

Stepwise Palladium-Catalyzed 1,4-Addition of Arylboronic Acids to Enones and Regioselective Baeyer–Villiger Oxidation for Enantioselective Synthesis of β -Diaryl Esters and (+)-(R)-Tolterodine

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Baeyer–Villiger oxidation of chiral β -diaryl ketones synthesized by 1,4-addition of arylboronic acids to α -aryl- α , β -unsaturated ketones catalyzed by a palladium(2+)-chiraphos complex provided optically active β -diaryl esters up to 98% ee. The protocol was applied to the synthesis of a potent competitive muscarinic receptor antagonist, (R)-tolterodine (**21**), which has a chiral center consisting of two aryl rings.

Enantioselective conjugate additions of nucleophiles to α , β -unsaturated carbonyl compounds are a versatile methodology for forming chiral carbon–carbon bonds.¹ We have reported that aryl- and 1-alkenylboronic acids undergo 1,4-addition to α , β -unsaturated carbonyl compounds in the presence of a rhodium(I) catalyst.² The protocol has been proved to be a general reaction for a wide range of selective carbon–carbon bond formations including enantioselective reactions using chiral rhodium–phosphine catalysts.³ We have also reported that dicationic palladium(II) complexes are excellent catalysts that allow 1,4-additions of organoboron,⁴ -silicon,⁵ and -bismuth⁶ compounds at temperatures lower than room temperature. Palladium(2+) complexes possessing bisphosphines bridged by two carbons, such as chiraphos (2,3-bis(diphenylphosphino)butane),^{6,7} dipamp (1,2-bis[(2-methoxyphenyl)(phenyl)phosphino]ethane),⁷ and Me-Duphos (1,2-bis(2,5-dimethylphospholano)benzene),⁸ were found to be effective for asymmetric versions of compounds of those elements. Among these three catalysts, the palladium–chiraphos complex exhibited high enantioselectivities for 1,4-additions of arylboronic acids to unsaturated ketones,^{7a,7b,7d,7f} aldehydes,^{7c} and *N*-acyl amides^{7e} having an aryl ring at the β -carbon. Chiral β -diaryl carbonyl compounds thus obtained are key intermediates in the syntheses of a potent competitive muscarinic receptor antagonist (R)-tolterodine,⁹ a potential therapeutic agent (+)-(R)-CDP 840,^{7c,10} and endothelin receptor antagonist.¹¹ Although chiral β -diaryl esters are often desirable for syntheses of those biologically active compounds, this protocol effective for ketones, aldehydes, and *N*-acyl amides does not work for unsaturated esters. The corresponding rhodium catalysts such as [Rh(nbd)₂]BF₄/chiraphos and [Rh(binap)(nbd)]BF₄ meet this purpose; however, enantioselectivities ranging from 89 to 90% ee can be improved to 95–98% ee by the use of palladium–chiraphos complex.^{7d} Another advantage of palladium(2+) catalysts is higher turnover number (TON) than those of rhodium complexes.^{7d,12} The palladium complexes achieved quantita-

tive yields with a less than 0.1 mol % catalyst loading in the presence of less than 1.5 equivalents of arylboronic acids.^{7d}

A difficulty in using palladium(2+) catalysts for unsaturated esters is the high stability of palladium(+) *C*-enolate compared to that of *O*-enolate. A proposed catalytic cycle shown in Figure 1 involves an equilibrium formation of *C*-enolate **4** and *O*-enolate **5** as the key intermediates for giving 1,4-addition products **6** via hydrolysis with water. Although ketone substrates selectively produced **6**, the ester derivatives resulted in Heck coupling **7**, thus suggesting slow formation of water-sensitive **5**. Thus, all attempts at using acrylates as the substrates of palladium-catalyzed 1,4-addition failed to give 1,4-addition products **6**.

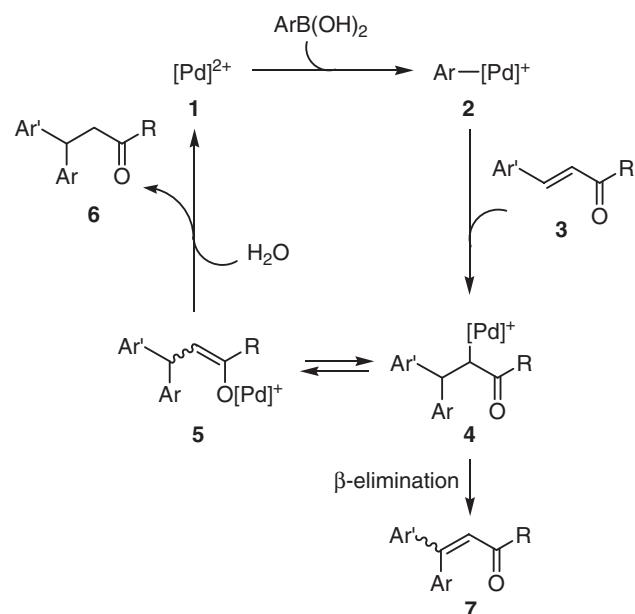
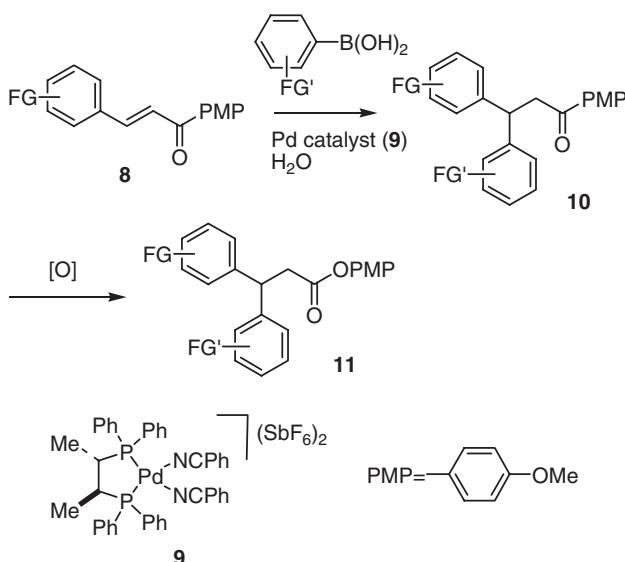


Figure 1. Catalytic cycle.

