The Chemotherapy of Schistosomiasis. Part VIII.¹ 2-Carboxy- and 2-Carbamoyl-4-aminophenyl Ethers

By R. F. Collins and M. Davis

Synthetic work in the series of schistosomicidal *p*-aminophenyl ethers has been continued with the introduction of carboxy-, carbamoyl-, alkylsulphonyl-, and some miscellaneous nuclear substituents for structure–activity studies.

Following the discovery that the insertion of a methoxy-group *ortho* to the ether linkage was beneficial in a series of schistosomicidal p-aminophenyl ethers,² alkyl and substituted alkyl groups were likewise introduced.¹ Extension of the synthetic work to p-aminophenyl ethers bearing carboxy-, carbamoyl-, methylsulphonyl-, and several other substituents is now recorded. The biological results are reported separately.³

The 2-hydroxymethyl-4-nitrophenyl ethers (I) from earlier work ¹ were readily oxidised with chromic acid to the 2-carboxy-ethers (III), alternatively obtained by nitration of the o-alkoxybenzoic acid (IV). Use of a limited quantity of chromic acid led also to the aldehyde (II; $R = n \cdot C_8 H_{17}$), which with orthoformic ester yielded the acetal (V). The acids (III) were converted into acid chlorides and thence into a number of aliphatic, aromatic, and heterocyclic amides (VI).

An amino-group para to the ether linkage is essential for anthelmintic activity in the simple aminophenyl ethers, and a meta isomer (VIII) of an active compound (VII; $R = n \cdot C_8 H_{17}$, R' = OH) was therefore made to confirm that this relationship held when a carboxyl group was present. It was prepared from ethyl p-acetamidosalicylate by alkylation with n-octyl bromide and subsequent hydrolysis of the ester-amide.

 1 Part VII, R. F. Collins and M. Davis, J. Chem. Soc. (C), 1966, 873

873.

² Part IV, R. F. Collins and M. Davis, J. Chem. Soc., 1961, 1863; Part VI, idem., J. Chem. Soc. (C), 1966, 366.

5-Nitro-2-octyloxybenzyl chloride ¹ (IX) was utilised for the synthesis of a phenylalanine derivative (XI) by condensation with acetamidomalonic ester, hydrolysis of

the intermediate (XI), and reduction. For the amidine (XIII; R = R' = H), 2-cyano-4-nitrophenol was condensed with octyl bromide and the nitro-ether was

³ R. F. Collins, (Mrs.) V. A. Cox, M. Davis, N. D. Edge, J. Hill, K. F. Rivett, and (Miss) M. A. Rust, in preparation.

⁴ R. F. Collins, M. Davis, N. D. Edge, and J. Hill, Brit. J. Pharmacol., 1958, 13, 238.

⁵ W. Grimme and H. Schmitz, Chem. Ber., 1954, 87, 179.

Org. 2197

reduced and acetylated. Treatment of the product (XII) with hydroxylamine afforded the amidoxime (XIII; R = Ac, R' = OH) which was hydrogenated and hydrolysed to the amidine.

The preparation of 5-acetamido-2-methoxybenzene-sulphinic acid from p-acetamidoanisole was recorded in Part VII,¹ and methylation gave the sulphone (XIV; R = Ac, R' = Me). Replacement of the methoxygroup by the n-octyloxy-group (which was more potent chemotherapeutically) was effected by hydrolysis with 50% aqueous hydrobromic acid, reacetylation of the aminophenol (XIV; R = R' = H) to protect the aminogroup, and alkylation with octyl bromide. Hydrolysis of the amide (XIV; R = Ac, R' = n-C₈H₁₇) then gave the desired base (XIV; R = H, R' = n-C₈H₁₇).

EXPERIMENTAL

Except where stated, light petroleum had b. p. 40—60°. Nitro-compounds.—The carboxylic acids listed in Table 1 were prepared by one of the following methods.

1-(2-Carboxy-4-nitrophenoxy)-n-octane (Method A). 1-(2-Hydroxymethyl-4-nitrophenoxy)-n-octane ¹ (108·5 g.) in acetone (2 l.) was oxidised by dropwise addition at room temperature of a solution of chromium trioxide (43·5 g.) in water (220 ml.) and sulphuric acid (38 ml.). After addition the mixture was stirred for 30 min. and poured into water (4 l.) and the solid was collected, washed with 2n-sulphuric acid, and partially dissolved in potassium hydrogen carbonate solution. The insoluble residue was recrystallised from methanol and yielded 1-(2-formyl-4-nitrophenoxy)-n-octane (24 g., 22%), m. p. 47—50° (Found: C, 64·2; H, 7·8; N, 5·1. $C_{15}H_{21}NO_4$ requires C, 64·5; H, 7·6; N, 5·0%).

