar. CH), 7.61 (1H, d, J 8.0, THF NH), 8.16 (1H, d, J 8.5, Trp NH), 8.68 (1H, d, J 8.0, Leu NH), 10.84 (1H, s, indole NH); $[a]_D = -35.8$, $[a]_{365} = -133.2$ (c = 1.0, CHCl₃, 20 °C); HPLC: 6.6 min (M1). m/z (FAB) 607.3139 (M + H⁺. C₃₃H₄₃N₄O₇ requires 607.3132).

Z-Trp-D-THF-D-Leu-Trp-D-Leu-Trp-2-(t-butyl-

diphenylsilyloxy)-ethylamide 30. 150 mg (0.25 mmol) of peptide **29** were C-deprotected according to GP 1 ($R_f = 0.28$ in CHCl₃-MeOH-HCO₂H), as well as 1.20 g (1.30 mmol) of tripeptide 21 were N-deprotected according to GP 2A (MeOH, rt, 2 h; $R_f =$ 0.38 in CHCl₃-MeOH-NH₃(aq.) 100 : 10 : 1). The crude carboxylic acid and a portion (1/4) of the crude amine were collected and coupled according to GP 4 (8.0 mm, 3 h). After aqueous work-up and FCC (40 g, 3% MeOH in CHCl₃) 319 mg (23.5 mmol, 94%) of peptide 30 were isolated as a colourless foam. $R_f = 0.65$ (CHCl₃–MeOH–HCO₂H 200 : 10 : 1); δ_H (400 MHz; DMSO- d_6 ; Me₄Si) 0.59 (3H, d, J 6.6, Leu δ -H₃), 0.65 $(3H, d, J 6.3, Leu \delta-H_3), 0.66(3H, d, J 6.6, Leu \delta-H_3), 0.70 (3H, d)$ d, J 6.3, Leu d-H₃), 0.84 (3H, d, J 6.6, THF 2'-H₃), 0.97 (9H, s, $3 \times t$ -Bu-H₃), 0.96–1.03 (1H, m, Leu γ -H), 1.11–1.21 (3H, m, THF 4-HH and Leu β -H₂), 1.22–1.36 (2H, m, THF 3HH and Leu γ -H), 1.48–1.58 (1H, m, THF 3-HH), 1.68–1.78 (1H, m, THF 4-HH), 1.88-2.05 (2H, m, Leu β -H₂), 2.83-2.99 (3H, m, 3 \times Trp β -*HH*), 3.05–3.20 (3H, m, 3 \times Trp β -H*H*), 3.20–3.29 (2H, m, AE 2-H₂), 3.62 (2H, t, J 6.6, AE 1-H₂), 3.65 (1H, dd, J 14.9 and 8.9, THF 5-H), 3.81 (1H, sextet, J 7.0, THF 1'-H), 4.15-4.31 (4H, m, 2 × Leu α -H, Trp α -H and THF 2-H), 4.47 (1H, dt, J 4.4 and 9.1, Trp α-H), 4.53 (1H, dt, J 5.6 and 8.5, Trp α-H), 4.92 (1H, d, J 13.5, Z-HH), 4.96 (1H, d, J 13.2, Z-HH), 6.89-6.96 (3H, m, 3 × ar. CH), 6.98–7.09 (6H, m, 5 × ar. CH and Trp NH), 7.11–7.14 (1H, m, ar. CH), 7.19–7.24 (2H, m, 2 × ar. CH), 7.25–7.34 (6H, m, $6 \times \text{ar. CH}$), 7.37–7.47 (6H, m, $6 \times \text{ar. CH}$), 7.51-7.60 (3H, m, 3 × ar. CH), 7.59-7.65 (4H, m, 4 × ar. CH), 7.83 (1H, d, J 7.6, Leu NH), 8.02, (1H, d, J 8.6, THF NH), 8.04 (1H, t, J 5.5, AE NH), 8.13 (1H, d, J 7.9, Trp NH), 8.20 (1H, d, J 8.3, Trp NH), 8.24 (1H, d, J 7.6, Leu NH), 10.73 (1H, d, J 1.9, indole NH), 10.76-10.83 (2H, m, indole NH); $[a]_D = -19.8$, $[a]_{365} = -70.8$ (c = 1.1, CHCl₃, 20 °C); HPLC: 13.2 (M1); m/z (ESI) 1381.9 (M + Na⁺. $C_{78}H_{94}N_{10}NaO_{10}Si$ requires 1381.7).