Table 1. Enantioselective Synthesis of β -Diaryl Ketones and Esters (Scheme 1)^{a)}

Entry	Enone 8	Arylboronic acid	Product	10			11		
				No	Yield/% ^{b)}	% ee	No	Yield/% ^{b)}	% ee
1				10a	99	95	11a	73	95
2				10b	86	95	11b	72	97
3				10c	74 ^{c)}	97	11c	67	95
4				10d	90	95	11d	0	0

a) All reactions were carried out at room temperature for 6 h in aqueous acetone in the presence of enone (1 mmol), ArB(OH)₂ (1.2 mmol), and [Pd(*S,S*-chiraphos)(PhCN)₂](SbF₆)₂ (*S,S*-chiraphos = (2*S,3S*)-(−)-bis(diphenylphosphino)butane) (**9**, 0.5 mol %). Chromatographic isolation of **10** was followed by Baeyer–Villiger oxidation with NaBO₃ in acetic acid at 30–50 °C. b) Isolated yields by chromatography. c) In MeOH–water (10/1).



[Pd(*S,S*-chiraphos)(PhCN)₂](SbF₆)₂
S,S-chiraphos = (2*S,3S*)-(−)-bis(diphenylphosphino)butane

Scheme 1. Stepwise 1,4-addition and Baeyer–Villiger oxidation for synthesis of optically active β -diaryl esters.

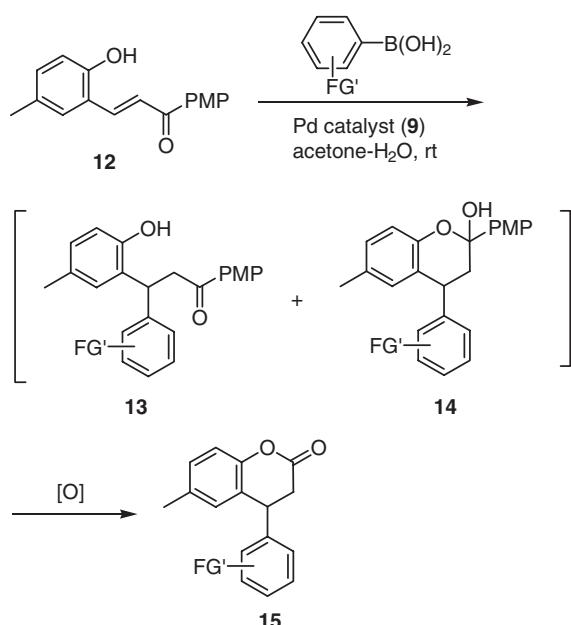
As a means to overcome this limitation of the palladium-catalyzed protocol, we report here stepwise 1,4-addition and Baeyer–Villiger oxidation for the synthesis of optically active β -diaryl esters (Scheme 1). We developed two selective methods for Baeyer–Villiger oxidation of *p*-methoxyphenyl 3,3-di-

arylpropanates and 4-aryldihydrocoumarins which provided a simple access to (*R*)-tolterodine (**21**) with high enantioselectivity.

Results and Discussion

Synthesis of Optically Active 3,3-Diarylpropanoic Esters. Traditional methods for conversion of ketones to the corresponding acids or esters are haloform reaction of methyl ketones or trihalomethyl ketones,¹³ oxidative cleavage of β -hydroxy ketones with (NH₄)₂Ce(NO₃)₆,¹⁴ and Baeyer–Villiger oxidation of unsymmetrical ketones.¹⁵ Thus, PhCH=CHCOR (R = Me, CF₃, C(OH)Me₂, CMe₃, and Ph) are potent substrates that can be finally transformed into desired acids or esters. However, palladium(2+) complex **9** failed to catalyze the additions of phenylboronic acid for trifluoromethyl, dimethyl(hydroxy)methyl, and *t*-butyl ketones, whereas methyl and phenyl ketones afforded excellent yields of 1,4-addition products. Thus, *p*-methoxyphenyl (PMP) enones **8** were finally chosen as the substrates for stepwise 1,4-addition and Baeyer–Villiger oxidation, giving **11** because of higher enantioselectivities of the palladium(2+)/chiraphos catalyst for arylketones than for methyl ketones.^{7b} Regioselective cleavage of the PMP ring rather than the primary alkyl group was previously reported by Rüedi and Hansen (Scheme 1).^{15b}

Asymmetric addition of arylboronic acids to four β -aryl ketones and oxidation of **10** to **11** with NaBO₃ are summarized in Table 1. Arylboronic acids possessing one or two alkoxy groups at para or meta carbons were smoothly added to β -aryl enones **8** at room temperature in the presence of 1.2 equivalents of arylboronic acids and 0.5 mol % of dicationic palla-

**Scheme 2.** Synthesis of optically active 4-aryldihydrocoumarins.

dium(II) catalyst **9** in aqueous acetone. The reaction easily achieved 74–99% yields with 95–97% ee. The efficiency of a rhodium–chiraphos catalyst for chalcone derivatives and the enantioselection mechanism proposed on the basis of theoretical calculation of the transition state have been previously reported.¹¹ Baeyer–Villiger oxidation of **10** suffered from low yields, resulting in complex mixtures. This is mainly due to steric hindrance of substituents on aryl groups around the carbonyl group because unsubstituted Ph₂CHCH₂COPMP resulted in 99% yield by traditional MCPBA oxidation. Oxidation of **10** with NaBO₃ resulted in the best yields among the representative methods reported for Baeyer–Villiger oxidation. Oxidation of **10a–10c** with 3 equivalents of NaBO₃ at 30–50 °C gave **11a–11c** in 67–73% yields, whereas MCPBA (5 equivalents) and trifluoroacetic acid (TFA) resulted in 38–48% yields (Entries 1–3). A combination of H₂O₂ and TFA or bis(trimethylsilyl)peroxide (TMSO)₂ and SnCl₄ gave a complex mixture from which it was difficult to isolate pure **11**. All attempts at oxidation of **10d** with these oxidants failed (Entry 4). The enantioselectivities of **10a–10c** remained perfectly intact during the NaBO₃ oxidation (Entries 1–3).

Synthesis of Optically Active 4-Aryldihydrocoumarins.

In a previous study on the synthesis of optically active 4-aryl-4*H*-chromenes, 1,4-addition of arylboronic acids to β -(2-hydroxyaryl)enones **12** provided chromanol **14** accompanied by a small amount of **13**, which were then led to single chromenes via acid-catalyzed dehydration.^{7d} Baeyer–Villiger oxidation of these intermediates **13** and **14** provided 4-aryldihydrocoumarins, which were previously synthesized by rhodium-catalyzed 1,4-addition of arylboronic acids to coumarins³⁰ (Scheme 2). Oxidation with MCPBA and NaBO₃ resulted in low yields, but (TMSO)₂ and SnCl₄ were found to be an excellent combination for selective Baeyer–Villiger oxidation. A one-pot two-step procedure without isolation of 1,4-addition intermediates afforded good yields of desired dihydrocoumarins **15** as the sole products with excellent enantioselectivities.

