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Enantioselective Total Synthesis of (+)-Ottelione A, (-)-Ottelione B, (+)-3-epi-Ottelione A and Preliminary Evaluation of Their Antitumor Activity

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Abstract: Enantioselective total synthesis of (+)-ottelione A (1) and (-)-ottelione B (2), novel and potent antitumor agents from a freshwater plant, and (+)-3-epi-ottelione A (3), the earlier proposed stereostructure of 1, was efficiently achieved starting from the known tricyclic compound 10. The synthesis involved the following key steps: i) coupling reactions of aldehydes 8 and 9 with the aromatic portion 7 (8+ $7\rightarrow15$ and $9+7\rightarrow27$), ii) base-induced

hemiacetal-opening/epimerization reactions of the cyclic hemiacetals 6 and 27 ($6\rightarrow17$ and $27a\rightarrow26a$), and iii) Corey-Winter's reductive olefination of the cyclic thiocarbonates 21 and 36 ($21\rightarrow22$ and $36\rightarrow37$). The present total synthe-

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sis fully established the absolute configuration of these natural products. The cell growth inhibition profile, COM-PARE analysis, and tubulin inhibitory assay of (+)-3-epi-ottelione A (3) and its O-acetyl derivative 24 demonstrated that these unnatural substances could be prominent lead compounds for the development of anticancer agents with a novel mode of action.

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- Supporting information for this article is available on the WWW under http://www.chemeurj.org/ or from the author: ¹H and ¹³C NMR spectra for all new compounds.

Introduction

In 1998, Ayyad and Hoye et al. reported the isolation and structural elucidation of two novel diastereomeric natural products, otteliones A (1) and B (2) (Figure 1), from the freshwater plant *Ottelia alismoides* collected from the Nile Delta, Egypt.^[1,2] These substances were found to exhibit prominent biological properties such as antitubercular^[3] and antitumor activities.^[1,2] Remarkably, these small-molecule

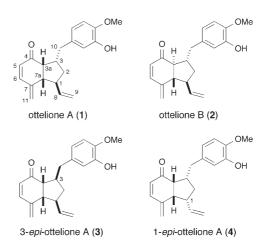


Figure 1. Structures of ottelione A (1), ottelione B (2), 3-epi-ottelione A (3), and 1-epi-ottelione A (4).

natural products were reported to display quite potent cytotoxicity at nanomolar to picomolar levels of IC50 values against a panel of 60 human tumor cell lines at the National Cancer Institute in the United States.^[1] It has also been reported that ottelione A inhibits tubulin polymerization into microtubules similarly to the well-known alkaloids colchicine, vincristine, and vinbrastine.[4] In addition, ottelione A has been shown to inhibit doxorubicin-resistant leukemia cells (P388/DOX) with an IC_{50} value of 1.0 ng mL⁻¹ (3.0 nm).^[1] Otteliones, therefore, are expected to be promising new leads for the development of cancer chemotherapeutic agents. However, detailed study of the biological properties of these natural products is severely limited by their scarcity. It has been reported that each 9.0 mg of ottelione A (1) and B (2) was isolated from 1.0 kg (dry weight) of the plant O. alismoides (each 0.0009 % isolated yield).^[1]

Structurally, otteliones A and B possess a novel bicyclic hydrindane skeleton with four contiguous asymmetric carbon centers, in which the rare and sensitive 4-methylene-2-cyclohexenone substructure is a special characteristic feature. [5] When these natural products were isolated, the relative configuration of ottelione B was elucidated as shown in formulation 2 (Figure 1) by the combination of molecular modeling and ¹H NMR studies; however, ottelione A could not be assigned unambiguously and both formulations 3 and 4 were proposed as possible stereostructures with more preference for 3.^[1] In 2000, scientists from Rhône-Poulenc Rorer (now Sanofi-Aventis) proposed an alternative stereostructure 1 for ottelione A, which was assigned based on the analysis of 2D-NMR spectra including COSY, HMQC, and NOESY experiments, [6] while its absolute configuration was not determined.

The remarkable biological properties, unique structural features, and limited availability from natural resources, as well as the necessity of confirming the absolute configuration, have made the otteliones exceptionally intriguing and timely targets for total synthesis. A number of synthetic approaches for otteliones A and B has been reported to date. [7] The first total synthesis of racemic (\pm)-otteliones A (1) and B (2) was reported by Mehta and Islam in 2002, [8] which verified their relative stereostructures. In the following year, Mehta and Islam^[9] and our groups^[10] independently accomplished the first enantioselective total synthesis of naturally occurring (+)-ottelione A (1) and (-)-ottelione B (2), leading to the establishment of their absolute configurations as depicted in Figure 1. In addition, we also achieved an enantioselective total synthesis of 3-epi-ottelione A (3),^[7f] which represents the earlier proposed formulation of ottelione A. Recently, Clive and Liu reported the elegant total synthesis of (+)-1 and (-)-2 in an enantioselective manner.[11] In this paper, we describe the enantioselective total synthesis of ottelione A (1), ottelione B (2), and 3-epiottelione A (3). In addition, the cell growth inhibition analysis and tubulin polymerization assay of unnatural 3-epi-ottelione A (3) and its O-acetyl derivative 24 along with ottelione A (1) are also described.

Results and Discussion

Synthesis of 3-epi-ottelione A (3), an earlier proposed stereostructure of ottelione A (1)

Synthetic plan: At the beginning of this project (April 2000), formulation **3** was considered as the most likely stereostructure for ottelione A (1); therefore, our initial synthetic efforts were targeted toward structure **3**. Our synthetic plan for 3-epi-ottelione A (3) is outlined in Scheme 1. The

Scheme 1. Synthetic plan for 3-epi-ottelione A (3). TBS = tert-butyldimethylsilyl, MOM = methoxymethyl.

key feature of this plan is the utilization of the highly and appropriately functionalized intermediate 8, which contains the requisite bicyclic hydrindane core framework possessing the correct stereochemistries at the C3, C3a, and C7a positions as well as the desirable functionalities for elaboration of the target molecule 3. Intermediate 8 should be accessible from intermediate 9 by epimerization at the C3 position. The 4-methylene-2-cyclohexenone substructure present in molecule 3 was expected to be highly sensitive; therefore, we decided to elaborate this substructure at the final stage of the synthesis. Since the C1 formyl group of 8 is masked as an internal hemiacetal moiety, intermediate 6 would be constructed through the coupling reaction of 8 with the aryllithium generated from bromobenzene derivative 7. Intermediate 6 would be converted into the target molecule 3 through the advanced key intermediate 5 by sequential functional group manipulation and deprotection or vice versa; the sequence involves base-induced hemiacetal-opening/epimerization at the masked C1 formyl group and subsequent construction of the 4-methylene-2-cyclohexenone substruc-

ture as the crucial steps. Intermediate **9** should in turn be accessed from the known tricyclic compound **10**, which was previously prepared in our laboratories by Diels–Alder cycloaddition between optically active cyclohexenone **11** and cyclopentadiene (**12**).^[12]

Synthesis of intermediate 9: First, as shown in Scheme 2, we pursued the synthesis of intermediate 9 starting from the known enantiomerically pure tricyclic compound 10. Thus, stereoselective reduction of the carbonyl group in 10 with NaBH₄ provided the expected alcohol 13 in 87% yield as a single stereoisomer. Subsequent Lemieux–Johnson oxidation $(OsO_4/NaIO_4)^{[13]}$ of 13 furnished the cyclic hemiacetal 9 in 75% yield through the intermediary dialdehyde 14. The stereochemistry at the C8 position in 9 was assigned based on NOESY experiments in the $500\,\text{MHz}$ ^1H NMR spectrum, in which a clear NOE interaction between C8-H and C2-H α was observed.

Scheme 2. Synthesis of intermediate 9. a) NaBH₄, THF/H₂O 10:1, 0°C, 87%; b) OsO₄, NaIO₄, $tBuOH/THF/H_2O$ 8:6:3, 0°C \rightarrow RT, 75%.

Synthesis of intermediate 5: Having obtained intermediate 9, we next performed the synthesis of intermediate 5 possessing the requisite carbon skeleton and correct stereochemistries at the C1, C3, C3a and C7a positions as shown in Scheme 3. Epimerization at the C3 position of 9 occurred smoothly and cleanly by treatment with 1,8-diazabicyclo-[5.4.0]undec-7-ene (DBU) in THF at ambient temperature for 2 h to give the desired β -formyl compound 8 in quantitative yield. The β -disposition of the C3 formyl group in 8 was confirmed by NOESY experiments, which showed clear NOE interactions between C3-H and C5-H, C2-H α .

The crucial coupling reaction of **8** with the aromatic portion **7** was efficiently achieved by halogen/metal exchange of 4-bromo-1-methoxy-2-(methoxymethoxy)benzene (**7**)^[14] with *n*BuLi in THF at -78 °C followed by the addition of **8** at the same temperature; the desired coupling product **15** was obtained in quantitative yield as an inseparable mixture of epimeric alcohols (9:1 by 500 MHz ¹H NMR). All attempts to directly remove the benzylic hydroxy group from **15** under conventional conditions such as Barton–McCombie deoxygenation [NaN(SiMe₃)₂, CS₂, MeI; (*n*Bu)₃SnH, AIBN] or Birch reductive deoxygenation (Li metal, liq.

Scheme 3. Synthesis of intermediate **5**. a) DBU, THF, RT, quant.; b) 4-bromo-1-methoxy-2-(methoxymethoxy)benzene **(7)**, nBuLi, THF, $-78\,^{\circ}C$; at $-78\,^{\circ}C$, add. **8**, quant.; c) Ac₂O, DMAP, pyridine, RT, quant.; d) Li, liq. NH₃, THF, $-78\,^{\circ}C$, 87%; e) DBU, toluene, reflux, 52% (see entry 3 in Table 1); f) Dess–Martin periodinane, CH₂Cl₂, RT, 90%; g) Ph₃P+CH₃Br⁻, tBuOK, benzene, RT \rightarrow reflux, 88%. DBU=1,8-diazabicyclo[5.4.0]undec-7-ene, DMAP=4-dimethylaminopyridine.

NH₃) were unsuccessful. This was attributed to the presence of reactive hemiacetal function in substrate **15**; therefore, we decided to protect the hydroxy group in the hemiacetal moiety. Thus, acetylation of the hydroxy groups in **15** followed by treatment of the resulting diacetates **16** with a large excess of Li metal (50 equiv) in liquid NH₃/THF at -78 °C for 5 min resulted in the formation of the desired deoxygenated lactol **6** in 87 % yield in two steps. In Birch reductive deacetoxylation, the best results were obtained with a large excess of Li metal at low temperature (-78 °C) in a short time (5 min).

We next examined the base-induced hemiacetal-opening/epimerization reaction of **6** to obtain **17** possessing the requisite stereochemistry at the C1 position. The result summarized in Table 1 deserves some comments. This one-pot two-step reaction via the intermediary **6 A** was best achieved by treatment of **6** with DBU (20 equiv) in refluxing toluene for 1.5 h (entry 3), which provided the desired product **17** in 52% yield along with 40% recovery of the starting material **6**. The choice of the appropriate reaction conditions such as equivalent of DBU and temperature is very crucial in this reaction. By employing lower amounts of DBU (2–10 equiv) (entries 1, 2), a lower yield of **17** (14–25%) was produced and a large amount of the starting material **6** (86–70% yield) was recovered unchanged. When a large excess of DBU (50 equiv) was used (entry 4), only un-

Table 1. Base-induced hemiacetal-opening/epimerization reaction of intermediate 6 leading to intermediate 17.

Entry	DBU [equiv]	Solvent	$T^{[a]}$ [°C]	t [h]	Yield [%] ^[b]	
					17	6
1	2	toluene	110	10	14	86
2	10	toluene	110	5	25	70
3	20	toluene	110	1.5	52	40
4	50	toluene	110	1.5	decomposition	
5	20	benzene	80	1.5	no reaction	
6	20	xylene	140	1.5	decomposition	

[a] Reflux temperature. [b] Isolated yield.

identified decomposition products were generated in the reaction mixture. In addition, when the reaction was carried out at a lower temperature (80°C) (entry 5), it did not proceed and substrate 6 was recovered quantitatively. At a higher temperature (140°C) (entry 6), the complete decomposition of the materials was observed. To continue the synthesis (cf. Scheme 3), compound 17 was further converted to the key intermediate 5 in 79% overall yield via a two-step sequence involving Dess-Martin oxidation followed by twofold Wittig methylenation of the resulting keto-aldehyde 18.

Completion of synthesis of (+)-3-epi-ottelione A (3): Having obtained the key intermediate 5 in an efficient way, we next investigated the synthesis of 3-epi-ottelione A (3), as shown in Scheme 4, which involves the crucial elaboration of the highly sensitive 4-methylene-2-cyclohexenone substructure. To this end, treatment of 5 with CF₃CO₂H in THF containing H₂O at 0°C effected simultaneous deprotection of both the acetonide moiety and the MOM group to provide the corresponding triol 19 in 76% yield. Subsequent chemoselective acetylation of the phenolic hydroxy group in 19 afforded the acetate 20 in 90% yield. Installation of the C5-C6 double bond was successfully achieved by employing Corey-Winter's protocol^[15] with some modifications to the reaction conditions. Thus, treatment of 20 with thiophosgene (2.0 equiv) in the presence of DMAP (5.0 equiv) followed by exposure of the resulting cyclic thiocarbonate 21 to refluxing triethyl phosphite, resulted in the formation of the requisite diene 22 in 73% yield in two steps. Subsequent selective removal of the TBS group in 22 was successfully attained via a one-pot two-step operation involving treatment with tetra-n-butylammonium fluoride (TBAF) followed by chemoselective acetylation of the liberated phenolic hydroxy group; the desired alcohol 23 was produced in 92% yield. Dess-Martin oxidation of 23 furnished dienone 24 possessing the sensitive 4-methylene-2-cyclohexenone substructure in 97% yield. Finally, the targeted 3-epi-ottelione A (3) was obtained in 98% yield upon deacetylation (K₂CO₃, MeOH, 0°C) of 24.

Scheme 4. Synthesis of 3-epi-ottelione A (3). a) CF₃CO₂H, THF/H₂O 10:1, 0°C, 76%; b) Ac₂O, 2M NaOH, iPrOH, RT, 90%; c) CSCl₂, DMAP, CH₂Cl₂, RT, 96%; d) (EtO)₃P, reflux, 68%; e) TBAF, THF, RT; Ac₂O, 92%; f) Dess-Martin periodinane, CH₂Cl₂, RT, 97%; g) K₂CO₃, MeOH, 0°C, 98%. TBAF = tetra-n-butylammonium fluoride.

The structure and stereochemistry of 3, the earlier proposed stereostructure of ottelione A (1), was unambiguously confirmed by extensive spectroscopic analysis including NOESY experiments in the 500 MHz ¹H NMR spectra. The selected NOESY correlation of 3 is depicted in Figure 2, wherein clear NOE interactions between H_{3a} and H_{2B}, H_{7a}, $H_{10\alpha}$, $H_{10\beta}$, between $H_{7\alpha}$ and $H_{2\beta}$, H_{8} , and between $H_{2\alpha}$ and H₃, H₁ are observed, revealing that both the C1 and C3 substituents are syn to the bridgehead protons H_{3a} and H_{7a} in the cis-fused hydrindane skeleton. To our disappointment,

OMe
$$\begin{array}{c} H_{10\beta} \\ OH_{3a} \\ OH \\ H_{3a} \end{array} \xrightarrow{\begin{array}{c} OH \\ H_{2\beta} \\ H_{2\alpha} \\ H_{3a} \end{array}} \xrightarrow{\begin{array}{c} OH \\ H_{2\alpha} \\ H_{3a} \\ H_{3a} \\ H_{3a} \end{array} \xrightarrow{\begin{array}{c} OH \\ H_{3a} \\ H_{3a} \\ H_{3a} \\ H_{3a} \\ \end{array}} \xrightarrow{\begin{array}{c} OH \\ H_{3a} \\ H_{3a} \\ H_{3a} \\ \end{array}} \xrightarrow{\begin{array}{c} OH \\ H_{3a} \\ H_{3a} \\ H_{3a} \\ \end{array}} \xrightarrow{\begin{array}{c} OH \\ H_{3a} \\$$

Figure 2. Selected NOESY correlation of 3-epi-ottelione A (3).

the 1H NMR spectral data of the synthesized compound 3 did not match that of the natural product ottelione A; however, we achieved the first total synthesis of 3 (3-epi-ottelione A), the earlier proposed formulation of ottelione A (1), in an enantiomerically pure form. [7f] Coincidentally, at the time we completed the projected synthesis of 3, the first total synthesis of (\pm)-ottelione A (1) and B (2) was reported by Mehta and Islam, [8] which verified their relative stereostructures as shown in Figure 1. Consequently, our synthetic efforts were directed toward formulations 1 and 2 (i.e., otteliones A and B). This is the subject of the following section.

