Structure–Activity Relationship of N-[2-(Dimethylamino)-6-[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propoxy]phenyl]-N-pentylurea and Analogues. Novel Potent Inhibitors of Acyl-CoA:Cholesterol O-Acyltransferase with Antiatherosclerotic Activity

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We have discovered N-butyl-N'-[2-(dimethylamino)-6-[3-(4-phenyl-1H-imidazol-1-yl)propoxy]-phenyl]urea (4), a novel, potent, and systemically bioavailable inhibitor of ACAT (acylCoA: cholesterol O-acyltransferase). The structure-activity relationships (SARs) of this lead compound 4 were investigated by systematic modification of four regions in the molecule. The compounds prepared in this study were tested for in vitro inhibitory activity toward both aortic and intestinal ACATs, and selected compounds were further tested for in vivo hypocholesterolemic activity. The studies not only resulted in the discovery of N-[2-(dimethylamino)-6-[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propoxy]phenyl]-N'-pentylurea (24), with potent activity and moderate plasma level after oral administration, but also revealed the SAR in each modified region. Four compounds (4, 13, 14, 24) were further selected for testing of in vivo antiatherosclerotic activity; 4, 13, and 24 reduced atherosclerotic plaque development to 38-45% of the control value in terms of area, while 14 did not have a significant antiatherosclerotic effect.

The entry of cholesterol into cells by receptor-mediated endocytosis of plasma lipoprotein and the subsequent intracellular fate of free and esterified cholesterol have received much attention from researchers interested in cholesterol metabolism and its role in pathological conditions such as coronary heart disease. AcylCoA:cholesterol O-acyltransferase (ACAT, EC 2.3.1.26) is one of the enzymes responsible for catalyzing the formation of cholesteryl ester.²⁻⁴ This enzyme plays an important role in the intestinal absorption of dietary cholesterol, because ACAT activity is greatest in the jejunum, where the majority of cholesterol absorption occurs, 3,5 and in vitro and ex vivo inhibition of ACAT activity is correlated with the inhibition of cholesterol absorption, at least in rats⁶ and rabbits. In addition, studies both in cultured cells^{8–12} and in arterial tissue 13-15 have suggested that ACAT activity is regulated and increased when cells are exposed to cholesterol-rich media. Since the intracellular accumulation of esterified cholesterol is one of the characteristic features of the atherosclerotic plaque, the regulation of ACAT is likely to be of great importance for the treatment of atherosclerosis.16

Several ACAT inhibitors have been shown to be effective in treating hypercholesterolemia^{7,17-24} and atherosclerotic plaque formation. The latter was assessed by measuring aortic cholesterol content and lesion severity^{18,21,25,26} in animal models during cholesterol feeding. Recently, it has been reported that a systemically bioavailable potent ACAT inhibitor prevented the accumulation of monocytes-macrophages, reduced the foam cell area, and blunted the development of thoracic aortic fatty streaklike lesions even in the absence of a plasma cholesterol lowering effect.²⁷ These results suggest that inhibition of ACAT in the arterial wall may well be considered a true antiatherosclerotic therapy. A nonabsorbable ACAT inhibitor (one that is not detected in plasma after oral administration) may retard the progression of atherosclerosis only by reducing serum cholesterol levels. However, a systemically bioavailable ACAT inhibitor may also delay or impede atherosclerosis by preventing the accumulation of esterified cholesterol in arterial tissue as well as by reducing serum cholesterol levels. We therefore searched for novel, potent, and systemically bioavailable ACAT inhibitors. Our screening program identified the benzophenone derivatives 2 and 3 as weak inhibitors of

2. R= NO₂ 3. R= NHCOC₄H₉

ACAT activity.²⁸ We then prepared a wide variety of derivatives of compounds 2 and 3 and evaluated their in vitro activity.²⁹ This led to the selection of N-butyl-N'-[2-(dimethylamino)-6-[3-(4-phenyl-1H-imidazol-1-yl)propoxy]phenyl]urea (4) as an example of a novel, potent, and systemically bioavailable inhibitor of ACAT.³⁰ The aim of the present studies was to search for analogues more potent than 4 in terms of in vitro ACAT inhibitory activity, in vivo hypocholesterolemic effect, and antiatherosclerotic effect, as well as to establish the SAR of 4 for in vitro ACAT inhibitory activity and in vivo hypocholesterolemic effect in four regions of the molecule. The results obtained, which led to the discovery of a candidate drug with potent hypocholesterolemic and antiatherosclerotic activities, are presented herein.

Scheme I. Preparation of Heterocycle Intermediates

Scheme II. Preparation of Target Molecules by Method A and Method E^a

^a Reagents: (a) (1) SOCl₂, (2) NaN₃/acetone-H₂O, (3) Δ/benzene, (4) NaOH, (5) HCl, (6) MeI, K₂CO₃/DMF; (b) (1) NaOMe/DMF, (2) HNMe₂/DMF; (c) MeI, K₂CO₃/DMF; (d) H₂, Pd-C/EtOAc; (e) alkyl isocyanate/CHCl₃; (f) (1) phenyl chloroformate/pyridine, (2) alkylamine/ toluene; (g) caproyl chloride, Et₃N/CHCl₃; (h) (1) HCO₂H, (2) MeI, K₂CO₃/DMF, (3) HCl/1,4-dioxane, (4) pentyl isocyanate/CHCl₃; (i) (1) HCO₂H, (2) BzlBr, NaH/DMF, (3) H₂, Pd-C/EtOA, (4) MeI, K₂CO₃/DMF, (5) HCl/1,4-dioxane, (6) H₂, Pd(OH)₂-C/EtOH, (7) pentyl isocyanate/ CHCl₃; (j) BBr₃/CH₂Cl₂; (k) (1) MOMCl, K₂CO₃/DMF, (2) MeI, K₂CO₃/DMF, (3) H₂, Pd-C/EtOAc, (4) pentyl isocyanate/CHCl₃, (5) HCl/THF; (l) Br₂/CHCl₃; (m) 5a-l, K₂CO₃, NaI/DMF; (n) 1-bromo-3-chloropropane, K₂CO₃/DMF; (o) Het-H, K₂CO₃, NaI/DMF; (p) LiAlH₄/THF.

Chemistry

The chloroalkylated heterocycles required for these syntheses were prepared as illustrated in Scheme I. Alkylation of arylated imidazoles,³¹ 4-phenyl-1,2,3-triazole,³² and 5-phenyltetrazole³³ with 1-bromo-ω-chloroalkanes in the presence of NaH provided a mixture of regioisomers, which were separated by flash column chromatography. The regioisomers of imidazole derivatives were identified from ¹H-NMR spectral data. In $CDCl_3$, the $ClCH_2CH_2CH_2$ protons of the 5-phenyl isomer 5f were shifted upfield, relative to the corresponding protons of the 4-phenyl isomer 5a, by the shielding effect of the benzene ring. 34,35 1,2,3-Triazole (5k) and tetrazole (51) derivatives were structurally identified by comparison of the spectral data with literature values for analogous compounds.36,37 Treatment of 5a with sodium thiobenzoate followed by reductive deprotection with LiAlH4 afforded 5-methyl-4-phenyl-1-(3-propylthio)imidazole (5m). Alkylation of 5-methyl-4-phenylimidazole with 3-chloropropionaldehyde diethyl acetal and subsequent separation of the regioisomers afforded the intermediate 5n. Deprotection of 5n with hydrochloric acid gave 3-(5-methyl-4phenyl-1*H*-imidazol-1-yl)propionaldehyde (50).³⁸

Scheme II shows the preparation of the target molecules by methods A and E. Synthesis of nitroanisole derivatives 6a-d was accomplished by the usual methods. In brief, 3-(dimethylamino)-2-nitroanisole (6a) was prepared from 3-methoxy-2-nitrobenzoic acid by Curtius rearrangement. hydrolysis of isocyanate to amine, and methylation. By the same method, 3-(dimethylamino)-6-methyl-2-nitroanisole (6b) was prepared from 3-methoxy-4-methyl-2-nitrobenzoic acid. The pyrimidine derivative 6c was obtained from 4,6-dichloro-5-nitropyrimidine by sequential treatment with sodium methoxide and dimethylamine. 4-(Dimethylamino)-3-nitroanisole (6d) was prepared from 4-methoxy-2-nitroaniline by methylation. Aniline derivatives 7a-d were obtained from the corresponding nitroanisoles 6a-d by catalytic hydrogenation.