Z-Ala-D-Val-Val-D-Val-Trp-D-THF-D-Leu-Trp-D-Leu-Trp-2- (*t*-butyldiphenylsilyloxy)-ethylamide 16. 319 mg (0.23 mmol) of peptide 30 were *N*-deprotected according to GP 2A (EtOH, 50 °C, 2h; $R_{\rm f}=0.18$ in CHCl₃–MeOH–HCO₂H 200 : 10 : 1) and 369 mg (0.70 mmol) of tetrapeptide 24 were *C*-deprotected according to GP 1. The crude amine and a portion (1/4) of the crude saponification product were then collected and coupled

according to GP 4 (3.0 mm, 3 h). After aqueous work-up, FCC (20 g, 3% MeOH in CHCl₃→5% MeOH in CHCl₃ + 0.5% HCO₂H) gave 231 mg (0.13 mmol, 57%) of pure ethylamide 16 as a colourless foam. $R_f = 0.28$ (CHCl₃-MeOH 20 : 1); δ_H (400 MHz; DMSO- d_6 ; Me₄Si) 0.38 (3H, d, J 6.6, Leu δ -H₃), 0.55 (3H, d, J 6.6, Leu δ-H₃), 0.57 (3H, d, J 6.6, Leu δ-H₃), 0.64 (6H, d, J 5.9, Leu δ -H₃ and Val γ -H₃), 0.72 (3H, d, J 6.3, Val γ -H₃), 0.74-0.83 (12H, m, 4 × Val γ -H₃), 0.86 (3H, d, J 6.9, THF 2'- H_3), 0.98 (9H, s, 3 × t-Bu- H_3), 0.95–1.05 (1H, m, Leu γ -H), 1.09-1.21 (3H, m, THF 4-HH and Leu β -H₂), 1.19 (3H, d, J7.3, Ala β -H₃), 1.22–1.35 (2H, m, THF 4-H*H*, and Val β -H), 1.45– 1.55 (1H, m, THF 3-HH), 1.63 (1H, sextet, J 7.2, Leu γ -H), 1.69–1.79 (1H, m, THF 4-HH), 1.90–2.11 (4H, m, 2 × Val β-H and Leu β -H₂), 2.79 (1H, dd, J 14.2 and 10.1, Trp β -HH), 2.84 $2.95 (2H, m, 2 \times \text{Trp }\beta\text{-}HH), 3.13-3.24 (2H, m, 2 \times \text{Trp }\beta\text{-}HH),$ 3.25–3.31 (3H, m, AE 2-H₂ and Trp β -HH), 3.64 (2H, t, J 6.4, AE 1-H₂), 3.64–3.74 (1H, m, THF 5-H), 3.82 (1H, t, J 7.1, Leu α-H), 3.92 (1H, sextet, J 8.3, THF 1'-H), 4.13 (1H, dt, J 14.5 and 7.3, Ala α -H), 4.17–4.27 (3H, m, Leu α -H, Val α -H and THF 2-H), 4.28–4.36 (2H, m, $2 \times \text{Val }\alpha$ -H), 4.38–4.51 (2H, m, $2 \times \text{Trp } \alpha\text{-H}$), 4.52–4.61 (1H, m, Trp $\alpha\text{-H}$), 4.98 (1H, d, J 12.9, Z-HH), 5.02 (1H, d, J 12.9, Z-HH), 7.26–7.35 (11H, m, Ala NH and $10 \times \text{ar. CH}$), 7.36–7.46 (10H, m, $10 \times \text{ar. CH}$), 7.47– 7.58 (5H, m, $5 \times$ ar. CH), 7.59–7.64 (6H, m, THF NH and $5 \times$ ar. CH), 7.66 (1H, d, J 8.9, Val NH), 7.68 (1H, d, J 7.3, NH), 7.82 (1H, d, J 6.9, Leu NH), 7.92 (1H, d, J 8.9, Val NH), 8.05 (1H, t, J 5.8, AE NH), 8.13 (1H, d, J 6.9, NH), 8.24 (1H, d, J 8.3, Trp NH), 8.26–8.30 (2H, m, Trp NH), 10.71–10.76 (2H, m, indole NH), 10.80 (1H, d, J 1.3, indole NH); HPLC: 10.9 min (250 × 4 mm Jupiter C 5, 70 \rightarrow 100% B in 25 min, A: H_2O ; B: acetonitrile-isopropanol 2 : 1; 0.7 ml min⁻¹); m/z (ESI) 1749.9 (M + Na⁺. $C_{96}H_{126}N_{14}O_{14}Si$ requires 1749.9).

H-Ala-D-Val-Val-D-Val-Trp-D-THF-D-Leu-Trp-D-Leu-Trp-2-(*t*-butyldiphenylsilyloxy)-ethylamide 16a. 210 mg (120 μmol) of peptide 16 were *N*-deprotected according to GP 2A (EtOH, 50 °C). After FCC (8 g; CHCl₃–MeOH 15 : 1—CHCl₃–MeOH–NH₃(conc.) 100 : 10 : 1) 131 mg (0.08 mmol, 69%) of the pure amine 16a were obtained. $R_{\rm f} = 0.09$ in CHCl₃–MeOH–HCO₂H 100 : 10 : 1; HPLC: 7.4 min (M4). m/z (ESI) 1616.1 (M + Na⁺. C₈₈H₁₂₀N₁₄NaO₁₂Si requires 1615.9).