The alkaline filtrate was acidified and the precipitate was collected and recrystallised from aqueous ethanol to yield the required acid (79 g., 69%), m. p. 100—102° (Found: C, 60·8; H, 7·4; N, 4·8. $C_{15}H_{21}NO_5$ requires C, 61·1; H, 7·2; N, 4·7%). Esterification with methanolic hydrogen chloride gave the methyl ester, which was used directly. 1-(2-Formyl-4-nitrophenoxy)-n-hexane, m. p. 66—70° (Found: C, 62·4; H, 7·0; N, 5·6. $C_{13}H_{17}NO_4$ requires C, 61·2; H, 6·8; N, 5·6%) was similarly obtained.

1-(2-Carboxy-4-nitrophenoxy)-n-heptane (Method B). A solution of 2-heptyloxybenzoic acid (49 g.) in acetic acid (147 ml.) was added to furning nitric acid (490 ml.) with stirring at 0°. After 20 min. the mixture was poured into

water and the product collected, and crystallised from acetic acid, giving the acid (54%), m. p. 93—95° (Found: C, 60·0; H, 6·9; N, 5·0. $C_{14}H_{19}NO_5$ requires C, 59·8; H, 6·8; N, 5·0%).

1-(2-Carbamoyl-4-nitrophenoxy)-n-hexane. 1-(2-Carboxy-4-nitrophenoxy)-n-hexane (15 g.) was refluxed with thionyl chloride (45 ml.) for 1 hr. and excess of reagent removed. The residue in benzene was poured into aqueous ammonia (d 0-880, 250 ml.) and ice (200 g.). The product was recrystallised twice from ethanol, giving the *amide* (78%), m. p. 117—122° (Found: N, 10-6. $C_{13}H_{18}N_2O_4$ requires N, $10\cdot5\%$). The other primary *amides* in Table 3 were prepared similarly. The substituted amides were prepared by adding the acid chloride to a solution of the amine in dry benzene or pyridine.

N-(5-Nitro-2-octyloxybenzoyl)aminoacetic acid. 1-(2-Carboxy-4-nitrophenoxy)-n-octane (20 g.) was converted with thionyl chloride into the acid chloride which was added slowly to a solution of glycine ($10\cdot 1$ g.) in 2N-aqueous sodium hydroxide (200 ml.), with vigorous shaking. The mixture was allowed to stand for 10 min., diluted to 11., filtered, and acidified with hydrochloric acid. The product, which separated overnight as a gum which slowly crystallised, was collected and recrystallised from methanol, giving the amide (65%), m. p. $119-122^\circ$ (Found: C, $58\cdot 1$; H, $6\cdot 8$; N, $7\cdot 9$. $C_{17}H_{24}N_2O_6$ requires C, $57\cdot 9$; H, $6\cdot 9$; N, $7\cdot 95\%$).

1-(2-Diethoxymethyl-4-nitrophenoxy)-n-octane. 1-(2-Formyl-4-nitrophenoxy)-n-octane (17·4 g.) in ethanol (70 ml.) was treated with ethyl orthoformate (9·3 g.) and concentrated hydrochloric acid (0·04 ml.). After 24 hr. the solution was made just alkaline with ethanolic potassium hydroxide, concentrated, and extracted with ether. The dried extract was distilled to give the acetal (72%), b. p. $170-172^{\circ}/0.04$ mm. (Found: C, $64\cdot4$; H, $8\cdot7$; N, $4\cdot1$. $C_{19}H_{31}NO_5$ requires C, $64\cdot6$; H, $8\cdot8$; N, $4\cdot0\%$).

Diethyl acetamido-(5-nitro-2-n-octyloxybenzyl)malonate. Diethyl acetamidomalonate (12 g.) and 5-nitro-2-n-octyloxybenzyl chloride 1 (15 g.) were added successively to a cooled, stirred solution of sodium (1·15 g.) in dry ethanol (100 ml.). After 30 min., the mixture was refluxed for 2 hr., diluted with water, and extracted with ether. Evaporation of the extract and crystallisation of the residue from light petroleum yielded the ester (35%), m. p. 69—71° (Found: C, 60·6; H, 7·9; N, 6·2. $C_{24}H_{36}N_2O_8$ requires C, 60·0; H, 7·55; N, 5·8%).

1-[2-(2-Amino-2-carboxyethyl)-4-nitrophenoxy]-n-octane [β-(5-nitro-2-octyloxyphenyl)-α-alanine]. The foregoing ester (8 g.) was refluxed with concentrated hydrochloric acid (150 ml.) for 44 hr. The mixture was allowed to cool and the hydrochloride of the product was filtered off, dissolved in water, and treated with aqueous sodium acetate solution. The product was recrystallised from aqueous methanol, yielding the amino-acid (80%), m. p. 205—207° (decomp.) (Found: C, 60·1; H, 7·4; N, 8·2. $C_{17}H_{26}N_2O_5$ requires C, 60·3; H, 7·7; N, 8·3%).