Treatment of 7a-d with either isocyanate or phenyl chloroformate followed by reaction with alkylamines gave urea derivatives 8aa-ai, 8ba, 8ca, and 8da. Acylation of 7a with caproyl chloride gave the amide derivative 8aj.

Scheme III. Preparation of 25 by Method Ba

CH₂Br

CH2-PPh2

 a Reagents: (a) B_2H_6/THF ; (b) $HNMe_2/DMF$; (c) PBr_3/DMF ; (d) $PPh_3/toluene$; (e) 50, EtONa/EtOH; (f) (1) $H_2,\ Pd-C/EtOAc$, (2) pentyl isocyanate/CHCl $_3$.

The N-methyl-N-phenyl-N'-pentylurea derivative 8ak was afforded from 7a by formylation, methylation, deprotection of the formyl group, and treatment with pentyl isocyanate. The other (methoxyphenyl)urea (8ea) was prepared from 4-methoxy-2-nitroaniline by formylation, benzylation, catalytic hydrogenation of the nitro group. methylation, deprotection, and reaction with pentyl isocyanate. Demethylation of the anisole derivatives 8aaak. 8ba. 8ca. 8da, and 8ea was accomplished with BBr₃ to give the corresponding phenol derivatives 9aa-ak, 9ba, 9ca, 9da, and 9ea. The other phenol derivative (9fa) was prepared from 2-amino-3-nitrophenol by O-methoxymethylation, N-dimethylation, catalytic hydrogenation, reaction with pentyl isocyanate, and deprotection. The 4,6dibromo derivative 9ga was afforded by treatment of 9ad with bromine.

The preparation of compounds by method A was accomplished by reaction of the above phenol derivatives 9aa-ak, 9ba, 9ca, 9da, 9ea, 9fa, and 9ga with chloroalkylated heterocycles 5a-l in the presence of potassium carbonate. The amine analogue 19 was prepared from compound 18 by reduction with LiAlH₄.

Preparation by method E was accomplished by treatment of 10, which was afforded by chloropropylation of 3-(dimethylamino)-2-(N'-pentylureido)phenol (9ad), with various heterocycles in the presence of potassium carbonate. In the case of the benzotriazole analogue 33, the position of the substituent was assigned on the basis of the ¹H-NMR spectral data, which showed a symmetrical pattern of the protons of the benzotriazole moiety.

In order to examine the effect of oxygen in the bridging portion of 24, compounds 25–27, listed in Table III, were prepared as illustrated in Schemes III–V. Scheme III shows the preparation of the carbon analogue 25 by method B. Reduction of 3-chloro-2-nitrobenzoic acid with diborane gave the benzyl alcohol derivative 47, which was heated with dimethylamine in an autoclave to provide 3-(dimethylamino)-2-nitrobenzyl alcohol (48). Bromination of 48 with phosphorus tribromide gave the benzyl bromide derivative 49. Formation of the phosphonium salt 50 was accomplished by reaction of 49 with triphenylphosphine. The Wittig reaction of 50 with 3-(5-methyl4-phenyl-1*H*-imidazol-1-yl)propionaldehyde (50) in the presence of sodium ethoxide in EtOH afforded the olefin derivative 51 as a cis-trans mixture. Catalytic hydroge-

Scheme IV. Preparation of 26 by Method Ca

$$\begin{array}{c}
COOH \\
NO_2 \\
CI
\end{array}$$

$$\begin{array}{c}
N(CH_3)_2 \\
NO_2 \\
CI
\end{array}$$

$$\begin{array}{c}
b \\
52
\end{array}$$

^a Reagents: (a) (1) SOCl₂, (2) NaN₃/acetone– H_2O , (3) NaOH, (4) HCl, (5) MeI, K_2CO_3/DMF ; (b) t-BuOK, 5m/DMF; (c) (1) H_2 , Pd–C/EtOAc, (2) pentyl isocyanate/CHCl₃.

Scheme V. Preparation of 27 by Method Da

^a Reagents: (a) (1) HCO₂H, (2) K₂CO₃, NaI, **5a**/DMF, (3) HCl/EtOH, (4) BzlBr, NaH/DMF; (b) HNMe₂/DMF; (c) (1) SnCl₂/HCl, (2) pentyl isocyanate/CHCl₃, (3) H₂, Pd(OH)₂-C/EtOH.

nation of 51 and subsequent treatment with pentyl isocyanate led to the carbon analogue 25.

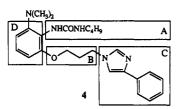
Scheme IV shows the preparation of the sulfur analogue 26 by method C. The Curtius rearrangement of 3-chloro-2-nitrobenzoic acid, hydrolysis of the isocyanate to amine, and methylation afforded the dimethylamine derivative 52. Treatment of 52 with 5-methyl-4-phenyl-1-(3-propyl-thio)imidazole (5m) in the presence of potassium butoxide afforded the intermediate 53. The desired analogue 26 was obtained by catalytic hydrogenation of 53 and subsequent treatment with pentyl isocyanate.

Scheme V shows the preparation of the nitrogen analogue 27 by method D. Formylation of 3-chloro-2-nitroaniline with formic acid, alkylation with 1-(3-chloropropyl)-5-methyl-4-phenylimidazole (5a) in the presence of NaH, deprotection with 36% hydrochloric acid in THF, and benzylation with benzyl bromide gave the intermediate 54. The intermediate 54 was heated with dimethylamine in an autoclave to provide the dimethylamine derivative 55. The nitrogen analogue 27 was obtained by reduction of 55 with SnCl₂, treatment with pentyl isocyanate, and hydrogenolysis of the benzyl group.

Biological Results and Structure-Activity Relationship

We wished to find a compound able to act on both intestinal and aortic ACATs to prevent cholesterol absorption in the intestine and the accumulation of cholesteryl esters in arterial wall. Since Lange et al.³⁹ had suggested the possibility of the existence of isozymes of ACAT, the synthesized compounds were evaluated for in vitro ACAT inhibitory activity according to the method of Heider et al.⁷ using two enzyme sources, rabbit aorta homogenate and rabbit small intestine microsomes. The potency of activity was expressed as the micromolar

Chart I. Four Structural Regions of Lead Compound 4 **Examined in SAR Studies**



concentration of a compound required to inhibit enzyme activity by 50% (IC₅₀).

Selected analogues were then tested for in vivo hypocholesterolemic activity. In vivo serum cholesterol lowering activity was assessed in male Sprague-Dawley rats (5-weeks-old) fed a hypercholesterolemic diet along with the compound administered orally once a day for 2 or 5 days. Effect was expressed as a percentage of the control value. For the compounds listed in Table VII, ED₃₀ values, the mg/kg dose of a compound required to decrease serum cholesterol level to 30% of the control value, were obtained.

To study structure-activity relationships, we divided the parent compound 4 into four regions, as illustrated in Chart I: (A) urea moiety, (B) bridging portion, (C) heterocycle moiety, (D) phenylurea nucleus. In region A, we examined the optimal size of the alkyl group. The results are shown in Table I. When the linear alkyl groups (4 and 11-14) were compared, lengthening the alkyl chain increased in vitro activities. The hexylurea derivative 14 showed the most potent in vitro agric activity, while the pentylurea compound 13 showed the most potent in vitro intestinal activity. Cyclic and branched analogues 15-17 were also evaluated. The in vitro activity was decreased compared with linear analogues. The in vivo hypocholesterolemic activity was increased in the order tert-butyl (17) < cyclopentyl (16) < n-butyl (4) < cyclohexyl (15) < n-butyl (4) < n-butyln-pentyl (13) < n-hexyl (14). The discrepancy between in vitro intestinal inhibitory activity and in vivo hypocholesterolemic effect (13 > 14 for in vitro intestinal inhibition and 13 < 14 for in vivo hypocholesterolemic activity) might be due not only to the difference in species employed (rabbit enzyme in vitro and rat in vivo) but also to many other factors, including absorption and metabolism. The order of in vivo activity correlated with the L lengths of STERIMOL parameters⁴⁰ for alkyl groups of 4 and 13-17. Although this correlation indicates that longer alkyl analogues may have more potent in vivo hypocholesterolemic activity, we also found in a preliminary experiment³⁰ that lengthening the alkyl chain of 4, 13, and 14 reduced the concentration of the compounds in plasma after oral administration; in particular, the plasma level of 14 was much lower than those of 4 and 13. Since our aim was to explore compounds which directly inhibited arterial ACAT after oral administration, we discontinued studies on compounds with alkyl chains longer than hexyl (14), which gave poor plasma levels. Replacement of the urea group (13) with an amide group (18) resulted in almost unchanged in vitro aortic activity, while the in vitro intestinal activity of 18 was about oneseventh as potent as that of 13, and 18 lacked in vivo hypocholesterolemic activity. When an amine group (19) was used, the activity nearly disappeared. Introduction of a methyl group on the nitrogen atom of the urea group (20 and 21) reduced the activity. The in vitro activity of the N-methyl isomer 20 was lower than that of the N'methyl isomer 21.