CHO-Val-Gly-Ala-D-Leu-Ala-D-Val-Val-D-Val-Trp-D-THF-D-Leu-Trp-D-Leu-Trp-2-(t-butyldiphenylsilyloxy)-ethylamide 6. 217 mg (0.54 mmol) of tetrapeptide 14 were C-deprotected according to GP 1 ($R_f = 0.21$ in CHCl₃-MeOH-HCO₂H 100: 10:1) and a portion (1/4) of the crude saponification product was coupled with amine 16a (121 mg, 76.0 µmol) according to GP 4 (4.0 mm, 3 h). After aqueous work-up, FCC (14 g, CHCl₃-MeOH-HCO₂H 150 : 10 : 1) gave 130 mg (66.3 μmol, 87%) of the pure peptide 6 as a colourless solid. Prior to single-channel measurements, purification of 10.0 mg of compound 6 by preparative HPLC (250 mm Rainin C 8, 75→90% B in 25 min, A: H₂O; B: acetonitrile-isopropanol 2:1; 10 ml min⁻¹) afforded 9.0 mg (90%) of pure product. $R_f = 0.36$ (CHCl₃-MeOH- $HCO_2H 100: 10: 1); \delta_H (500 MHz; DMSO-d_6; Me_4Si) 0.21 (3H,$ d, J 6.4, Leu δ-H₃), 0.35 (3H, d, J 6.7, Leu δ-H₃), 0.38 (3H, d, J 6.4, Leu δ -H₃), 0.45 (6H, d, J 5.7, 2 × Leu δ -H₃), 0.50–0.73 (30H, m, Leu δ -H₃, THF 2'-H₃ and 8 × Val γ -H₃), 0.74–0.87 (11H, m, $3 \times t$ -Bu-H₃ and $2 \times \text{Leu } \gamma$ -H), 0.88–1.17 (11H, m, $2 \times t$ Ala β-H₃, 2 × Leu β-H₂ and Leu γ-H), 1.19–1.49 (4H, m, THF 4-HH, Val β-H and Leu β-H₂), 1.50–1.63 (1H, m, THF 3-HH), 1.70–1.91 (5H, m, THF 3-HH, THF 4-HH and 3 × Val β -H), 2.53–2.78 (3H, m, 3 × Trp β -HH), 2.91–3.13 (5H, m, 3 × Trp β -HH and AE 2-H₂), 3.45 (2H, t, J 6.2, AE 1-H₂), 3.49–3.60 (3H, m, Gly α -H₂ and Leu α -H), 3.61–3.80 (2H, m, Leu a-H and THF 1'-H), 3.95–4.17 (9H, m, $4 \times \text{Val } \alpha$ -H, THF 2-H and 5-H, Leu α -H and 2 × Ala α -H), 4.18–4.33 (2H, m, 2 × Trp α -H), 4.34–4.43 (1H, m, Trp α -H), 6.67–6.78 (3H, m, 3 × ar. CH), 6.79–7.00 (6H, m, $6 \times \text{ar. CH}$), 7.01–7.14 (3H, m, $3 \times \text{ar.CH}$),

7.15–7.28 (6H, m, 6 × ar.CH), 7.29–7.48 (9H, m, 7 × ar.CH, THF NH and NH), 7.51 (1H, d, J 5.8 Leu NH), 7.60–7.98 (8H, m, 8 × NH), 8.00–8.19 (5H, m, 3 × Trp NH, Gly NH and AE NH), 10.55 (2H, s, indole NH), 10.62 (1H, s, indole NH); HPLC: 7.6 min (250 × 4 mm Jupiter C 5, 65—75% B in 25 min, A: H₂O; B: acetonitrile–isopropanol 2 : 1; 0.7 ml min⁻¹); m/z (ESI): 1984.0 (M + Na⁺. C₁₀₅H₁₄₈N₁₈NaO₁₇Si requires 1984.1).