Synthesis of ottelione A (1) and ottelione B (2)

Synthetic plan: Our redesigned synthetic plan for ottelione A (1) and B (2) is outlined in Scheme 5, based on the synthesis of 3-epi-ottelione A (3) described above. We envisaged that the highly and appropriately elaborated intermediate 27 would be converted to the target molecules 1 and 2

Scheme 5. Synthetic plan for ottelione A (1) and ottelione B (2).

via the intermediates **26** and **25** by a methodology similar to the synthesis of 3-epi-ottelione (3) from intermediate **9** (cf. Scheme 1 and two previous sections). Epimerization at the C3a position in ottelione A (1) would result in ottelione B (2). Intermediate **27** would, in turn, be produced through the coupling reaction of the common intermediate **9** and the aromatic portion **7**.

Synthesis of intermediate 26: As shown in Scheme 6, we initially investigated the synthesis of the advanced key intermediate 26a possessing all four correct stereogenic centers (C1, C3, C3a, and C7a) and the proper functionalities for elaboration of the target molecules 1 and 2. The coupling reaction of the aldehyde 9 with the aryllithium generated in

Scheme 6. Synthesis of intermediate **26a**. a) 4-bromo-1-methoxy-2-(methoxymethoxy)benzene **(7)**, nBuLi, THF,-78°C; at -78°C, add. **9**, 80% for **27a**, 20% for **27b**; b) Li, liq. NH₃, THF, -78°C, 73% for **27a** \rightarrow **28**, 75% for **27b** \rightarrow **28**; c) DBU, toluene, reflux, no reaction (recovery of **28**) for **28** \rightarrow **29**; 30% (65% based on recycling four times) for **27a** \rightarrow **26a**; no reaction (recovery of **27b**) for **27b** \rightarrow **26b**.

situ from 7 proceeded cleanly and smoothly to furnish the desired products 27a (80% yield) and 27b (20% yield) as a mixture of epimeric alcohols separated by column chromatography on silica gel. The stereochemistry at the benzylic C10 position in the coupling products **27a**,**b** was tentatively assigned based on the well-known Felkin-Anh model, while the stereogenic center disappeared at a later stage of the synthesis. The C10 hydroxy group in both 27a and 27b was removed under Birch reduction conditions (Li metal, liquid NH₃/THF, -78°C) to obtain the deoxygenated hemiacetal 28 in 73 and 75% yield, respectively. The crucial base-induced hemiacetal-opening/epimerization reaction of 28 leading to the hydroxy aldehyde 29 was attempted next under similar conditions (DBU, toluene, reflux) to those employed in the Section on intermediate 5 (cf. $6\rightarrow 17$, Scheme 3 and Table 1); however, the reaction did not proceed and the starting material 28 was recovered unchanged. We reasoned that substrate 28 might be thermodynamically much more stable than the hemiacetal-opened hydroxy aldehyde 28a. Therefore, we next examined the hemiacetal-opening/epimerization reaction employing hemiacetals 27 a,b as the alternative substrates. Thus, in the case of 27a, the expected hemiacetal-opening/epimerization reaction proceeded under the same conditions (DBU, toluene, reflux, 1.5 h) described above, which resulted in the formation of the desired product 26a (30% yield) along with the starting material 27a (60% yield). Since prolonged reaction time caused decomposition of 26a and/or 27a, the reaction was terminated when an approximately 1:2 mixture of 26 a/27 a was generated. After recycling the recovered starting material 27a four times, we could obtain 65% yield of the desired compound 26a. On the contrary, the same treatment of the minor product 27b turned out to be unsuccessful and the starting material was recovered unchanged. Therefore, the projected synthesis was conducted forward using only the major coupling product 27a.

From these results, it is evident that the stereochemistry of the C10 hydroxy group in substrates 27a,b plays an important role in the hemiacetal-opening/epimerization reaction. The difference in reactivity between 27a and 27b is not clear, but can be rationalized by the tentative mechanistic route depicted in Scheme 7. Thus, in the case of 27a, the formation of six-membered hemiacetal 27a-II would participate in the equilibrium event between 27a and 27a-I;^[16] this may bring about an equilibrium shift to 27a-I, facilitating the production of the desired 26a upon epimerization at C1 in 27a-I. On the other hand, in the case of 27b, the formation of the six-membered hemiacetal 27b-II would be precluded due to a severe steric interaction between the aromatic ring and the C4-OTBS group in the intermediary 27b-I; this phenomenon would prohibit the desired production of 26b.

Synthesis of intermediate 25: In the next stage, we pursued the synthesis of the key intermediate **25** as shown in Scheme 8. Thus, reduction of the formyl group in compound **26a** with LiAlH₄ followed by complete acetylation of the

Scheme 7. Tentative mechanistic route in the hemiacetal-opening/epimerization reaction of **27a.b**.

Scheme 8. Synthesis of intermediate **25**. a) LiAlH₄, THF, 0°C, 96%; b) Ac₂O, DMAP, pyridine, RT, 95%; c) Li, liq. NH₃, THF, -78°C, 98%; d) Dess–Martin periodinane, CH₂Cl₂, RT, 90%; e) Ph₃P+CH₃Br⁻, tBuOK, benzene, RT \rightarrow reflux, 95%.

three hydroxy groups in the resulting alcohol **30** furnished triacetate **31** in 91% overall yield in two steps. Subsequent reaction of **31** with a large excess of Li metal (100 equiv) in liquid NH₃/THF at -78°C resulted in the desired deoxygenated diol **32** in 98% yield. Simultaneous oxidation of the primary and secondary hydroxy groups in **32** with Dess–Martin periodinane provided the corresponding keto-aldehyde **33** in 90% yield, which was then subjected to twofold Wittig methylenation to give the requisite key intermediate **25** in 95% yield.

Completion of the total synthesis of (+)-ottelione A (1) and (-)-ottelione B (2): The final route that led to the completion of the total synthesis of the targeted ottelione A (1) and ottelione B (2) is summarized in Scheme 9. The key intermediate 25 was initially converted to the triene 37 in four steps in 57% overall yield by the same reaction sequence described for the synthesis of 3-epi-ottelione A (3) from diene 5 (cf. $5\rightarrow 19\rightarrow 20\rightarrow 21\rightarrow 22$, Scheme 4). Thus, deprotection of both the acetonide moiety and the MOM group in 25 by exposure to CF₃CO₂H/H₂O at 0°C followed by chemoselective acetylation of the phenolic hydroxy group in the resulting triol 34 afforded acetate 35 in 78% yield in two steps. Subsequent treatment of the diol 35 with thiophosgene in the presence of DMAP furnished the cyclic thiocarbonate 36 in 93% yield, which was then heated in triethyl phosphite, resulted in the desired diene 37 in 78% yield. Treatment of 37 with TBAF effected simultaneous deprotection of the TBS and acetyl groups to give the diol 38 in 83% yield. Chemoselective dichloroacetylation of the phenolic hydroxy group in 38 furnished the dichloroacetate 39 in 79% yield. It is noteworthy that selection of this dichloroacetyl group proved useful at a later deprotection stage (cf. $40 \rightarrow 1$) (vide infra). Finally, compound 39 was efficiently converted to ottelione A (1) in 86% overall yield via a two-step sequence of reactions involving Dess-Martin oxi-

TBSO OME

OMOM

TBSO OMOM

A)

$$R^{1}O$$
 $R^{1}O$
 $R^{$

Scheme 9. Synthesis of ottelione A (1) and ottelione B (2). a) CF₃CO₂H, THF/H₂O 10:1, 0°C, 86%; b) Ac₂O, 2_M NaOH, *i*PrOH, RT, 91%; c) CSCl₂, DMAP, CH₂Cl₂, RT, 93%; d) (EtO)₃P, reflux, 78%; e) TBAF, THF, RT, 83%; f) (CHCl₂CO)₂O, pyridine, CH₂Cl₂, RT, 79%; g) Dess–Martin periodinane, CH₂Cl₂, RT, 95%; h) 50% aqueous NaHCO₃/MeCN 1:1, RT, 90%; i) *t*BuOK, *t*BuOH, RT, 79% (1/2 23:77 by 500 MHz ¹H NMR); isolation of 2 by HPLC, 23%.

dation of the C4 hydroxy group and removal of the dichloroacetyl group in the resulting dienone 40 by brief exposure to 50% aqueous NaHCO3 in MeCN at ambient temperature. In the last deprotection step $(40\rightarrow 1)$, selection of the dichloroacetyl group proved important because this protecting group could be smoothly and cleanly removed under mild conditions without appreciable epimerization at the C3a position. In our preliminary experiment, we had prepared an O-acetyl variant of 40 (R=Ac); unfortunately, all attempts to remove the O-acetyl group from this compound under standard conditions (e.g., aq. NaOH, aq. KOH in MeOH or THF; K2CO3, NaOMe in MeOH) resulted in partial epimerization at the C3a position. We emphasize here that a only small amount of contamination of the C3a-epimerized product [ottelione B (2)] should be precluded for accurately measuring the optical rotation of ottelione A (1), because the absolute value of $[\alpha]_D$ for ottelione B (vide infra) is much greater than that for ottelione A.

The spectroscopic properties (IR, ¹H and ¹³C NMR, MS) of the synthetic sample **1** were identical to those of the natural product **1**. Comparison of the optical rotation [synthetic **1**, $[\alpha]_D^{25} = +17.3(c=0.55 \text{ in CHCl}_3)$; natural **1**, $[\alpha]_D^{25} = +14.0 (c=0.87 \text{ in CHCl}_3)^{[17,18]}$ established the absolute configuration of (+)-ottelione A (**1**) to be (1S,3S,3aR,7aS) as shown in Scheme 9.

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Conversion of ottelione A (1) to ottelione B (2) was successfully achieved via epimerization at the C3a position by exposure to tBuOK in tBuOH at room temperature for 2 h, which resulted in a 23:77 mixture of 1 and 2. Isolation of 2 from this mixture by column chromatography on silica gel was not effective; therefore, we performed the isolation using HPLC (DAICEL CHIRALPAK AD-H) to yield an entirely pure sample of 2, whose spectroscopic properties (IR, ¹H and ¹³C NMR, MS) were identical to those of natural 2. The optical rotation of a pure synthetic sample of 2 $\{[\alpha]_D^{25} = -333.0(c=0.18 \text{ in CHCl}_3)\}$ was essentially identical to that of natural 2 (contaminated with a small amount of 1, **2/1** 85:15) {[α]_D²⁵ = -276 (c = 0.20 in CHCl₃)},^[17,18] indicating that natural 2 possesses (1S,3S,3aS,7aS) absolute configuration as depicted in Scheme 9.

Preliminary biological evaluation of (+)-ottelione A (1), (+)-3-epi-ottelione A (3), and (+)-O-acetyl-3-epi-ottelione A (24)

Biological evaluation of unnatural 3-epi-ottelione A (3) and O-acetyl-3-epi-ottelione A (24) along with (-)-ottelione A (1) is of great interest from the viewpoint of structure-activity relationships. To this end, we used a panel of 39 human cancer cell lines (termed JFCR39) coupled to a drug activity database^[19] comparable to the panel developed by the National Cancer Institute. [20] The cell growth inhibition profiles against the JFCR39 (herein termed fingerprints) of more than 60 standard anticancer drugs were compared using COMPARE analysis, [19a,c] which showed that this is an information-rich approach for identifying the molecular target of a new compound, as described by Paull et al. [20a] This system can be used to predict the molecular target or the mode of action of test compounds by assessing the correlation coefficient between the fingerprints mediated by such test compounds and various reference compounds with known modes of action.[21]

The growth inhibitory activity of 1, 3, and 24 was evaluated by JFCR39 in the Japanese Foundation for Cancer Research.^[19] The number of cell lines and their origin (organ) are as follows: five breast, six central nervous system (brain), one melanoma, five ovary, two kidney, six stomach, and two prostate cancers. Dose-response curves were measured at five different concentrations (10^{-10} – 10^{-6} M, one log interval) for each compound, and the concentration causing 50% cell growth inhibition (GI₅₀) was compared with the control. The results are presented in Table 2 as log GI₅₀ values. It was evident that the test compounds exhibited extremely potent cytotoxic activity against almost all of the 39 cell lines [log $GI_{50} = -10.0 (1.0 \times 10^{-10} \text{ M}) - -6.0 (1.0 \times 10^{-10} \text{ M})$ 10⁻⁶ M)]. The order of the potency was estimated by MG-MID value (mean value of log GI₅₀ over all cell lines tested) to be 1(-8.44) > 3(-8.26) = 24(-8.22). The delta values (the difference in log GI₅₀ value of the most sensitive cell and MG-MID value) and the range values (the difference in log GI₅₀ values of the most sensitive cell and the least sensitive cell) were 1.52 and 4.00 for 1, 1.31 and 4.00 for 3, and

Table 2. Growth inhibition of compounds 1, 3, and 24 against a panel of 39 human cancer cell lines.

		$\log \operatorname{GI}_{50}^{[a]}[M]$			
Origin of cancer	Cell line	1	3	24	
breast	HBC-4	-8.46	-8.39	-8.32	
	BSY-1	$-10.0^{[c]}$	-9.51	-9.40	
	HBC-5	-9.74	-8.11	-6.51	
	MCF-7	-8.47	-8.16	-8.13	
	MDA-MB-	-9.13	-8.49	-8.47	
	231				
central nervous system	U-251	-8.40	-8.42	-8.35	
(brain)					
	SF-268	-8.42	-8.43	-8.60	
	SF-295	-8.34	-8.66	-8.40	
	SF-539	-9.48	-8.95	-9.46	
	SNB-75	-8.55	-7.47	-6.62	
	SNB-78	-9.36	-8.71	-8.62	
colon	HCC2998	8.66	-8.55	-8.66	
Colon	KM-12	-9.41	-9.30	-9.33	
	HT-29	-6.40	-6.22	-6.42	
	HCT-15	-8.46	-8.45	-8.61	
	HCT-116	-9.14	-8.30	-8.38	
lung	NCI-H23	-8.49	-8.34	-8.28	
8	NCI-H226	-8.84	-7.49	-6.77	
	NCI-H522	-9.67	$-9.56^{[c]}$	$-9.55^{[c]}$	
	NCI-H460	-8.48	-8.19	-8.13	
	A549	-7.55	-7.94	-8.23	
	DMS273	-9.56	-8.53	-8.65	
	DMS114	-9.40	-8.50	-8.65	
melanoma	LOX-IMVI	-9.03	-8.44	-8.75	
ovary	OVCAR-3	-9.80	-8.53	-8.53	
	OVCAR-4	-8.44	-7.09	-7.17	
	OVCAR-5	-8.09	-7.08	-7.25	
	OVCAR-8	-8.50	-8.31	-8.42	
	SK-OV-3	-8.45	-8.41	-8.45	
kidney	RXF-631 L	-7.02	$-6.00^{[b]}$	-6.71	
maney	ACHN	$-6.00^{[b]}$	-8.55	-8.35	
stomach	St-4	-6.29	$-6.00^{[b]}$	$-6.00^{[b]}$	
Stomath	MKN1	-9.12	-8.71	-8.91	
	MKN7	-8.58	8.90	-9.11	
	MKN28	-9.14	-9.27	-9.24	
	MKN45	-8.26	-7.12	-7.20	
	MKN74	-9.41	-9.21	-8.43	
prostate	DU-145	$-6.00^{[b]}$	-8.52	-8.50	
produce	PC-3	$-6.00^{[b]}$	-9.18	-9.15	
$MG-MID^{[d]}$	103	-8.44	-8.26	-8.22	
delta ^[e]		1.52	1.31	1.32	
range ^[f]		4.00	3.56	3.55	
range -		4.00	5.50	5.55	

[a] Log concentration that induces 50% inhibition of cell growth compared to control. [b] The least sensitive cell. [c] The most sensitive cell. [d] Mean value of log GI₅₀ over all cell lines tested. [e] The difference in log GI50 value of the most sensitive cell and MG-MID value. [f] The difference in log GI50 value of the most sensitive cell and the least sensitive

1.32 and 3.55 for **24** (effective value: delta ≥ 0.5 as well as range ≥ 1.0), respectively, indicating that all of these compounds showed pronounced selective cytotoxic activity.