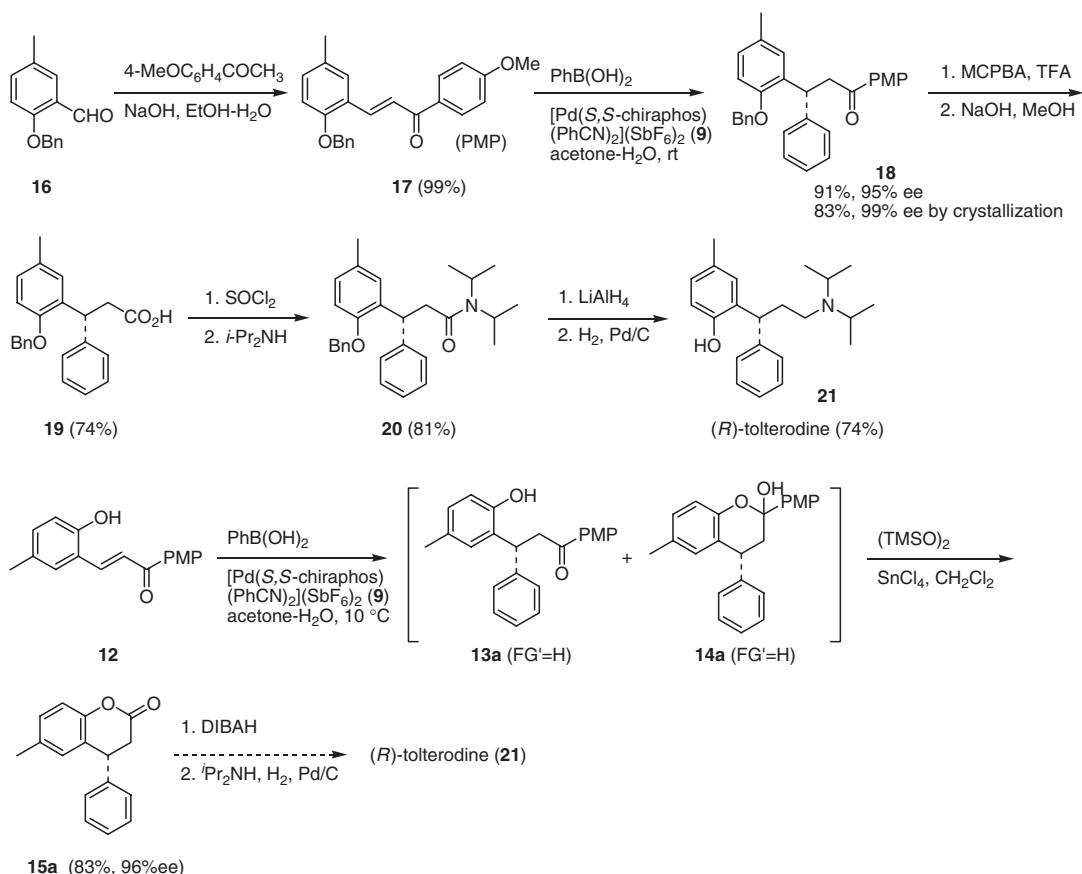
Table 2. Enantioselective Synthesis of 4-Aryldihydrocoumarins (Scheme 2)^{a)}

Entry	Arylboronic acid	Product 15	Yield /% ^{b)}	% ee
1	Ph-B(OH) ₂		15a 83	96
2	MeO-Ph-B(OH) ₂		15b 75	98
3	4-((2-methoxyethyl)oxy)phenyl-B(OH) ₂		15c 70	97
4	4-(2,6-dimethylphenyl)-Ph-B(OH) ₂		15d 74	97

a) All reactions were carried out at 10 °C for 24 h in aqueous acetone in the presence of enone (0.5 mmol), ArB(OH)₂ (0.75 mmol), and [Pd(*S,S*-chiraphos)(PhCN)₂](SbF₆)₂ (**9**, 0.5 mol %). The crude product **13/14** thus obtained was directly treated with (TMSO)₂ (1 mmol) and SnCl₄ (0.5 mmol) in CH₂Cl₂. b) Isolated yields by chromatography.

The synthesis of **13/14** via asymmetric addition of arylboronic acids to **12** was followed by oxidation with (TMSO)₂ at room temperature in the presence of SnCl₄ to give optically active 4-aryldihydrocoumarins (Table 2). Total yields of this two-step synthesis of **15** were 70–83%, which were higher than those obtained by an analogous procedure described for the synthesis of 3,3-diarylpropanoic esters (Table 1). Palladium(2+)-chiraphos catalyst **9** again achieved high enantioselectivities in a range of 96 to 98% ee.

Enantioselective synthesis of a potent competitive muscarinic receptor antagonist, (*R*)-tolterodine (**21**), has attracted much attention as a target for enantioselective synthesis of a chiral center consisting of two aryl rings. The synthesis has been achieved by diastereoselective cycloaddition of *o*-quinone methide with a chiral enol ether,¹⁶ 1,4-addition of aryl-copper reagent to cinnamic amide of chiral oxazolidinone,¹⁷ catalytic 1,4-addition of arylboronic acids to coumarins³⁰ and hydrogenation of 4-arylcoumarin.¹⁸ The present method provided an alternative convenient and practical access to (*R*)-tolterodine (**21**) with high enantioselectivity (Scheme 3). The desired enone **17** synthesized from **16** was subjected to 1,4-addition of phenylboronic acid (1.2 equiv) with the (*S,S*)-chiraphos complex **9** (0.5 mol %). Chromatographic separation of **18** resulted in 95% ee, but crystallization from THF furnished enan-



Scheme 3. Synthesis of (R)-tolterodine.

tiomerically pure **18** in 83% yield. Baeyer–Villiger oxidation of **18** with MCPBA and TFA in toluene followed by saponification of the ester group afforded an acid **19** which was previously led to (*R*)-tolterodine (**21**) in two steps. Conversion into an *N,N*-diisopropylamide **20** in 81% yield was followed by reduction of the carbonyl group and deprotection of the benzyl group to furnish optically pure tolterodine (**21**) in 74% yield.¹⁷ The formation of (*R*)-**18** from the (*S,S*)-chiraphos complex **9** was finally established by the specific rotation reported for **21** ([α]_D²³ +23.9° (c 0.58, MeOH)).^{3a,9b} Thus, the product was produced by the same mode of face selection as that proposed on the basis of theoretical calculation.^{7b}

The alternative protocol provides a simpler access to (*R*)-tolterodine. A one-pot, two-step procedure of stepwise 1,4-addition and Baeyer–Villiger oxidation directly gave a key intermediate **15a** in 96% ee, which has been previously converted into (*R*)-tolterodine (**21**) in two steps.^{3o}

Conclusion

In conclusion, we have demonstrated the efficiency of a Pd(2+)-chiraphos complex for the synthesis of β -diaryl ketones and esters that provides a simple access to (*R*)-tolterodine.