In order to examine the effect of length and heteroatom in the bridging portion between the phenylimidazole and phenylurea moieties, the compounds listed in Tables II and III were prepared. When the length of the bridging portion (13, 22, and 23) was varied, a propoxy group (13) was found to give the greatest activity. The in vitro intestinal activity of the ethoxy compound 22 was equal to that of the butoxy compound 23, while the in vitro aortic activity of 22 was half that of 23. When oxygen (24) was replaced with carbon (25), sulfur (26), and nitrogen (27), aortic activity was decreased in the order sulfur (26) > oxygen (24) > carbon (25) > nitrogen (27), and intestinal activity was decreased in the order oxygen (24) > sulfur (26) > carbon (25) > nitrogen (27). The in vivo hypocholesterolemic effect of the oxygen analogue 24 was more potent than that of the carbon analogue 25, correlating with the in vitro intestinal activities. Although the activity of the sulfur analogue was equal to that of the oxygen analogue, on the basis of the ease of synthesis, we selected the oxygen analogue.

Next, the heterocyclic ring moiety was examined, since variation of the heterocycle may lead to a change in molecular properties such as basicity or hydrophilicity, and such a change of molecular properties may affect absorption or distribution, which is expected to influence in vivo activity. The results are shown in Table IV.

Although replacement of the imidazole (13) with 1,2,3triazole (28), tetrazole (29), or piperazine (30) left the in vitro activity almost unchanged, it strongly increased the in vivo hypocholesterolemic activity. Piperidine (31) reduced the in vitro activity, while the in vivo activity was maintained. Phenethylamine (32), which is a ring-opened analogue of a heterocycle, reduced both in vitro and in vivo activities. Replacement of phenylated heterocycles (13 and 28) with benzo analogues (33 and 34) reduced the potency. Although compounds 28-31 were all potent in vivo, they showed lower plasma concentrations after oral administration compared with 13 in a preliminary pharmacokinetic study.³⁰ For this reason, the imidazole analogue was selected for further deviation.

Introduction of a substituent on the phenyl group of the phenylimidazole moiety (36 and 37) or replacement of the phenyl group with a naphthyl group (38) decreased in vitro aortic activity, but increased in vivo hypocholesterolemic activity. In particular, compound 37, which is a more lipophilic compound, possessed more potent in vivo hypocholesterolemic activity than 13. However, 37 was not detectable in plasma after oral administration in the preliminary experiment.³⁰ As regards substitution position, the 2-phenylimidazole analogue 35 was less potent than the 4-phenylimidazole analogue (13). Introduction of a methyl group at the 2-position (39) of the imidazole ring of 36 reduced in vivo hypocholesterolemic activity. Introduction at the 5-position (24) of the imidazole ring of 13 strongly increased in vivo hypocholesterolemic activity. When 24 was compared with the regioisomer 40. 24 had slightly more potent activity than 40.

Effects of substituents in the phenylurea nucleus (D region) are shown in Table V. Introduction of substituents (41 and 42) into the phenylurea nucleus and replacement of phenyl with pyrimidine (43) decreased the activity. To examine the effect of the location of the propoxy phenylimidazole moiety on the phenylurea nucleus, 44-46 were synthesized (Table VI). The 3-position derivative (24) was the most potent and the activity decreased in the

Table I. Physical Data and Biological Activities of N-Alkyl-N'-[2-(dimethylamino)-6-[3-(4-phenyl-1H-imidazol-1-yl)propoxy]phenyl]ureas, Amide, and Amine (Variable A Region)

| | | | | | ACAT | inhibn . | |
|-----|---|--------------------|---------|--|--------------------|---------------|-----------------------|
| no. | R | $yield^a$ (method) | mp (°C) | $formula^b$ | aorta ^c | $intestine^d$ | cholesterol loweringe |
| 1 | CL-277082 | - | | - | 0.90 | 0.18 | |
| 11 | $NHCONHC_2H_5$ | 53 (A) | 115-116 | $C_{23}H_{29}N_5O_2$ | >10 | 0.87 | +2 |
| 12 | NHCONHC ₃ H ₇ | 51 (A) | 121-123 | $C_{24}H_{31}N_5O_2$ | 7.23 | 0.66 | \mathbf{NT}^f |
| 4 | NHCONHC ₄ H ₉ | 36 (A) | 98-100 | $C_{25}H_{33}N_5O_2$ | 0.81 | 0.14 | -37 |
| 13 | NHCONHC5H11 | 51 (A) | 110-111 | $C_{26}H_{35}N_5O_2$ | 0.16 | 0.031 | -51 |
| 14 | NHCONHC ₆ H ₁₃ | 48 (A) | 124-126 | $C_{27}H_{37}N_5O_2$ | 0.050 | 0.060 | -57 |
| 15 | NHCONHcyclo-C ₆ H ₁₁ | 40 (A) | 129-131 | $C_{27}H_{35}N_5O_2$ | 1.07 | 0.073 | -42 |
| 16 | NHCONHcyclo-C ₅ H ₉ | 55 (A) | 112-113 | $C_{26}H_{33}N_5O_2$ | 3.53 | 0.24 | -24 |
| 17 | NHCONHtert-C ₄ H ₉ | 51 (A) | 127-128 | $C_{25}H_{33}N_5O_2$ | 0.99 | 0.13 | -12 |
| 18 | $NHCOC_5H_{11}$ | 34 (A) | g | $C_{26}H_{34}N_4O_2^h$ | 0.10 | 0.20 | +3 |
| 19 | NHC_6H_{13} | 91 ⁱ | g | $C_{27}H_{36}N_4O^h$ | 3.3 | 11 | \mathbf{NT}^f |
| 20 | N(CH ₃)CONHC ₅ H ₁₁ | 33 (A) | 108-109 | $C_{27}H_{37}N_5O_2$ | 0.40 | 0.93 | \mathbf{NT}^f |
| 21 | NHCON(CH ₃)C ₅ H ₁₁ | 19 (A) | 90-92 | $\mathbf{C}_{27}\mathbf{H}_{37}\mathbf{N}_5\mathbf{O}_2$ | 0.28 | 0.55 | NT [/] |

^a Yield (%) of final step. ^b Satisfactory elemental analyses were obtained for C, H, N unless otherwise indicated. ^c IC₅₀ (μ M) for the enzyme derived from rabbit anota homogenate. ^d IC₅₀ (μ M) for the enzyme derived from rabbit intestine microsomes. ^e Compounds were administered to rats at 50 mg/kg once a day for 2 days. Cholesterol lowering activity expressed as percentage of control value ($n \ge 4$). ^f Not tested. ^g Compound obtained as an oil. ^h Elemental analysis was not performed. Spectral data were consistent with the indicated structure. See Experimental Section. ⁱ Compound 19 was prepared from 18 by reduction with LiAlH₄ as illustrated in Scheme II.

