

Bioorganic & Medicinal Chemistry Letters 12 (2002) 2291-2294

Design and Synthesis of Orally Bioavailable Inhibitors of Inducible Nitric Oxide Synthase. Part 1: Synthesis and Biological Evaluation of Dihydropyridin-2-imines

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Received 19 April 2002; accepted 7 June 2002

Abstract—Dihydropyridin-2-imines were synthesized and biologically evaluated both in vitro and in vivo using a nitric oxide inhibition assay. Compounds 1, 4, 5 and 7–11 exhibited potent activity in the inducible nitric oxide (iNOS) inhibition assay. Of these 5, 6, 9 and 10 showed 5- to 11-fold increases in isoform selectivity. Compounds 1, 5, 9 and 10 showed potent inhibitory activity in the NOx accumulation assay in mice. Compounds 1 and 5 also showed good bioavailability (BA) when given orally. \bigcirc 2002 Elsevier Science Ltd. All rights reserved.

Introduction

Nitric oxide (NO) is a mediator of normal and pathophysiological processes.^{1–3} NO is synthesized by two major classes of nitric oxide synthase (NOS): constitutive and inducible. The constitutive enzymes require $Ca^{2+}/calmodulin$ for their activity and are further divided into the neuronal and endothelial isoforms. The endothelial isoform (eNOS) is found predominantly in the vascular endothelium and it produces low concentrations of NO, which reduce the blood pressure and inhibit platelet aggregation. NO produced by the neuronal enzyme (nNOS) appears to function as a neurotransmitter which regulates neuronal transmission. The inducible isoform is found in activated macrophages as well as many other cells and it produces NO, which plays a role in the host defense system. The overexpression of NO which destroys functional tissue during acute and chronic inflammation is caused by this inducible NOS (iNOS). Identification of potent and selective inhibitors for the above-mentioned iNOS has been the subject of intense interest because of their therapeutic potential in the treatment of diseases mediated by

the overexpression of NO.^{4,5} The natural substrate for NOS, arginine, has been the obvious basis for the molecular design of NOS inhibitors.

Structural analogues of L-arginine such as NG-methyl-L-arginine (L-NMA),^{6,7} N^G-nitro-L-arginine,⁸ N-imino-ethyl-L-ornithine (L-NIO),⁹ and L-thiocitrulline¹⁰ have been shown to inhibit the various forms of NOS. Much pharmacology and biology data have been determined utilizing some of the early relatively nonspecific inhibitors of arginine analogues as described above. A number of non-amino acid inhibitors of iNOS such as aminoguanidine,¹¹ isothioureas,¹² 2-amino-5,6-dihydro-6-methyl-4*H*-1,3-thiazine,¹³ 2-iminopiperidine¹⁴ and 2-aminopyridine¹⁵ have also been described. As illustrated in Chart 1, our molecular design started with a structural hybridization of 2-amino-4-methylpyridine 13^{15} and (4R)-4-methylpiperidin-2-imine $14.^{14}$ This modification led us to the discovery of 4-methyl-5,6dihydropyridin-2-imine 1, which is structurally new and possesses no asymmetric center. A series of 5,6-dihydropyridin-2-imines were synthesized and biologically evaluated with the expectation of identifying a new selective inhibitor of human iNOS (hiNOS). We now report a series of 5,6-dihydropyridin-2-imines which are structurally new inhibitors of iNOS.

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Chart 1. Molecular design of 5,6-dihydropyridin-2-imine.