Experimental

General. All experiments were carried out under nitrogen atmosphere. HPLC analysis was directly performed with chiral stationary phase columns using Chiralcel IA, IB, OD-H, AD,

AD-H, OJ-H, and OB-H purchased from Daicel Co., Ltd. Phenyl-, 3,5-dimethyl-4-methoxyphenyl-, and 3,4-dimethoxyphenylboronic acid were purchased from Wako Chemical Co., Ltd. and Lancaster Co., Ltd.

Preparation of Enones (8 and 12). **8a** was commercially available. Other β -arylenones were synthesized by reported procedures.¹⁹ To a solution of ArCHO (30 mmol) and *p*-methoxyacetophenone (30 mmol) in ethanol (15 mL) was slowly added aqueous NaOH (39 mmol for **8** or 90 mmol for **12**) in water (30 mL). The resulting mixture was stirred overnight at 40 °C. The mixture was acidified with aqueous HCl and extracted with diethyl ether. The product was isolated by crystallization.

8b: 89% yield; ¹H NMR (400 MHz, CDCl₃) δ 8.03 (d, *J* = 8.8 Hz, 2H), 7.78 (d, *J* = 15.6 Hz, 1H), 7.60 (d, *J* = 8.8 Hz, 2H), 7.43 (d, *J* = 15.6 Hz, 1H), 6.98 (d, *J* = 8.8 Hz, 2H), 6.93 (d, *J* = 8.8 Hz, 2H), 3.89 (s, 3H), 3.86 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 188.67, 163.20, 161.44, 143.73, 131.27, 130.63, 130.48, 127.73, 119.45, 114.31, 113.72, 55.41, 55.33; MS (*m/z*) 77 (18.5), 92 (14.4), 133 (10.6), 135 (30.6), 225 (20.0), 237 (17.8), 253 (33.7), 268 (M⁺, 100.0); Exact mass calcd for C₁₇H₁₆O₃: 268.1100. Found: 268.1082.

8c: 91% yield; ¹H NMR (400 MHz, CDCl₃) δ 8.02 (d, *J* = 9.1 Hz, 2H), 7.73 (d, *J* = 15.6 Hz, 1H), 7.38 (d, *J* = 15.6 Hz, 1H), 7.17 (s, 1H), 7.12 (d, *J* = 8.1 Hz, 1H), 6.98 (d, *J* = 9.1 Hz, 2H), 6.84 (d, *J* = 8.1 Hz, 1H), 6.03 (s, 2H), 3.89 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 188.58, 163.32, 149.71, 148.36, 143.80, 131.25, 130.70, 129.55, 125.00, 119.90, 113.80, 108.64, 106.62, 101.58, 55.48; MS (*m/z*) 77 (11.8), 89 (12.3), 122 (13.0), 135 (31.2), 145 (11.9), 175 (10.2), 251 (10.5), 267 (13.1), 282 (M⁺, 100); Exact mass calcd for C₁₇H₁₄O₄: 282.0892. Found: 282.0901.

8d: 52% yield; ¹H NMR (400 MHz, CDCl₃) δ 8.03 (d, J = 8.8 Hz, 2H), 7.80 (d, J = 15.6 Hz, 1H), 7.58 (d, J = 8.1 Hz, 2H), 7.51 (d, J = 15.6 Hz, 1H), 7.27 (d, J = 8.1 Hz, 2H), 6.95 (d, J = 8.8 Hz, 2H), 3.89 (s, 3H), 2.95 (sept, J = 6.8 Hz, 1H), 1.27 (d, J = 6.8 Hz, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 188.36, 162.85, 151.23, 143.58, 132.25, 130.77, 130.29, 128.01, 126.57, 120.49, 113.32, 54.50, 33.64, 23.30; MS (m/z) 77 (10.7), 135 (28.2), 237 (100.0), 238 (19.2), 265 (35.0), 280 (M⁺, 56.6); Exact mass calcd for C₁₉H₂₀O₂: 280.1463. Found: 280.146.

12: 63% yield; ¹H NMR (400 MHz, CDCl₃) δ 8.16 (d, J = 16.1 Hz, 1H), 8.06 (d, J = 8.7 Hz, 2H), 7.68 (d, J = 16.1 Hz, 1H), 7.39 (s, 1H), 7.07 (d, J = 8.5 Hz, 1H), 6.98 (d, J = 8.7 Hz, 2H), 6.82 (d, J = 8.5 Hz, 1H), 6.60 (s, 1H), 3.89 (s, 3H), 2.30 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 190.09, 163.39, 153.68, 140.24, 132.34, 131.23, 130.99, 129.91, 129.40, 122.20, 122.03, 116.54, 113.79, 55.49, 20.46; MS (m/z) 77 (22.3), 92 (12.5), 108 (22.9), 135 (100.0), 161 (10.7), 251 (47.4), 268 (M⁺, 31.7); Exact mass calcd for C₁₇H₁₆O₃: 268.1010. Found: 268.1088.

General Procedure for 1,4-Addition (Table 1). A solution of [Pd(*S,S*-chiraphos)(PhCN)₂](SbF₆)₂ (**9**, 0.5 mol %), enone substrate (0.5 mmol) and ArB(OH)₂ (0.6 mmol) in acetone (3.0 mL) and H₂O (0.3 mL) was stirred at room temperature for 6–12 h. The product was extracted with diethyl ether and the extract was then concentrated in vacuo. Chromatography on silica gel with hexane/EtOAc afforded the desired **10**.