The *compounds* of Table 5 were prepared by condensation of the sodium or potassium salt of the phenol with the appropriate halide.

Amines.—Except where indicated, the amines in Tables 2, 4, and 6 were prepared by catalytic reduction of the corresponding nitro-compounds.

1-[4-Amino-2-(thiazol-2-ylcarbamoyl)phenoxy]-n-octane. A solution of 1-[4-nitro-2-(thiazol-2-ylcarbamoyl)phenoxy]-n-octane (7·45 g.) in dimethylformamide (15 ml.) was added to a stirred, refluxing mixture of reduced iron powder

J. Chem. Soc. (C), 1966

TABLE 1

2-Carboxy-4-nitrophenyl ethers
$$O_2N$$
 O_2N O_2N O_2N O_2N

	Yield						und (%	6)	Required (%)			
								_				
n	R	Method	(%)	М. р.	Solvent a	С	H	N	Formula	C	H	N
2	Me	\mathbf{B}	67	128130°	MeOH	$53 \cdot 2$	$5 \cdot 1$	$6 \cdot 3$	$C_{10}H_{11}NO_{5}$	$53 \cdot 4$	4.9	$6 \cdot 3$
3	Me	\mathbf{A}	61	117.5 - 119	AcOH	$55 \cdot 3$	5.85	$5 \cdot 9$	$C_{11}H_{13}NO_5$	$55 \cdot 2$	$5 \cdot 5$	5.9
4	Me	\mathbf{B}	55	108110	EtOAc	$57 \cdot 1$	$6 \cdot 1$	5.8	$C_{12}H_{15}NO_{5}$	56.9	6.0	$5 \cdot 5$
5	Me	A	88	117-118.5	MeOH-H ₂ O	58.7	6.8	$5\cdot 2$	$C_{13}H_{17}NO_5$	58.4	$6 \cdot 4$	$5 \cdot 2$
8	Me	\mathbf{B}	58	130 - 131	EtOAc	$62 \cdot 4$	$7 \cdot 6$	4.9	$C_{16}H_{23}NO_5$	$62 \cdot 1$	7.5	4.5
9	Me	Α	66	67—7 0	Pet	$62 \cdot 9$	7.8	$5 \cdot 4$	$C_{17}H_{25}NO_5$	$63 \cdot 1$	7.8	$4 \cdot 3$
5	Ph	Α	54	167 - 169	EtOH	$65 \cdot 6$	5.8	4.45	$C_{18}H_{19}NO_5$	65.7	5.8	$4 \cdot 3$
5	Phthalimido	A b	82	153155	AcOH	60.25	$4 \cdot 6$	$7 \cdot 1$	$C_{20}H_{18}N_2O_7$	60.3	4.6	$7 \cdot 0$
		c	89	119-121	EtOH	61.4	5.0	$7 \cdot 0$	$\mathrm{C_{21}H_{20}N_{2}O_{7}}$	$61 \cdot 2$	$4 \cdot 9$	6.8
5	$O \cdot C_6 H_3 (CO_2 H) \cdot NO_2 - 1, 2, 4$	\mathbf{B}_{-}	61	182 - 185	AcOH	$52 \cdot 4$	$4 \cdot 6$	$6 \cdot 4$	$\mathrm{C_{19}H_{18}N_{2}O_{10}}$	$52 \cdot 5$	4.15	6.45
		\mathbf{B}^{d}	14	136 - 138	EtOAc	54.0	4.8	$6 \cdot 4$	$C_{21}H_{22}N_2O_{10}$	$54 \cdot 5$	4.8	$6 \cdot 1$

^a Solvent for crystn.; Pet = light petroleum (b. p. 80—100°). ^b Oxidation carried out at 100°. ^c Methyl ester, prepared using methanolic hydrogen chloride. ^d Dimethyl ester.