^{10,11}[D-THF]-gA 5. 30.0 mg (15.3 μmol) of peptide 6 were dissolved in 27 ml CH₂Cl₂. A solution of 5% HF in CH₃CN (2.7 ml) was then added dropwise and the mixture was stirred for 45 min at room temperature. After dilution with 30 ml CH₂Cl₂ the mixture was carefully washed with sat. NaHCO₃ (2 × 10 ml) and brine, and dried over MgSO₄. Evaporation gave 33.0 mg of the crude peptide 5, which was purified by preparative HPLC (250 mm Rainin C 8, 60→75% B in 25 min, A: H₂O; B: acetonitrile-isopropanol 2 : 1; 10 ml min⁻¹). 22.0 mg (12.7 μmol, 83%) of pure product were obtained as a colourless solid. $R_{\rm f} = 0.16 \; (\text{CHCl}_3 - \text{MeOH-HCO}_2 \text{H} \; 100 : 10 : 1); \; \delta_{\rm H} \; (500 \; \text{MHz};$ DMSO- d_6 ; Me₄Si) 0.33 (3H, d, J 6.8, Leu δ -H₃), 0.47 (3H, d, J 6.4, Leu δ-H₃), 0.51 (3H, d, J 6.8, Leu δ-H₃), 0.59 (6H, d, J 6.8, $2 \times \text{Leu } \delta - \text{H}_3$, 0.63–0.87 (30H, m, $8 \times \text{Val } \gamma - \text{H}_3$, THF 2'-H₃ and Leu δ -H₃), 0.88–0.97 (2H, m, 2 × Leu γ -H), 1.04–1.29 (13H, m, $2 \times \text{Ala } \beta\text{-H}_3$, $3 \times \text{Leu } \beta\text{-H}_2$, and Leu γ -H), 1.35–1.62 (5H, m, Leu β -H₂, Leu γ -H, Val β -H and THF 4-HH), 1.65–1.77 (1H, m, THF 3-HH), 1.83-2.03 (5H, m, THF 4-HH, THF 3-HH and $3 \times \text{Val } \beta$ -H), 2.64–2.90 (3H, m, $3 \times \text{Trp } \beta$ -HH), 3.05–3.25 (5H, m, $3 \times \text{Trp }\beta\text{-H}H$ and AE 1-H₂), 3.33 (1H, d, J 6.0, AE 2-HH), 3.37 (1H, d, J 6.4, AE 2-HH), 3.60–3.73 (3H, m, Gly α -H₂ and Leu α -H), 3.74–3.94 (2H, m, Leu α -H and THF 1'-H), 4.05– 4.30 (9H, m, $4 \times \text{Val }\alpha\text{-H}$, $2 \times \text{Ala }\alpha\text{-H}$, Leu $\alpha\text{-H}$, THF 2-H and 5-H), 4.31–4.44 (2H, m, 2 × Trp α -H), 4.46–4.57 (1H, m, Trp α -H), 4.63 (1H, t, J 5.7, OH), 6.82–6.92 (3H, m, 3 × ar. CH), 6.93–7.00 (3H, m, $3 \times \text{ar. CH}$), 7.01–7.08 (3H, m, $3 \times \text{ar. CH}$), 7.23 (3H, d, J 7.9, 3 × ar. CH), 7.47 (3H, dd, J 19.2 and 7.9, 3 × ar. CH), 7.53-7.62 (2H, m, 1H, NH and Leu NH), 7.68 (1H, d, J 7.9, Leu NH), 7.75-7.91 (4H, m, THF NH, AE NH and 2 × NH), 7.94 (1H, d, J 6.8, NH), 8.06 (2H, d, J 7.9, NH), 8.13-8.31 (5H, m, 3 × Trp NH, Gly NH and NH), 10.68 (1H, d, J 0.8, indole NH), 10.70 (1H, d, J 1.9, indole NH), 10.75 (1H, d, J 1.9, indole NH); HPLC: 6.8 min (250 \times 4 mm Jupiter C 5, 65→75% B in 25 min, A: H₂O; B: acetonitrile-isopropanol 2:1; 0.7 ml min⁻¹.); m/z (ESI) 1745.9 (M + Na⁺. C₈₉H₁₃₀- $N_{18}NaO_{17}$ requires 1746.0).

Z-Ala-D-Leu-Ala-D-Val-Val-D-Val-Trp-D-THF-D-Leu-Trp-D-Leu-Trp-2-(t-butyldiphenylsilyloxy)-ethylamide 33. 210 mg (0.60 mmol) of dipeptide 18 were C-deprotected according to GP 1. A portion (1/2) of the crude saponification product was coupled with amine 16a (220 mg 0.14 mmol) according to GP 4 (3.0 mm, 3 h) and after aqueous work-up FCC (20g, 4% MeOH in CHCl₃ + 0.4% HCO₂H) gave 215 mg (0.11 mmol, 79%) of pure peptide 33 as a colourless solid. $R_f = 0.49 \text{ (CHCl}_3\text{-MeOH-}$ NH₃(conc.) 100 : 10 : 1); $\delta_{\rm H}$ (500 MHz; DMSO- d_6 ; Me₄Si) 0.40 $(3H, d, J6.6, \text{Leu }\delta-H_3), 0.54 (3H, d, J6.4, \text{Leu }\delta-H_3), 0.58 (3H,$ d, J 6.4, Leu δ -H₃), 0.65 (6H, d, J 5.5, 2 × Leu δ -H₃), 0.70–0.82 (18H, m, Leu δ -H₃ and 5 × Val γ -H₃), 0.84 (3H, d, J 6.9, Val γ -H₃), 0.89 (3H, d, J 6.4, THF 2'-H₃), 0.98 (9H, s, 3 × t-Bu-H₃), 0.95–1.05 (1H, m, Leu γ-H), 1.19 (6H, d, J 7.1, 2 × Ala β-H₃), 1.09-1.22 (3H, m, THF 3-HH and Leu β -H₂), 1.23-1.36 (2H, m, THF 4-HH and Leu γ-H), 1.42–1.58 (4H, m, Val β-H, THF 3-HH and Leu β-H₂), 1.59–1.68 (1H, m, Leu γ-H), 1.71–1.80 (1H, m, THF 4-HH), 1.92–2.06 (4H, m, 2 \times Val β -H and Leu β -H₂), 2.80 (1H, dd, J 13.9 and 10.8, Trp β -HH), 2.85–2.95 (2H, m, 2 × Trp β-HH), 3.14–3.24 (3H, m, 3 × Trp β-HH), 3.25– 3.20 (2H, m, AE 2-H₂), 3.65 (2H, t, J 6.5, AE 1-H₂), 3.73 (1H, dd, J 14.8 and 7.5 THF 5-H), 3.84 (1H, t, J 7.2, Leu α-H), 3.93 (1H, dd, J 14.4 and 6.9, THF 1'-H), 4.02-4.09 (1H, m, Ala α -H), 4.17–4.33 (7H, m, 2 × Leu α -H, Ala α -H, THF 2-H and 3H and Val α -H), 4.41–4.51 (2H, m, 2 × Trp α -H), 4.54–4.60