10a: 99% yield, 95% ee; Daicel Chiralcel IB with hexane/2-propanol = 9/1, flow = 1.0 mL min⁻¹, wavelength = 254 nm, *t*_R = 13.9 min (minor) and 15.5 min (major). The spectral data have been previously reported.¹¹

10b: 86% yield, 95% ee; [α]_D²³ +2.1° (c 0.33, CDCl₃), Daicel Chiralcel IA with hexane/2-propanol/dichloromethane = 4/2/1, flow = 0.5 mL min⁻¹, wavelength = 254 nm, *t*_R = 13.8 min (minor) and 15.6 min (major); ¹H NMR (400 MHz, CDCl₃) δ 7.92 (d, J = 8.8 Hz, 2H), 7.17 (d, J = 8.8 Hz, 2H), 6.90 (d, J = 8.8 Hz, 2H), 6.81 (d, J = 8.8 Hz, 2H), 6.77–6.75 (m, 3H), 4.72 (t, J = 7.3 Hz, 1H), 3.84 (s, 3H), 3.81 (s, 3H), 3.79 (s, 3H), 3.75 (s, 3H), 3.61 (t, J = 6.6 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 196.35, 162.97, 157.49, 148.34, 146.93, 136.75, 136.15, 129.87, 129.73, 128.18, 117.78, 113.39, 113.23, 111.03, 110.61, 55.35, 55.32, 54.98, 54.72, 44.45, 44.25; IR (neat) 1598, 1509, 1244, 1169, 1141, 1025, 986, 831, 807, 545 cm⁻¹; MS (m/z) 135 (12.9), 257 (100), 406 (M⁺, 18.7); Exact mass calcd for C₂₅H₂₆O₅: 406.1780. Found: 406.1283.

10c: 74% yield, 97% ee; [α]_D²² +4.8° (c 0.17, CDCl₃), Daicel Chiralcel IA with hexane/2-propanol/dichloromethane = 4/2/1, flow = 0.5 mL min⁻¹, wavelength = 254 nm, *t*_R = 15 min (minor) and 19 min (major); ¹H NMR (400 MHz, CDCl₃) δ 7.92 (d, J = 8.9 Hz, 2H), 6.91 (d, J = 8.9 Hz, 2H), 6.82–6.67 (m, 6H), 5.88 (s, 2H), 4.68 (t, J = 7.3 Hz, 1H), 3.85 (s, 3H), 3.82 (s, 3H), 3.80 (s, 3H), 3.60 (d, J = 7.3 Hz, 1H), 3.59 (d, J = 7.3 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 196.14, 163.00, 148.38, 147.22, 147.02, 145.44, 138.03, 136.45, 129.87, 129.66, 120.03, 118.66, 113.25, 110.95, 110.61, 107.85, 107.68, 100.40, 55.35, 55.34, 54.99, 44.92, 44.11; IR (neat) 1598, 1509, 1486, 1235, 1169, 1141, 1027, 930, 832, 809 cm⁻¹; MS (m/z) 135 (15.6), 271 (100), 420 (M⁺, 20.7); Exact mass calcd for C₂₅H₂₄O₆: 420.1573. Found: 420.1579.

10d: 90% yield, 95% ee; [α]_D²² +3.4° (c 0.22, CDCl₃), Daicel Chiralcel OD-H with hexane/2-propanol = 9/1, flow = 0.7 mL min⁻¹, wavelength = 254 nm, *t*_R = 19.6 min (minor) and 27.9 min (major); ¹H NMR (400 MHz, CDCl₃) δ 7.91 (d, J =

8.3 Hz, 2H), 7.22–7.12 (m, 3H), 7.10 (d, J = 7.8 Hz, 2H), 6.90–6.85 (m, 3H), 6.81 (s, 1H), 6.69 (d, J = 7.8 Hz, 1H), 4.75 (t, J = 6.8 Hz, 1H), 3.83 (s, 3H), 3.73 (s, 3H), 3.65 (dd, J = 6.8, 2.9 Hz, 2H), 2.85 (sept, J = 6.8 Hz, 1H), 1.19 (d, J = 6.8 Hz, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 196.12, 162.94, 159.11, 146.25, 145.68, 140.98, 129.87, 129.71, 128.95, 127.13, 126.09, 119.76, 113.60, 113.21, 110.67, 54.98, 54.63, 45.19, 43.93, 33.13, 23.49; IR (neat) 1675, 1598, 1253, 1167, 1130, 984, 830, 780, 703, 580 cm⁻¹; MS (m/z) 135 (100), 145 (15.6), 211 (11.1), 239 (52.7), 240 (10.0), 253 (16.1), 388 (M⁺, 50.9); Exact mass calcd for C₂₆H₂₈O₃: 388.2038. Found: 388.2032.

Baeyer–Villiger Oxidation (Table 1). Baeyer–Villiger oxidation of **10** was carried out by a modified procedure.²⁰ A solution of 1,4-adduct **10** (0.5 mmol) and NaBO₃·4H₂O (2.5 mmol) in acetic acid (3 mL) was stirred at 30–50 °C to room temperature for 24 h. The product was extracted with diethyl ether, washed with brine and concentrated in vacuo. Chromatography on silica gel with hexane/EtOAc afforded **11**.

11a: 73% yield, 95% ee; [α]_D²³ +6.3° (c 0.10, CDCl₃), Daicel Chiralcel IB with hexane/2-propanol = 19/1, flow = 1.0 mL min⁻¹, wavelength = 254 nm, *t*_R = 15.4 min (minor) and 16.5 min (major); ¹H NMR (400 MHz, CDCl₃) δ 7.33–7.29 (m, 4H), 7.27–7.20 (m, 2H), 6.90 (d, J = 8.0 Hz, 1H), 6.86–6.74 (m, 4H), 6.73–6.66 (m, 2H), 4.62 (t, J = 8.3 Hz, 1H), 3.77 (s, 3H), 3.75 (s, 3H), 3.26 (d, J = 8.3 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 170.79, 159.73, 157.19, 144.68, 143.99, 142.90, 129.61, 128.65, 127.73, 126.77, 122.16, 120.04, 114.34, 113.84, 111.77, 55.54, 55.17, 47.32, 40.81; IR (neat) 1752, 1503, 1236, 1191, 1128, 1033, 842, 768, 743, 699 cm⁻¹; MS (m/z): 124 (100.0), 165 (13.2), 197 (30.3), 239 (15.3), 362 (M⁺, 4.3); Exact mass calcd for C₂₃H₂₂O₄: 362.1518. Found: 362.1502.