Table II. Physical Data and Biological Activities of N-[2-(Dimethylamino)-6-(4-phenyl-1H-imidazol-1-yl)-alkoxy]phenyl]-N-pentylureas (Variable B Region)

| | | $vield^a$ | ACAT inhibn | | | |
|-----|---|-----------|-------------|----------------------|--------|---------------|
| no. | n | (method) | mp (°C) | formula ^b | aortac | $intestine^d$ |
| 22 | 2 | 48 (A) | 138-140 | $C_{25}H_{33}N_5O_2$ | 1.4 | 0.17 |
| 13 | 3 | 51 (A) | 110-111 | $C_{26}H_{35}N_5O_2$ | 0.16 | 0.031 |
| 23 | 4 | 51 (A) | 142-144 | $C_{27}H_{37}N_5O_2$ | 0.78 | 0.16 |

a-d See corresponding footnotes of Table I.

order 3-position (24) > 4-position (44) > 5-position (45) > 6-position (46).

On the basis of in vitro ACAT inhibitory activity, in vivo hypocholesterolemic activity, and the results of the preliminary pharmacokinetic study, the lead compound 4 and the more potent derivatives 13, 14, and 24 were selected for in vivo antiatherosclerotic testing. The in vitro ACAT inhibitory activity and in vivo hypocholesterolemic activity of these compounds are summarized in Table VII. In vivo antiatherosclerotic activity (Table VIII) was evaluated in New Zealand white rabbits (NZW) fed an atherogenic diet for 12 weeks by mixing each compound with the diet for the last 4 weeks. The antiatherosclerotic effect was assessed from the development of fatty streaks (surface involvement) and the cholesterol content in the aortic wall. The plasma cholesterol level after administration of 4, 13, 14, and 24 was decreased by 51%, 66%, 62%, and 67%, respectively. Compounds 4, 13, 14, and 24 diminished the surface involvement (in terms of area) by 38%, 40%, 20%, and 45%, respectively, and decreased cholesterol content by 59%, 56%, 23%, and 68%, respectively. Compounds 4, 13, and 24 exhibited significant antiatherosclerotic effect, while 14 did not. Although the reason for the poor antiatherosclerotic effect of 14 is not clear, we observed a reduction in food consumption and body weight gain in 14-treated rabbits. These changes may reflect some toxic effect of 14 and could have contributed to its poor antiatherosclerotic effect.

Conclusion

To develop a novel, potent, and systemically bioavailable ACAT inhibitor, many analogues of compound 4 were synthesized and evaluated as ACAT inhibitors. Structureactivity relationship studies (see Chart I) revealed the following features: (i) In the A region of the molecule, a urea binding pattern was desirable, and introduction of substituents on the nitrogen atom of the urea group reduced the activity. Alkyl groups longer than butyl were necessary for potent activity. It was also found that in vivo hypocholesterolemic activity correlated with the L length of STERIMOL parameters of alkyl groups. (ii) In the B region, the n-propoxy moiety was optimal in length and the oxygen atom could be replaced with carbon or sulfur. (iii) In the C region, 4-phenylimidazole, 4-phenyl-1,2,3-triazole, 5-phenyltetrazole, and 4-phenylpiperaine had potent activity. Among the imidazole derivatives, introduction of a substituent at the para position on the phenyl ring or substitution with a methyl group at the 5-position of the imidazole ring of the 4-phenylimidazole moiety resulted in enhanced in vivo hypocholesterolemic activity. (iv) In the D region, 2,6-disubstituted phenylurea was the optimal nucleus. Introduction of substituents into the phenyl nucleus and replacement of phenyl with pyrimidine decreased the activity. On the basis of these SARs and the biological test data, compound 24 was identified as a potent and orally bioavailable inhibitor of ACAT with hypocholesterolemic and antiatherosclerotic activity and was selected as a candidate for further preclinical studies.

Experimental Section

The purity of each product was checked by thin-layer chromatography (TLC) on silica gel plates (Kieselgel 60 F_{254} , thickness 0.25 mm). Column chromatography was performed

Table III. Physical Data and Biological Activities of N-[2-(Dimethylamino)-6-[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)heteropropyl]phenyl]-N'-pentylureas (Variable B Region)

| | | | | | ACA | T inhibn | |
|-----|--------|-----------------------------|---------|---|--------|------------------------|-----------------------------------|
| no. | X | yield ^a (method) | mp (°C) | $formula^b$ | aortac | intestine ^d | cholesterol lowering ^e |
| 24 | 0 | 69 (A) | 128-130 | C ₂₇ H ₃₇ N ₅ O ₂ | 0.11 | 0.043 | -49 |
| 25 | CH_2 | 57 (B) | 117-119 | $C_{28}H_{39}N_5O$ | 0.14 | 0.17 | -37 |
| 26 | S | 44 (C) | g | $C_{27}H_{37}N_5OS^h$ | 0.058 | 0.056 | \mathbf{NT}^f |
| 27 | NH | 28 (D) | g | $\mathbf{C}_{27}\mathbf{H}_{38}\mathbf{N}_{6}\mathbf{O}^{h}$ | 2.2 | 0.47 | \mathbf{NT}^f |

a-d-/-h See corresponding footnotes of Table I. e Compounds were administered to rats at 10 mg/kg once a day for 5 days. Cholesterol lowering activity expressed as percentage of control value $(n \ge 4)$.

on silica gel (Merck; particle size 0.063-0.200 mm for normal chromatography and 15 μ m for flash chromatography). All melting points (mp) were determined on a Yanagimoto micromelting point apparatus and are uncorrected. 1H-NMR spectra were measured on a JEOL JNM-FX90Q (90 MHz) instrument. Chemical shifts are reported in δ units from tetramethylsilane as an internal standard; coupling constants (J) are reported in hertz. Mass spectra were obtained on a JEOL HX100 mass spectrometer and data are tabulated as m/e. Elemental analyses were performed at the Analytical Chemistry Section of Eisai Tsukuba Research Laboratories.

1-(3-Chloropropyl)-5-methyl-4-phenylimidazole (5a). To a suspension of NaH (55% oil suspension, 50.0 g, 1.14 mol) in DMF was added a solution of 5-methyl-4-phenylimidazole (150 g, 0.95 mol) in DMF at 40-50 °C. The mixture was stirred at 50 °C for 1 h and then added dropwise to a solution of 1-bromo-3-chloropropane in DMF at 50-60 °C over 1 h. The reaction mixture was stirred at 60 °C for 3 h, poured into water, and extracted with EtOAc. The organic layer was dried over MgSO4, and evaporated to remove the solvent. The residue was chromatographed with benzene-acetone (3:1) to give 121 g (54%) of **5a** as an oil: ¹H NMR (CDCl₃) δ 2.16 (2H, quintet, J = 5.6), 2.20 (3H, s), 3.50 (2H, t, J = 5.6), 4.08 (2H, t, J = 5.6), 7.08-7.69 (6H, t, J = 5.6)

1-(3-Chloropropyl)-4-methyl-5-phenylimidazole (5f): 1H NMR (CDCl₃) δ 1.88 (2H, quintet, J = 5.6), 2.20 (3H, s), 3.35 (2H, t, J = 5.6), 4.07 (2H, t, J = 5.6), 7.23-7.52 (6H, m).

5-Methyl-4-phenyl-1-(3-propylthio)imidazole (5m). A solution of 1-(3-chloropropyl)-5-methyl-4-phenylimidazole (5a, 5.00 g, 21.3 mmol) and potassium thiobenzoate (3.75 g, 21.3 mmol) in DMF was stirred at 60 °C for 1 h. The mixture was poured into water and extracted with EtOAc. The organic layer was washed with 1 N NaOH, dried over MgSO4, and evaporated. The residue was purified by column chromatography (benzeneacetone 2:1) to give 4.9 g (68%) of 5-methyl-4-phenyl-1-(3thiobenzoylpropyl)imidazole as an oil.

A mixture of the oil (1.26 g, 3.75 mmol) and LiAlH₄ (142 mg, 3.74 mmol) in THF was stirred for 1 h, and then water (0.14 mL). 15% NaOH (0.14 mL), and water (0.42 mL) were added consecutively at 0 °C. The resulting precipitate was filtered off and the filtrate was evaporated in vacuo to give 5m as an oil. The product was used without further purification.