(1H, m, Trp α -H), 4.99 (1H, d, J12.6, Z-HH), 5.02 (1H, d, J12.1, Z-HH), 6.90–6.97 (3H, m, 3 × ar. CH), 6.99–7.05 (3H, m, 3 × ar. CH), 7.08 (2H, s, 2 × ar. CH), 7.13 (1H, s, ar. CH), 7.27–7.37 (8H, m, 8 × ar. CH), 7.38–7.47 (7H, m, Ala NH and 6 × ar. CH), 7.51 (1H, d, J7.6, ar. CH), 7.56 (2H, t, J7.5, 2 × ar. CH), 7.56–7.60 (1H, m, NH), 7.61–7.67 (5H, m, 4 × ar. CH and THF NH), 7.71 (1H, d, J7.3, NH), 7.85 (1H, d, J6.9, Leu NH), 7.91 (1H, d, J9.2, NH), 7.98 (1H, d, J7.3, NH), 8.08 (1H, t, J5.8, AE NH), 8.11–8.18 (2H, m, NH), 8.25–8.36 (3H, m, Trp NH), 10.73–10.78 (2H, m, indole NH), 10.82 (1H, s, indole NH); HPLC: 11.8 min (M4); m/z (ESI): 1933.8 (M + Na $^+$. $C_{105}H_{142}N_{16}NaO_{16}Si$ requires 1934.0).

H-Ala-D-Leu-Ala-D-Val-Val-D-Val-Trp-D-THF-D-Leu-Trp-D-Leu-Trp-2-(t-butyldiphenylsilyloxy)-ethylamide 33a. 200 mg (104 μmol) of peptide 33 were N-deprotected according to GP 2A (EtOH, 50 °C). After filtration and evaporation FCC (25 g, CHCl₃–MeOH–NH₃(conc.) 100 : 10 : 1) gave 125 mg (70.0 μmol, 67%) of amine 33a. $R_{\rm f}$ = 0.15 (CHCl₃–MeOH–HCO₂H 100 : 10 : 1). HPLC: 9.9 min (250 × 4 mm Dynamax C8, 60→100% B in 25 min, A: H₂O + 0.1% TFA; B:acetonitrile—isopropanol 2 : 1 + 0.1% TFA; 0.7 ml min⁻¹). m/z (ESI): 1799.7 (M + Na⁺. C₉₇H₁₃₆N₁₆NaO₁₄Si requires 1800.0).