11b: 72% yield, 97% ee; [α]_D²³ +2.4° (c 0.10, CDCl₃), Daicel Chiralcel AD-H with hexane/2-propanol = 9/1, flow = 0.5 mL min⁻¹, wavelength = 254 nm, *t*_R = 70 min (major) and 79 min (minor); ¹H NMR (400 MHz, CDCl₃) δ 7.14 (d, J = 8.0 Hz, 2H), 6.81–6.70 (m, 7H), 6.66–6.61 (m, 2H), 4.48 (t, J = 8.3 Hz, 1H), 3.79 (s, 3H), 3.76 (s, 3H), 3.72 (s, 3H), 3.69 (s, 3H), 3.14 (d, J = 8.3 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 170.90, 158.26, 157.19, 148.90, 147.68, 143.99, 136.02, 135.44, 128.58, 122.16, 119.31, 114.36, 113.97, 111.18, 111.13, 55.86, 55.82, 55.53, 55.24, 46.14, 41.32; IR (neat) 1751, 1505, 1243, 1190, 1127, 1027, 838, 810, 760, 539 cm⁻¹; MS (m/z): 124 (25.9), 257 (100.0), 258 (17.8), 270 (31.0), 298 (35.0), 422 (M⁺, 23.6); Exact mass calcd for C₂₅H₂₆O₆: 422.1729. Found: 422.1747.

11c: 67% yield, 95% ee; [α]_D²³: +3.6° (c 0.19, CDCl₃), Daicel Chiralcel IA with hexane/2-propanol = 9/1, flow = 1.0 mL min⁻¹, wavelength = 254 nm, *t*_R = 46.5 min (major) and 53.2 min (minor); ¹H NMR (400 MHz, CDCl₃) δ 6.86–6.71 (m, 10H), 5.91 (s, 2H), 4.52 (t, J = 8.3 Hz, 1H), 3.86 (s, 3H), 3.84 (s, 3H), 3.76 (s, 3H), 3.19 (d, J = 8.3 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 170.75, 157.21, 148.94, 147.83, 147.78, 146.24, 143.96, 137.34, 135.68, 122.14, 120.44, 119.21, 114.38, 111.13, 111.12, 108.19, 108.16, 100.97, 55.86, 55.84, 55.53, 46.58, 41.20; IR (neat) 1503, 1481, 1237, 1191, 1127, 1029, 910, 811, 761, 728 cm⁻¹; MS (m/z) 124 (31.7), 271 (100), 284 (29.9), 312 (53.3), 436 (31.2); Exact mass calcd for C₂₅H₂₄O₇: 436.1522. Found: 436.1535.

General Procedure for Synthesis of 4-Aryldihydrocoumarins (Table 2). A solution of [Pd(*S,S*-chiraphos)(PhCN)₂]-
(SbF₆)₂ (**9**, 0.5 mol %), enone **12** (0.5 mmol), and ArB(OH)₂ (0.75 mmol) in acetone (3.0 mL) and H₂O (0.3 mL) was stirred

at 10 °C for 24 h. The mixture was passed through a short pad of silica gel with diethyl ether as eluent. The filtrate was concentrated in vacuo to give crude **13/14**.

Baeyer–Villiger oxidation of crude **13/14** was carried out by a reported procedure.²¹ The residue thus obtained was dissolved in CH₂Cl₂ (5 mL) and treated with (TMSO)₂ (0.5 mmol). To this mixture was slowly added SnCl₄ (0.5 mmol) (1.0 M solution in heptane) at 0 °C and the mixture was stirred for 15 min at 0 °C and at room temperature for 1 h. Additional (TMSO)₂ (0.5 mmol) was then added and stirred for 3 h. The product was extracted with diethyl ether, washed with saturated K₂CO₃ in water and concentrated in vacuo. Chromatography on silica gel with hexane/EtOAc afforded **15**.

15a and **15b** have been previously reported.³⁰

15c: 70% yield, 97% ee; [α]_D²³ +9.4° (c 0.9, CDCl₃), Daicel Chiralcel IB with hexane/2-propanol/dichloromethane = 50/1/3, flow = 0.5 mL min⁻¹, wavelength = 254 nm, *t*_R = 29 min (minor) and 33 min (major); ¹H NMR (400 MHz, CDCl₃) δ 7.09 (d, *J* = 8.3 Hz, 1H), 7.01 (d, *J* = 8.3 Hz, 1H), 6.80 (s, 1H), 6.76 (d, *J* = 8.8 Hz, 1H), 6.65–6.55 (m, 2H), 5.95 (s, 2H), 4.22 (t, *J* = 6.6 Hz, 1H), 3.02 (dd, *J* = 16, 5.9 Hz, 1H), 2.94 (dd, *J* = 16, 7.6 Hz, 1H), 2.27 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 167.79, 149.52, 148.21, 146.96, 134.33, 134.29, 129.31, 128.58, 125.33, 120.80, 116.84, 108.61, 107.70, 101.18, 40.41, 37.36, 20.74; IR (neat) 1487, 1443, 1239, 1198, 1164, 1150, 1118, 1036, 927, 812 cm⁻¹; MS (*m/z*): 152 (12.6), 182 (15.6), 210 (19.6), 239 (81.7), 264 (41.8), 282 (M⁺, 100); Exact mass calcd for C₁₇H₁₄O₄: 282.08921. Found: 282.08827.

15d: 74% yield, 97% ee; [α]_D²³ +2.6° (c 0.45, CDCl₃), Daicel Chiralcel IA with hexane/2-propanol = 9/1, flow = 0.5 mL min⁻¹, wavelength = 254 nm, *t*_R = 12 min (minor) and 14 min (major); ¹H NMR (400 MHz, CDCl₃) δ 6.99 (d, *J* = 8.3 Hz, 1H), 6.92 (d, *J* = 8.3 Hz, 1H), 6.72 (s, 1H), 6.68 (s, 2H), 4.09 (t, *J* = 6.8 Hz, 1H), 3.62 (s, 3H), 2.92 (dd, *J* = 16, 6.1 Hz, 1H), 2.86 (dd, *J* = 16, 7.1 Hz, 1H), 2.18 (s, 3H), 2.16 (s, 6H); ¹³C NMR (100 MHz, CDCl₃) δ 167.98, 156.17, 149.47, 135.67, 134.19, 131.41, 129.11, 128.64, 127.68, 125.47, 116.07, 59.56, 40.08, 37.16, 20.69, 16.11; IR (neat) 1764, 1488, 1223, 1200, 1154, 1128, 1010, 924, 894, 815 cm⁻¹; MS (*m/z*): 195 (10.3), 223 (37.0), 239 (81.1), 253 (19.1), 263 (28.0), 278 (31.8), 296 (M⁺, 100); Exact mass calcd for C₁₉H₂₀O₃: 296.1413. Found: 296.1400.

