

Initiator-Dependent Chemoselective Addition of THF Radical to Aldehyde and Aldimine and Its Application to a Three-Component Reaction

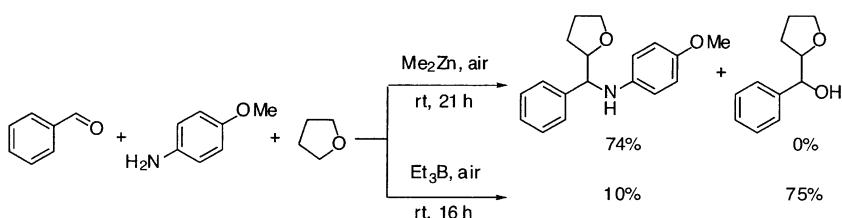
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ABSTRACT



The product distribution of the three-component reaction of aldehydes, arylamines, and THF was dependent on a radical initiator, preferentially giving the corresponding THF adducts of imines with dimethylzinc and adducts of aldehyde with triethylborane.

Chemoslectivity has been one of the goals of organic chemistry. Especially differentiation of C=O and C=N double bonds as nucleophilic targets has been the topic of recent allylation¹ and Mannich-type reactions.² However, to the best of our knowledge, there has been no report on such differentiation for radical reactions.³ We describe herein that a THF radical⁴ undergoes chemoselective addition to aldehydes and aldimines depending on the choice of radical-generating agents, dimethylzinc and triethylborane. The chemoselectivity was well demonstrated in the three-component reaction of aldehydes, amines, and THF.

(1) (a) Akiyama, T.; Iwai, J.; Onuma, Y.; Kagoshima, H. *Chem. Commun.* **1999**, 2191–2192. (b) Akiyama, T.; Iwai, J. *Synlett* **1998**, 273–274. (c) Nakamura, H.; Iwama, H.; Yamamoto, Y. *J. Am. Chem. Soc.* **1996**, 118, 6641–6647. (d) Nakamura, H.; Iwama, H.; Yamamoto, Y. *J. Chem. Soc., Chem. Commun.* **1996**, 1459–1460.

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(3) For initiator-dependent chemoselective radical addition to C=O and C=C double bonds, see: Devin, P.; Fensterbank, L.; Malacria, M. *Tetrahedron Lett.* **1999**, 40, 5511–5514.

We have recently reported radical addition reaction of ethers with aldimines, in which methyl radicals, generated from dimethylzinc as an initiator and molecular oxygen, abstract an α -hydrogen of ethers to generate ether radicals,

(4) For the reaction of an ether radical, see: (a) Yoshimitsu, T.; Arano, Y.; Nagaoka, H. *J. Org. Chem.* **2003**, 68, 625–627. (b) Hirano, K.; Sakaguchi, S.; Ishii, Y. *Tetrahedron Lett.* **2002**, 43, 3617–3620. (c) Mosca, R.; Fagnoni, M.; Mella, M.; Albini, A. *Tetrahedron* **2001**, 57, 10319–10328. (d) Kim, S.; Kim, N.; Chung, W.-J.; Cho, C. H. *Synlett* **2001**, 937–940. (e) Hirano, K.; Iwahama, T.; Sakaguchi, S.; Ishii, Y. *Chem. Commun.* **2000**, 2457–2458. (f) Iwahama, T.; Sakaguchi, S.; Ishii, Y. *Chem. Commun.* **2000**, 613–615. (g) Yoshimitsu, T.; Tsunoda, M.; Nagaoka, H. *Chem. Commun.* **1999**, 1745–1746. (h) Inoue, A.; Shinokubo, H.; Oshima, K. *Synlett* **1999**, 1582–1584. (i) Alves, M. J.; Gilchrist, T. L.; Sousa, J. H. *J. Chem. Soc., Perkin Trans. 1* **1999**, 1305–1310. (j) Xiang, J.; Evarts, J.; Rivkin, A.; Curran, D. P.; Fuchs, P. L. *Tetrahedron Lett.* **1998**, 39, 4163–4166. (k) Xiang, J.; Jiang, W.; Fuchs, P. L. *Tetrahedron Lett.* **1997**, 38, 6635–6638. (l) Xiang, J.; Jiang, W.; Gong, J.; Fuchs, P. L. *J. Am. Chem. Soc.* **1997**, 119, 4123–4129. (m) Xiang, J.; Fuchs, P. L. *J. Am. Chem. Soc.* **1996**, 118, 11986–11987. (n) Gong, J.; Fuchs, P. L. *J. Am. Chem. Soc.* **1996**, 118, 4486–4487. (o) Clark, A. J.; Rooke, S.; Sparey, T. J.; Taylor, P. C. *Tetrahedron Lett.* **1996**, 37, 909–912. (p) Ishida, A.; Sugita, D.; Itoh, Y.; Takamuku, S. *J. Am. Chem. Soc.* **1995**, 117, 11687–11694. (q) Jung, J. C.; Choi, H. C.; Kim, Y. H. *Tetrahedron Lett.* **1993**, 34, 3581–3584. (r) Fontana, F.; Minisci, F.; Yan, Y. M.; Zhao, L. *Tetrahedron Lett.* **1993**, 34, 2517–2520. (s) Matthews, D. P.; McCarthy, J. R. *J. Org. Chem.* **1990**, 55, 2973–2975. (t) Gevorgyan, V.; Priede, E.; Liepins, E.; Gavars, M.; Lukevics, E. *J. Organomet. Chem.* **1990**, 393, 333–338. (u) Russell, G.