Table 1. Inhibitory activities of 1–12 against hiNOS and heNOS andtheir isoform selectivity e

Structure	Compd (synthetic method)	hiNOS IC ₅₀ (µM) ^a	$\begin{array}{c} he NOS \ IC_{50} \\ (\mu M)^a \end{array}$	Selectivity heNOS/hiNOS
R N N H	1 $R = Me$ (A) 2 $R = Et^{c}$	0.20 1.2	0.23 1.5	1.2 1.3
N N H	3°	0.70	0.44	0.6
NH H	4 (B)	0.31	0.32	1.0
R N N H	5 R = Me (B) 6 R = H (C)	0.42 2.8	2.1 31.0	5.0 11.1
NH H	7 (B) ^d	0.51	0.52	1.0
R N NH	8 R = Me (A) ^d 9 R = n -Pr (A) ^d 10 R = Allyl (A) ^d	0.18 0.06 0.05	0.24 0.58 0.36	1.3 9.7 7.2
	11 $n = 1$ (A) ^d 12 $n = 2$ (A) ^d	0.17 1.4	0.51 7.4	3.0 5.3

^aPreparation of partially purified enzyme and determination of K_i values. Human eNOS was overexpressed in Sf-21 cells, by infecting the cells with baculovirus carrying heNOS cDNA. The hiNOS was overexpressed in A549 by stimulation with LPS (10 µg/mL) plus cytokines (10 ng/mL TNF- α , 5 ng/mL IL-1 β and 100 ng/mL interferon- γ). Human eNOS and iNOS were partially purified by chromatography on 2', 5'-ADP-Sepharose gels. NOS activity was determined by the method for the conversion of [¹⁴C]-L-arginine to L-citrulline with a minor modification. The conversion rates for various concentrations of the test compounds and L-arginine were measured. Dixon and Lineweaver–Bulk plots were constructed to determine the K_i values and the mode of inhibition.

^bSelectivity was evaluated as the rate of the IC_{50} values for heNOS and hiNOS. ^cCompounds **2** and **3** were prepared according to a modified Birch reduction of 2-amino-4-methylpyridine and 2-amino-4-ethylpyridine, respectively.¹⁶ ^dSynthesized as a mixture of racemates.

eAll compounds were prepared as their hydrochlorides

Chemistry

Synthesis of the compounds described in Table 1 is outlined in Schemes 1–3. As described in Scheme 1, the common intermediates, substituted 5,6-dihydropyridin-2-ones 18a-i, were prepared by the conventional Beckmann rearrangement reaction of oximes prepared from the corresponding cyclopentenones 15a-i via 16a-i and 17a-i (Method A: 22-48% yield) or directly from 15a-i (Method B: 44-60% yield), respectively. The common intermediates 18a-i were used to prepare 1 and 8-12 as well as 4-5 and 7 according to Methods A and B, respectively. As described in Scheme 2, another intermediate 21 used to prepare 6 was obtained by the oxidation of 20 with a 70% yield, which was prepared by the Beckmann rearrangement of an oxime prepared from the cyclopentanone 19 with a 52% yield (Method C). As described in Scheme 3, 18a-i and 21 were converted to the test compounds 1 and 4–12 (8–70% yield) via their corresponding ethyl imidates 22a-i and 23, respectively.

Results and Discussion

As described in Table 1, the compounds were evaluated for their ability to inhibit the two isoforms of human NOS. As expected, 4-methyl-5,6-dihydropyridin-2-imine 1 demonstrated potent inhibitory activity against both the human iNOS (hiNOS) and human eNOS (heNOS) isoforms without any selectivity.



Scheme 1. Synthesis of 5,6-dihydropyridin-2-ones. Method A: (a) NH₂OH·HCl, NaOAc, MeOH, reflux; (b) *p*-TsCl, pyridine, 0° C; (c) concd HCl, MeOH, refux (18a, 18e–i: 22–48% in three steps); Method B: (d) NaN₃, TFA, reflux (18b–d: 44–60%).

Compd	\mathbf{R}_1	R_2	R ₃	R_4	R_5
a	Н	Me	Н	Н	Н
b	Me	Me	Н	Н	Н
с	Н	Me	Me	Me	Н
d	Н	Me	(H,	Me)	Н
e	Н	Me	Н	Н	Me
f	Н	Me	Н	Н	<i>n</i> -Pr
g	Н	Me	Н	Н	Allyl
ĥ	Н	Me	Н	-(C	H ₂) ₃ -
i	Н	Me	Н	-(C	H ₂) ₄ -



Scheme 2. Synthesis of 5,5-dimethyl-5,6-dihydropyridin-2-one (21). Method C: (a) NH₂OSO₃H, HCO₂H, reflux (52%); (b) DDQ, BSTFA, dioxane, reflux (70%).