Synthesis of 16. To a mixture of 5-methylsalicylaldehyde (50 mmol) and K₂CO₃ (55 mmol) in ethanol (100 mL) and water (50 mL) was slowly added benzyl chloride (55 mmol) and the mixture was then stirred overnight at reflux. The product was obtained by crystallization. 77% yield; ¹H NMR (400 MHz, CDCl₃) δ 10.53 (s, 1H), 7.66 (s, 1H), 7.46–7.30 (m, 6H), 6.95 (d, *J* = 8.4 Hz, 1H), 5.17 (s, 2H), 2.31 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 189.93, 159.15, 136.54, 136.24, 130.44, 128.68, 128.42, 128.19, 127.25, 124.84, 113.08, 70.54, 20.26; MS (*m/z*): 65 (10.3), 91 (100.0), 135 (8.2), 226 (M⁺, 7.5); Exact mass calcd for C₁₅H₁₄O₂: 226.0994. Found: 226.0987.

Synthesis of 17. To a solution of **16** (77 mmol) and *p*-methoxyacetophenone (77 mmol) in ethanol (18 mL) was slowly added aqueous NaOH (100 mmol, 36 mL). The mixture was stirred overnight at 40 °C and then acidified with hydrochloric acid at room temperature. The product was obtained by crystallization. 99% yield; ¹H NMR (400 MHz, CDCl₃) δ 8.00 (d, *J* = 15.6 Hz, 1H), 7.85 (d, *J* = 8.8 Hz, 2H), 7.74 (d, *J* = 15.6 Hz, 1H), 7.49 (d, *J* = 6.3 Hz, 2H), 7.45–7.35 (m, 4H), 7.14 (d, *J* = 8.3 Hz, 1H), 6.91 (d, *J* = 8.3 Hz, 1H), 6.88 (d, *J* = 8.8 Hz, 2H), 5.14

(s, 2H), 3.88 (s, 3H), 2.33 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 162.63, 155.69, 139.65, 136.20, 131.35, 130.89, 130.29, 129.78, 128.23, 127.63, 127.36, 123.52, 122.68, 113.26, 111.96, 70.15, 54.95, 31.10, 22.17, 19.93; MS (*m/z*): 77 (11.2), 91 (100), 135 (89.0), 160 (16.0), 251 (47.3), 358 (M⁺, 12.5); Exact mass calcd for C₂₄H₂₂O₃: 358.1569. Found: 358.1581.

Synthesis of 18. The procedure shown in general for 1,4-addition (Table 1) gave **18**. 91% yield; 95% ee. [α]_D²³ +13.3° (c 0.28, CDCl₃), Daicel Chiralcel AD with hexane/2-propanol = 9/1, flow = 1.0 mL min⁻¹, wavelength = 254 nm, *t*_R = 17 min (minor) and 205 min (major); ¹H NMR (400 MHz, CDCl₃) δ 7.89 (d, *J* = 8.0 Hz, 2H), 7.31–7.12 (m, 10H), 6.98 (s, 1H), 6.93 (d, *J* = 8.3 Hz, 1H), 6.82 (d, *J* = 7.3 Hz, 2H), 6.77 (d, *J* = 8.3 Hz, 1H), 5.16 (t, *J* = 7.3 Hz, 1H), 5.01 (d, *J* = 14.6 Hz, 1H), 4.98 (d, *J* = 14.6 Hz, 1H), 3.83 (s, 3H), 3.65 (d, *J* = 7.3 Hz, 2H), 2.23 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 196.98, 163.19, 153.81, 143.74, 137.26, 132.60, 130.32, 130.17, 129.84, 129.04, 128.38, 128.18, 128.14, 127.68, 127.62, 127.36, 125.94, 113.54, 112.05, 70.14, 55.39, 43.38, 40.40, 20.72; IR (neat) 1659, 1598, 1571, 1501, 1450, 1312, 1263, 1238, 1178, 1120, 1044, 1022, 994 cm⁻¹; MS (*m/z*): 91 (30.7), 135 (100), 241 (11.7), 345 (18.6), 345 (18.6), 436 (M⁺, 1.8); Exact mass calcd for C₃₀H₂₈O₃: 436.2038. Found: 436.2059.

Syntheses and spectral data of **19**, **20**, and (*R*)-tolterodine (**21**) have been previously reported.¹⁷

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References

- For reviews, see: a) B. E. Rossiter, N. M. Swingle, *Chem. Rev.* **1992**, 92, 771. b) K. Tomioka, Y. Nagaoka, in *Comprehensive Asymmetric Catalysis*, ed. by E. N. Jacobsen, A. E. Pfalz, H. Yamamoto, S.-V. Berlin, **1999**, Chap. 31.1. c) M. P. Sibi, S. Manyem, *Tetrahedron* **2000**, 56, 8033. d) N. Krause, A. Hoffmann-Röder, *Synthesis* **2001**, 171. e) J. Christoffers, G. Koripelly, A. Rosiak, M. Rössle, *Synthesis* **2007**, 1279.
- M. Sakai, H. Hayashi, N. Miyaura, *Organometallics* **1997**, 16, 4229.
- a) Y. Takaya, M. Ogasawara, T. Hayashi, M. Sakai, N. Miyaura, *J. Am. Chem. Soc.* **1998**, 120, 5579. b) R. A. Batey, A. N. Thadani, D. V. Smil, *Org. Lett.* **1999**, 1, 1683. c) T. Hayashi, T. Senda, Y. Takaya, M. Ogasawara, *J. Am. Chem. Soc.* **1999**, 121, 11591. d) S. Sakuma, M. Sakai, R. Itooka, N. Miyaura, *J. Org. Chem.* **2000**, 65, 5951. e) M. T. Reetz, D. Moulin, A. Gosberg, *Org. Lett.* **2001**, 3, 4083. f) J. Ramnauth, O. Poulin, S. Bratovanov, S. S. Rakshit, S. P. Maddaford, *Org. Lett.* **2001**, 3, 2571. g) R. Itooka, Y. Iguchi, N. Miyaura, *Chem. Lett.* **2001**, 722. h) R. Amengual, V. Michelet, J.-P. Genêt, *Tetrahedron Lett.* **2002**, 43, 5905. i) M. Pucheault, S. Darses, J.-P. Genêt, *Tetrahedron Lett.* **2002**, 43, 6155. j) T. Hayashi, M. Takahashi, Y. Takaya, M. Ogasawara, *J. Am. Chem. Soc.* **2002**, 124, 5052. k) R. Itooka, Y. Iguchi, N. Miyaura, *J. Org. Chem.* **2003**, 68, 6000. l) Y. Ma, C. Song, C. Ma, Z. Sun, Q. Chai, M. B. Andrus, *Angew. Chem., Int. Ed.* **2003**, 42, 5871. m) D. F. Cauble, J. D. Gipson, M. J. Krische, *J. Am. Chem. Soc.* **2003**, 125, 1110. n) L. Navarre, M. Pucheault, S. Darses, J.-P. Genêt, *Tetrahedron Lett.* **2005**, 46, 4247. o) G. Chen, N. Tokunaga, T. Hayashi, *Org. Lett.* **2005**, 7, 2285. p) Y. Yamamoto, K. Kurihara, N. Sugishita, K. Oshita, D. Piao, N. Miyaura, *Chem. Lett.* **2005**, 34, 1224. q) R. T. Stemmle, C. Bolm, *J. Org. Chem.* **2005**, 70,