Table 2

4-Amino-2-carboxyphenyl ethers Found (%) Required (%) Yield Derivб ć М. р. \mathbf{R} Solvent a H N Hal Formula Η \mathbf{N} Hal ative (%)145—148° C₁₀H₁₃NO₃ Me Base 46 **EtOAc** 61.76.57.061.57.2C₁₁H₁₅NO₃ C₁₂H₁₇NO₃ 3 82105 - 106**EtOH** $63 \cdot 4$ $63 \cdot 15$ MeBase 64 **EtOH** 7.7**4 5** Me Base 115-117 $64 \cdot 2$ 7.8 $6 \cdot 4$ 64.56.38.4 5.7 5.9 Base 70 103-105 $\rm EtOH\text{-}H_2O$ 66.0 $C_{13}H_{19}NO_3$ 65.88.1 Me 80 103---105 67.08.4 5.8 C14H21NO3 67.0 $8 \cdot 4$ 5.6Base Pet 6 MeMeOH 8.8 $5 \cdot 3$ Base 83 97---99 68.2 $C_{15}H_{23}NO_3$ 8.7567.95.3Me 11.1 11.370 173 - 175**EtOH** 4.5Me ester, 4.4HCl 103-105 Base EtOH $8.8 \ 5.0$ $9 \cdot 0$ 69.05 $C_{16}H_{25}NO_3$ 68.8 5.0109—111 9 Me Base 79 **EtOH** 69.99.54.8 $C_{17}H_{27}NO_3$ 69.69.34.8 $C_{16}H_{21}NO_3$ $C_{20}H_{20}N_2O_5$ 99—101 165—167 5 Ph Base 82**EtOH** $72 \cdot 1$ 6.9 $4 \cdot 3$ $72 \cdot 2$ $7 \cdot 1$ 4.75 Phthalimido Base 75 $\mathbf{NMe_2\text{-}CHO}$ $65 \cdot 4$ 5.97.7 $65 \cdot 2$ 5.57.6Me ester, 63 6.5 $C_{21}H_{22}N_2O_5$, HCl (decomp.) HC1 241-244 Aq. HCl $O \cdot C_6H_3(CO_2H) \cdot NH_2-1,2,4$ 2HCl $6 \cdot 1 \quad 15 \cdot 7 \quad C_{19} H_{22} N_2 O_6, HC1$ 6.3 15.85(decomp.) ${
m Me_2~ester}, \ {
m 2HCl}$ $5.9 \ 15.1 \ C_{21}H_{26}N_2O_6, 2HCl$ 5.9 14.95

Solvent for crystn.; Pet = light petroleum.
 2-Ethoxyethanol-ethyl ether.

TABLE 3

180 (decomp.)

2-Carbamoyl-4-nitrophenyl ethers

				Yield	Found (%)						Required (%)		
n	\mathbf{R}	$\mathbf{R'}$	R''	(%)	M. p.	Solvent a	С	H	\mathbf{N}	Formula	С	Η	N
7	Me	H	Н	74 6	140141°	EtOH	$61 \cdot 2$	7.7	9.45	$C_{15}H_{22}N_2O_4$	61.2	7.5	9.5
9	Me	H	H	85	106 - 109	EtOH	63.2	8.25	6.9	$C_{17}H_{26}N_2O_4$	$63 \cdot 3$	$8 \cdot 1$	8.7
5	Phthalimido	\mathbf{H}	H	97	172 - 175	AcOH	60.2	4.8	10.4	$C_{20}H_{19}N_3O_6$	60.5	4.8	10.6
7	Me	Et	Et	100	c					$C_{19}H_{30}N_2O_4$			
7	Me		$-[CH_2]_4$	100	9395	C_6H_6			$8 \cdot 2$	$C_{19}H_{28}N_2O_4$			8.05
7	Me	\mathbf{H}	[CH ₂] ₂ ·NEt ₂ ,HCl	86	133 - 135				9.65	$C_{21}H_{35}N_3O_4$, HCl ^d			9.8
7	Me	\mathbf{H}	Ph	_	109 - 111	Pet	67.8	6.95	$7 \cdot 6$	$C_{21}H_{26}N_2O_4$	$68 \cdot 1$	$7 \cdot 1$	$7 \cdot 6$
7	Me	\mathbf{H}	Pyrid-2-yl	72	104 - 105	EtOH			11.6	$C_{20}H_{25}N_3O_4$		_	11.3
7	${f Me}$	Η	Thiazol-2-yl	47	121 - 124	E	_		$11 \cdot 1$	C ₁₈ H ₂₃ N ₃ O ₄ S ^e		_	$11 \cdot 1$
7	Me	H	5-Methylthiazol-2-yl	76	130 - 133	EtOH			10.9	$C_{19}H_{25}N_3O_4S^f$			10.7
7	Me	Н	Pyrimid-2-yl	50	142 - 145	EtOH				$C_{19}H_{24}N_4O_4$			
7	${ m Me}$	Н	Pyrazin-2-yl	30	137 - 139	$E-H_2O$			14.6	$C_{19}H_{24}N_4O_4$		_	15.0

^a Solvent for crystn; E=2-ethoxycthanol. ^b Also prepared (53%) from the methyl ester and ammonia. ^c Not crystalline. sed without purification. ^d Found: Cl, 8·2; Reqd.: Cl, 8·3%. ^e Found: S, 8·5; Reqd.: S, 8·5%. ^f Found: S, 8·3; Reqd.: Used without purification. S, 8.2%. 9 Not analysed.