consequently giving the ether adducts of aldimines in high yields.^{5–7} For example, the reaction of benzaldehyde *N*-4-methoxyphenylimine (*N*-PMP-imine) **1** with THF was conducted in the presence of 6 equiv of dimethylzinc under constant air bubbling (10 mL/h), producing THF adduct **3a** in 83% yield after 5 h at room temperature (Table 1, entry

Table 1. Initiator-Dependent Chemoselective Addition of THF Radical to Benzaldehyde PMP-imine **1** and Benzaldehyde **2a**

entry	1/2a	initiator (equiv)	t (h)	product	yield (%)
1	1	Me ₂ Zn (6)	5	3a	83
2	1	Et ₃ B (12)	18	3a ^a	63
3	2a	Me ₂ Zn (12)	120	4	<8
4	2a	Et ₃ B (12)	6	4	81

^a An ethyl adduct of **1** was obtained in 26% yield.

1).⁸ As a logical extension of our approach, we then examined the reaction of benzaldehyde **2a** as a radical acceptor, instead of aldimine **1**. However, the addition reaction of a THF radical with **2a** was sluggish under the dimethylzinc–air conditions, producing THF adduct **4** in less than 8% yield even after 5 days (entry 3). This poor efficiency is in dramatic contrast to the reaction of **2a** with the use of triethylborane–air as a radical initiator (a modified Nagaoka procedure),^{4a,g} giving **4** in 81% yield after 6 h (entry 4).^{9,10} It is also important to note that the reaction of **1** under the triethylborane (12 equiv)–air conditions gave **3a** in 63% yield after prolonged reaction time of 18 h (entry 2).¹¹

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(6) For radical reactions with dialkylzinc, see: (a) Bazin, S.; Feray, L.; Siri, D.; Naubron, J.-V.; Bertrand, M. *P. Chem. Commun.* **2002**, 2506–2507. (b) Bertrand, M. P.; Feray, L.; Gastaldi, S. *C. R. Chim.* **2002**, *5*, 623–638. (c) Bertrand, M. P.; Coantic, S.; Feray, L.; Nouguier, R.; Perfetti, P. *Tetrahedron* **2000**, *56*, 3951–3961. (d) Miyabe, H.; Konishi, C.; Naito, T. *Org. Lett.* **2000**, *2*, 1443–1445. (e) Miyabe, H.; Ushiro, C.; Ueda, M.; Yamakawa, K.; Naito, T. *J. Org. Chem.* **2000**, *65*, 176–185. (f) Bertrand, M. P.; Feray, L.; Nouguier, R.; Perfetti, P. *J. Org. Chem.* **1999**, *64*, 9189–9193. (g) Ryu, I.; Araki, F.; Minakata, S.; Komatsu, M. *Tetrahedron Lett.* **1998**, *39*, 6335–6336.

(7) For reviews of radical addition to C=N bonds, see: (a) Ishibashi, H.; Sato, T.; Ikeda, M. *Synthesis* **2002**, 695–713. (b) Friestad, G. K. *Tetrahedron* **2001**, *57*, 5461–5496. (c) Naito, T. *Heterocycles* **1999**, *50*, 505–541. (d) Fallis, A. G.; Brinza, I. M. *Tetrahedron* **1997**, *53*, 17543–17594.

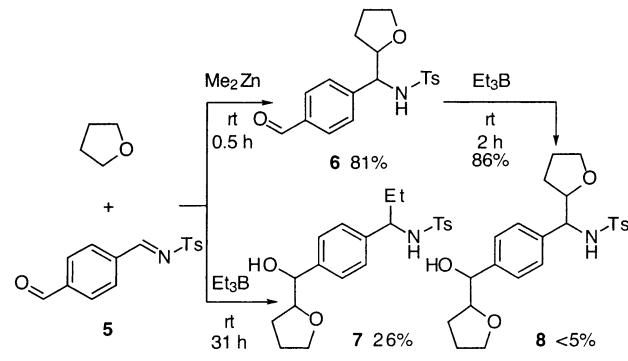
(8) Diastereomer ratios of the products described in this paper were generally between 3:2 and 1:1 except for **3g** (34:66), **4** (84:16), and **7** (7:7:43:43).