Succ-1-(1'-t-butyldiphenylsilyloxy-^{10,11}[D-THF]-gramicidin **A)-4-(Val-Gly-OMe)-diamide 34.** 23.5 mg (44.0 μmol) of benzyl ester 17 were C-deprotected according to GP 2A. The crude carboxylic acid and 40.0 mg (22.5 µmol) of amine 33a were then collected and coupled according to GP 4 (1.0 mm, 3 h). After aqueous work-up FCC (4 g, 4% MeOH in CHCl₃→6% MeOH in CHCl₃ + 0.6% HCO₂H) gave 35.0 mg (16.0 mmol, 73%) of pure diamide 34 as a colourless solid. $R_f = 0.35$ (CHCl₃-MeOH–HCO₂H 100 : 10 : 1); δ_H (500 MHz; DMSO-d₆; Me₄Si) $0.40 (3H, d, J 6.6, Leu \delta-H_3), 0.55 (3H, d, J 6.4, Leu \delta-H_3), 0.58$ $(3H, d, J 6.4, Leu \delta-H_3), 0.65 (6H, d, J 6.2, 2 \times Leu \delta-H_3), 0.72$ (3H, d, J 6.6, Leu δ-H₃), 0.74–0.82 (15H, m, 5 × Val γ -H₃), 0.83-0.93 (18H, m, $5 \times \text{Val } \gamma$ -H₃ and THF 2'-H₃), 0.98 (9H, s, $3 \times t$ -Bu-H₃), 0.99–1.04 (1H, m, Leu γ -H), 1.08–1.34 (13H, m, $2 \times \text{Ala } \beta\text{-H}_3$, THF 3-HH and 4-HH, $2 \times \text{Leu } \beta\text{-H}_2$ and Leu γ -H), 1.41–1.60 (4H, m, Leu β-H₂, THF 3-HH and Val β-H), 1.61-1.69 (1H, m, Leu γ -H), 1.71-1.81 (1H, m, THF 4-HH), 1.91–2.07 (4H, m, 4 × Val β -H), 2.36–2.47 (4H, m, succ 3-H₂ and 4-H₂), 2.75–2.83 (1H, m, Trp β -HH), 2.85–2.96 (2H, m, 2 × Trp β-HH), 3.13–3.30 (5H, m, 3 × Trp βHH and AE 2-H₂), 3.62 (3H, s, OCH₃), 3.65 (2H, t, J 6.5, AE 1-H₂), 3.67–3.89 (6H, m, 2 \times Gly α -H₂, THF 5-H and Leu α CH), 3.90–3.96 (1H, m, THF 1'-H), 4.06–4.34 (10H, m, $5 \times \text{Val } \alpha$ -H, $2 \times \text{Leu } \alpha$ -H, $2 \times \text{Ala}$ α -H, and THF 2-H), 4.40–4.50 (2H, m, 2 × Trp α -H), 4.54–4.61 (1H, m, Trp α -H), 6.90–6.97 (3H, m, 3 × ar. CH), 6.99–7.05 (3H, m, 3 × ar. CH), 7.08 (2H, s, 2 × ar. CH), 7.13 (1H, d, J 1.4, ar. CH), 7.28-7.32 (3H, m, 3 × ar. CH), 7.38-7.47 (6H, m, 6 × ar. CH), 7.51 (1H, d, J 7.6, ar. CH), 7.56 (2H, t, J 7.1, ar. CH), 7.59–7.64 (5H, m, $4 \times$ CH and NH), 7.65 (1H, d, J 8.5, THF NH), 7.71 (1H, d, J 6.7, NH), 7.83-7.93 (4H, m, Leu NH and 3 \times NH), 7.96 (1H, d, J 8.9, NH), 8.05–8.11 (2H, m, 2 \times NH), 8.12–8.17 (1H, m, NH), 8.20–8.37 (4H, m, 4 × NH), 8.39 (1H, t, J 5.84, Gly NH), 10.75 (1H, s, indole NH), 10.77 (1H, s, indole NH), 10.82 (1H, s, indole NH). HPLC: 9.8 min (M4); m/z (ESI): 2226.4 (M + Na⁺. $C_{116}H_{166}N_{20}NaO_{21}Si$ requires 2226.2).

Succ-(10,11 [D-THF]-gA-TBDPS)₂ 7. 39.0 mg (17.7 μmol) of methyl ester 34 were *C*-deprotected according to GP 1 (R_f = 0.20 in CHCl₃–MeOH–HCO₂H 100 : 10 : 1; HPLC: 9.8 min, 250 × 4 mm Jupiter C 5, 70—100% B in 25 min, A: H₂O + 0.1% TFA; B: acetonitrile–isopropanol 2 : 1 + 0.1% TFA; 0.7 ml min⁻¹). The crude carboxylic acid and 40.0 mg (22.5 μmol) of amine 33a were collected and coupled according to GP 5 (1.0 mm, 12 h). After aqueous work-up 64.0 mg of crude diamide 7 were isolated and purified directly by preparative HPLC (250 mm Rainin C 4, 60—100% B in 25 min, A: H₂O + 0.1% TFA;