Scheme 3. Synthesis of 5,6-dihydropyridine-2-imines (1, 4–12) Reagents: (a) $Et_3O^+BF_4^-$, CH_2Cl_2 ; (b) NH_3 , EtOH; (c) 4 M HCl dioxane (8–70% in three steps)

Compd	\mathbf{R}_1	R_2	R ₃	R_4	R ₅
a	Н	Me	Н	Н	Н
b	Me	Me	Н	Н	Н
c	Н	Me	Me	Me	Н
d	Н	Me	(H,Me)		Н
e	Н	Me	Н	Н	Me
f	Н	Me	Н	Н	<i>n</i> -Pr
g	Н	Me	Н	Н	Allyl
ĥ	Н	Me	Н	-(C	CH ₂) ₃ -
i	Н	Me	Н	-(C	CH ₂) ₄ -

Replacement of the 4-methyl group with a 4-ethyl group afforded 2^{16} with a nearly 6-fold reduction in the inhibitory activity and still no isoform selectivity. Shifting the double bond of 1 from position 3,4- to 4,5- provided 3^{16} with reduced inhibitory activity against both of the isoforms as well as a reduced selectivity. Based on this information, further chemical modification was carried out by fixing the double bond in position 3,4- as illustrated in compounds 4-12. Introduction of another methyl group into position 3 of 1 afforded 4 with little change in the inhibitory activity or isoform selectivity. Introduction of a *gem*-dimethyl group into position 5 of 1 provided 5 with a 2-fold reduction in the inhibitory activity against hiNOS and 10-fold reduction against heNOS. As a result, the isoform selectivity was 5 times more than that of 1. Removal of the 4-methyl group from 5 afforded 6 with a 14-fold reduction in the potency of the hiNOS inhibition and a 135-fold reduction in the heNOS inhibition. As a result, the isoform selectivity (heNOS/hiNOS) of 6 increased to nearly 10 times greater than that of 1. Introduction of another methyl group into position 5 of 1 afforded 7 with a nearly 2-fold reduction in potency against both the hiNOS and heNOS isoforms but with the isoform selectivity retained. Introduction of another methyl group into position 6 of 1 afforded 8 which retained the inhibitory activity for both isoforms as well as the selectivity. Replacement of the 6-methyl group of 8 with a 6-propyl group afforded 9 with a 3-fold increase in potent inhibitory activity against hiNOS and a 2.4-fold reduction against heNOS resulting in an isoform selectivity which was 7.5-fold higher than 8. A similar result was obtained in the corresponding allyl derivative **10**.

Finally, two of the *cis*-fused bicyclic analogues were prepared and biologically evaluated as inhibitors of NOS. The *cis*-fused 5,6-bicyclic analogue 11 demonstrated nearly the same potency of hiNOS inhibition as compared with 1 but with a 2.2-fold decrease in the potency of heNOS inhibition, while the isoform selectivity of 11 was higher than that of 1. This was also true for the *cis*-fused 6,6-bicyclic analogue 12, although its IC_{50} values for the isoforms were much less potent than those of 11.

Among the tested compounds described in Table 1, the compound 1, which is a structurally representative compound of this series with potent iNOS inhibitory activity was selected together with 5, 9 and 10 for further biological and pharmacodynamic evaluations in animal models because of their in vitro potency and isoform selectivity, as well as their mouse acute toxicity (MTD: maximum tolerated dose) and structural features. Compounds 9 and 10 exhibited greater potency and more isoform selectivity than the above-mentioned compounds 1 and 5. As shown in Table 2, compounds 9 and 10 demonstrated higher acute toxicity than 1 and 5, as evaluated by the ratio between their maximum tolerated doses (MTDs) and the ID₅₀ values of NOx¹⁷ accumulation (MTD/NOx).