Next, the pattern of differential cytotoxicity was analyzed using COMPARE analysis, [19a,c,21] which indicated that (+)ottelione A (1) was acting via a mechanism similar to vincristine (r=0.816) that is widely used in cancer chemotherapy as a prominent tubulin polymerization inhibitor. Interestingly, the modes of action for 3-epi analogues 3 and 24 did not correlate with that shown by any other anticancer drugs

developed to date (3/r=0.481, 24/r=0.422), suggesting that these compounds may represent new leads for anticancer agents with a novel action mechanism.

In order to confirm that compounds 1, 3, and 24 actually inhibit the polymerization of tubulin, we employed the two-step bioassay for tubulin inhibitors established by the Screening Committee of New Anticancer Agents in Japan.^[22] The results summarized in Table 3 clearly show

Table 3. Inhibitory activity of compounds 1, 3, and 24 against tubulin polymerization.

	1	3	24	Vincristine[b]
EC[a] [M]	$\geqslant 10^{-10}$	$> 10^{-10}$	$\geq 10^{-9}$	$\geq 10^{-9}$

[a] Effective concentration that induces inhibition of tubulin polymerization. [b] Positive control as a representative tubulin polymerization inhibitor.

that all of these compounds exhibit potent inhibitory activity against tubulin polymerization. Compound ${\bf 1}$ was found to exhibit ≈ 10 times more potent activity compared to vincristine, a reference compound. Furthermore, the potency of 3-epi analogues ${\bf 3}$ and ${\bf 24}$ was similar or superior to that of vincristine. Considering the results obtained by both the COMPARE analysis and tubulin inhibitory assay, 3-epi analogues ${\bf 3}$ and ${\bf 24}$ could be promising candidates or potential lead compounds for the development of novel anticancer agents targeting tubulin, albeit the binding sites are unknown at present.

Conclusion

We initially achieved the total synthesis of (+)-3-epi-ottelione A (3), the earlier proposed stereostructure of (+)-ottelione A (1) starting from the known, readily available tricyclic compound 10. Subsequently, by applying this approach, we accomplished the total synthesis of (+)-ottelione A (1) and (-)-ottelione B (2); this synthesis resulted in the establishment of their absolute configurations. Importantly, the routes explored have potential for preparing various types of ottelione analogues in enantiomerically pure forms due to their generality and flexibility. Preliminary biological evaluation of the synthetic (+)-ottelione A (1), (+)-3-epi-ottelione A (3), and (+)-O-acetyl-3-epi-ottelione A (24) revealed that all of the test compounds exhibited significant tumor growth inhibitory activity as well as extremely potent tubulin inhibitory potency. It was evident that unnatural 3epi analogues 3 and 24 displayed a unique tumor growth inhibitory profile. On the basis of the present study, further investigations concerning structure-activity relationships and in vivo antitumor activity are currently under way and will be reported in due course.

Experimental Section

General techniques: All reactions involving air- and moisture-sensitive reagents were carried out using oven dried glassware and standard syringe-septum cap techniques. Routine monitorings of reaction were carried out using glass-supported Merck silica gel 60 F₂₅₄ TLC plates. Flash column chromatography was performed on Kanto Chemical Silica Gel 60N (spherical, neutral 40–50 μm) with the solvents indicated.

All solvents and reagents were used as supplied with following exceptions. THF was freshly distilled from Na metal/benzophenone under argon. Toluene was distilled from Na metal under argon. CH₂Cl₂, benzene, and pyridine were distilled from CaH₂ under argon.

Measurements of optical rotations were performed with a JASCO P-1020 automatic digital polarimeter. Melting points were taken on a Yanaco MP-3 micro melting point apparatus and are uncorrected. $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectra were measured with a Bruker DRX-500 (500 MHz) spectrometer. Chemical shifts were expressed in ppm using Me₄Si (δ =0) as an internal standard. The following abbreviations are used: singlet (s), doublet (d), triplet (t), multiplet (m), and broad (br). Copies of $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectra for all new compounds are shown in the Supporting Information. Infrared (IR) spectral measurements were carried out with a JASCO FT/IR-5300 spectrometer. Low-resolution mass (MS) spectra was measured on a Shimadzu GCMS-QP2010. High-resolution mass (HRMS) spectra was measured on a JEOL MStation JMS-700 mass spectrometer. Elemental analyses were performed with a Perkin Elmer 2400II apparatus.

$(1R,\!4S,\!4aR,\!5R,\!6R,\!7R,\!8S,\!8aS) - 5 - tert - \textbf{Butyldimethylsiloxy-}6,\!7 - O - \textbf{isopropylidenedioxy-}1,\!4,\!4a,\!5,\!6,\!7,\!8,\!8a - \textbf{octahydro-} endo-\\ \textbf{methanonaphthalen-}8 - \textbf{ol}$

(13): NaBH₄ (153 mg, 4.0 mmol) was added to a stirred solution of (1R,4S,4aS,5R,6S,7S,8aS)-5-tert-butyldimethylsiloxy-6,7-O-isopropylidene $dioxy-1,4,4a,5,6,7,8,8a-octahydro-\textit{endo}-methanonaphthalen-8-one \quad \textbf{(10)}^{[12]}$ (737 mg, 2.0 mmol) in THF/H₂O 10:1 (22 mL) at 0 °C. After 30 min, the reaction was quenched with saturated aqueous NH₄Cl (5 mL) at 0 °C. The reaction mixture was extracted with Et₂O $(2 \times 100 \text{ mL})$. The organic layer was washed successively with saturated aqueous NaHCO3 (2× 50 mL) and brine (2×50 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 10:1→4:1) to give 13 (644 mg, 87%) as a white solid. Recrystallization from hexane/Et₂O afforded colorless needles. M.p. 147.0–147.7 °C; $[\alpha]_D^{20} = -25.1$ (c = 0.98 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.10$ (s, 6H), 0.90 (s, 9H), 1.23–1.34 (m, 4H), 1.38-1.49 (m, 4H), 2.58-2.79 (m, 2H), 2.80-3.05 (m, 2H), 3.80-4.30 (m, 4H), 6.03 (s, 1H), 6.09 ppm (s, 1H); 13 C NMR (125 MHz, CDCl₃): $\delta =$ -4.95, -4.71, 18.0, 23.8, 25.8, 25.9, 26.6, 44.9, 51.7, 78.1, 107.7, 134.5 ppm; IR (KBr): $\tilde{v} = 3426$, 2934, 2858, 2363, 1630, 1381, 1251, 1207, 1167, 1111, 1064, 898, 862, 837, 777 cm⁻¹; MS (CI): m/z: 367 $[M+H]^+$; elemental analysis calcd (%) for C20H34O4Si: C 65.53, H 9.35; found: C 65.57, H

(2S,2aS,4R,4aS,5R,6R,7R,7aS,7bR)-5-tert-Butyl dimethyl siloxy-2-hydroxy-6,7-(O-isopropylidenedioxy) decahydroindeno [7,1-bc] furan-4-carbal decahydroindeno [7

hyde (9): OsO₄ in tBuOH (0.04 m solution, 1.35 mL, 55 mmol) and NaIO₄ (467 mg, 2.2 mmol) were added successively to a stirred solution of 13 (200 mg, 0.55 mmol) in tBuOH/THF/H₂O 8:6:3 (10 mL) at 0 °C, and the mixture was allowed to warm up to room temperature. After 12 h, the reaction was quenched with 20% aqueous NaS2O3 (10 mL), and the resulting mixture was extracted with Et₂O (2×50 mL). The combined extracts were washed with saturated aqueous NaHCO3 (2×25 mL) and brine (2× 25 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 5:1) to give 9 (163 mg, 75%) as a white solid. Recrystallization from hexane/Et₂O afforded colorless needles. M.p. 120.7-121.5 °C; $[\alpha]_D^{20} = +19.0 \ (c = 0.99 \ \text{in CHCl}_3)$; ¹H NMR (500 MHz, CDCl₃): $\delta = 0.06$ (s, 3H), 0.12 (s, 3H), 0.88 (s, 9H), 1.34 (s, 3H), 1.44 (s, 3H), 2.07-2.15 (m, 1H), 2.18-2.25 (m, 1H), 2.26 (d, 1H, J=2.3 Hz), 2.74-2.91(m, 3H), 2.91-2.97 (m, 1H), 4.17 (dd, 1H, J=6.7, 4.5 Hz), 4.23 (s, 1H),4.46 (dd, 1H, J=12.2, 6.7 Hz), 4.47 (dd, 1H, J=12.2, 6.7 Hz), 5.27 (d, 1H, J=2.3 Hz), 9.93 ppm (s, 1H); 13 C NMR (125 MHz, CDCl₃): $\delta=$ -4.43, -4.37, 18.2, 24.3, 25.9, 27.0, 30.2, 40.6, 41.9, 52.0, 55.8, 69.4, 73.6, 75.7, 76.0, 103.1, 107.7, 202.9 ppm; IR (KBr): $\bar{v}=3433$, 2955, 2928, 2854, 1714, 1469, 1371, 1246, 1221, 1151, 1103, 1057, 1012, 956, 898, 860, 831, 775, 511 cm⁻¹; HRMS (EI): m/z: calcd for $C_{20}H_{34}O_6Si$: 398.2125, found 398.2124 $[M]^+$.

(2S,2aS,4S,4aS,5R,6R,7R,7aS,7bR)-5-tert-Butyldimethylsiloxy-2-hydroxy-6,7-(O-isopropylidenedioxy)decahydroindeno[7,1-bc]furan-4-carbalde-

hyde (8): DBU (0.50 mL, 3.3 mmol) was added to a stirred solution of 9 (657 mg, 1.6 mmol) in dry THF (20 mL) at room temperature under argon. After 2 h, the mixture was diluted with Et₂O (100 mL) and the organic layer was washed successively with 3% aqueous HCl (2×50 mL), saturated aqueous NaHCO₃ (2×50 mL), and brine (50 mL), then dried over Na2SO4. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 2:1) to give 8 (657 mg, quant.) as a white solid. Recrystallization from hexane/ CH₂Cl₂ afforded colorless needles. M.p. 108.7–110.0 °C; $[\alpha]_D^{20} = -25.0$ (c =0.93 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.06$ (s, 3H), 0.11 (s, 3H), 0.87 (s, 9H), 1.36 (s, 3H), 1.48 (s, 3H), 1.72–1.79 (m, 1H), 1.95–2.05 (m, 1H), 2.31 (d, 1H, J=2.2 Hz), 2.46 (ddd, 1H, J=12.5, 7.8, 4.6 Hz), 2.59-2.67 (m, 1H), 2.83-2.88 (m, 1H), 3.00-3.02 (m, 1H), 3.89 (dd, 1H, J=7.8, 4.6 Hz), 4.13 (dd, 1H, J=7.8, 6.2 Hz), 4.49 (d, 1H, J=6.2 Hz), 4.56 (d, 1H, J=7.3 Hz), 5.22 (d, 1H, J=2.2 Hz), 9.50 ppm (d, 1H, J=3.2 Hz) 3.7 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = -4.51$, -4.31, 18.1, 25.5, 25.8, 28.1, 31.4, 42.7, 49.5, 50.3, 51.7, 71.9, 76.4, 76.6, 76.8, 103.3, 107.5, 202.0 ppm; IR (KBr): \tilde{v} =3391, 2937, 2891, 1707, 1471, 1386, 1246, 1219, 1089, 1014, 904, 829, 779, 667 cm⁻¹; MS (CI): m/z: 399 [M+H]⁺; elemental analysis cald (%) for C₂₀H₃₄O₆Si: C 60.27, H 8.64; found: C 60.25, H

(2S,2aS,4S,4aR,5R,6R,7R,7aS,7bR)-5-tert-Butyldimethylsiloxy-4-[1-hydroxy[4-methoxy-3-(methoxymethoxy)phenyl]methyl]-6,7-(O-isopropylidenedioxy)decahydroindeno[7,1-bc]furan-2-ol (15): nBuLi in nhexane (1.55 M solution, 3.10 mL, 4.8 mmol) was added dropwise to a stirred solution of 4-bromo-1-methoxy-2-(methoxymethoxy)benzene (7) (1.50 g, 5.2 mmol) in dry THF (50 mL) at -78 °C under argon. After 30 min, a solution of 8 (510 mg, 1.3 mmol) in dry THF (25 mL) was added dropwise to the above mixture at -78°C. After 1 h, the reaction was quenched with saturated aqueous NH₄Cl (10 mL), and the reaction mixture was extracted with Et₂O (2×100 mL). The organic layer was washed with saturated aqueous NaHCO3 (2×30 mL) and brine (2×30 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 2:1→1:1) to give 15 (724 mg, quant.) as an inseparable mixture of two diastereomers (9:1) as a colorless viscous liquid. ¹H NMR (500 MHz, CDCl₃): δ = 0.16 (s, 3H), 0.18 (s, 3H), 0.94 (s, 9H), 1.34 (s, 3H), 1.46 (s, 3H), 1.31-1.39 (m, 1H), 1.90-1.94 (m, 1H), 2.12-2.20 (m, 1H), 2.17 (s, 1H), 2.31-2.36 (m, 1H), 2.47 (d, 1H, J=4.4 Hz), 2.54–2.57 (m, 1H), 2.93–2.99 (m, 1H), 3.51 (s, 3H), 3.86 (s, 3H), 3.94 (dd, 1H, J=7.4, 4.1 Hz), 4.25–4.30 (m, 1H), 4.40 (dd, 1H, J=8.4, 2.3 Hz), 4.42 (dd, 1H, J=7.4, 2.3 Hz), 5.01-5.05 (m, 1H), 5.09 (s, 1H), 5.19-5.24 (m, 2H), 6.85 (d, 1H, J=8.4 Hz), 6.92 (dd, 1H, J=8.4, 1.9 Hz), 7.08 ppm (d, 1H, J=1.9 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = -4.07$, -3.82, 18.5, 23.7, 26.1, 26.4, 32.7, $36.6,\ 42.1,\ 52.1,\ 52.4,\ 53.4,\ 55.9,\ 56.1,\ 68.6,\ 74.9,\ 76.0,\ 95.6,\ 103.6,\ 107.4,$ 111.7, 114.9, 120.5, 136.6, 146.7, 149.4 ppm; IR (neat): $\tilde{v} = 3420$, 2934, 2858, 1720, 1608, 1512, 1464, 1381, 1257, 1155, 1078, 1005, 912, 837, 779 cm $^{-1}$; HRMS (FAB): m/z: calcd for $C_{29}H_{45}O_{9}Si$: 565.2833, found 565.2831 [M-H]+.