9925. r) K. Vandyck, B. Matthys, M. Willen, K. Robeyns, L. van Meervelt, J. van der Eycken, *Org. Lett.* **2006**, *8*, 363. s) K. Kurihara, N. Sugishita, K. Oshita, D. Piao, Y. Yamamoto, N. Miyaura, *J. Organomet. Chem.* **2007**, *692*, 428. For reviews, see: t) Y. Yamamoto, T. Nishikata, N. Miyaura, *J. Synth. Org. Chem., Jpn.* **2006**, *64*, 1112. u) T. Hayashi, K. Yamasaki, *Chem. Rev.* **2003**, *103*, 2829.
- 4 a) T. Nishikata, Y. Yamamoto, N. Miyaura, *Angew. Chem., Int. Ed.* **2003**, *42*, 2768. b) T. Nishikata, Y. Yamamoto, N. Miyaura, *Organometallics* **2004**, *23*, 4317.
- 5 T. Nishikata, Y. Yamamoto, N. Miyaura, *Chem. Lett.* **2003**, *32*, 752.
- 6 T. Nishikata, Y. Yamamoto, N. Miyaura, *Chem. Commun.* **2004**, 1822.
- 7 a) T. Nishikata, Y. Yamamoto, N. Miyaura, *Chem. Lett.* **2005**, *34*, 720. b) T. Nishikata, Y. Yamamoto, I. D. Gridnev, N. Miyaura, *Organometallics* **2005**, *24*, 5025. c) T. Nishikata, Y. Yamamoto, N. Miyaura, *Tetrahedron Lett.* **2007**, *48*, 4007. d) T. Nishikata, Y. Yamamoto, N. Miyaura, *Adv. Synth. Catal.* **2007**, *349*, 1759. e) T. Nishikata, Y. Yamamoto, N. Miyaura, *Chem. Lett.* **2007**, *36*, 1442. f) T. Nishikata, Y. Kobayashi, K. Kobayashi, Y. Yamamoto, N. Miyaura, *Synlett* **2007**, 3055.
- 8 a) F. Gini, B. Hessen, A. J. Minnaard, *Org. Lett.* **2005**, *7*, 5309. b) F. Gini, B. Hessen, B. L. Feringa, A. J. Minnaard, *Chem. Commun.* **2007**, 710.
- 9 a) L. Nilvebrant, K.-E. Andersson, P.-G. Gillberg, M. Stahl, B. Sparf, *Eur. J. Pharmacol.* **1997**, *327*, 195. b) S. Sörgel, N. Tokunaga, K. Sasaki, K. Okamoto, T. Hayashi, *Org. Lett.* **2008**, *10*, 589.
- 10 a) J. E. Lynch, W.-B. Choi, H. R. O. Churchill, R. P. Volante, R. A. Reamer, R. G. Ball, *J. Org. Chem.* **1997**, *62*, 9223. b) R. P. Alexander, G. J. Warrelow, M. A. W. Eaton, E. C. Boyd, J. C. Head, J. R. Porter, J. A. Brown, J. T. Reuberson, B. Hutchinson, P. Turner, B. Boyce, D. Barnes, B. Mason, A. Cannell, R. J. Taylor, A. Zomaya, A. Millican, J. Leonard, R. Morphy, M. Wales, M. Perry, R. A. Allen, N. Gozzard, B. Hughes, G. Higgs, *Bioorg. Med. Chem. Lett.* **2002**, *12*, 1451. c) V. K. Aggarwal, I. Bae, H.-Y. Lee, J. Richardson, D. T. Williams, *Angew. Chem., Int. Ed.* **2003**, *42*, 3274.
- 11 T. Itoh, T. Mase, T. Nishikata, T. Iyama, H. Tachikawa, Y. Kobayashi, Y. Yamamoto, N. Miyaura, *Tetrahedron* **2006**, *62*, 9610.
- 12 a) R. Amengual, V. Michelet, J.-P. Genet, *Synlett* **2002**, 1791. b) F.-X. Chen, A. Kina, T. Hayashi, *Org. Lett.* **2006**, *8*, 341.
- 13 a) R. C. Fuson, B. A. Bull, *Chem. Rev.* **1934**, *15*, 275. b) P. Kaszynski, J. Michel, *J. Org. Chem.* **1988**, *53*, 4593.
- 14 C. Palomo, M. Oiarbide, J. M. García, A. González, E. Arceo, *J. Am. Chem. Soc.* **2003**, *125*, 13942.
- 15 a) S. Kobayashi, H. Tanaka, H. Amii, K. Uneyama, *Tetrahedron* **2003**, *59*, 1547. b) G. Rüedi, H.-J. Hansen, *Helv. Chim. Acta* **2004**, *87*, 1968.
- 16 C. Selenski, T. R. R. Pettus, *J. Org. Chem.* **2004**, *69*, 9196.
- 17 P. G. Andersson, H. E. Schink, K. Österlund, *J. Org. Chem.* **1998**, *63*, 8067.
- 18 F. Ulgheri, M. Marchetti, O. Piccolo, *J. Org. Chem.* **2007**, *72*, 6056.
- 19 R. M. Kellogg, J. W. Nieuwenhuijzen, K. Pouwer, T. R. Vries, Q. B. Broxterman, R. F. P. Grimbergen, B. Kaptein, R. M. L. Crois, E. de Wever, K. Zwaagstra, A. C. van der Laan, *Synthesis* **2003**, 1626.
- 20 A. McKillop, J. A. Tarbin, *Tetrahedron* **1987**, *43*, 1753.
- 21 S. Matsubara, K. Takai, H. Nozaki, *Bull. Chem. Soc. Jpn.* **1983**, *56*, 2029.