3-(5-Methyl-4-phenyl-1H-imidazol-1-yl)propionaldehyde (50). 3-(5-Methyl-4-phenyl-1*H*-imidazol-1-yl)propionaldehyde diethyl acetal (5n) was prepared by the same procedure as reported for 5a. A solution of 5n (5.00 g, 17.4 mmol) and 7 N HCl (10 mL) in THF was stirred at room temperature for 1 h. The mixture was poured into water, neutralized with saturated sodium hydrogen carbonate, and extracted with EtOAc. The organic layer was dried over MgSO₄ and evaporated to give 890 mg (24%) of 50 as an oil, which was used without further purification.

2-(Dimethylamino)-6-(pentylureido)phenol (9ad). A mixture of 3-methoxy-2-nitrobenzoic acid (98.6 g, 0.50 mol) and thionyl chloride (500 mL) was heated under reflux for 3 h. The excess thionyl chloride was removed under reduced pressure and the residue was washed with hexane to give 3-methoxy-2nitrobenzoyl chloride.

A solution of the acid chloride in acetone (500 mL) was added dropwise to a solution of sodium azide (40.0 g, 0.60 mol) in water (290 mL) at below 10 °C over 1 h. The mixture was stirred at room temperature for 30 min and poured into water (1 L). The resulting precipitate was collected and washed with water (500 $mL \times 2$) to give 3-methoxy-2-nitrobenzoyl azide.

The benzoyl azide was refluxed in benzene to give 3-methoxy-2-nitrophenyl isocyanate until nitrogen gas evolution ceased. The reaction mixture was cooled, 5 N NaOH (200 mL) was added at 0 °C, and the whole was stirred at room temperature for 1 h. Then 10 N HCl (200 mL) was added to the solution at 0 °C, and the reaction mixture was heated at 80 °C for 2 h. The benzene layer was separated, and the water layer was extracted with EtOAc. The combined organic layer was washed with brine, dried over MgSO₄, and evaporated. The resulting solid was recrystallized from EtOAc-hexane to give 63.8 g (76%) of 3-methoxy-2-nitroaniline.

To a suspension of the aniline derivative (110 g, 0.65 mmol) and potassium carbonate (455 g, 3.2 mol) in DMF was added methyl iodide (410 mL, 6.5 mol) while stirring. The mixture was stirred for 24 h at 50 °C. Water was then added, and the mixture was extracted with EtOAc. The organic layer was washed with brine, dried over MgSO₄, and evaporated to give 88.3 g (69%) of 3-methoxy-2-nitrodimethylaniline (6a).

A solution of 6a (88.3 g, 0.45 mol) in EtOAc was hydrogenated over 10% palladium on carbon (1.5 g) at 1 atm for 1 h. The catalyst was filtered off and the filtrate was evaporated in vacuo to give 74.6 g (99%) of 2-(dimethylamino)-6-methoxyaniline (7a).

A solution of 7a (74.6 g, 0.45 mol) and pentyl isocyanate (51 g, 0.45 mol) in CHCl₃ was heated under reflux for 24 h. The solvent was removed in vacuo. The solid residue was recrystallized from EtOAc to give 124 g (99%) of N-[2-(dimethylamino)-6-methoxyphenyl]-N'-pentylurea (8ad).

To a solution of 8ad (81.3 g, 0.29 mol) in CH₂Cl₂ was added BBr₃ (2 M CH₂Cl₂ solution, 370 mL, 0.74 mol) at 0 °C. The mixture was heated under reflux for 2 h and then poured into water, neutralized with sodium hydrogen carbonate, and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO₄, and evaporated to give 69.8 g (91%) of 9ad as an amorphous solid. The compound was used without further purification: ¹H NMR (CDCl₃) δ 0.90 (3H, t, J = 7.5), 1.15–1.60 (6H, m), 2.62 (6H, s), 3.30 (2H, q, J = 7.5), 5.00 (1H, brs), 6.62 (1H, dd, J = 8.0, 1.5), 6.71 (1H, dd, J = 8.0, 1.5), 6.98 (1H, t, J)= 8.0), 7.62 (1H, brs), 10.0 (1H, s).

Method A. N-[2-(Dimethylamino)-6-[3-(5-methyl-4-phenyl-1*H*-imidazol-1-yl)propoxy]phenyl]-*N*-pentylurea (24). A solution of 1-(3-chloropropyl)-5-methyl-4-phenylimidazole (5a, 8.84 g, 37.7 mmol) in DMF was added dropwise to a suspension of 2-(dimethylamino)-6-(pentylureido)phenol (9ad, 10.0 g, 37.7 mmol), sodium iodide (100 mg), and potassium carbonate (10.4 g, 76.5 mmol) in DMF at 60 °C over 2 h. The mixture was stirred at 60 °C for 2 h. The reaction mixture was poured into water and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO₄, and evaporated. The residue was purified by silica gel column chromatography (benzene-acetone

Table IV. Physical Data and Biological Activities of N-[2-(Dimethylamino)-6-[3-(heterocycle)propoxy]phenyl]-N'-pentylureas (Variable C Region)

| | | | - 0 | Het | ACA | T inhibn | |
|-----|----------------------------|-----------------------------|----------------------|------------------------|--------|---------------|-----------------------------------|
| no. | Het | yield ^a (method) | mp (°C) | $formula^b$ | aortac | $intestine^d$ | cholesterol lowering ^e |
| 13 | -N_N Ph | 51 (A) | 110–111 | $C_{26}H_{35}N_5O_2$ | 0.16 | 0.031 | -21 |
| 28 | -NN-NPh | 49 (A) | 12 9–1 31 | $C_{25}H_{34}N_6O_2$ | 0.18 | 0.032 | -32 |
| 29 | -N, N N | 55 (A) | 146–147 | $C_{24}H_{33}N_7O_2\\$ | 0.23 | 0.023 | -48 |
| 30 | -N $N-Ph$ | 37 (E) | 148-150 | $C_{27}H_{41}N_5O_2\\$ | 0.18 | 0.017 | -44 |
| 31 | -N $-Ph$ | 66 (E) | 12 9- 131 | $C_{28}H_{42}N_4O_2$ | 0.47 | 0.34 | -27 |
| 32 | $-N$ p_h | 30 (E) | 171-173 | $C_{25}H_{38}N_4O_2$ | 3.44 | 0.96 | -18 |
| 33 | -N,N | 41 (E) | 1 49- 151 | $C_{23}H_{32}N_6O_2$ | 0.29 | 0.77 | -3.4 |
| 34 | -N | 35 (E) | 137-139 | $C_{24}H_{33}N_5O_2$ | 0.62 | 0.18 | -12 |
| 35 | N Ph | 31 (A) | 125–126 | $C_{26}H_{35}N_5O_2$ | 0.48 | 0.38 | -12/ |
| 36 | -N [↑] N OMe | 41 (A) | 98-101 | $C_{27}H_{37}N_5O_3$ | 1.33 | 0.059 | -36 |
| 37 | -N^N Bu | 22 (A) | 111-114 | $C_{30}H_{43}N_5O_2$ | 0.43 | 0.054 | -46 |
| 38 | -N_N | 33 (A) | 71–75 | $C_{30}H_{37}N_5O_2$ | 1.55 | 0.065 | -30 |
| 39 | -N N OMe | 53 (A) | 140–143 | $C_{28}H_{39}N_5O_3$ | 0.73 | 0.16 | -15 |
| 24 | Ne Ph | 69 (A) | 128-130 | $C_{27}H_{37}N_5O_2$ | 0.11 | 0.043 | -49 |
| 40 | -N ^N N Ph Me | 45 (A) | 119–121 | $C_{27}H_{37}N_5O_2$ | 0.21 | 0.085 | -46 |

a-d See corresponding footnotes of Table I. ^e Compounds were administered to rats at 10 mg/kg once a day for 5 days, unless otherwise indicated. Cholesterol lowering activity expressed as percentage of control value ($n \ge 4$). ^f The compound was administered to rats at 30 mg/kg once a day for 5 days.

2:1), and recrystallization from EtOAc–hexane gave 12.0 g (69%) of **24**: mp 128–130 °C; ^1H NMR (CDCl $_3$) δ 0.90 (3H, t, J=6.0), 1.16–1.71 (6H, m), 2.21 (2H, quintet, J=5.0), 2.28 (3H, s), 2.78 (6H, s), 3.19 (2H, q, J=5.0), 3.95 (2H, t, J=6.0), 4.13 (2H, t, J=6.0), 6.19 (1H, brs), 6.70 (1H, brt), 6.40–7.60 (9H, m). Anal. (C $_{27}\text{H}_{37}\text{N}_5\text{O}_2$) C, H, N.