(2R,2aS,4S,4aR,5R,6R,7R,7aS,7bR)-2-Acetoxy-5-tert-butyldimethylsiloxy-4-[1-acetoxy-[4-methoxy-3-(methoxymethoxy)phenyl]methyl]-6,7-(O-isopropylidenedioxy)decahydroindeno[7,1-bc]furan (16): (CH₃CO)₂O (1.17 mL, 12 mmol) and 4-dimethylaminopyridine (DMAP) (15.0 mg, 0.12 mmol) were added to a stirred solution of 15 (700 mg, 1.2 mmol) in pyridine (12 mL) at room temperature. After 3 h, the mixture was diluted with Et₂O (120 mL). The organic layer was washed with 3% aqueous HCl (4×30 mL), saturated aqueous NaHCO₃ (2×30 mL), and brine (30 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo gave a residue, which was purified by column chromatography (hexane/EtOAc 2:1) to give 16 (804 mg, quant.) as a colorless viscous liquid. 1 H NMR (500 MHz, CDCl₃): δ =0.10 (s, 6H), 0.91 (s, 9 H), 1.29 (s, 3 H), 1.34 (s, 3H), 1.78–1.85 (m, 1 H), 2.02 (s, 3 H), 2.08 (s, 3 H), 2.03–2.07 (m,

1H), 2.10–2.16 (m, 1H), 2.43–2.50 (m, 1H), 2.83–2.92 (m, 2H), 3.50 (s, 3H), 3.83 (t, 1H, J=4.4 Hz), 3.86 (s, 3H), 4.17 (dd, 1H, J=6.9, 4.4 Hz), 4.36 (dd, 1H, J=6.9, 1.9 Hz), 4.44 (dd, 1H, J=6.9, 2.0 Hz), 5.20 (s, 2H), 5.67 (d, 1H, J=6.9 Hz), 5.92 (s, 1H), 6.83 (d, 1H, J=8.3 Hz), 6.88 (dd, 1H, J=8.3, 1.9 Hz), 7.06 ppm (d, 1H, J=1.9 Hz); 13 C NMR (125 MHz, CDCl₃): δ = –4.82, –4.45, 18.3, 21.3, 21.3, 24.4, 25.9, 26.9, 31.0, 41.7, 49.0, 51.1, 55.9, 56.1. 71.0, 73.9, 75.7, 76.8, 77.2, 79.0, 95.6, 103.7, 107.7, 111.5, 114.5, 120.6, 132.8, 140.6, 149.3, 170.2, 170.3 ppm; IR (neat): \bar{v} =2934, 1741, 1514, 1464, 1373, 1234, 1155, 1134, 1080, 1005, 837, 777, 733 cm⁻¹; HRMS (FAB): m/z: calcd for $C_{33}H_{50}NaO_{11}Si$: 673.3020, found 673.3022 $[M+Na]^+$.

(2S,2aS,4R,4aR,5R,6R,7R,7aS,7bR)-5-tert-Butyldimethylsiloxy-4-[4-methoxy-3-(methyoxymethoxy)benzyl]-6,7-(O-isopropylidenedioxy)decahydroindeno[7,1-bc]furan-2-ol (6): A solution of 16 (420 mg, 0.62 mmol) in dry THF (20 mL) was added dropwise to a stirred solution of Li metal (215 mg, 31 mmol) in liquid NH₃ (40 mL) at -78 °C under argon. After 5 min, the reaction was quenched with saturated aqueous NH₄Cl (10 mL) at the same temperature. The mixture was then allowed to stand at room temperature for 4 h in order to evaporate off excess NH₃. The mixture was extracted with EtOAc (3×100 mL) and the extracts were washed with saturated aqueous NaHCO3 (2×60 mL) and brine (60 mL), then dried over Na2SO4. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 2:1) to give 6 (299 mg, 87%) as a white solid. Recrystallization from hexane/ Et₂O afforded colorless needles. M.p. 112.8–113.5 °C; $[a]_D^{20} = -36.6$ (c= 0.77 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.10$ (s, 3 H), 0.13 (s, 3H), 0.91 (s, 9H), 1.36 (s, 3H), 1.47 (s, 3H), 1.43-1.49 (m, 1H), 1.62 (ddd, 1H, J=13.5, 6.3, 3.0 Hz), 1.90–1.95 (m, 1H), 1.96–2.03 (m, 1H), 2.07 (dd, 1H, J=13.1, 10.4 Hz), 2.15 (d, 1H, J=2.1 Hz), 2.65-2.70 (m, 1 H), 2.93–2.99 (m, 1 H), 3.34 (dd, 1 H, J=13.1, 3.7 Hz), 3.52 (s, 3 H), 3.85 (s, 3H), 3.93 (dd, 1H, J=7.7, 4.23–4.27 Hz), 4.25 (m, 1H), 4.46–4.49 (m, 2H), 5.07 (d, 1H, J=2.1 Hz), 5.20 (s, 2H), 6.79 (dd, 1H, J=8.2, 1.9 Hz), 6.80 (d, 1H, J=8.2 Hz), 6.91 ppm (d, 1H, J=1.9 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = -4.50$, -4.42, 18.2, 25.2, 25.9, 27.8, 36.8, 41.1, 42.4, 43.8, 49.0, 49.5, 55.9, 56.2, 72.8, 76.3, 76.9, 77.8, 95.7, 103.4, 107.3, 111.7, 117.4, 122.5, 134.6, 146.1, 148.0 ppm; IR (KBr): $\tilde{v} = 3449$, 2955, 1516, 1257, 1132, 1082, 1006, 837, 779 cm⁻¹; MS (EI): m/z: 550 [M]+; elemental analysis calcd (%) for C₂₉H₄₆O₈Si: C 63.24, H 8.42; found: C 63.39, H 8.55.

(1R,3R,3aR,4R,5R,6R,7S,7aR)-4-tert-Butyldimethylsiloxy-7-hydroxy-5,6-O-isopropylidenedioxy-3-[4-methoxy-3-(methoxymethoxy)benzyl]octahydroinden-1-carbaldehyde (17): DBU (2.6 mL, 17 mmol) was added to a stirred solution of 6 (480 mg, 0.87 mmol) in dry toluene (10 mL) at room temperature under argon. The mixture was heated at reflux for 1.5 h. After cooling, the mixture was directly subjected to column chromatography on silica gel eluting with EtOAc in order to remove excess DBU. The fractions containing 17 and starting material 6 were collected and then concentrated in vacuo. The residue was purified by column chromatography (hexane/EtOAc 4:1) to give less polar 17 (250 mg, 52 %, colorless viscous liquid) and more polar starting material 6 (192 mg, 40 %). 17: $[\alpha]_{D}^{20} = +7.09 \ (c=0.92 \text{ in CHCl}_3); {}^{1}\text{H NMR } (500 \text{ MHz, CDCl}_3): \delta=0.18$ (s, 6H), 0.93 (s, 9H), 1.35 (s, 3H), 1.46 (s, 3H), 1.32-1.40 (m, 1H), 1.96-2.03 (m, 1H), 2.13-2.19 (m, 1H), 2.26-2.35 (m, 1H), 2.40 (dd, 1H, J=13.4, 9.5 Hz), 2.66–2.73 (m, 1 H), 2.85 (dd, 1 H, J = 13.4, 9.5 Hz), 3.04–3.12 (m, 1H), 3.50 (s, 3H), 3.86 (s, 3H), 3.89 (ddd, 1H, <math>J = 12.5, 5.0, 2.9 Hz),4.01-4.04 (m, 1H), 4.20 (d, 1H, J=12.5 Hz), 4.33 (dd, 1H, J=7.1, 2.8 Hz), 4.53 (dd, 1 H, J = 7.1, 2.9 Hz), 5.20 (s, 2 H), 6.76 (dd, 1 H, J = 8.2,1.9 Hz), 6.81 (d, 1H, J=8.2 Hz), 6.97 (d, 1H, J=1.9 Hz), 9.73 ppm (d, 1 H, J = 1.9 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = -4.33$, -3.85, 18.0, 23.6, 25.7, 26.2, 32.9, 38.5, 40.0, 41.4, 41.7, 52.3, 55.9, 56.1, 67.0, 69.8, 75.5, 76.3, 95.6, 108.3, 111.7, 117.1, 122.5, 132.9, 146.5, 148.2, 203.1 ppm; IR (neat): $\tilde{v} = 3418$, 2932, 1722, 1514, 1263, 1157, 1026, 839 cm⁻¹; HRMS (EI): m/z: calcd for $C_{29}H_{46}O_8Si$: 550.2962, found 550.2939 [M]⁺.

(1R,3R,3aR,4R,5R,6S,7aR)-4-tert-Butyldimethylsiloxy-5,6-O-isopropylidenedioxy-3-[4-methoxy-3-(methoxymethoxy)benzyl]-7-oxooctahydroinden-1-carbaldehyde (18): Dess-Martin periodinane (865 mg, 2.0 mmol) was added to a stirred solution of 17 (375 mg, 0.68 mmol) in dry CH₂Cl₂ (7 mL) at room temperature. After 1 h, the reaction was quenched with

10% aqueous Na₂S₂O₃ (3 mL), and the resulting mixture was extracted with EtO₂ (2×50 mL). The organic layer was washed with saturated aqueous NaHCO3 (2×20 mL) and brine (20 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 3:1) to give 18 (336 mg, 90%) as a colorless viscous liquid. $[a]_D^{20} = +17.8$ (c=1.19 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.03$ (s, 3 H), 0.04 (s, 3 H), 0.82 (s, 9H), 1.34 (s, 3H), 1.30–1.39 (m, 1H), 1.57 (s, 3H), 1.95–2.02 (m, 1H), 2.24-2.33 (m, 1 H), 2.58 (dd, 1 H, J=13.5, 8.2 Hz), 2.70 (dd, 1 H, J=13.5, 6.7 Hz), 2.85 (ddd, 1H, J = 11.8, 7.9, 1.7 Hz), 2.99 - 3.07 (m, 1H), 3.42 (dd, 1H)1 H, J = 11.8, 10.4 Hz), 3.51 (s, 3 H), 3.65 (dd, 1 H, J = 3.5, 1.7 Hz), 3.86 (s, 3H), 4.26 (d, 1H, J=7.4 Hz), 4.41 (dd, 1H, J=7.4, 3.5 Hz), 5.21 (s, 2H), 6.75 (dd, 1H, J=8.2, 1.9 Hz), 6.82 (d, 1H, J=8.2 Hz), 6.96 (d, 1H, J= 1.9 Hz), 9.82 ppm (d, 1 H, J=1.2 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta=$ -4.67, -4.48, 17.8, 23.9, 25.5, 26.1, 33.0, 40.5, 42.7, 46.4, 46.5, 53.2, 56.0, 56.1, 69.3, 76.77, 77.3, 80.5, 95.6, 111.8, 112.0, 117.0, 122.5, 132.9, 146.6, 148.2, 202.5, 206.0 ppm; IR (neat): $\tilde{v} = 2932$, 1728, 1512, 1466, 1383, 1261, 1209, 1157, 1076, 1006, 839, 777 cm⁻¹; HRMS (EI): m/z: calcd for $C_{29}H_{44}O_8Si: 548.2805$, found 548.2806 [M]⁺.

(1S,3R,3aR,4R,5R,6R,7aS)-4-tert-Butyldimethylsiloxy-5,6-O-isopropylidenedioxy-3-[4-methoxy-3-(methoxymethoxy)benzyl]-7-mehtylene-1-(vinyl)octahydroindene (5): Wittig reagent $(Ph_3P=CH_2)$ in benzene solution was first prepared as follows: a suspension of Ph₃P+CH₃Br- (1.0 g, 2.8 mmol) and tBuOK (314 mg, 2.8 mmol) in dry benzene (6 mL) were heated at reflux for 4 h under argon, and the solution was cooled to room temperature. A solution of the Wittig reagent in benzene (1.0 mL, 0.47 mmol) was added very slowly to a stirred solution of 18 (257 mg, 0.47 mmol) in dry benzene (20 mL) at room temperature under argon. After the first methylenation of the C1-formyl group was completed (monitored by TLC), a solution of the Wittig reagent in benzene (4 mL, 1.9 mmol) was added once again and the resulting mixture was heated under reflux for 30 min to pursue the second methylenation of the C7carbonyl group. After cooling, the reaction was quenched with saturated aqueous NH₄Cl (5 mL), and the mixture was extracted with Et₂O (2× 50 mL). The organic layer was washed with brine (2×20 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 20:1) to give 5 (224 mg, 88 %) as a colorless viscous liquid. $[a]_D^{20} = +9.75$ (c=0.77 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.03$ (s, 3H), 0.04 (s, 3H), 0.83 (s, 9H), 1.08–1.19 (m, 1H), 1.35 (s, 3H), 1.49 (s, 3H), 1.78–1.84 (m, 1H), 2.18-2.26 (m, 1H), 2.26-2.32 (m, 1H), 2.45 (dd, 1H, J=13.4, 8.9 Hz), 2.46–2.54 (m, 1H), 2.61–2.68 (m, 1H), 2.78 (dd, 1H, J=13.4, 5.5 Hz), 3.50 (s, 3H), 3.64–3.67 (m, 1H), 3.85 (s, 3H), 4.18 (dd, 1H, J =7.4, 3.0 Hz), 4.61 (d, 1H, J=7.4 Hz), 4.94 (dd, 1H, J=10.3, 1.0 Hz), 5.00– 5.08 (m, 3H), 5.18–5.23 (m, 2H), 5.81 (ddd, 1H, J=17.4, 10.3, 7.3 Hz), 6.77 (dd, 1H, J=8.2, 1.9 Hz), 6.80 (d, 1H, J=8.2 Hz), 6.97 ppm (d, 1H, $J=1.9~{\rm Hz}$); $^{13}{\rm C~NMR}$ (125 MHz, CDCl₃): $\delta=-4.73, -4.37, 17.9, 24.0,$ 25.6, 26.3, 40.5, 41.0, 41.1, 42.8, 45.0, 48.3, 55.9, 56.1, 70.2, 78.3, 79.1, 95.6, 108.7, 111.6, 112.2, 113.6, 117.2, 122.5, 134.1, 143.0, 145.6, 146.3, 147.8 ppm; IR (neat): $\tilde{v} = 2928$, 2856, 1512, 1462, 1379, 1261, 1209, 1155, 1080, 1030, 902, 837, 810, 775 cm⁻¹; HRMS (EI): m/z: calcd for $C_{31}H_{48}O_6Si: 544.3220$, found 544.3224 [M]+.

 $(1S,\!3R,\!3aR,\!4R,\!5R,\!6R,\!7aS) - 4 \textit{-}tert - \text{Butyldimethylsiloxy-3-} (3 - \text{hydroxy-4-me-}) - 4 - \text{hydroxy-4-me-} (3 - \text{hydroxy-4-me-}) - 4 - \text{hydroxy-4$ thoxybenzyl)-7-methylene-1-(vinyl)octahydroinden-5,6-diol (19): A solution of CF₃CO₂H/H₂O 10:1 (11 mL) was added to a stirred solution of 5 (230 mg, 0.42 mmol) in THF (1 mL) at 0 °C. After 10 min, the mixture was neutralized with $6\,\text{M}$ NaOH and then extracted with EtOAc ($3\times$ 40 mL). The organic layer was washed with saturated aqueous NaHCO₃ (2×20 mL) and brine (2×20 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 3:1) to give 19 (148 mg, 76%) as a colorless viscous liquid. $[\alpha]_D^{20} = +15.1$ (c = 0.93 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.08$ (s, 3H), 0.11 (s, 3H), 0.91 (s, 9H), 1.14 (ddd, 1 H, J = 13.1, 10.1, 6.9 Hz), 1.70 (d, 1 H, J = 5.9 Hz), 1.91 (d, 1 H, J = 7.4 Hz), 2.06-2.15 (m, 2H), 2.35-2.43 (m, 1H), 2.47-2.60 (m, 3H), 2.73-2.82 (m, 1H), 3.72-3.75 (m, 1H), 3.81-3.85 (m, 1H), 3.86 (s, 3H), 4.61-4.65 (m, 1 H), 4.87 (d, 1 H, J = 17.0 Hz), 4.93 (d, 1 H, J = 10.2 Hz), 4.99 (s,1H), 5.18-5.20 (m, 1H), 5.52 (s, 1H), 5.57 (ddd, 1H, J=17.0, 10.2, 7.9 Hz), 6.62 (dd, 1 H, J = 8.0, 1.8 Hz), 6.73 (d, 1 H, J = 1.8 Hz), 6.74 ppm

(d, 1 H, J = 8.0 Hz); 13 C NMR (125 MHz, CDCl₃): δ = -4.78, -4.43, 17.8, 25.8, 39.5, 39.8, 43.1, 45.7, 48.9, 52.6, 55.9, 68.1, 74.1, 75.7, 110.5, 111.4, 114.2, 114.9, 120.1, 134.6, 141.7, 144.7, 145.3, 145.4 ppm; IR (neat): \tilde{v} = 3429, 2930, 2856, 1591, 1512, 1442, 1273, 1076, 875, 835, 775, 505, 430, 407 cm⁻¹; HRMS (FAB): m/z: calcd for $C_{26}H_{40}NaO_5Si$: 483.2543, found, 483.2546 [M+Na]⁺.