TABLE 4

			4-A	mino-2-ca	arbam	oylphenyl	ethers H_2N		O·[C ·N R	H ₂] _n ·R ′R″				
								Fo	und	(%)		Requ	iired	(%)
				Deriv-	Yield					(707				(707
n	R	$\mathbf{R'}$	R''	ative	(%)	M. p.	Solvent a	C	\mathbf{H}	N	Formula	Ċ	H	N.
5	Me	\mathbf{H}	H	Base	86	9698°	EtOH-H ₂ O	66.5	$8 \cdot 6$	11.5	$C_{13}H_{20}N_{2}O_{2}$	$66 \cdot 1$	8.5	11.9
				$Acetyl^b$		184 - 185	EtOH	64.7	$7 \cdot 7$	10.2	$C_{15}H_{22}N_2O_3$	$64 \cdot 7$	8.0	10.1
7	Me	H	H	Base	91	105-106	EtOH-H,O	67.6	8.8	10.75	$C_{15}H_{24}N_2O_2$	$68 \cdot 1$	$9 \cdot 1$	10.6
9	Me	Н	H	Base	87	103106	EtOH-H.O	70.2	9.7	9.7	$C_{17}^{13}H_{28}N_{2}O_{2}$	69.8	9.7	9.7
5	Phthalimido	Н	H	Base	89	150-152	NMe, CHO-EtOH	65.3	5.8	11.5	$C_{20}^{17}H_{21}^{28}N_{3}O_{4}$	65.4		11.4
•	2			Acetyl b		198	EtOH	64.8		10.1	$C_{22}H_{23}N_3O_5$	64.5		10.3
7	Me	Et	Et	HCl	82	171-174				8.1	C ₁₉ H ₃₂ N ₂ O ₂ ,HCl ^c			7.9
•	1.10			Acetyl b		7475	PhMe-Pet (b. p.	70.0	9.4	7.5	$C_{21}H_{34}N_2O_3$	69.6	9.5	7.7
							60-80°)			• •	021-34-1203	00 0		• •
7	Me		$-[CH_2]_4$	Base	72	d	,	71.2	9.8	8.8	$C_{19}H_{30}N_2O_2$	71.7	9.5	8.8
-			234	Acetyl b		105-106	PhMe-Pet (b. p.	70.1	8.8	8.0	$C_{21}^{13}H_{32}^{30}N_{2}^{2}O_{3}^{2}$	70.0	9.0	7.8
				J			60—80°) ` ¹				21 32 2 3			
7	Me	H	CH.·CO.H	HCl	71	> 310	Aq. HCl			$8 \cdot 1$	C ₁₇ H ₂₆ N ₂ O ₄ ,HCl e			7.8
7	Me	Η	[CH2]2·NEt2	Base	83	7578	Pet. (b. p. 80—			11.6	$C_{21}^{11}H_{37}^{20}N_{3}^{2}O_{3}^{3}$			11.6
•	1/20		[2]22				100°)				-21373-2			
				Acetyl b		168 - 169	EtOH '	68.4	9.7	10.6	$C_{23}H_{39}N_3O_3$	$68 \cdot 1$	9.7	10.4
7	Me	Η	Ph	Base	89	107-109	EtOH	74.2		8.35		74.1		8.2
7	Me	Ĥ	Pyrid-2-yl	Base	88	7779	MeOH-H ₂ O	70.2			$C_{20}H_{27}N_3O_2$	70.3	8.0	12.3
•	2.20		- J J-							•	2027302		- 0	0

^a Solvent for crystn.; Pet = light petroleum. ^b Prepared by Mr. D. L. Pain. ^c Found: Cl, 10·0. Reqd.: Cl, 10·0%. ^d B. p. 210—220°/0·03 mm. ^e Found: Cl, 9·3. Reqd.: Cl, 9·9%.

TABLE 5

Miscellaneous 2-substituted 4-nitrophenyl ethers

			Found (%)						Required (%)			
n	R	\mathbb{R}'	Yield (%)	М. р.	Solvent a	ć	H	N	Formula	ć	H	N
7	Me	Cl	62	$24-25^{\circ}$	Pet (b. p. 40—60°)			4.95	C ₁₄ H ₂₀ ClNO ₃ b			4.9
5	Phthalimido	Cl	46	115 - 116	AcOH				C19H17ClN2O5 6			$7 \cdot 2$
7	Me	NHAc	47	75—78	EtOH	61.6	7.9	$9 \cdot 3$	$C_{16}H_{24}N_{2}O_{4}$	$62 \cdot 3$	7.8	$9 \cdot 1$
7	Me	CN	39	45-47 d	Pet (b. p. 80—100°)	$65 \cdot 4$	7.7	10.2	$C_{15}H_{20}N_2O_3$	$65 \cdot 2$	$7 \cdot 3$	10.1

a Solvent for crystn.; Pet = light petroleum.
 b Found: Cl, 12·5. Reqd.: Cl, 12·4%.
 c Found: Cl, 9·1. Reqd.: Cl, 9·1%.
 d B. p. 165—166°/0·05 mm.