1-[*N*-[2-(Dimethylamino)-6-[3-(4-phenyl-1*H*-imidazol-1-yl)propoxy]phenyl]carbamoyl]pentane (18): MS m/e (FAB) 435 (M + H)+; ¹H NMR (CDCl₃) δ 0.90 (3H, t, J = 6.0), 1.08–1.90

(6H, m), 2.19 (2H, quintet, J = 7.0), 2.63 (6H, s), 3.36 (2H, t, J = 6.5), 3.89 (2H, t, J = 7.0), 4.11 (2H, t, J = 7.0), 6.41-7.80 (11H, m).

2-(Hexylamino)-3-[3-(4-phenyl-1*H*-imidazol-1-yl)propoxy]dimethylaniline (19). To a solution of 18 (330 mg, 0.76 mmol) in THF was slowly added LiAlH₄ (40 mg). The mixture was stirred for 1 h at room temperature, and 1 N NaOH solution was added to it. The resulting precipitate was filtered off and the filtrate was evaporated in vacuo. The residue was purified by column chromatography (benzene-acetone 4:1) to give

Table V. Physical Data and Biological Activities of N-[2-(Dimethylamino)-6-[3-(4-phenyl-1H-imidazol-1-yl)propoxy]aryl]-N'-pentylureas (Variable D Region)

| | | | | | | | | ACA | T inhibn |
|-----|------------------------|---------|----------------|-----------------|-----------------|----------------------|--------------------------|--------|------------------------|
| no. | X | Y | \mathbf{R}_1 | \mathbf{R}_2 | yielda (method) | mp (°C) | $formula^b$ | aortac | intestine ^d |
| 13 | CH | CH | Ph | H | 51 (A) | 110-111 | $C_{26}H_{35}N_5O_2$ | 0.16 | 0.031 |
| 41 | $\mathbf{C}\mathbf{H}$ | CCH_3 | Ph | H | 50 (A) | 11 9- 121 | $C_{27}H_{37}N_5O_2$ | 0.25 | 0.41 |
| 42 | \mathbf{CBr} | CBr | 4-OCH₃Ph | Ħ | 65 (A) | 124-126 | $C_{27}H_{35}N_5O_3Br_2$ | 1.3 | 2.4 |
| 43 | N | N | Ph | \mathbf{CH}_3 | 17 (A) | e | $C_{25}H_{35}N_7O_2$ | 11 | 5.5 |

a-d See corresponding footnotes of Table I. e See footnote g of Table I. See footnote h of Table I.

Table VI. Physical Data and Biological Activities of 24 and the Regioisomers (Variable D Region)

| | | $vield^a$ | , | | ACA | T inhibn |
|-----|----------|-----------|---------|---|--------------------|---------------|
| no. | position | (method) | mp (°C) | $formula^b$ | aorta ^c | $intestine^d$ |
| 24 | 3 | 69 (A) | 128-130 | C ₂₇ H ₃₇ N ₅ O ₂ | 0.11 | 0.043 |
| 44 | 4 | 6 (A) | | $C_{27}H_{37}N_5O_2$ | 0.41 | 0.41 |
| 45 | 5 | 22 (A) | 125-126 | $C_{27}H_{37}N_5O_2$ | 1.5 | 2.4 |
| 46 | 6 | 25 (A) | 178-180 | $C_{27}H_{37}N_5O_2$ | 3.2 | 5.8 |

a-d See corresponding footnotes of Table I.

Table VII. Biological Activities of Selected Compounds

| | | | ACAT inhibn | | hypocholesterolemic activity |
|-----|-----------------|---|-------------|------------------------|---------------------------------------|
| no. | R | n | aortaa | intestine ^b | $(\mathbf{ED}_{30},\mathbf{mg/kg})^c$ |
| 4 | Н | 3 | 0.81 | 0.14 | 75 |
| 13 | H | 4 | 0.16 | 0.031 | 14 |
| 14 | H | 5 | 0.050 | 0.060 | 1.3 |
| 24 | \mathbf{CH}_3 | 4 | 0.11 | 0.043 | 4.5 |

^a See footnote c of Table I. ^b See footnote d of Table I. ^c The values are the effective dose required to decrease serum total cholesterol to 30% of the control value, calculated from the dose-response curve derived from experiments with at least 3 doses (n = 4 or 5). Each compound was administered orally as a suspension in 2% Sefuzol for 5 days.

300 mg (91%) of 19: MS m/e (FAB) 421 (M + H)+; ¹H NMR $(CDCl_3) \delta 0.90 (3H, t, J = 6.5), 1.25-1.47 (8H, m), 2.19 (2H, quintet,$ J = 7.0), 2.63 (6H, s), 3.30 (2H, q, J = 6.5), 3.89 (2H, t, J = 7.0), 4.11 (2H, t, J = 7.0), 6.41-7.80 (11H, m).

4-(Dimethylamino)-6-[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propoxy]-5-(N-pentylureido)pyrimidine (43): MS m/e(FAB) 466 (M + H)⁺; ¹H NMR (CDCl₃) δ 0.87 (3H, t, J = 6.2), 1.12-1.78 (6H, m), 2.19 (2H, quintet, J = 6.0), 2.33 (3H, s), 3.10(6H, s), 3.17 (2H, q, J = 6.8), 3.83 (2H, t, J = 6.0), 3.95 (2H, t, J = 6.0)J = 6.0), 5.58 (1H, brt), 6.65 (1H, brs), 7.21-7.71 (7H, m).

Method B. N-[2-(Dimethylamino)-6-[4-(5-methyl-4-phenyl-1H-imidazol-1-yl)butyl]phenyl]-N'-pentylurea (25). To a solution of 3-chloro-2-nitrobenzoic acid (25.0 g, 124 mmol) in THF was added a solution of B₂H₆ (1 M THF solution, 300 mL) at 0 °C. The mixture was stirred at room temperature for 2 h, then MeOH was added until gas evolution ceased. After 10 min, the mixture was poured into water and extracted with EtOAc. The organic layer was washed with aqueous sodium hydrogen

Table VIII. Effect of Selected Compounds on Plasma Cholesterol Level, Atheromatous Lesion Development, and Esterified Cholesterol Content in New Zealand White Rabbits Fed Cholesterol-Containing Atherogenic Diet

| | · | plasma total | aortic atheromate | ous lesion index |
|-----------------|----------|----------------------------------|-----------------------|------------------------|
| | | cholesterol ^c (mg/dL) | % surface involvement | cholesterol content |
| control | (n = 12) | 2428 ± 113 | 38.2 ± 5.0 | 32.9 ± 3.5 |
| 4 a | (n = 9) | 1192 ± 127^d | 23.5 ± 3.2^{g} | $13.6 	 2.7^d$ |
| 13^{a} | (n = 8) | 830 ± 76^{d} | 22.8 ± 4.0^{g} | 14.4 ± 3.3^{e} |
| 14a | (n=7) | 924 ± 151^d | 30.6 ± 5.8 | 25.2 ± 8.7 |
| 24 ^b | (n=9) | 800 ± 84^d | 21.0 ± 1.9 | 10.4 ± 2.1^d |

^a Compounds 4, 13, and 14 were administered at 0.2% (w/w) for the first 4 days of drug treatment, and then the dosage was changed to 0.1% (w/w). b Compound was dosed at 0.1% (w/w). c Values are mean
SEM. Statistical comparisons against control values were performed by means of Student's t test. d p < 0.001. e p < 0.01. / p< 0.02. $^g p < 0.05$.

carbonate and brine, dried over MgSO4, and evaporated to give 20.5 g (88%) of 3-chloro-2-nitrobenzyl alcohol (47).

A mixture of 47 (20.0 g, 107 mmol) and dimethylamine (50% water solution, 50 mL) in DMF was heated at 120 °C in an autoclave for 4 h. The reaction mixture was evaporated to give a brown oil. This crude oil was purified by silica gel column chromatography (hexane-EtOAc 4:1) to give 15.9 g (76%) of 3-(dimethylamino)-2-nitrobenzyl alcohol (48).