(1S,3R,3aR,4R,5R,6R,7aS)-3-(3-Acetoxy-4-methoxybenzyl)-4-tert-butyldimethylsiloxy-7-methylene-1-(vinyl)octahydroinden-5.6-diol (20): 2 M NaOH (0.42 mL, 0.84 mmol) and (CH₃CO)₂O (79 μL, 0.84 mmol) were added dropwise to a stirred solution of 19 (140 mg, 0.30 mmol) in 2-propanol (3.5 mL) at room temperature. After 30 min, the mixture was diluted with EtOAc (50 mL). The organic layer was washed with H₂O (2× 15 mL) and brine (15 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo gave a residue, which was purified by column chromatography (hexane/EtOAc 2:1) to give 20 (138 mg, 90 %) as a colorless viscous liquid. $[\alpha]_D^{20} = +6.16$ (c=0.97 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.04$ (s, 3H), 0.05 (s, 3H), 0.89 (s, 9H), 1.14 (ddd, 1H, J =12.9, 10.3, 7.3 Hz), 1.94-1.99 (m, 2H), 2.08-2.14 (m, 1H), 2.14-2.18 (m, 1H), 2.29 (s, 3H), 2.31-2.39 (m, 1H), 2.46-2.54 (m, 2H), 2.67 (dd, 1H, J = 13.5, 6.7 Hz), 2.72–2.81 (m, 1H), 3.58–3.62 (m, 1H), 3.79–3.84 (s, 3H), 3.82 (m, 1H), 4.56-4.60 (m, 1H), 4.87 (d, 1H, J=17.1 Hz), 4.93 (d, 1H, J=17.1 Hz)J=10.2 Hz), 4.96 (s, 1 H), 5.18–5.22 (m, 1 H), 5.56 (ddd, 1 H, J=17.1, 10.2, 7.9 Hz), 6.84 (d, 1H, J=2.1 Hz), 6.87 (d, 1H, J=8.3 Hz), 6.98 ppm (dd, 1 H, J = 2.1, 8.3 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = -4.90$, -4.40, 17.8, 20.7, 25.7, 39.7, 39.9, 42.5, 45.2, 48.9, 52.5, 55.9, 68.9, 74.1, 75.5, 111.3, 112.3, 114.3, 123.3, 126.9, 134.1, 139.3, 141.5, 145.5, 149.1, 169.3 ppm; IR (neat): $\tilde{v} = 3452$, 2930, 2856, 1768, 1639, 1512, 1460, 1369, 1263, 1205, 1122, 1066, 1018, 902, 835, 775, 408 cm⁻¹; HRMS (FAB): *m/z*: calcd for C₂₈H₄₂NaO₆Si: 525.2648, found, 525.2634 [M+Na]⁺.

(3aR,4R,4aR,5R,7S,7aS)-5-(3-Acetoxy-4-methoxybenzyl)-4-tert-butyldimethylsiloxy-8-methylene-7-(vinyl)octahydroindeno[5,6-d,1,3]dioxol-2-

thione (21): A solution of CSCl₂ (50 µL, 0.66 mmol) in dry CH₂Cl₂ (1.5 mL) was added dropwise to a stirred solution of 20 (165 mg, 0.33 mmol) in dry CH₂Cl₂ (6 mL) containing DMAP (200 mg, 1.7 mmol) at room temperature. After 1 h, silica gel (2.0 g) was added to the reaction mixture, and the solvent was carefully evaporated off in vacuo. The resulting solid was charged on the top of a silica gel column chromatography, and elution with hexane/EtOAc 10:1 gave 21 (172 mg 96%) as a colorless viscous liquid. $[a]_D^{20} = +5.04$ (c = 0.91 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.01$ (s, 3H), 0.02 (s, 3H), 0.81 (s, 9H), 1.15–2.24 (m, 1H), 1.87-1.94 (m, 1H), 2.11-2.16 (m, 1H), 2.16-2.24 (m, 1H), 2.31 (s, 3H), 2.48-2.55 (m, 1H), 2.57 (d, 1H, J=18.3, 8.0 Hz), 2.61-2.72 (m, 2H), 3.66-3.70 (m, 1H), 3.81 (s, 3H), 4.81 (dd, 1H, J=8.6, 3.2 Hz), 5.00(d, 1H, J=10.2 Hz), 5.07 (d, 1H, J=17.2 Hz), 5.19 (d, 1H, J=2.5 Hz), 5.28 (d, 1H, J=8.6 Hz), 5.32 (d, 1H, J=2.5 Hz), 5.78 (ddd, 1H, J=17.2, 7.6, 10.2 Hz), 6.82 (d, 1 H, J = 2.1 Hz), 6.90 (d, 1 H, J = 8.3 Hz), 6.97 ppm (dd, 1 H, J = 8.3, 2.1 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = -4.78$, -4.69, 17.8, 20.7, 25.5, 40.4, 40.5, 41.1, 44.4, 48.8, 56.0, 68.0, 77.2, 81.9, 85.5, 112.5, 114.6, 116.7, 123.2, 126.9, 133.2, 139.6, 140.6, 141.7, 149.5, 168.9, 191.0 ppm; IR (neat): $\tilde{v} = 2930$, 2856, 1768, 1512, 1460, 1367, 1329, 1269, 1203, 1093, 995, 916, 833, 777 cm⁻¹; HRMS (EI): m/z: calcd for $C_{29}H_{40}O_6SSi: 544.2315$, found 544.2303 [M]⁺.

siloxy-7-methylene-1-vinyl-1,2,3,3a,7,7a-hexahydroindene (22): A solution of 21 (153 mg, 0.28 mmol) in (EtO)₃P (20 mL) was heated at reflux for 2 h under argon. After cooling, excess (EtO)₃P was removed through short column chromatography eluting with hexane. The combined fractions were concentrated in vacuo to afford a residue, which was purified by column chromatography (hexane/EtOAc 100:1) to give 21 (89.0 mg, 68%) as a colorless viscous liquid. $[\alpha]_D^{20} = +103.3$ (c=1.06 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.04$ (s, 3H), 0.07 (s, 3H), 0.88 (s, 9H), 1.15 (ddd, 1 H, J = 13.1, 8.9, 5.7 Hz), 1.98–2.02 (m, 1 H), 2.09 (dt, 1 H, J = 13.1, 8.2 Hz), 2.26 (dd, 1 H, J = 13.5, 7.6 Hz), 2.30 (s, 3 H), 2.46–2.54 (m, 2H), 2.66-2.75 (m, 1H), 2.71 (dd, 1H, J=13.5, 8.3 Hz), 3.80 (s, 3H), 4.06 (t, 1H, J=5.0 Hz), 4.81 (s, 1H), 4.88 (d, 1H, J=17.0 Hz), 4.94 (d, 1H, J=17.0 Hz)J=10.2 Hz), 4.99 (s, 1H), 5.65–5.73 (m, 2H), 6.09 (d, 1H, J=9.8 Hz), 6.85 (d, 1H, J=2.1 Hz), 6.86 (d, 1H, J=8.3 Hz), 6.99 ppm (dd, 1H, J=8.3, 2.1 Hz); 13 C NMR (125 MHz, CDCl₃): $\delta = -4.80, -4.05, 18.0, 20.6,$

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25.8, 38.8, 40.4, 42.4, 47.1, 47.1, 50.0, 55.9, 66.6, 112.2, 114.1, 114.7, 123.1, 126.9, 129.8, 130.1, 134.7, 139.4, 142.1, 143.1, 149.0, 169.0 ppm; IR (neat): $\bar{\nu}$ =2928, 2854, 1770, 1512, 1460, 1367, 1265, 1201, 1124, 1028, 885, 835, 773, 432, 414 cm⁻¹; HRMS (FAB): m/z: calcd for $C_{28}H_{41}O_4Si$: 469.2774, found 469.2771 [M+H] $^+$.

(1S, 3R, 3aR, 4S, 7aS) - 3 - (Acetoxy-4-methoxybenzyl) - 7 - methylene-1-vinyl-10 - (Acetoxy-4-methylene-1-vinyl-10 - (Acetoxy-4-methylene-1-vin1,2,3,3a,7,7a-hexahydroinden-4-ol (23): Tetra-n-butylammonium fluoride (TBAF) in THF (1 m solution, 0.90 mL, 0.90 mmol) was added to a stirred solution of 22 (105 mg, 0.22 mmol) in THF (5 mL) at room temperature. After 16 h, (CH₃CO)₂O (62 µL, 0.66 mmol) was added very slowly to the reaction mixture. After 10 min, the reaction was quenched with saturated aqueous NH₄Cl (1 mL), and the resulting mixture was extracted with EtOAc (2×50 mL). The combined extracts were washed with saturated aqueous NaHCO3 (2×20 mL) and brine (20 mL), then dried over Na₂SO₄. Concentration of solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 4:1) to give 23 (73 mg, 92%) as a colorless viscous liquid. $[\alpha]_D^{20} = +61.8$ (c = 0.78 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 1.18-1.35$ (m, 2H), 2.02–2.06 (m, 1H), 2.15-2.22 (m, 1H), 2.30 (s, 3H), 2.50-2.59 (m, 2H), 2.60-2.69 (m, 3H), 3.80 (s, 3H), 4.03 (t, 1H, J=5.1 Hz), 4.89 (s, 1H), 4.91 (d, 1H, J=5.1 Hz)J=17.1 Hz), 4.96 (d, 1 H, J=10.2 Hz), 5.04 (s, 1 H), 5.68 (ddd, 1 H, J=10.2 Hz) 17.1, 10.2, 7.8 Hz), 5.84 (dd, 1 H, J = 9.8, 5.1 Hz), 6.18 (d, 1 H, J = 9.8 Hz), 6.87 (d, 1H, J=8.3 Hz), 6.90 (d, 1H, J=2.1 Hz), 7.02 ppm (dd, 1H, J= 8.3, 2.1 Hz); 13 C NMR (125 MHz, CDCl₃): δ = 20.7, 38.9, 40.3, 42.4, 46.7, 46.7, 50.7, 55.9, 65.7, 112.2, 114.6, 116.0, 123.1, 127.0, 128.5, 131.4, 134.2, 139.5, 141.4, 142.8, 149.2, 169.1 ppm; IR (neat): $\tilde{v} = 3516$, 1766, 1639, 1512, 1442, 1369, 1267, 1205, 1122, 1014, 904, 810 cm⁻¹; HRMS (EI): m/z:

calcd for $C_{22}H_{26}O_4$: 354.1831, found 354.1843 [M]⁺.

(1S,3R,3aR,7aS)-3-(3-Acetoxy-4-methoxybenzyl)-7-methylene-1-vinyl-1,2,3,3a,7,7a-hexahydroinden-4-one (24): Dess-Martin periodinane (81.0 mg, 0.51 mmol) was added in small portions to a stirred solution of 23 (90.0 mg, 0.25 mmol) in dry CH₂Cl₂ (6 mL) at room temperature. After 30 min, the reaction was quenched with 20% aqueous Na₂S₂O₃ (1.5 mL), and the mixture was extracted with EtOAc (2×25 mL). The combined extracts were washed with brine (2×15 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 5:1) to give 24 (74.9 mg, 84%) as a white cloudy oil. $[\alpha]_D^{20} = +18.0 \ (c = 0.67 \text{ in CHCl}_3);$ ¹H NMR (500 MHz, CDCl₃): $\delta = 1.20$ (ddd, 1H, J = 13.0, 10.6, 7.5 Hz), 1.99 (dt, 1H, J = 13.0, 7.7 Hz), 2.24 (m, 1H), 2.30 (s, 3H), 2.57 (dd, 1H, 1H)J=8.2, 3.6 Hz), 2.61 (dd, 1 H, J=13.9, 8.6 Hz), 2.72 (dd, 1 H, J=10.6, 8.2 Hz), 2.84 (dd, 1H, J=13.0, 7.0 Hz), 2.90–2.99 (m, 1H), 3.80 (s, 3H), 4.87 (d, 1H, J=17.0 Hz), 5.00 (d, 1H, J=10.2 Hz), 5.21 (s, 1H), 5.36 (s, 1H), 5.62 (ddd, 1H, J=17.0, 10.2, 8.1 Hz), 5.90 (d, 1H, J=9.4 Hz), 6.88 (d, 1H, J=8.3 Hz), 6.92 (d, 1H, J=2.1 Hz), 6.95 (d, 1H, J=9.4 Hz), 7.08 ppm (dd, 1 H, J = 8.3, 2.1 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = 20.7$, 37.6, 41.2, 41.7, 48.7, 50.2, 53.7, 55.9, 112.3, 116.0, 121.6, 123.4, 126.5, 133.2, 139.5, 140.3, 140.5, 145.7, 149.4, 169.1, 199.7, 127.1 ppm; IR (neat): $\tilde{v} = 2934, 1766, 1664, 1581, 1512, 1442, 1369, 1267, 1203, 1124, 1028, 906,$ 812, 779 cm $^{-1}$; HRMS (EI): m/z: calcd for $C_{22}H_{24}O_4$: 352.1675, found 352.1674 [M]+.

(1S,3R,3aR,7aS)-3-(3-Hydroxy-4-methoxybenzyl)-7-methylene-1-vinyl-1,2,3,3a,7,7a-hexahydroinden-4-one (3) (3-epi-ottelione A): K_2CO_3 (29.0 mg, 0.21 mmol) was added in small portions to a stirred solution of 24 (74.0 mg, 0.21 mmol) in MeOH (4 mL) at 0 °C. After 30 min, the mixture was diluted with EtOAc (30 mL). The organic layer was washed successively with 3% aqueous HCl (2×6 mL), saturated aqueous NaHCO₃ (2×6 mL) and brine (3 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo gave a residue, which was purified by column chromatography (hexane/EtOAc 4:1) to give 3 (63.8 mg, 98%) as a white cloudy oil. $[a]_{\rm D}^{20} = +15.6$ (c = 0.32 in CHCl₃); $^{1}{\rm H~NMR}$ (500 MHz, CDCl₃): $\delta = 1.20$ (ddd, 1H, J = 17.7, 10.3, 7.4 Hz), 1.98 (dt, 1H, J = 13.1, 7.8 Hz), 2.19–2.28 (m, 1H), 2.54–2.60 (m, 2H), 2.73 (dd, 1H, J=10.6, 8.4 Hz), 2.79 (dd, 1 H, J = 13.7, 7.1 Hz), 2.90–2.98 (m, 1 H), 3.85 (s, 3 H), 4.87 (d, 1H, J=17.0 Hz), 4.99 (d, 1H, J=10.2 Hz), 5.21 (s, 1H), 5.36 (s, 1H), 5.53 (s, 1H), 5.64 (ddd, 1H, J=17.0, 10.2, 8.1 Hz), 5.88–5.92 (m, 1H), 6.73 (dd, 1H, J=8.1, 1.9 Hz), 6.77 (d, 1H, J=8.1 Hz), 6.81 (d, 1H, J=1.9 Hz), 6.94 ppm (d, 1 H, J=9.9 Hz); ¹³C NMR (125 MHz, CDCl₃):

 $\delta\!=\!37.5,\ 41.3,\ 41.9,\ 48.7,\ 50.2,\ 53.7,\ 56.0,\ 110.6,\ 115.4,\ 115.9,\ 120.3,\ 121.5,\ 126.5,\ 133.9,\ 140.5,\ 140.7,\ 144.9,\ 145.3,\ 145.6,\ 199.8\ ppm;\ IR\ (neat):\ \tilde{\nu}\!=\!3423,\ 2920,\ 1655,\ 1589,\ 1510,\ 1440,\ 1273,\ 1234,\ 1130,\ 1028,\ 912,\ 804,\ 760\ cm^{-1};\ HRMS\ (EI):\ \emph{m/z}:\ calcd\ for\ C$_{20}H_{22}O_3:\ 310.1569,\ found\ 310.1565\ [\emph{M}]^+.$

(2S,2aS,4R,4aR,5R,6R,7R,7aS,7bR)-5-tert-Butyldimethylsiloxy-4-[1-hydroxy-[4-methoxy-3-(methoxymethoxy)phenyl]methyl]-6,7-(O-isopropylidenedioxy)decahydroindeno[7,1-bc]furan-2-ol (27a,b): nBuLi in nhexane (1.55 m solution, 4.50 mL, 7.0 mmol) was added dropwise to a stirred solution of 4-bromo-1-methoxy-2-(methoxymethoxy)benzene (7) (2.30 g, 7.7 mmol) in dry THF (40 mL) at -78 °C under argon. After 1 h, a solution of 9 (930 mg, 2.3 mmol) in dry THF (30 mL) was added to the above mixture at -78 °C. After 1 h, the reaction was quenched with saturated aqueous NH₄Cl (10 mL), and the resulting mixture was extracted with Et₂O (2×100 mL). The organic layer was washed with saturated aqueous NaHCO₃ (2×30 mL) and brine (30 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 2:1 \rightarrow 1:1) to give 27a (1.06 g, 80 %, more polar) and 27b (263 mg, 20 %, less polar).