TABLE 6

Miscellaneous 2-substituted 4-aminophenyl ethers Found (%) Required (%) Yield \overline{c} ć $\vec{\mathbf{N}}$ R'М. р. \mathbf{H} Н R Derivative Solvent a N (%) Formula Cl Base 58 b 57—58° MeOH $5 \cdot 2$ $C_{14}H_{22}CINO$ σ 7 Me 5.5C₁₉H₁₉ClN₂O₃ ^d C₁₉H₃₃NO₃, C₂₀H₁₈O₈ C₁₆H₂₆N₂O₂ C₁₇H₂₈N₂O₃ Phthalimido Cl Me CH(OEt)₂ 68 93 - - 94**EtOH** Base 7.85 7 7.8 $66 \cdot 4$ 7.37.2592 - 94 71 - 72 $2 \cdot 2$ 66.0Diptolate * 44 Et_2O $2 \cdot 0$ EtOH-H2O NHAc 80 $\mathbf{68 \cdot 7} \quad \mathbf{9 \cdot 4}$ $69 \cdot 1$ 7 7 MeBase 10.0 9.410.1 CH₂CH(NH₂)·CO₂H Base 75 217 - 219Me NMe₂·CHO 9.2 $9 \cdot 1$ 62^{b} Base 29---31 $11 \cdot 2$ $C_{15}H_{22}N_2O$ Acetyl 83 9 95 - - 9770.8 8.5 9.3 $C_{17}H_{24}N_2O_2$ 70.8 $8 \cdot 4$ 9.77 Me COMe HCl 50 h 120 - 1254.9C₁₆H₂₅NO₂,HCl ⁱ 4.7C₁₈H₂₇NO₃ C₁₇H₂₈N₄OS ^k Acetyl 71^{j} 99-101 EtOH 70.6 9.1 4.770.84.6 118-119 C₆H₆ Thiosemicarbazone

^a Solvent for crystn. ^b Nitro-compound reduced using iron-acetic acid. ^c Found: Cl, 13·65. Reqd.: Cl, 13·9%. ^d Found: Cl, 9·9. Reqd.: Cl, 9·9%. ^e Diptolate = di-p-toluoyl-p-tartrate. ^f B. p. 165—175°/0·02 mm. ^f By acetylation of the amine with acetic anhydride-sodium acetate. ^h By hydrolysis of the acetyl derivative with aq. hydrochloric acid. ^f Found: Cl, 11·4. Reqd.: Cl, 11·8%. ^f From potassium 4-acetamido-2-acetylphenoxide and octyl bromide. ^k Found: S, 9·5. Reqd.: S, 9·5%.

(5.35 g.) in water (7.12 ml.), ethanol (25 ml.), and concentrated hydrochloric acid (1 drop). After 18 hr. sodium hydrogen carbonate (0.72 g.) was added, and the mixture stirred and refluxed for a further 1/4 hr., then filtered hot, and the residue was washed with hot ethanol. The combined solutions were cooled and filtered, and the product was recrystallised from ethanol, giving the amine (58%), m. p. 123-123.5° (Found: C, 62.1; H, 7.3; N, 12.2. $C_{18}H_{25}N_3O_2S$ requires C, 62·2; H, 7·25; N, 12·1%). Similarly prepared were 1-[4-amino-2-(5-methylthiazol-2ylcarbamoyl)phenoxy]-n-octane (61%), m. p. 104-106° (from light petroleum, b. p. 80—100°) (Found: C, 62.9; H, 7.4; N, 11.7. C₁₉H₂₇N₃O₂S requires C, 63.1; H, 7.5; N, 11.6%); 1-[4-amino-2-(pyrimid-2-ylcarbamoyl)phenoxy]n-octane (42%), m. p. 136—139° (from benzene) (Found: C, 66·8; H, 7·2; N, 16·2. $C_{19}H_{26}N_4O_2$ requires C, 66·6; H, 7.65; N, 16.4%); and 1-[4-amino-2-(pyrazin-2-ylcarbamoyl)phenoxy]-n-octane (64%), m. p. 135-137° (from ethanol) (Found: C, 66-7; H, 7-7; N, 16-6. $C_{19}H_{26}N_4O_2$ requires C, 66.6; H, 7.65; N, 16.4%).