To a solution of 48 (15.9 g, 81.3 mmol) in DMF (20 mL) was slowly added phosphorus tribromide (2.82 mL) at 0 °C for 30 min. The mixture was stirred at room temperature for 1 h and then poured into water. The mixture was neutralized with 1 N NaOH and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO₄, and evaporated. The residue was chromatographed on silica gel with hexane-EtOAc (8:1) to give 10.3 g (49%) of 3-(dimethylamino)-2-nitrobenzyl bromide (49).

A mixture of 49 (10.3 g, 39.8 mmol) and triphenylphosphine (11.4 g, 43.7 mmol) in toluene (200 mL) was heated under reflux for 1 h. The mixture was cooled to room temperature and allowed to stand for 5 h. The resulting precipitate was collected by suction filtration, washed with cold toluene, and dried in vacuo to give 20.0 g (96%) of [3-(dimethylamino)-2-nitrobenzyl]triphenylphosphonium bromide (50).

To a solution of 50 (2.17 g, 4.17 mmol) and 3-(5-methyl-4phenyl-1H-imidazol-1-yl)propionaldehyde (50, 890 mg, 4.17 mmol) in EtOH was added a solution of sodium ethoxide (21 wt% ethanol solution, 1.3 mL). The mixture was stirred under reflux for 2 h, and then the dark mixture was cooled and concentrated in vacuo. The residue was diluted with Et₂O (100 mL) and extracted with 2 N HCl (50 mL × 2). The combined aqueous layer was basified with 2 N NaOH and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO₄, and evaporated. The residue was purified by column chromatography (benzene-acetone 3:1) to give 980 mg (63%) of 3-[4-(5-methyl-4-phenyl-1H-imidazol-1-yl)-1-butenyl]-2-nitro-N,N-dimethylaniline (51).

A solution of 51 (980 mg, 2.61 mmol) in EtOAc was hydrogenated over 10% palladium on carbon at 5 kg/cm^2 for 4 h. The catalyst was filtered off and the filtrate was evaporated to give 880 mg (97%) of 2-(dimethylamino)-6-[4-(5-methyl-4-phenyl-1H-imidazol-1-yl)butyl]aniline, which was used without further purification.

A mixture of the aniline derivative (880 mg, 2.53 mmol) and pentyl isocyanate (430 mg, 3.81 mmol) in CHCl₃ was refluxed for 10 h. The excess pentyl isocyanate and solvent were removed under reduced pressure. The residue was purified by column chromatography (benzene-acetone 3:1), and the resulting solid was recrystallized from EtOAc-hexane to give 670 mg (57%) of 25: mp 117-119 °C; ¹H NMR (CDCl₃) δ 0.87 (3H, t, J = 6.8), 1.26 (4H, m), 1.43 (2H, quintet, J = 7.2), 1.63 (2H, quintet, J = 7.2), 1.79 (2H, quintet, J = 7.2), 2.37 (3H, s), 2.66 (2H, t, J = 7.2), 2.68 (6H, s), 3.18 (2H, q, J = 6.8), 3.86 (2H, t, J = 7.2), 5.24 (1H, brt), 6.00 (1H, brs), 6.86-7.65 (9H, m). Anal. ($C_{28}H_{39}N_5O$) C, H, N.

Method C. N-[2-(Dimethylamino)-6-[3-(5-methyl-4-phen-yl-1*H*-imidazol-1-yl)propylthio]phenyl]-N-pentylurea (26). 3-Chloro-2-nitro-N,N-dimethylaniline (52) was prepared by the same procedure as reported for 3-methoxy-2-nitro-N,N-dimethylaniline (6a).

A mixture of 52 (1.97 g, 9.82 mmol), 5-methyl-4-phenyl-1-(3-propylthio)imidazole (5m, 2.28 g, 9.83 mmol), and potassium tert-butoxide (1.21 g, 10.8 mmol) was stirred at 50 °C for 1 h. The mixture was poured into water and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO₄, and evaporated to remove the solvent. The residue was purified by column chromatography (benzene-acetone 2:1) to give 690 mg (18%) of 3-[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propylthio]-2-nitro-N,N-dimethylaniline (53).

A solution of 53 (690 mg, 1.74 mmol) in EtOAc was hydrogenated over 10% palladium on carbon (1.5 g) at 1 atm for 5 h. The catalyst was filtered off and the filtrate was evaporated to give 190 mg (30%) of 2-(dimethylamino)-6-[[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propyl]thio]aniline.

A solution of the above aniline derivative (190 mg, 0.52 mmol) and pentyl isocyanate (88 mg, 0.78 mmol) in CHCl₃ was heated under reflux for 10 h. The mixture was evaporated under reducing conditions to remove the excess reagent and solvent. The residue was chromatographed on silica gel (benzene-acetone 1:1) to give 110 mg (44%) of **26** as an oil: MS m/e (FAB) 480 (M + H)⁺; ¹H NMR δ 0.89 (3H, t, J = 6.4), 1.29 (4H, m), 1.48 (2H, m), 2.20 (2H, quintet, J = 6.8), 2.36 (3H, s), 2.77 (6H, s), 2.86 (2H, t, J = 6.8), 3.21 (2H, q, J = 6.8), 4.02 (2H, t, J = 6.8), 6.34 (1H, brs), 6.47 (1H, brs), 6.97–7.64 (9H, m).

Method D. N-[2-(Dimethylamino)-6-[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propylamino]phenyl]-N-pentylurea (27). A mixture of 3-chloro-2-nitroaniline (6.73 g, 39.0 mmol) and formic acid (50 mL) was heated under reflux for 1 h. The excess formic acid was removed in vacuo, and the precipitate was recrystallized from Et₂O-hexane to give 6.33 g (81%) of 3-chloro-2-nitro-N-formylaniline.

A mixture of the formylaniline derivative (6.33 g, 31.6 mmol), potassium carbonate (8.71 g), sodium iodide (100 mg), and 5a (6.96 g, 31.6 mmol) in DMF was stirred at 60 °C for 3 h. The reaction mixture was cooled and diluted with water until all solids were dissolved. The mixture was extracted with EtOAc. The organic layer was washed with brine, dried over MgSO₄, and evaporated. The residue was purified by flash chromatography (benzene-acetone 1:1 and 1:2) to give 6.13 g (50%) of 3-chloro-2-nitro-N-formyl-N-[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propyllaniline.

A solution of the above compound (4.78 g, 12.43 mmol) and 2 N HCl (20 mL) in 1,4-dioxane (10 mL) was heated under reflux for 30 min. The mixture was poured into ice water, basified with 1 N NaOH, and extracted with EtOAc. The organic layer was dried over MgSO₄ and evaporated to give 4.43 g (100%) of 3-chloro-2-nitro-N-[3-(5-methyl-4-phenyl-1*H*-imidazol-1-yl)propyl]aniline as an oil.

A solution of the above oil (4.43 g, 12.4 mmol) and benzyl bromide (2.4 g, 14.0 mmol) in DMF was added dropwise to a suspension of NaH (55% oil suspension, 650 mg, 14.9 mmol) in DMF at 0 °C over 1 h. The mixture was stirred at room temperature for 1 h, poured into water, and extracted with EtOAc. The organic layer was dried over MgSO₄, and evaporated to give 5.55 g (100%) of 3-chloro-2-nitro-N-benzyl-N-[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propyl]aniline (54).

A solution of 54 (5.55 g, 12.4 mmol) and dimethylamine (50% water solution, 22.4 mL) in DMF was heated at 200 °C for 3 h in an autoclave. The reaction mixture was cooled to room temperature, poured into water, and extracted with EtOAc. The organic layer was dried over MgSO₄, and evaporated to remove the solvent. The residue was purified by column chromatography (benzene-acetone 3:1) to give 2.20 g (39%) of 3-(dimethylamino)-2-nitro-N-benzyl-N-[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propyllaniline (55).

To a solution of 55 (2.20 g, 5.60 mmol) in 36% HCl (10 mL) was added $SnCl_2$ · H_2O (3.79 g) at 0 °C. The reaction mixture was stirred at room temperature for 1 h, poured into ice water, basified with 5 N NaOH, and extracted with CHCl₃. The organic layer was dried over MgSO₄, and evaportaed to give 880 mg (37%) of 2-[N-benzyl-N-[[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propyl]-amino]]-6-(dimethylamino)aniline.