B: acetonitrile-isopropanol 2 : 1 + 0.1% TFA; 14 ml min⁻¹). Treatment of the collected fractions according to GP 6 gave 36.0 mg (9.1 µmol, 61%) of pure diamide 7 as a colourless solid. $\delta_{\rm H}(500 \text{ MHz}; \text{ DMSO-}d_6; \text{ Me}_4\text{Si}) 0.40 (6\text{H}, d, J 6.6, 2 \times \text{Leu})$ δ -H₃), 0.54 (6H, d, J 6.4, 2 × Leu δ -H₃), 0.58 (6H, d, J 6.4, 2 × Leu δ -H₃), 0.65 (12H, d, J 6.0, $4 \times \text{Leu } \delta$ -H₃), 0.72 (6H, d, J 6.19, $2 \times$ Leu δ-H₃), 0.74–0.82 (30H, m, $10 \times \text{Val } \gamma$ -H₃), 0.83–0.87 (18H, m, $6 \times \text{Val } \gamma$ -H₃), 0.89 (6H, d, J 6.6, $2 \times \text{THF } 2'$ -H₃), 0.98 (18H, s, $6 \times t$ -Bu-H₃), 0.99–1.05 (2H, m, $2 \times \text{Leu } \gamma$ -H), 1.09–1.38 $(26H, m, 4 \times Ala \beta-H_3, 4 \times Leu \beta-H_2, 2 \times THF 4- HH, 2 \times THF)$ 3-HH and 2 × Leu γ-H), 1.39–1.59 (8H, m, 2 × Leu β-H₂, 2 × THF 3-H*H* and 2 × Val β-H), 1.61–1.68 (2H, m, 2 × Leu γ -H), 1.72-1.80 (2H, m, $2 \times \text{THF 4-H}H$), 1.90-2.06 (6H, m, $6 \times \text{Val}$ β-H), 2.36–2.47 (4H, m, succ 2-H₂ and 3-H₂), 2.76–2.84 (2H, m, $2 \times \text{Trp }\beta$ -HH), 2.85–2.95 (4H, m, $4 \times \text{Trp }\beta$ -HH), 3.14–3.31 $(6H, m, 6 \times Trp β-HH)$, 3.52 $(2H, s, 2 \times Gly α-HH)$, 3.61–3.68 $(6H, m, 2 \times AE 2-H, and 2 \times Gly \alpha-HH), 3.69-3.78 (4H, m, 2 \times AE 2-H, and 2 \times Gly \alpha-HH)$ THF 5-H and 2 × Leu α -H), 3.81–3.88 (2H, m, 2 × Leu α -H), 3.89-3.97 (2H, m, $2 \times \text{THF 1'-H}$), 4.07-4.13 (2H, m, $2 \times \text{Val } \alpha$ -H), 4.15–4.33 (14H, m, 6 × Val α-H, 2 × THF 2-H, 4 × Ala α-H and 2 × Leu α -H), 4.39–4.51 (4H, m, 4 × Trp α -H), 4.54–4.61 $(2H, m, 2 \times Trp \alpha - H), 6.90 - 6.97 (6H, m, 6 \times ar. CH), 6.99 - 7.06$ $(6H, m, 6 \times ar. CH)$, 7.08 $(4H, s, 4 \times ar. CH)$, 7.13 $(2H, s, 2 \times ar. CH)$ CH), 7.30 (6H, d, J 8.0, 6 × ar. CH), 7.37–7.47 (12H, m, 12 × ar. CH), 7.51 (2H, d, J 7.8, 2 × ar. CH), 7.56 (4H, t, J 6.5, 4 × ar. CH), 7.59–7.67 (12H, m, 8 × ar. CH, NH and THF NH), 7.68– 7.73 (2H, m, $2 \times NH$), 7.82–7.94 (8H, m, $6 \times NH$ and $2 \times Leu$ NH), 8.00-8.11 (6H, m, $4 \times$ NH and $2 \times$ AE NH), 8.12-8.18 $(2H, m, 2 \times NH), 8.21-8.39$ (8H, m, 6 × Trp NH and 2 × Gly NH), 10.73–10.79 (4H, m, 4 \times indole NH), 10.82 (2H, s, 2 \times indole NH); HPLC: 18.1 min (M4); m/z (ESI): 1998.0 (M + $2Na^{+}$. $C_{212}H_{298}N_{36}Na_{2}O_{34}Si$ requires 1997.1).