4-Acetamido-2-n-octyloxybenzoic acid. A mixture of the potassium salt from 5-acetamido-2-methoxycarbonylphenol ⁵ (23·3 g.) and n-octyl bromide (36 g.) in dimethylformamide (40 ml.) was heated at 150—160° for 30 min., cooled, and poured into water. The oily product, extracted with ether, was dissolved in methanol (150 ml.) and the solution was treated with aqueous 2N-potassium hydroxide (56 ml.). The mixture was heated on a steam-bath (15 min.), cooled, diluted with water (200 ml.), and acidified with hydrochloric acid. The product (55%) was recrystallised from methanol. The ether had m. p. 119—121° (Found: C, 66·4; H, 8·4; N, 4·6. C₁₇H₂₅NO₄ requires C, 66·4; H, 8·2; N, 4·6%).

1-(4-Amino-2-octyloxy)benzoic acid. A solution of 1-(4-acetamido-2-octyloxy)benzoic acid (15·85 g.) in aqueous 2N-potassium hydroxide (77·4 ml.) and water (75 ml.) was refluxed for $6\frac{1}{2}$ hr., filtered, cooled, and made just acid with dilute acetic acid. The product (52%) was crystallised from carbon tetrachloride; the amine had m. p. 67·5—71·5° (Found: C, 68·5; H, 8·8; N, 5·2. $C_{15}H_{23}NO_3$ requires C, 67·9; H, 8·7; N, 5·3%).

5-Acetamido-2-methoxyphenyl methyl sulphone. 5-Acetamido-2-methoxybenzenesulphinic acid 1 (7 g.) was dissolved in a solution of sodium acetate trihydrate (3·7 g.) in water (10 ml.), and methyl iodide (5 ml.) in acetone (50 ml.) was added. The mixture was refluxed for 2 hr., concentrated, and cooled, yielding the sulphone (74%), m. p. 216—218° (Found: N, 5·7; S, 13·3. $C_{10}H_{13}NO_4S$ requires N, 5·8; S, $13\cdot2\%$).

5-Amino-2-methoxybenzenesulphinic acid hydrochloride A solution of 5-acetamido-2-methoxybenzenesulphinic acid (6·4 g.) in 11·7n-potassium hydroxide (6 ml.) and water (10 ml.) was refluxed for 3 hr., cooled, and acidified with concentrated hydrochloric acid, giving the salt (100%), m. p. 205—215° (decomp.) (Found: Cl, 16·1; N, 5·6. $C_7H_9NO_3S$, HCl requires Cl, 15·9; N, 6·3%).

5-Amino-2-methoxyphenyl methyl sulphone. 5-Acetamido-2-methoxyphenyl methyl sulphone (3 g.) was refluxed for 2 hr., in a mixture of concentrated hydrochloric acid (7.5 ml.) and water (7.5 ml.). The basic product (56%) was crystallised from ethanol, yielding the amine, m. p. $103-104^{\circ}$ (Found: N, 6.75; S, $16\cdot3$. $C_8H_{11}NO_3S$ requires N, 7.0; S, $15\cdot9\%$).

o-Methoxyphenyl methyl sulphone. The foregoing amine was diazotised (sodium nitrite-aqueous hydrochloric acid)

and the diazonium solution was reduced with hypophosphorous acid, giving the sulphone (54%), m. p. 88—90° (Found: C, 51·85; H, 5·65; S, 17·3. Calc. for $C_8H_{10}O_3S$: C, 51·6; H, 5·4; S, 17·2%), not depressed by a sample prepared by methylation of o-methoxybenzenesulphinic acid (lit., 6 m. p. 95°).

5-Amino-2-hydroxyphenyl methyl sulphone hydrobromide. 5-Acetamido-2-methoxyphenyl methyl sulphone (29.5 g.) was refluxed for 24 hr. with 50% w/v hydrobromic acid (90 ml.). The solution was cooled giving the salt (84%), m. p. 160—170° (Found: N, 5·3; S, 12·0. C₇H₉NO₃S,HBr requires N, 5.2; S, 11.9%). Acetylation with acetic anhydride-sodium acetate gave 5-acetamido-2-acetoxyphenyl methyl sulphone, m. p. 172-175° (Found: C, 48.8; H, 4.8; S, 12.0. $C_{11}H_{13}NO_5S$ requires C, 48.7; H, 4.8; S, 11.8%). 5-Acetamido-2-hydroxyphenyl methyl sulphone. Crude 5-acetamido-2-acetoxyphenyl methyl sulphone (15.5 g.) was warmed and stirred for 1 hr. with 2N-sodium hydroxide (100 ml.); the mixture was filtered and the filtrate was acidified with hydrochloric acid, giving the phenol (28%), m. p. 223-224° (Found: C, 47·1; H, 4·4; N, 5·8. $C_9H_{11}NO_4S$ requires C, 47.15; H, 4.8; N, 6.1%). The overall yield from 5-acetamido-2-methoxyphenyl methyl sulphone, without purification of the intermediates, was