A solution of the above aniline derivative (800 mg, 2.20 mmol) and pentyl isocyanate (250 mg) in CHCl₃ was refluxed for 5 h and then evaporated to remove excess reagent and solvent. The residue was chromatographed on silica gel with benzene—acetone (1:1) to give 470 mg (45%) of N-[2-[N-benzyl-N-[3-(5-methyl-4-phenyl-1H-imidazol-1-yl)propyl]amino]-6-(dimethylamino)-phenyl]-N-pentylurea.

A solution of the above compound (470 mg, 0.87 mmol) in EtOH was hydrogenated over 20% palladium hydroxide on carbon at 5 kg/cm² for 2 h. The catalyst was filtered off and the filtrate was evaporated. The residue was purified by column chromatography (CH₂Cl₂-MeOH 10:1) to give 110 mg (28%) of 27 as an oil. MS m/e (FAB) 463 (M + H)+; ¹H NMR δ 0.87 (3H, t, J=6.5), 1.20-1.54 (6H, m), 2.10 (2H, quintet, J=7.2), 2.39 (3H, s), 2.68 (6H, s), 3.18 (2H, q, J=7.3), 3.23 (2H, t, J=7.2), 4.05 (2H, t, J=7.2), 4.78 (1H, brt), 5.00 (1H, brs), 6.16 (1H, brt), 6.39 (1H, dd, J=7.6, 1.0), 6.49 (1H, dd, J=7.6, 1.0), 7.08 (1H, t, J=7.6), 7.26 (1H, t, J=4.8), 7.40 (2H, t, J=8.4), 7.64 (2H, dd, J=8.4, 1.2).

Method E. N-[2-(Dimethylamino)-6-[3-(4-phenylpiperazine)propoxy]phenyl]-N-pentylurea (30). A mixture of 3-(dimethylamino)-2-(N-pentylureido)phenol (9ad, 5.00 g, 18.9 mmol), 1-bromo-3-chloropropane (4.46 g), and potassium carbonate (6.5 g) in DMF was stirred at 50 °C for 3 h. The mixture was poured into water, and the resulting precipitate was collected by suction filtration. This product was recrystallized from EtOAc to give 5.73 g (89%) of N-[2-(3-chloropropoxy)-6-(dimethylamino)phenyl]-N-pentylurea (10).

To a suspension of 10 (1.00 g, 3.03 mmol), potassium carbonate (810 mg, 5.79 mmol), and sodium iodide (100 mg, 0.667 mmol) in DMF was added a solution of 1-phenylpiperazine (490 mg, 3.03 mmol) in DMF at 60 °C. The mixture was stirred at 60 °C for 2 h, poured into water, and extracted with EtOAc. The organic layer was washed with brine, dried over MgSO₄, and evaporated. The residue was crystallized from benzene-hexane and the product was recrystallized from EtOAc to give 500 mg (37%) of 30: mp 148-150 °C; ¹H NMR (CDCl₃) δ 0.91 (3H, t, J = 6.0), 1.15-1.74 (6H, m), 2.01 (2H, quintet, J = 6.5), 2.58 (2H, t, J = 6.5), 2.64 (4H, m), 3.20 (6H, m), 4.09 (2H, t, J = 6.5), 6.34 (1H, brs), 6.51-7.38 (9H, m). Anal. ($C_{27}H_{41}N_5O_2$) C, H, N.

Assay of ACAT Activity. The intestinal microsomes were prepared according to the method of Field et al.⁴ Briefly, the intestinal mucosa of an NZW rabbit was homogenized in a buffered sucrose solution (0.1 N sucrose, 0.05 M KCl, 0.04 M KH₂PO₄, 0.03 M EDTA, pH 7.4) in a glass Dounce homogenizer. The whole homogenate was spun at 10000g for 20 min at 4 °C. The resulting supernatant was centrifuged at 105000g for 1 h at 4 °C. The microsomal pellet thus obtained was rehomogenized in cold buffered sucrose solution and used as intestinal ACAT.

Arterial ACAT was prepared as follows. The aortae of cholesterol-fed NZW rabbits were homogenized in 154 mM potassium phosphate buffer (pH 7.4) containing 2.4 mg/mL BSA. The whole homogenate was spun at 300g for 5 min to remove cellular debris. The resulting supernatant was used as arterial ACAT.

ACAT activity was determined essentially by the method described by Heider et al. Endogenous cholesterol (of microsomal fraction or homogenate) and exogenous [1-14C]oleoyl-CoA were used as the substrates. The reaction mixture for intestinal ACAT consisted of 250 μ L of K₂HPO₄ buffer (pH 7.4) containing

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2.4 mg/mL BSA and 40 μ M [1-14C]oleoyl-CoA, and the reaction mixture for arterial ACAT consisted of 250 μ L of the same buffer containing 2.4 mg/mL BSA and 5 μ M [1-14C]oleoyl-CoA. Then. $30 \mu L$ of test compound solution was added, and the mixture was preincubated at 37 °C for 3 min before the addition of 20 μ L of the enzyme (8 mg protein/mL). The reaction mixture was incubated for a certain time (5 min for aorta and 2 min for intestine), and then the reaction was terminated by the addition of 3 mL of chloroform/methanol (2:1; v/v) and 500 μ L of 0.04 N HCl. After shaking, the chloroform phase was taken to dryness and the residue was separated by thin-layer chromatography on plastic sheets (TLC plastic sheets, silica gel 60, Merck Co.,) using heptane-diethyl ether-acetic acid (90:30:1; v/v/v). Cholesteryl oleate was visualized with iodine vapor, and the cholesterol ester zone was cut out and placed directly in a scintillation vial for counting. The values are expressed as percent inhibition or IC₅₀ values. Calculation of IC₅₀ values was performed with data from triplicate assay tubes at each drug concentration.

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Serum Cholesterol Level. Male Sprague-Dawley rats (5weeks-old) were fed a standard rat chow, MF (Oriental Yeast Co., Ltd., Tokyo, Japan), containing 1% cholesterol and 0.5% cholic acid. At the same time, the drug suspended in 2% Sefuzol (Nikko Chemical, Tokyo, Japan) was administered orally once a day. Rats were starved for 5 h after the last administration and bled via the abdominal aorta while under ether anesthesia. Serum concentration of total cholesterol was determined enzymatically using the Iatrolipo TC kit (Iatron Lab., Tokyo, Japan). After the last day of the test, the control rat serum cholesterol levels were in the range of 180-230 mg/dL, while the normal rat serum cholesterol levels were in the range of 50-60 mg/dL. The values Heider, J. G.; Pickens, C. E.; Kelly, L. A. Role of Acyl-CoA: Cholesterol Acyltransferase in Cholesterol Absorption and Its Inhibition by 57-118 in the Rabbit. J. Lipid Res. 1983, 24, 1127-

Cholesterol-Fed Rabbits. Male New Zealand white rabbits

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weighing 2.5-3.0 kg were fed standard chow (Oriental, ORC-400) supplemented with 0.5% cholesterol and 0.5% olive oil. All rabbits received 100 g of food/day and water ad libitum during the course of the study. After 8 weeks, the rabbits were randomized to standardize plasma cholesterol levels. The grouped animals were placed on the following diets: control = cholesterol-containing diet with 0.5% olive oil; drug treatment = control diet with a test compound. The dosage in the drugtreated rabbits during the subsequent 4 weeks of the study was 30 mg/kg per day. Blood samples were obtained from the ear vein in the presence of EDTA, and plasma total cholesterol level was measured enzymatically using Iatrolipo TC kit (Iatron Lab., Tokyo, Japan). At the termination of the experiment, the rabbits were anesthetized with pentobarbital and exsanguinated from the aortic arch, and the thoracic aorta was separated at 5 mm proximal to the outlet of the first intercostal artery. These tissues were then longitudinally cut into two halves of about equal size. One half was fixed in buffered formalin and stained with Sudan IV. The positively stained areas were measured with an image analyzer (Nachet). Values are mean \pm SEM percent of 7-12 animals per group. The other half was used for the determination of cholesterol content by use of a TLC-FID analyzer (Iatron). Values are mean \pm SEM mg/g of wet tissue.

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