27a: colorless oil; $[\alpha]_D^{20} = +20.1$ (c=0.74 in CHCl₃); 1 H NMR (500 MHz, CDCl₃): $\delta=0.17$ (s, 3 H), 0.24 (s, 3 H), 0.95 (s, 9 H), 1.26 (s, 6 H), 1.81 (brs, 1 H), 1.88 (brs, 1 H), 1.98–2.08 (m, 1 H), 2.27 (brs, 1 H), 2.29 (d, 1 H, J=2.4 Hz), 2.47–2.56 (m, 1 H), 2.76–2.87 (m, 2 H), 3.51 (s, 3 H), 3.88 (s, 3 H), 3.99 (t, 1 H, J=3.4 Hz), 4.20–4.25 (m, 1 H), 4.42–4.49 (m, 1 H), 4.47 (dd, 1 H, J=7.1, 1.6 Hz), 4.82–4.89 (m, 1 H), 5.23 (s, 2 H), 5.27 (d, 1 H, J=2.5 Hz), 6.88 (d, 1 H, J=8.3 Hz), 7.03 (dd, 1 H, J=8.3, 1.9 Hz), 7.22 ppm (d, 1 H, J=1.9 Hz); 13 C NMR (125 MHz, CDCl₃): $\delta=-4.07$, -3.82, 18.5, 23.7, 26.1, 26.4, 32.7, 36.6, 42.1, 52.1, 52.4, 53.4, 55.9, 56.1, 68.6, 73.9, 74.9, 76.0, 95.6, 103.6, 107.4, 111.7, 114.9, 120.5, 136.6, 146.7, 149.4 ppm; IR (neat): $\bar{\nu}=3423$, 2932, 2856, 1606, 1510, 1464, 1385, 1259, 1211, 1155, 1132, 1080, 1012, 868, 835, 775 cm⁻¹; HRMS (EI): m/z: calcd for $C_{29}H_{46}O_9$ Si: 566.2911, found 566.2904 [M] $^+$.

27b: colorless oil; $[\alpha]_D^{20} = +12.5$ (c=0.74 in CHCl₃); ^1H NMR (500 MHz, CDCl₃): $\delta=0.20$ (s, 3 H), 0.24 (s, 3 H), 0.97 (s, 9 H), 1.26 (s, 3 H), 1.36–1.44 (m, 1 H), 1.44 (s, 3 H), 1.67–1.78 (m, 1 H), 1.77 (m, 1 H), 2.15–2.18 (m, 1 H), 2.36–2.40 (m, 1 H), 2.51–2.58 (m, 1 H), 2.63–2.72 (m, 1 H), 2.84–2.93 (m, 1 H), 3.51 (s, 3 H), 3.84–3.90 (m, 3 H), 4.36 (m, 1 H), 4.48 (d, 1 H, J=7.0 Hz), 4.54 (d, 1 H, J=6.4 Hz), 4.55–4.60 (m, 1 H), 4.86 (d, 1 H, J=10.7 Hz), 5.18 (s, 1 H), 5.21 (s, 2 H), 6.87 (d, 1 H, J=8.3 Hz), 6.97 (d, 1 H, J=7.7 Hz), 7.19 ppm (s, 1 H); 13 C NMR (125 MHz, CDCl₃): $\delta=-3.81$, -3.66, 18.5, 23.7, 25.8, 26.2, 26.5, 32.6, 36.4, 42.0, 52.6, 53.1, 68.3, 72.4, 73.8, 74.5, 75.1, 95.6, 103.8, 107.5, 111.6, 114.8, 120.0, 137.6, 146.4, 149.3 ppm; IR (neat): $\bar{v}=3435$, 2934, 1510, 1464, 1383, 1261, 1157, 1078, 1049, 1006, 868, 835, 775 cm⁻¹; HRMS (FAB): m/z: calcd for C₂₉H₄₆O₉Si: 566.2911, found 566.2914 [M]⁺.

(1R,3R,3aR,4R,5R,6R,7S,7aS)-4-tert-Butyldimethylsiloxy-7-hydroxy-3-[(1S)-hydroxy[4-methoxy-3-(methoxymethoxy)phenyl]methyl]-5,6-(O-iso-propylidenedioxy)octahydroinden-1-carbaldehyde (26a): DBU (4.70 mL, 32 mmol) was added to a stirred solution of 27a (900 mg, 1.6 mmol) in dry toluene (30 mL) at room temperature under argon. The mixture was heated at reflux for 1.5 h. After cooling, the mixture was directly subjected to column chromatography eluting with EtOAc in order to remove excess DBU. The fractions containing 26a and starting material 27a were collected and then concentrated in vacuo. The residue was purified by column chromatography (hexane/EtOAc 2:1→1:1) to give less polar 26a (270 mg, 30 %) as a colorless viscous liquid and more polar starting material 27a (540 mg, 60%).

26a: $[\alpha]_D^{20} = -34.3 \ (c = 0.98 \ \text{in CHCl}_3); \ ^1\text{H NMR} \ (500 \ \text{MHz}, \ \text{CDCl}_3); \ \delta = 0.23 \ (s, 6 \ \text{H}), 0.31 \ (s, 3 \ \text{H}), 0.97 \ (s, 9 \ \text{H}), 1.24 \ (d, 6 \ \text{H}, J = 2.0 \ \text{Hz}), 2.01 - 2.12 \ (m, 1 \ \text{H}), 2.22 - 2.43 \ (m, 3 \ \text{H}), 3.02 - 3.09 \ (m, 1 \ \text{H}), 3.37 - 3.43 \ (m, 1 \ \text{H}), 3.50 \ (s, 3 \ \text{H}), 3.82 - 3.89 \ (m, 2 \ \text{H}), 3.87 \ (s, 3 \ \text{H}), 4.02 \ (ddd, 1 \ \text{H}, J = 10.8, 6.6, 4.1 \ \text{Hz}), 4.14 - 4.21 \ (m, 1 \ \text{H}), 4.45 \ (br, 1 \ \text{H}), 4.73 \ (br, 1 \ \text{H}), 6.87 \ (d, 1 \ \text{H}, J = 8.3 \ \text{Hz}), 6.99 \ (dd, 1 \ \text{H}, J = 8.3, 2.0 \ \text{Hz}), 7.18 \ (d, 1 \ \text{H}, J = 2.0 \ \text{Hz}), 9.73 \ \text{ppm} \ (s, 1 \ \text{H}); \ ^{13}\text{C NMR} \ (125 \ \text{MHz}, \ \text{CDCl}_3); \ \delta = -4.33, -3.84, 14.2, 18.1, 21.0, 23.8, 25.9, 26.1, 50.7, 55.9, 56.2, 60.3, 75.7, 75.9, 77.2, 95.6, 108.5, 111.8, 114.9, 120.1, 135.9, 146.7, 203.0 \ \text{ppm}; \ \text{IR} \ (\text{neat}); \ \tilde{v} = 3422, 2934, 2860, 1718, 1512, 1383, 1261, 1211, 1155, 1078, 1022, 864, 839, 781 \ \text{cm}^{-1}; \ \text{HRMS} \ (\text{EI}); \ m/z : \ \text{calcd for } \ \text{C}_{29} \ \text{H}_{46} \ \text{O}_{9} \ \text{Si}: 566.2911, \ \text{found } 566.2916 \ [M]^+.$

(1R,3R,3aS,4S,5R,6R,7R,7aR)-7-tert-Butyldimethylsiloxy-1-[(1S)-hydroxy[4-methoxy-3-(methoxymethoxy)phenyl]methyl]-3-hydroxymethyl-5,6-(O-isopropylidenedioxy)octahydroinden-4-ol (30): A solution of 26 a (306 mg, 0.54 mmol) in dry THF (6 mL) was added dropwise to a stirred suspension of LiAlH₄ (41.0 mg, 1.1 mmol) in dry THF (4 mL) at 0 °C under argon. After 1 min, H_2O (40 μL), 15% aqueous NaOH (40 μL), H₂O (0.12 mL), and Na₂SO₄ (500 mg) were added successively to the reaction mixture at 0°C, and the resulting mixture was further stirred for 30 min at room temperature. The mixture was filtrated through a pad of Celite, and the filtrate was concentrated in vacuo to afford a residue, which was purified by column chromatography (hexane/EtOAc 1:1-> EtOAc) to give **30** (294 mg, 96%) as a colorless viscous liquid. $[\alpha]_D^{20}$ -36.6 (c=0.83 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ =0.22 (s, 3H), 0.27 (s, 3H), 0.96 (s, 9H), 1.26 (s, 6H), 1.63 (br, 1H), 1.74 (br, 1H), 1.91-1.99 (m, 1H), 2.31-2.39 (m, 2H), 2.46-2.54 (m, 1H), 2.68 (br, 1H), 3.43 (m, 1H), 3.51 (s, 3H), 3.48–3.56 (m, 1H), 3.55–3.62 (m, 1H), 3.83–3.90 (m, 1H), 3.87 (s, 3H), 4.04 (br, 1H), 4.17-4.22 (m, 1H), 4.42 (dd, 1H, <math>J=7.3, 5.2 Hz), 4.80–4.84 (m, 1 H), 5.23 (s, 2 H), 6.87 (d, 1 H, J = 8.3 Hz), 7.01 (dd, 1H, J=8.3, 2.0 Hz), 7.19 ppm (d, 1H, J=2.0 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = -4.31$, -3.99, 18.1, 24.0, 25.9, 26.1, 26.2, 30.2, 40.0, 43.7, 49.0, 55.9, 56.2, 67.3, 67.4, 72.0, 73.0, 76.4, 76.5, 95.5, 108.5, 111.5, 114.5, 120.0, 136.4, 146.5, 149.1 ppm; IR (neat): $\tilde{v} = 3435$, 2932, 1512, 1383, 1259, 1155, 1078, 1018, 864, 837, 779 cm⁻¹; HRMS (EI): *m/z*: calcd for $C_{29}H_{48}O_9Si: 568.3068$, found 568.3085 [M]⁺.

(1R,3R,3aS,4S,5R,6R,7R,7aR)-4-Acetoxy-3-acetoxymethyl-7-tert-butyldimethylsiloxy-1-[(1S)-acetoxy[4-methoxy-3-(methoxymethoxy)phenyl]methyl]-5,6-(O-isopropylidenedioxy)octahydroindene (31): (CH₃CO)₂O (1.50 mL, 16 mmol) and DMAP (19.0 mg, 0.16 mmol) were added to a stirred solution of 30 (294 mg, 0.52 mmol) in pyridine (5 mL) at room temperature. After 1 h, the mixture was diluted with Et₂O (50 mL). The organic layer was washed successively with 3% aqueous HCl (2×10 mL), saturated aqueous NaHCO₃ (2×10 mL) and brine (10 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo gave a residue, which was purified by column chromatography (hexane/EtOAc 1:1) to give 31 (341 mg, 95%) as a colorless viscous liquid. $[a]_D^{20} = -38.8$ (c=1.16 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.20$ (s, 3 H), 0.33 (s, 3 H), 0.96 (s, 9H), 1.13 (s, 3H), 1.21 (s, 3H), 1.62 (ddd, 1H, J=12.6, 8.6, 3.8 Hz), 1.99 (s, 3H), 2.03 (s, 3H), 2.09 (s, 3H), 1.96-2.13 (m, 2H), 2.31-2.37 (m, 1H), 2.65-2.76 (m, 1H), 2.76-2.84 (m, 1H), 3.51 (s, 3H), 3.86 (s, 3H), 3.84-3.89 (m, 1 H), 3.94-4.03 (m, 2 H), 4.06 (dd, 1 H, J=6.3, 2.4 Hz), 4.21(dd, 1H, J=6.3, 4.9 Hz), 5.20–5.26 (m, 2H), 5.43 (dd, 1H, J=8.4, 4.7 Hz), 5.82 (d, 1 H, J = 10.6 Hz), 6.86 (d, 1 H, J = 8.3 Hz), 7.05 (dd, 1 H, J=8.3, 1.9 Hz), 7.23 ppm (d, 1 H, J=1.9 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = -3.96$, -3.89, 18.1, 20.9, 21.2, 24.7, 26.0, 26.4, 31.9, 37.3, 39.7, 41.8, 46.0, 55.9, 56.2, 68.2, 68.8, 69.3, 73.5, 76.6, 77.1, 95.6, 108.1, 111.3, 115.4, 121.7, 132.5, 146.5, 149.6, 169.9, 170.4, 171.3 ppm; IR (neat): \tilde{v} = 2935, 2856, 1738, 1608, 1516, 1466, 1371, 1238, 1157, 1371, 1238, 1157, 1132, 1078, 1024, 869, 837, 775, 607, 518, 412 cm⁻¹; HRMS (EI): *m/z*: calcd for $C_{35}H_{54}O_{12}Si: 694.3385$, found 694.3380 [M]⁺.

(1S,3R,3aS,4S,5R,6R,7R,7aR)-7-tert-Butyldimethylsiloxy-3-hydroxymethyl-5,6-O-isopropylidenedioxy-1-[4-methyoxy-3-(methoxymethoxy)benzyl]octahydroinden-4-ol (32): A solution of 31 (340 mg, 0.49 mmol) in dry THF (15 mL) was added dropwise to a stirred solution of Li metal (100 mg, 9.0 mmol) in liquid NH₃ (30 mL) at -78 °C under argon. After 5 min, the reaction was quenched with saturated aqueous NH₄Cl (5 mL) at the same temperature. The mixture was then allowed to stand at room temperature for 4 h in order to evaporate off excess NH3. The mixture was extracted with EtOAc (2×100 mL) and the extracts were washed with saturated aqueous NaHCO₃ (2×50 mL) and brine (50 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 1:1) to give 32 (265 mg, 98%) as a colorless viscous liquid. $[\alpha]_D^{20}$ $-27.7(c=0.91 \text{ in CHCl}_3)$; ¹H NMR (500 MHz, CDCl₃): $\delta=0.23$ (s, 3H), 0.26 (s, 3H), 0.97 (s, 9H), 1.34 (s, 3H), 1.44 (s, 3H), 1.42-1.49 (m, 1H), 1.64-1.71 (m, 1H), 1.81 (br, 1H), 2.26-2.38 (m, 2H), 2.50-2.57 (m, 2H), 2.61-2.69 (m, 1 H), 2.88 (dd, 1 H, J=13.3, 4.6 Hz), 3.43 (dd, 1 H, J=10.3, 7.3 Hz), 3.51 (m, 1H), 3.52 (s, 3H), 3.86 (s, 3H), 3.90-4.04 (m, 2H), 4.06 (t, 1H, J=3.0 Hz), 4.28 (dd, 1H, J=7.2, 2.9 Hz), 4.50–4.55 (m, 1H), 5.19-5.23 (m, 2H), 6.77 (dd, 1H, J=8.2, 1.9 Hz), 6.82 (d, 1H, J=8.2 Hz),

6.99 ppm (d, 1 H, J=1.9 Hz); 13 C NMR (125 MHz, CDCl₃): δ =-4.46, -3.56, 18.1, 24.0, 25.9, 26.4, 33.8, 35.4, 40.1, 40.6, 42.4, 42.8, 55.9, 56.1, 67.1, 67.3, 71.6, 76.1, 76.3, 95.5, 108.4, 111.6, 116.9, 122.1, 134.3, 146.3, 147.8 ppm; IR (neat): $\bar{\nu}$ =3433, 2932, 2858, 1512, 1466, 1383, 1261, 1211, 1155, 1134, 1078, 1016, 923, 868, 837, 808, 779 cm $^{-1}$; HRMS (EI): m/z: calcd for $C_{29}H_{48}O_8Si$: 552.3118, found 552.3129 [M] $^+$.