In order to assess compounds 1, 5, 9 and 10 for their ability to inhibit iNOS in vivo, mice were given a single compound subcutaneously (sc) 3 h after lipopoly-saccharide (LPS) injection. The plasma NOx accumulation from 3 to 6 h after the LPS injection was then determined. As described in Table 2, these test compounds 1, 5, 9 and 10 inhibited NOx accumulation in plasma and their ID₅₀ values were 0.16, 0.28, 0.36 and 0.16 mg/kg, sc, respectively. To assess the acute toxicity, the MTDs were determined. As shown in Table 2, the MTDs of 1 and 5, when a single intravenous (iv) dose

 Table 2.
 Pharmacological evaluation of 1, 5, 9 and 10

Compd	Mouse iNOS	NOx	MTD	MTD/NOx
	$\frac{IC_{50}}{(\mu M)^a}$	$\frac{ID_{50}}{(mg/kg, sc)^b}$	mg/kg, iv ^c	
1	0.022	0.16	20	130
5	0.10	0.28	20	71
9	0.10	0.36	5	14
10	0.025	0.16	5	31
l-NMMA	3.5	26	3000	120

^aEnzyme assay with recombinant mouse iNOS. Recombinant mouse iNOS was purchased from Cayman Chemical (Cat. No. 60862) and the inhibitory activities of the test compounds were evaluated by measuring the conversion rate from [14C]-L-arginine to [14C]-L-citrulline, and then the IC₅₀ values were determined.

^bThe ID₅₀ value was determined from a log-logit transformation of the dose-response curves (1, 5, 9 and 10; 0.1, 0.3, 1 mg/kg, sc L-NMMA; 10, 30 and 100 mg/kg, sc). The ID₅₀ value was defined as the dose of test compound that produced a 50% inhibition in the NOx accumulation induced by LPS treatment alone. The MTD was defined as the maximum dose at which no death was observed within 24 h after an intravenous injection administration. The doses used were 5, 10, 20, 30, 40, and 50 mg/kg for 1, 5, 9 and 10 and 1000, 2000, 3000, 4000, and 5000 mg/kg for L-NMMA.

^cInhibition of NOx accumulation and the maximum tolerated dose (MTD) in mice. The test compounds or saline were administered subcutaneously 3 h after the LPS (10 mg/kg, iv) injection into 7 week old Balb/c mice (Charles River Japan, Inc.). Blood was collected by venipuncture from the abdominal aorta under light anesthesia at 6 h after LPS treatment. Plasma was obtained by centrifugation and the concentration of accumulated NOx over 3 h was determined by the method described below. To evaluate the acute toxicity, the MTD (iv maximum dose where no death was observed within 24 h after the administration) of the test compound was determined.

was given to normal mice, were 20 and 20 mg/kg, respectively. The MTD values of all four compounds were lower than that of L-NMMA (3000 mg/kg), although the ratio of MTD/ID₅₀ for NOx accumulation in mice was 130 for 1, 71 for 5, 14 for 9 and 31 for 10, respectively. The oral bioavailability of 1 and 5 in rats was excellent (1: 88% and 5: 84%).

In summary, we have explored the structure-activity relationship (SAR) for a series of substituted 5,6-dihydropyridin-2-imines and obtained significant increases in the isoform selectivity and potency of iNOS inhibition. 4-Methyl and 4-methyl-5,5-dimethyl substituents provided compounds 1 and 5 which were potent and/or selective inhibitors of the iNOS isoform. Compounds 9

and 10 exhibited greater potent inhibitory activity and isoform selectivity than those of 1 and 5 but their acute toxicity in mice was also higher. Full details will soon be reported in *Bioorganic Medicinal Chemistry*.

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