Succ-1-(10,11 [D-THF-]-gA-TBDPS)-4-(gA-TBDPS) 8. 70.0 mg (29.6 µmol) of methyl ester 35 were C-deprotected according to GP 1. The crude carboxylic acid and 40.0 mg (22.5 µmol) of amine 33a were collected and coupled according to GP 5 (1.0 mm, 12 h). After aqueous work-up 81.0 mg of crude diamide 8 were obtained and purified by preparative HPLC (250 mm Rainin C 4, 80→90% B in 10 min, then 90→100% B in 15 min. A: H₂O + 0.1% TFA; B: acetonitrile-isopropanol 2:1 + 0.1% TFA; 14 ml min⁻¹). Treatment of the collected fractions according to GP 6 gave 43.0 mg (10.5 µmol, 46%) of pure diamide 8 as a colourless solid. $\delta_{\rm H}$ (500 MHz; DMSO- d_6 ; Me₄Si) 0.40 (3H, d, J 6.4, Leu δ-H₃), 0.49 (6H, d, J 6.4, 2 × Leu δ -H₃), 0.51–0.62 (15H, m, 5 × Leu δ -H₃), 0.61–0.68 (9H, m, 3 × Leu δ-H₃), 0.72 (3H, d, J 6.2, Leu d-H₃), 0.74–0.91 (54H, m, $2 \times$ Leu δ-H₃ and 16 × Val γ -H₃), 0.92–0.95 (3H, m, THF 2'-H₃), $0.97 \text{ (9H, s, } 3 \times t\text{-Bu-H}_3), 0.98 \text{ (9H, s, } 3 \times t\text{-Bu H}_3), 1.09-1.32$ (26H, m, 4 × Ala β -H₃, 5 × Leu β -H₂ and 4 × Leu γ -H), 1.42– 1.59 (8H, m, $2 \times \text{Leu }\beta\text{-H}_2$, $2 \times \text{Val }\beta\text{-H}$, THF 4-HH and THF 3-HH), 1.61–1.67 (1H, m, Leu γ -H), 1.74–1.82 (1H, m, Leu γ -H), 1.90–2.05 (7H, m, 6 × Val β-H and Leu γ -H), 2.19 (1H, t, J 7.3, THF 3-HH), 2.25-2.32 (1H, m, THF 4-HH), 2.36-2.47 (4H, m, succ 2-H, and 3-H₂) 2.74–2.95 (6H, m, $3 \times \text{Trp }\beta$ -H₂), 3.06-3.29 (8H, m, $3 \times \text{Trp } \beta$ -H₂ and AE 2-HH), 3.35-3.43 (2H, m, AE 2-HH), 3.46-3.50 (2H, m, Trp β-H₂), 3.58-3.68 (6H, m, $2 \times \text{C-2 H}_2$ and Gly α -H₂), 3.69–3.75 (2H, m, $2 \times \text{Leu } \alpha$ -H), 3.82-3.88 (1H, m, Leu α -H), 3.90-3.95 (2H, m, THF 1'-H and Leu α -H), 4.05–4.16 (5H, m, 2 × Val α -H, Gly α -H₂ and Leu α -H), 4.17–4.34 (13H, m, 4 × Ala α -H, Leu α -H, THF 5-H and 2-H and 6 × Val α -H), 4.41–4.61 (7H, m, 7 × Trp α -H), 6.88– 6.97 (6H, m, $6 \times ar$. CH), 6.99–7.05 (6H, m, $6 \times ar$. CH), 7.08 $(6H, s, 6 \times ar. CH)$, 7.13 (1H, s, ar. CH), 7.25–7.32 $(7H, m, 7 \times ar. CH)$ ar. CH), 7.36-7.47 (12H, 12 × ar. CH), 7.49-7.58 (7H, m, 7 × ar. CH), 7.59–7.63 (12H, m, 10 × ar. CH and 2 × NH), 7.64–7.67 (1H, m, THF NH), 7.68–7.82 (3H, m, NH), 7.85 (1H, d, J 5.7, Leu NH), 7.87–7.96 (7H, m, 7 × NH), 7.98 (1H, d, J 5.7, Leu NH), 8.01-8.18 (8H, m, $7 \times$ NH and Trp NH), 8.21-8.31 (6H, m, Gly NH and 5 × Trp NH), 8.33 (1H, d, J 6.4, Trp NH), 10.71

(1H, s, indole NH), 10.76 (5H, s, $5 \times \text{indole NH}$), 10.82 (1H, s, indole NH). HPLC: 10.6 min (10 × 250 mm Rainin C 4, $80 \longrightarrow 90\%$ B in 10 min, then $90 \longrightarrow 100\%$ B in 15 min. A: H₂O + 0.1% TFA; B: acetonitrile–isopropanol 2 : 1 + 0.1% TFA; 3.3 ml min⁻¹); m/z (ESI) 2077.4 (M + 2Na⁺. C₂₂₂H₃₀₈N₃₈Na₂O₃₄Si₂ requires 2076.2).

Abbreviations

Amino acid residues

For the schemes the common one letter code was used, where capital letters denote L- and small letters D-configuration. V = L-Val, G = Gly, A = L-Ala, l = D-Leu, v = D-Val, W = L-Trp.

Z: carboxy-benzyl; TBDPS: t-butyl-diphenyl-silyl; HOBt: 1-hydroxy-benzotriazole monohydrate; EDC: N-ethyl-N'-(3-dimethylamino-propyl)-carbodiimide hydrochloride; HBTU: O-(benzotriazol-1'-yl)-1,1,3,3-tetramethyl-uronium hexafluorophosphate; DIEA: di-iso-propyl-ethylamine; HATU: O-(7'-aza-benzotriazol-1'-yl)-1,1,3,3-tetramethyl-uronium hexafluorophosphate; HOAt: 1-hydroxy-7-aza-benzotriazole; FCC: flash column chromatography; TFE: trifluorethanol; DMSO: dimethylsulfoxide; DMF: dimethylformamide; DIBAL-H: diisobutyl aluminium hydride; MCPBA: m-chloroperbenzoic acid.

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