1-(4-Acetamido-2-methylsulphonylphenoxy)-n-octane. 5-Acetamido-2-hydroxyphenyl methyl sulphone (2·5 g.), anhydrous potassium carbonate (2·5 g.), n-octyl bromide (1·8 g.), and ethanol (10 ml.) were refluxed for 18 hr. with stirring on the steam-bath. The product was recrystallised from aqueous ethanol, yielding the octyl ether (78%), m. p. $106-107^{\circ}$ (Found: N, 4·0; S, 9·4. $C_{17}H_{27}NO_4S$ requires N, 4·1; S, 9·4%). Hydrolysis with aqueous hydrochloric acid afforded 1-(4-amino-2-methylsulphonylphenoxy)-n-octane (91%), m. p. $68-70^{\circ}$ (from aqueous methanol) (Found: N, 4·7; S, $11\cdot0$. $C_{15}H_{25}NO_3S$ requires N, 4·7; S, $10\cdot7\%$).

1-(4-Dimethylamino-2-carbamoylphenoxy)-n-octane, m. p. 78—79 [from light petroleum (b. p. 80—100°)] (Found: C, 70·2; H, 10·2; N, 9·6. $C_{17}H_{28}N_2O_2$ requires C, 69·8; H, 9·6; N, 9·6%) was prepared (42%) by pyrolysis of the methiodide, m. p. 110—120° (decomp.) (from acetone), itself obtained (62%) from the primary amine and methyl iodide-sodium carbonate in ethanol.

5-Acetamido-2-octyloxybenzamidoxime. Sodium hydride (50% suspension in oil, 5·05 g.) was added cautiously to 2-ethoxyethanol (50·5 ml.) cooled in ice, followed by a solution of hydroxylamine hydrochloride (7·71 g.) in 2-ethoxyethanol (150 ml.) and 5-acetamido-2-octyloxyphenyl cyanide (Table 6; 30·3 g.). The mixture was heated at 100° for 18 hr., and evaporated in vacuo. The residue was treated with water and the product, isolated with ether, was crystallised twice from ethyl acetate, giving the amidoxime (66%), m. p. 145—147° (Found: C, 63·8; H, 8·4; N, 12·9. $C_{17}H_{27}N_3O_3$ requires C, 63·5; H, 8·5; N, 13·1%). The material insoluble in ethyl acetate was 5-acetamido-2-octyloxybenzamide, m. p. 194—196° (Found: C, 66·9; H, 8·9; N, 9·1. $C_{17}H_{26}N_2O_3$ requires C, 66·6; H, 8·5; N, 9·15%).

5-Acetamido-2-octyloxybenzamidine. A solution of the foregoing amidoxime (9·3 g.) in methanol (200 ml.) was hydrogenated over Raney nickel at $60^\circ/70$ lb. per sq. in. The residue obtained on evaporation was crystallised from aqueous hydrochloric acid, yielding the amidine hydrochloride (73%), m. p. 90—95° (Found: N, 12·2. $C_{17}H_{27}N_3O_2$,HCl

⁶ M. E. Heppenstall and S. Smiles, J. Chem. Soc., 1938, 899.

Org. 2201

requires N, 12·3%). Hydrolysis with hydrochloric acid afforded 5-amino-2-octyloxybenzamidine dihydrochloride (90%), m. p. 179—183° (Found: N, 12·2; Cl, 21·0. $\rm C_{15}H_{25}N_3O$,2HCl requires N, 12·5; Cl, 21·2%).

3-Methoxy-4-n-nonyloxyaniline. The corresponding nitrocompound, previously described (Part IV²) as an oil, has now been crystallised, m. p. 47·5—49° (Found: C, 65·3; H, 8·6; N, 4·8. Calc. for $C_{16}H_{25}NO_4$: C, 65·1; H, 8·5; N, 4·75%). The amine formed on reduction had m. p. 65·5—67° (lit.,² 71—73°) (Found: C, 72·8; H, 10·2; N, 5·5. Calc. for $C_{16}H_{27}NO_2$: C, 72·5; H, 10·2; N, 5·3%).

THE RESEARCH LABORATORIES,
MAY AND BAKER LTD., DAGENHAM,
ESSEX. [6/780 Received, June 23rd, 1966]