(1R,3S,3aR,4R,5R,6S,7aR)-4-tert-Butyldimethylsiloxy-5,6-O-isopropylidene-3-[4-methoxy-3-(methoxymethoxy)benzyl]-7-oxooctahydroinden-1carbaldehyde (33): Dess-Martin periodinane (609 mg, 1.4 mmol) was added in small portions to a stirred solution of 32 (265 mg, 0.48 mmol) in dry CH₂Cl₂ (10 mL) at room temperature. After 30 min, the reaction was quenched with 10% aqueous Na₂S₂O₃ (3 mL), and the resulting mixture was extracted with Et₂O (2×50 mL). The organic layer was washed successively with saturated aqueous NaHCO3 (2×20 mL) and brine (20 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 6:1) to give 33 (237 mg, 90 %) as a colorless viscous oil. [α]_D²⁰=+24.5 (c= 2.50 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.16$ (s, 3 H), 0.22 (s, 3H), 0.86 (s, 9H), 1.34 (s, 3H), 1.57 (s, 3H), 1.82-1.92 (m, 2H), 2.13-2.23 (m, 1H), 2.57 (dd, 1H, J=13.5, 10.0 Hz), 2.81 (dd, 1H, J=13.5, 5.3 Hz),3.13-3.18 (m, 1H), 3.47-3.54 (m, 1H), 3.51 (s, 3H), 3.68 (dd, 1H, J=9.6, 5.1 Hz), 3.86 (s, 3 H), 4.09–4.12 (m, 1 H), 4.33 (d, 1 H, J = 7.4 Hz), 4.40 (dd, 1H, J=7.4, 3.1 Hz), 5.21 (s, 2H), 6.75 (dd, 1H, J=8.2, 1.9 Hz), 6.82 (d, 1H, J=8.2 Hz), 6.97 (d, 1H, J=1.9 Hz), 9.63 ppm (s, 1H); 13 C NMR (125 MHz, CDCl₃): $\delta = -4.07$, -3.56, 17.9, 24.0, 25.6, 25.9, 30.5, 35.2, 43.3, 44.1, 44.8, 52.1, 55.9, 56.1, 68.8, 77.2, 80.4, 95.5, 111.7, 112.0, 116.7, 121.9, 133.2, 146.4, 148.0, 201.9, 206.2 ppm; IR (neat): $\tilde{v} = 2932$, 2858, $1728,\,1514,\,1466,\,1383,\,1263,\,1209,\,1157,\,1134,\,1060,\,1006,\,925,\,858,\,837,$ 808, 777, 538, 468, 405 cm⁻¹; HRMS (EI): m/z: calcd for $C_{29}H_{44}O_8Si$: 548.2805, found 548.2801 [M]+.

(1S,3S,3aR,4R,5R,6R,7aS)-4-tert-Butyldimethylsiloxy-5,6-O-isopropylidenedioxy-3-[4-methoxy-3-(methoxymethoxy)benzyl]-7-methylene-1-(vinyl)octahydroindene (25): Wittig reagent (Ph₃P=CH₂) in benzene solution was first prepared as follows: a suspension of Ph₃P+CH₃Br⁻(1.22 g, 3.4 mmol) and tBuOK (380 mg, 3.4 mmol) in dry benzene (10 mL) were heated at reflux for 4 h under argon, and the solution was cooled to room temperature. A solution of the Wittig reagent in benzene (1.0 mL, 0.34 mmol) was added very slowly to a stirred solution of 33 (187 mg, 0.34 mmol) in dry benzene (8 mL) at room temperature under argon. After the first methylenation of the C1-formyl group was completed (monitored by TLC), a solution of the Wittig reagent in benzene (4.0 mL, 1.36 mmol) was added once again and the resulting mixture was heated under reflux for 30 min to pursue the second methylenation of the C7-carbonyl group. The reaction was quenched with saturated aqueous NH₄Cl (5 mL), and the mixture was extracted with Et₂O (2×50 mL). The organic layer was washed with brine (2×30 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 20:1→10:1) to give 25 (176 mg, 95 %) as a colorless viscous liquid. $[\alpha]_D^{20} = -21.9$ (c=0.93) in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.14$ (s, 3H), 0.18 (s, 3H), 0.88 (s, 9H), 1.35 (s, 3H), 1.41 (ddd, 1H, J=12.2, 8.4, 3.6 Hz), 1.45 (s, 3H), 1.91–2.00 (m, 1H), 2.36–2.45 (m, 1H), 2.53 (dd, 1H, J=13.3, 10.3 Hz), 2.58–2.63 (m, 1H), 2.72–2.78 (m, 1H), 2.83 (dd, 1H, J=13.3, 5.2 Hz), 2.85-2.93 (m, 1H), 3.51 (s, 3H), 3.84 (s, 3H), 3.95-3.98 (m, 1H), 4.15 (dd, 1H, J = 7.3, 2.4 Hz), 4.62 (d, 1H, J = 7.3 Hz), 4.84-4.88 (m, 1H),4.93-4.99 (m, 1H), 5.14-5.17 (m, 1H), 5.21-5.24 (s, 2H), 5.22 (m, 1H), 5.88 (ddd, 1H, J=17.8, 10.2, 7.8 Hz), 6.78-6.81 (m, 2H), 7.00 ppm (d, 1 H, J=1.3 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta=-4.15$, -3.57, 18.0, 24.2, 25.8, 26.4, 36.2, 38.2, 42.0, 43.0, 44.8, 49.2, 55.9, 56.1, 70.0, 78.5, 78.8, 95.6, 108.6, 111.6, 111.7, 114.1, 117.0, 122.2, 134.6, 145.8, 146.1, 146.3, 147.8 ppm; IR (neat): $\tilde{\nu}$ = 2932, 2858, 1637, 1512, 1464, 1381, 1259, 1209, 1157, 1134, 1078, 1059, 1030, 1005, 908, 864, 835, 810, 775 cm⁻¹; HRMS (EI): m/z: calcd for $C_{31}H_{48}O_6Si$: 544.3220, found 544.3224 [M]⁺.

(15,35,3aR,4R,5R,6R,7aS)-4-tert-Butyldimethylsiloxy-3-(3-hydroxy-4-methoxybenzyl)-7-methylene-1-vinyloctahydroinden-5,6-diol (34): A solution of TFA/H₂O 10:1 (11 mL) was added dropwise to a stirred solution of 25 (177 mg, 0.33 mmol) in THF (1 mL) at 0°C. After 10 min, the reaction mixture was neutralized with 6 M NaOH and extracted with EtOAc

(3×40 mL). The organic layer was washed with saturated aqueous NaHCO₃ (2×20 mL) and brine (30 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 3:1) to give 34 (129 mg, 86%) as a colorless viscous liquid. $[\alpha]_D^{20} = +19.2$ (c = 1.97 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.22$ (s, 3 H), 0.26 (s, 3 H), 0.98 (s, 9 H), 1.59 (ddd, 1 H, J=15.8, 10.0, 5.7 Hz), 1.74 (d, 1 H, J=6.1 Hz), 1.93 (d, 1H, J=7.5 Hz), 1.96–2.00 (m, 1H), 2.27–2.33 (m, 1H), 2.41–2.51 (m, 2H), 2.69 (dd, 1H, J=13.6, 8.9 Hz), 2.77-2.86 (m, 2H), 3.86 (s, 3H), 3.95-3.99 (m, 1 H), 4.26 (t, 1 H, J=4.0 Hz), 4.70-4.74 (m, 1 H), 4.80-4.85(m, 1H), 4.86–4.90 (m, 1H), 4.95 (s, 1H), 5.15 (t, 1H, J=1.8 Hz), 5.54 (s, 1H), 5.62 (ddd, 1H, J=18.3, 10.1, 8.3 Hz), 6.66 (dd, 1H, J=8.2, 2.1 Hz), 6.76 (d, 1H, J=8.2 Hz), 6.78 ppm (d, 1H, J=2.1 Hz); 13 C NMR (125 MHz, CDCl₃): $\delta = -4.05, -3.37, 18.1, 26.1, 35.5, 37.1, 43.2, 44.2,$ 47.5, 54.3, 56.0, 68.2, 72.7, 75.4, 110.6, 110.8, 113.4, 114.6, 119.6, 135.3, 142.8, 144.7, 145.4, 145.8 ppm; IR (neat): $\tilde{v} = 3422$, 2932, 2858, 1639, 1591, 1512, 1442, 1259, 1130, 1062, 1037, 904, 873, 835, 775, 706 cm⁻¹; HRMS (FAB): m/z: calcd for $C_{26}H_{41}O_5Si$: 461.2723, found 461.2697 $[M+H]^+$.

(1S,3S,3aR,4R,5R,6R,7aS)-3-(3-Acetoxy-4-methoxybenzyl)-4-tert-butyldimethylsiloxy-7-methylene-1-(vinyl)octahydroinden-5,6-diol (35): NaOH (0.40 mL, 0.80 mmol) and (CH₃CO)₂O (76 μL, 0.80 mmol) were added dropwise to a stirred solution of 34 (128 mg, 0.28 mmol) in 2-propanol (2 mL) at room temperature. After 30 min, the reaction mixture was diluted with EtOAc (70 mL). The organic layer was washed with H₂O (2×20 mL) and brine (2×20 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo gave a residue, which was purified by column chromatography (benzene/EtOAc 10:1) to give 35 (127 mg, 91%) as a colorless viscous liquid. $[\alpha]_D^{20} = +20.9$ (c = 0.80 in CHCl₃); ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3): \delta = 0.21 \text{ (s, 3H)}, 0.24 \text{ (s, 3H)}, 0.97 \text{ (s, 9H)}, 1.56-1.63$ (m, 1H), 1.78 (d, 1H, J=5.8 Hz), 1.92 (d, 1H, J=7.6 Hz), 1.95-2.00 (m, 1H)1H), 2.27-2.33 (m, 1H), 2.31 (s, 3H), 2.39-2.49 (m, 2H), 2.72 (dd, 1H, J=13.7, 8.8 Hz), 2.82 (dd, 1H, J=13.7, 6.3 Hz), 2.77–2.87 (m, 1H), 3.80 (s, 3H), 3.95-3.99 (m, 1H), 4.22-4.26 (m, 1H), 4.68-4.73 (m, 1H), 4.80-4.85 (m, 1 H), 4.88 (dd, 1 H, J = 10.0, 1.3 Hz), 4.94 (s, 1 H), 5.14–5.17 (m, 1H), 5.62 (ddd, 1H, J=17.1, 10.0, 8.3 Hz), 6.85 (d, 1H, J=2.1 Hz), 6.87 (d, 1H, J=8.3 Hz), 7.00 ppm (dd, 1H, J=8.3, 2.1 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = -4.06$, -3.36, 18.1, 20.7, 26.1, 35.3, 37.1, 43.1, 44.2, 47.5, 54.3, 55.9, 68.2, 72.7, 75.4, 110.9, 112.4, 113.5, 122.6, 126.4, 134.6, 139.5, 142.7, 145.7, 149.1, 168.3 ppm; IR (neat): $\tilde{v} = 3449$, 2955, 2932, 2858, 1768, 1639, 1512, 1464, 1442, 1369, 1263, 1205, 1122, 1062, 1035, 902, 873, 835, 775, 706 cm⁻¹; HRMS (FAB): *m/z*: calcd for $C_{28}H_{43}O_6Si: 503.2829$, found 503.2832 [M+H]+.

(3aR,4R,4aR,5S,7S,7aS)-5-(3-Acetoxy-4-methoxybenzyl)-4-tert-butyldimethylsiloxy-8-methylene-7-(vinyl)octahydroindeno[5,6-d,1,3]dioxol-2thione (36): A solution of CSCl₂ (18 µL, 0.24 mmol) in dry CH₂Cl₂ (0.5 mL) was added dropwise to a stirred mixture of 35 (60.0 mg, 0.12 mmol) in dry CH₂Cl₂ containing DMAP (72.0 mg, 0.60 mmol) at room temperature. After 30 min, silica gel (1.0 g) was added to the reaction mixture, and the solvent was carefully evaporated off in vacuo. The resulting solid was charged on the top of a silica gel column chromatography, and elution using hexane/EtOAc 10:1 gave 36 (60.3 mg, 93%) as a colorless viscous liquid. $[\alpha]_D^{20} = +25.3$ (c = 0.60 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ = 0.16 (s, 3H), 0.22 (s, 3H), 0.87 (s, 9H), 1.44–1.50 (m, 1H), 1.88-1.96 (m, 1H), 2.31 (s, 3H), 2.30-2.34 (m, 1H), 2.40-2.52 (m, 2H), 2.79-2.85 (m, 1H), 2.86-2.93 (m, 2H), 3.80 (s, 3H), 4.15-4.18 (m, 1H), 4.85 (dd, 1H, J=8.5, 2.4 Hz), 4.91 (d, 1H, J=10.1 Hz), 4.94– 4.99 (m, 1H), 5.28-5.33 (m, 2H), 5.47 (d, 1H, J=2.5 Hz), 5.85 (ddd, 1H, J=2.5 Hz)J=17.2, 10.1, 7.8 Hz), 6.82 (d, 1H, J=2.1 Hz), 6.89 (d, 1H, J=8.3 Hz), 6.97 ppm (dd, 1H, J=8.3, 2.1 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta=$ -4.41, -3.49, 17.9, 20.7, 25.6, 35.7, 38.0, 42.5, 43.0, 43.9, 49.6, 55.9, 67.6, 82.1, 85.3, 112.5, 112.6, 118.8, 122.7, 126.5, 133.5, 139.6, 141.1, 144.6, 149.4, 169.0, 190.8 ppm; IR (neat): $\tilde{v} = 3076$, 2955, 2932, 2858, 1768, 1639, 1583, 1512, 1464, 1442, 1369, 1331, 1267, 1205, 1163, 1124, 1089, 1064, 1032, 995, 958, 904, 866, 837, 777, 758 cm⁻¹; HRMS (EI): m/z: calcd for $C_{29}H_{40}O_6SSi: 544.2315$, found 544.2316 [M]⁺.

(15,35,3aR,45,7aS)-3-(3-Acetoxy-4-methoxybenzyl)-4-tert-butyldimethyl-siloxy-7-methylene-1-vinyl-1,2,3,3a,7,7a-hexahydroindene (37): A solution of 36 (48.0 mg, 88 μ mol) in (EtO)₃P (8.8 mL) was heated at reflux for 2 h

under argon. After cooling, excess (EtO)₃P was removed using short column chromatography eluting with hexane. The combined fractions were concentrated in vacuo to afford a residue, which was purified by column chromatography (benzene/EtOAc 100:1) to give 37 (32.2 mg, 78%) as a colorless viscous liquid. $[\alpha]_D^{20} = +98.4$ (c=1.07 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 0.11$ (s, 3 H), 0.16 (s, 3 H), 0.92 (s, 9 H), 1.50-1.57 (m, 1H), 1.93 (ddd, 1H, J=12.7, 10.1, 8.3 Hz), 2.13 (m, 1H), 2.30 (s, 3 H), 2.38-2.50 (m, 2 H), 2.70 (dd, 1 H, J=13.7, 9.3 Hz), 2.78-2.87(m, 1H), 2.96 (dd, 1H, J=13.7, 6.0 Hz), 3.80 (s, 3H), 4.45 (t, 1H, J=13.74.9 Hz), 4.84–4.92 (m. 3H), 4.95 (s. 1H), 5.73 (ddd, 1H, J=16.9, 10.0, 8.8 Hz), 5.83 (dd, 1H, J=9.7, 4.9 Hz), 6.13 (d, 1H, J=9.8 Hz), 6.85–6.89 (m, 2H), 7.01 ppm (dd, 1H, J=8.3, 2.0 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta = -4.08$, -3.29, 18.1, 20.7, 26.0, 35.7, 43.9, 45.9, 47.9, 49.6, 55.9, 65.5, 112.2, 113.5, 114.3, 122.8, 126.6, 129.9, 131.1, 135.4, 139.5, 143.3, 144.0, 148.9, 169.0 ppm; IR (neat): $\tilde{v} = 2928$, 2857, 1771, 1512, 1464, 1368, 1262, 1204, 1125, 1030, 882, 835, 773 cm⁻¹; HRMS (EI): m/z: calcd for C₂₈H₄₀O₄Si: 468.2696, found 468.2700 [M]+