(9) For reviews of radical reaction with organoborane, see: (a) Yorimitsu, H.; Shinokubo, H.; Oshima, K. *Synlett* **2002**, 674–686. (b) Olliver, C.; Renaud, P. *Chem. Rev.* **2001**, *101*, 3415–3434.

(10) For radical addition to an aldehyde with triethylborane–air, see: (a) Charayron, M.; Devin, P.; Fensterbank, L.; Malacria, M. *Synlett* **2000**, 83–85. (b) Devin, P.; Fensterbank, L.; Malacria, M. *Tetrahedron Lett.* **1998**, *39*, 833–836. See also refs 3, 4a, and 4g.

The initiator-dependent chemoselectivity was more highlighted by the reaction of imino-aldehyde **5** having both carbonyl and imino groups (Scheme 1). Under the dimethyl-

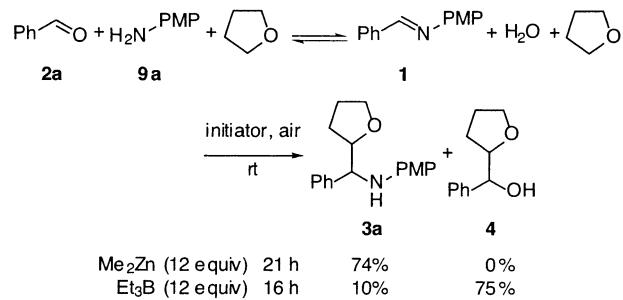
Scheme 1. Reaction of THF with Imino-aldehyde **5**



ylzinc–air conditions, the THF radical preferentially reacted with the imino group to provide mono-THF adduct **6** in 81% yield, and no other products resulting from the reaction of THF with the aldehyde moiety were observed. In contrast, under the triethylborane conditions, the addition of THF occurred mainly to the carbonyl group, giving **7** as a major product in 26% yield and di-THF adduct **8** in less than 5% yield. Mono-THF adduct **6** further underwent the addition reaction with THF under the triethylborane–air conditions, giving **8** in 86% yield.

The chemoselectivity was more clearly demonstrated by the reaction of a THF solution of a mixture of **2a** and *p*-anisidine **9a**, those establishing equilibrium with an imine and water (Scheme 2). A solution of 1 mmol each of **2a** and

Scheme 2. Chemoselective THF Radical Addition Reaction of an Equilibrium Mixture of **2a** and **9a**



9a in 22 mL of THF was treated with 12 mmol of an initiator (1.0 M solution in hexane) under a continuous air stream (0.5 mL/h) at room temperature. By the dimethylzinc

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initiation, THF-imine adduct **3a** was preferentially isolated in 74% yield after 21 h, whereas THF-aldehyde adduct **4** was obtained in 75% yield along with **3a** (10%) and **9a** (65%) after 16 h under the triethylborane initiation. It is also interesting to note that the reaction of the equilibrium mixture in the presence of both initiators was not clean and gave a mixture of **3a** (13%), **4** (12%), ethyl adduct of aldimine (7%), and aldimine **1** (56% recovery) after 12 days.

The utility of the imine-selective reaction with the use of dimethylzinc initiator was further demonstrated by the following three-component reaction of various aldehydes, arylamines, and THF (Table 2).¹² A mixture of benzaldehyde

robenzaldehyde **2b** and **9a** and **3e** in 57% yield from *p*-anisaldehyde **2c** and **9a** (entries 4 and 5). Acyclic and cyclic aliphatic aldehydes **2d** and **2e** were also applicable, giving **3f** and **3g** in 50 and 44% yields, respectively (entries 6 and 7). Thus, the three-component reaction can be generalized.

The initiator-dependent chemoselective addition reaction can be generalized and is interesting. Initiation by dimethylzinc favors THF-imine addition, and triethylborane initiation provides a THF-aldehyde addition product. Mechanistic details for the origin of the chemoselectivity are the focus of our ongoing study.

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Supporting Information Available: Experimental procedures and characterization data of the products. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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entry	2	R ¹	9	R ²	t (h)	3	yield (%)
1	2a	Ph	9a	PMP	22	3a	74
2	2a	Ph	9b	Ph	16	3b	66
3	2a	Ph	9c	4-CF ₃ C ₆ H ₄	22	3c	47
4	2b	4-ClC ₆ H ₄	9a	PMP	22	3d	64
5	2c	PMP	9a	PMP	45	3e	57
6	2d	PhCH ₂ CH ₂	9a	PMP	46	3f	50
7	2e	c-C ₆ H ₁₁	9a	PMP	138	3g	44

2a with aniline **9b** and 4-trifluoromethylaniline **9c** in THF provided THF adduct **3b** in 66% yield and THF adduct **3c** in 47% yield (entries 2 and 3). The reaction also proceeded smoothly whether aldehydes had an electron-withdrawing or -donating group, giving **3d** in 64% yield from 4-chloro-

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