 $(1S,\!3S,\!3aR,\!4S,\!7aS) - 3 - (3 - Hydroxy - 4 - methoxybenzyl) - 7 - methylene - 1 - vinyl-property - 1 - vinyl-property - 2 - vinyl-$ 1,2,3,3a,7,7a-hexahydroinden-4-ol (38): Tetra-n-butylammonium fluoride (TBAF) in THF (1 M solution, 0.86 mL. 0.86 mmol) was added to a stirred solution of 37 (40.0 mg, $85 \, \mu mol$) in THF (2 mL) at room temperature. After 48 h, the reaction was quenched with saturated aqueous NH₄Cl (1 mL), and extracted with EtOAc (2×30 mL). The combined extracts were washed successively with saturated aqueous NaHCO3 (2×10 mL) and brine (10 mL), then dried over Na2SO4. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (benzene/EtOAc 4:1) to give 38 (22.1 mg, 83%) as a colorless viscous liquid. [α]_D²⁰ = +74.6 (c = 0.35 in CHCl₃); ¹H NMR (500 MHz, $CDCl_3$): $\delta = 1.05$ (br, 1H), 1.65 (ddd, 1H, J = 16.1, 10.1, 6.0 Hz), 1.99–2.03 (m, 1H), 2.04-2.09 (m, 1H), 2.44 (dd, 1H, J=10.8, 6.3 Hz), 2.47-2.54 (m, 1H), 2.04-2.09 (m, 1H), 2.44 (dd, 1H, J=10.8, 6.3 Hz), 2.47-2.54 (m, 1H), 2.04-2.09 (m, 1H), 2.44 (dd, 1H, J=10.8, 6.3 Hz), 2.47-2.54 (m, 1H), 2.48 (dd, 1H, J=10.8, 6.3 Hz), 2.49-2.54 (m, 1H), 2.49 (dd, 1H, J=10.8, 6.3 Hz), 2.49-2.54 (m, 1H), 2.49 (dd, 1H, J=10.8, 6.3 Hz), 2.49-2.54 (m, 1H), 2.49 (dd, 1H, J=10.8, 6.3 Hz), 2.49-2.54 (m, 1H), 2.49 (dd, 1H, J=10.8, 6.3 Hz), 2.49-2.54 (m, 1H), 2.49 (dd, 1H), 2.41 H), 2.55–2.61 (m, 1 H), 2.87 (dd, 1 H, J = 13.7, 8.3 Hz), 2.98 (dd, 1 H, J = 13.7, 8.3 Hz), 2.98 (dd, 1 H, J = 13.7, 8.3 Hz) 13.7, 7.2 Hz), 3.86 (s, 3 H), 4.36 (br, 1 H), 4.83-4.88 (m, 2 H), 4.90 (dd, $1\,\mathrm{H},\,J\!=\!10.0,\,1.9\,\mathrm{Hz}),\,4.99\!-\!5.03\,\,(\mathrm{m},\,1\,\mathrm{H}),\,5.54\,\,(\mathrm{br},\,1\,\mathrm{H}),\,5.70\,\,(\mathrm{ddd},\,1\,\mathrm{H},\,1)$ J = 18.7, 10.0, 8.8 Hz, 5.93 (dd, 1 H, J = 9.7, 5.6 Hz), 6.19 (d, 1 H, J =9.7 Hz), 6.73 (dd, 1 H, J = 8.2, 1.9 Hz), 6.76 (d, 1 H, J = 8.2 Hz), 6.84 ppm (d, 1H, J=1.9 Hz); ¹³C NMR (125 MHz, CDCl₃): $\delta=36.7$, 37.9, 44.4, 44.9, 48.6, 49.9, 55.9, 64.3, 110.4, 113.9, 114.7, 115.8, 119.9, 128.1, 131.5, 135.8, 142.0, 143.7, 144.5, 145.3 ppm; IR (neat): $\tilde{v} = 3510$, 2924, 1591, 1512, 1442, 1273, 1130, 1008, 904, 802, 761 cm⁻¹; HRMS (EI): m/z: calcd for C₂₀H₂₄O₃: 312.1725, found 312.1729 [M]+.

(1S, 3S, 3aR, 4S, 7aS) - 3 - (3-Dichloroacetoxy-4-methoxybenzyl) - 7-methylene-methoxybenzyl) - 7-methylene-methylene-methoxybenzyl) - 7-methylene-methoxybenzyl) - 7-methylene-methylene-methoxybenzyl) - 7-methylene-methoxybenzyl) - 7-methylene-methoxybenzyl) - 7-methylene-methylene-methylene-methylene-methylene-methylene-methylene-methylene-methylene-methylene1-vinyl-1,2,3,3a,7,7a-hexahydroinden-4-ol (39): A $(CHCl_2CO)_2O$ (10 μL , 65 μmol) in dry CH_2Cl_2 (0.2 mL) was added very slowly to a stirred solution of 38 (17.0 mg, 54 µmol) in dry CH₂Cl₂ (0.8 mL) containing pyridine (9.0 µL, 0.1 mmol) at room temperature. The reaction mixture was diluted with EtOAc (10 mL), and the organic layer was washed with brine (3 mL), and dried over Na2SO4. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (benzene/EtOAc 20:1) to give 39 (18.2 mg, 79%) as a white cloudy oil. $[\alpha]_D^{20} = +71.5$ (c = 0.37 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): $\delta = 1.02$ (d, 1H, J = 6.3 Hz), 1.64 (ddd, 1H, J = 19.0, $10.1,\ 6.0\ Hz),\ 1.99-2.03\ (m,\ 1\,H),\ 2.05-2.10\ (m,\ 1\,H),\ 2.43-2.62\ (m,\ 3\,H),$ 2.93 (dd, 1H, J=13.9, 8.6 Hz), 3.03 (dd, 1H, J=13.9, 6.9 Hz), 3.82 (s, 3 H), 4.30-4.36 (m, 1 H), 4.83-4.85 (m, 1 H), 4.86-4.89 (m, 1 H), 4.92 (dd, 1H, J=9.9, 1.8 Hz), 5.01–5.04 (m, 1H), 5.70 (ddd, 1H, J=18.7, 9.9, 8.8 Hz), 5.91-5.96 (m, 1 H), 6.18 (s, 1 H), 6.20 (d, 1 H, J = 9.6 Hz), 6.91 (d, J = 9.6 Hz)1 H, J=8.4 Hz), 7.01 (d, 1 H, J=2.0 Hz), 7.14 ppm (dd, 1 H, J=8.4, 2.0 Hz); 13 C NMR (125 MHz, CDCl₃): $\delta = 35.9$, 37.8, 44.3, 44.9, 48.6, 49.9, 56.1, 64.0, 64.3, 112.6, 114.0, 116.0, 121.9, 127.6, 128.0, 131.6, 135.4, 138.8, 141.9, 143.5, 148.6, 162.5 ppm; IR (neat): $\tilde{v} = 3568$, 2926, 2345, 1782, 1512, 1269, 1141, 1028, 815, 407 cm $^{-1}$; HRMS (EI): m/z: calcd for $C_{22}H_{24}Cl_2O_4$: 422.1052, found 422.1050 [M]+.

(15,35,3aS,7aS)-3-(3-Dichloroacetyl-4-methoxybenzyl)-1,2,3,3a,7,7a-hexahydro-7-methylene-1-(vinyl)inden-4-one (40): Dess–Martin periodinane (34.0 mg, $81~\mu$ mol) was added to a stirred solution of 39 (17.0 mg, $40~\mu$ mol) in dry CH₂Cl₂ (1.2 mL) at room temperature. After 15 min, the reaction was quenched with 20% aqueous Na₂S₂O₃ (0.5 mL), and the mixture was extracted with EtOAc (2×15 mL). The combined extracts

were washed with brine (10 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 4:1) to give **40** (16.0 mg, 95%) as a white cloudy oil. [α]_D²⁰=+22.4 (c=0.83 in CHCl₃); ¹H NMR (500 MHz, CDCl₃): δ =1.58–1.63 (m, 1H), 1.69–1.77 (m, 1H), 2.47 (m, 1H), 2.54–2.63 (m, 2H), 2.73–2.78 (m, 1H), 2.78–2.83 (m, 1H), 3.16 (dd, 1H, J=13.6, 6.9 Hz), 3.80 (s, 3H), 4.90–4.95 (m, 1H), 4.96–5.00 (m, 1H), 5.29 (d, 1H, J=1.1 Hz), 5.32 (s, 1H), 5.75 (ddd, 1H, J=18.3, 10.1, 8.2 Hz), 5.92 (dd, 1H, J=8.4 Hz), 6.96 (d, 1H, J=2.1 Hz), 7.11 ppm (dd, 1H, J=8.4, 2.1 Hz); ¹³C NMR (125 MHz, CDCl₃): δ =35.5, 35.6, 46.3, 48.7, 49.1, 51.0, 56.1, 64.0, 112.6, 114.5, 120.1, 122.2, 128.0, 128.3, 134.8, 138.7, 141.6, 142.5, 145.8, 148.8, 162.5, 200.2 ppm; IR (neat): $\bar{\nu}$ =2932, 1784, 1660, 1512, 1442, 1267, 1217, 1141, 1028, 912, 814 cm⁻¹; HRMS (EI): m/z: calcd for $C_{22}H_{22}Cl_2O_4$: 420.0895, found 420.0888 [M]+.

(1S,3S,3aR,7aS)-1,2,3,3a,7,7a-Hexahydro-3-(3-hydroxy-4-methoxybenzyl)-7-methylene-1-(vinyl)inden-4-one (1) [(+)-ottelione A]: Compound 40 (13.0 mg, 31 $\mu mol)$ was dissolved in a solution of 50 % aqueous NaHCO $_3/$ CH₃CN 1:1 (2.0 mL), and the mixture was stirred for 1 h at room temperature. The reaction mixture was diluted with EtOAc (20 mL), and the organic layer was washed with brine (2×5 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (benzene/EtOAc 20:1) to give $\boldsymbol{1}$ (8.6 mg, 90%) as white cloudy oil. $[a]_D^{25} = +17.3$ (c=0.55 in CHCl₃). The ¹H NMR, ¹³C NMR, IR, and MS spectra (see below) are compatible with those of natural (+)-ottelione A. ¹H NMR (500 MHz, CDCl₃): $\delta = 1.52$ – 1.61 (m, 1H), 1.69-1.76 (m, 1H), 2.48-2.54 (m, 2H), 2.58-2.66 (m, 1H), 2.78 (t, 1H, J=8.2 Hz), 2.86 (dd, 1H, J=8.2, 5.9 Hz), 3.01-3.09 (m, 1H), 3.85 (s, 3H), 4.91-4.99 (m, 2H), 5.30-5.33 (m, 2H), 5.53 (br, 1H), 5.76 (ddd, 1 H, J = 18.2, 10.1, 8.2 Hz), 5.93 (d, 1 H, J = 10.0 Hz), 6.68 (dd, 1 H, J=8.2, 2.0 Hz), 6.75 (d, 1H, J=8.2 Hz), 6.79 (d, 1H, J=2.0 Hz), 6.99 ppm (d, 1 H, J = 9.9 Hz); 13 C NMR (125 MHz, CDCl₃): δ = 35.6, 35.8, 46.3, 48.5, 49.0, 51.4, 55.9, 110.4, 114.4, 114.9, 119.9, 120.2, 128.3, 135.1, 141.8, 142.9, 144.6, 145.2, 145.9, 200.4 ppm; IR (neat): $\tilde{v} = 3439$, 2926, 1658, 1589, 1510, 1440, 1275, 1236, 1130, 1030, 958, 800, 760 cm⁻¹; HRMS (EI): m/z: calcd for $C_{20}H_{22}O_3$: 310.1569, found 310.1571 $[M]^+$.

(1S,3S,3aS,7aS)-1,2,3,3a,7,7a-Hexahydro-3-(3-hydroxy-4-methoxybenzyl)-7-methylene-1-(vinyl)inden-4-one (2) [(+)-ottelione B]: tBuOK (8.0 mg, 0.13 mmol) was added in small portions to a stirred solution of ottelione A (1) (8.0 mg, 26 μmol) in tBuOH (1 mL) at room temperature. After 5 h, the reaction mixture was diluted with EtOAc (20 mL). The organic layer was washed with saturated aqueous NH₄Cl (2×4 mL), saturated aqueous NaHCO₃ (2×4 mL) and brine (4 mL), then dried over Na₂SO₄. Concentration of the solvent in vacuo afforded a residue, which was purified by column chromatography (hexane/EtOAc 4:1) to give a mixture of 2 and 1 (6.3 mg, 79%, 2/1 77:23 determined by 500 MHz ¹H NMR) as a colorless viscous liquid. Isolation of 2 from this mixture was performed by means of HPLC (DAICEL CHIRALPAK AD-H, i.d. 4.6×250 mm, hexane/2-propanol 4:1, 0.5 mL min⁻¹; measurement of UV 254 nm absorbance, t_R -2: 34.7 min, t_R -1: 29.6 min) to give 2 (1.8 mg, 23%) as a white crystal, m.p. 142.2–143.0 °C; $[\alpha]_D^{25} = -330.0$ (c = 0.18 in CHCl₃). The ¹H NMR, ¹³C NMR, IR, and MS (FAB) spectra (see below) are compatible with those of natural (–)-ottelione B. ^{1}H NMR (500 MHz, CDCl₃): $\delta = 1.59$ (ddd, 1H, J = 13.8, 10.2, 8.2 Hz), 1.78 (ddd, 1H, J = 13.8, 9.8, 6.7 Hz), 2.27 (dd, 1 H, J=13.9, 9.8 Hz), 2.34 (dd, 1 H, J=13.9, 10.2 Hz), 2.46-2.56 (m, 2H), 2.69-2.77 (m, 1H), 3.14 (dd, 1H, J=13.4, 3.6 Hz), 3.86 (s, 3H), 4.98-5.01 (m, 1H), 5.06-5.11 (m, 1H), 5.29-5.32 (m, 1H), 5.43-5.45 (m, 1 H), 5.52 (br, 1 H), 5.74 (ddd, 1 H, J=18.2, 10.2, 8.0, Hz), 5.94 (d, 1H, J=9.7 Hz), 6.69 (dd, 1H, J=8.1, 2.0 Hz), 6.76 (d, 1H, J=8.1 Hz), 6.80 (d, 1H, J=2.0 Hz), 6.98–7.02 ppm (m, 1H); 13 C NMR (125 MHz, CDCl₃): $\delta = 36.9$, 37.9, 40.6, 44.5, 50.5, 55.9, 58.3, 110.5, 114.6, 115.3, 117.1, 120.5, 128.7, 134.1, 141.5, 144.8, 144.9, 145.2, 147.6, 200.6 ppm; IR (KBr): $\tilde{v} = 3395$, 2930, 1664, 1589, 1512, 1440, 1275, 1176, 1126, 1035, 916, 808, 761 cm⁻¹; HRMS (FAB): m/z: calcd for C₂₀H₂₃O₃: 311.1647, found 311.1633 [*M*+H]⁺.

Cell-growth inhibition assay:^[19] This experiment was carried out at the Cancer Chemotherapy Center, Japanese Foundation for Cancer Research. The screening panel consisted of the following 39 human cancer

cell lines (HCC panel): breast cancer HBC-4, BSY-1, HBC-5, MCF-7, and MDA-MB-231; brain cancer U-251, SF-268, SF-295, SF-539, SNB-75, and SNB-78; colon cancer HCC2998, KM-12, HT-29, HCT-15, and HCT-116; lung cancer NCI-H23, NCI-H226, NCI-H522, NCI-H460, A549, DMS273, and DMS114; melanoma LOX-IMVI; ovarian cancer OVCAR-3, OVCAR-4, OVCAR-5, OVCAR-8, and SK-OV-3; renal cancer RXF-631 L and ACHN; stomach cancer St-4, MKN1, MKN7, MKN28, MKN45, and MKN74; prostate cancer DU-145 and PC-3. The GI₅₀ (50 % cell growth inhibition) value for these cell lines was determined by using the sulforhodamine B colorimetric method.

Tubulin polymerization assay: [22] Ability of the test compounds to interfere with the polymerization dynamics of tubulin was examined by the two-step bioassay. In brief, the nerve growth factor (NGF)/rat pheochromocytoma (PC12) cell system was utilized in the first step to examine whether the test compounds had the ability to affect the polymerization dynamics of tubulin. In the second step, effect of the test compounds on the microtuble structures was examined in HT1080 human fibrosarcoma cells to distinguish whether they inhibited the polymerization of tubulin (to destabilize microtubule structures) or the depolymerization of tubulin (to stabilize microtubule structures).

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