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Highly Enantioselective Total Synthesis of (+)-Isonitramine

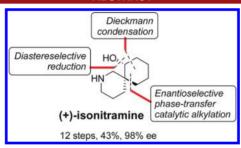
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ABSTRACT



A new efficient enantioselective synthetic method of (+)-isonitramine is reported. (+)-Isonitramine was obtained in 12 steps (98% ee and 43% overall yield) from δ -valerolactam via enantioselective phase-transfer catalytic alkylation, Dieckman condensation, and diastereoselective reduction as key steps.

There are a number of biologically important natural products¹ containing optically active 2-azaspirocycle structures, such as polyzonimine, ^{1a} nitropolyzonamine, ^{1a} horsfiline, ^{1b} and spirotryprostatin B. ^{1c} Among these compounds, *Nitraria* alkaloids, ²(+)-nitramine (1),

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(+)-isonitramine (2), and (-)-sibirine (3), have chiral quaternary carbon centers on the 2-azaspiro[5,5]undecane-7-ol skeletons. The structural similarity with the neurotoxic (-)-histrionicotoxin has faciliated the development of efficient synthetic routes of *Nitraria* alkaloids (Figure 1). Recently, it was reported that the extracts of *Nitraria* plants have antiproliferative effects on cancer cell lines through the apoptosic pathway.³

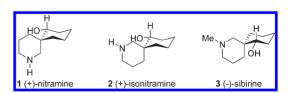


Figure 1. Nitraria alkaloids.

Although numerous enantioselective synthetic methods for *Nitraria* alkaloids have been reported thus far, ⁴ most of these approaches employed chiral auxiliaries, ^{4a,d,j} chiral substrates, ^{4b,g} or enzymatic resolution, ^{4e,f,h,l} which cannot be readily applied to large scale production. In addition,

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only one reported catalytic method uses an organopalladium catalyst for the establishment of quaternary carbon center (6*R*); however, the enantioselectivity is only 86% ee. ⁴ⁱ As one of our research programs for the development of new therapeutics for the treatment of cancer, we need to develop very efficient and highly enantioselective synthetic methods for *Nitraria* alkaloids. Herein, we present a novel approach to synthesize (+)-isonitramine (2), a representative *Nitraria* alkaloid, via asymmetric phase-transfer catalytic (PTC) alkylation, which is regarded as one of the most efficient synthetic processes for large scale production from the view points of economical and environmental aspects.⁵

Scheme 1. Enantioselective PTC α -Alkylation of Lactams

Very recently, we reported on the highly enantioselective PTC alkylations of α -tert-butoxycarbonylactams (Scheme 1). The alkylation of N(1)-methyl-2-oxo-pyrrolidine-3-carboxylic acid tert-butyl ester or N(1)-benzhydryl-2-oxo-piperidine-3-carboxylic acid tert-butyl ester under PTC conditions in the presence of chiral quaternary ammonium salts successfully afforded the corresponding α -alkylated lactams in high chemical (up to 99%) and optical yields (up to 98% ee), which is potencially applicable to the synthesis of 2-azaspirocycles by further cyclization. Given this, we attempted to apply our novel method to the synthesis of a representative *Nitraria* alkaloid, (+)-isonitramine (2).

As shown in a retrosynthetic analysis (Scheme 2), the C(7R) chirality can, in principle, be induced by diastereoselective reduction of **6**, which can be obtained by Dieckmann condensation⁷ of **5**. Optically active (R)-**5** can be derived from the enantioselective phase-transfer catalytic alkylation of N(1)-benzhydryl-2-oxo-piperidine-3-carboxylic acid *tert*-butyl ester (**4**).

First, the substrate (4) for PTC alkylation was prepared from δ -valerolactam (7) in two steps (Scheme 3). *N*-Diphenylmethylation of 4 under PTC reaction conditions in the presence of TBAB as a catalyst, followed by *tert*-butylcarboxylation using Boc₂O under LiHMDS basic conditions, provided lactam 4 in good yield.

Since the phase-transfer catalytic alkylation of **4** with unactivated alkyl halides usually produced poor chemical

Scheme 2. Retrosynthetic Analysis

Scheme 3. Enantioselective PTC Alkylation

yields, we changed the synthetic route of **5** through allylation, followed by chain elongation. However, the allylation still resulted in unsatisfactory enantioselectivity (87% ee) as reported previously; thus, we attempted to improve the enantioselectivity. We previously improved the enantioselectivity in the monoallylation of the *N*,*N*-diphenylmalonamide ester system during the total synthesis of (–)-paroxetine by switching the "allyl bromide" with "2-bromoallyl bromide". Therefore, we used bromoallyl bromide as an alternative electrophile. The phase-transfer catalytic allylation of **4** was carried out with 2-bromoallyl

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Scheme 4. Synthesis of (+)-Isonitramine^{a,b}

^a The enantiopurity was determined by HPLC analysis of **11**, **12**, and **16** using a chiral column (Chiralpak AD-H) with hexanes/2-propanol as the eluent. ^bAbsolute configuration was assigned by comparison of the specific optical rotation value of the (+)-isonitramine (**2**) with the literature value. ^{4f}

bromide under the previously reported optimal reaction conditions⁶ [(S,S)-3,4,5-Trifluorophenyl-NAS bromide (9, 5 mol %), and solid KOH (5.0 equiv) in toluene at -40 °Cl to afford 11. Surprisingly, the enantioselectivity was dramatically increased (87% ee to 98% ee), as well as the chemical yield (87% to 95%). These findings indicate that the bulkier electrophile was better able to selectively approach the re-face of the enolate of 4 when complexed with quaternary ammonium catalyst 10, resulting in C(6R)chirality with higher enantioselectivity. The debromination of 11 using tributyltin hydride in the presence of a catalytic amount of AIBN in toluene gave 12 (95%) (Scheme 4). For the elongation of the carbon chain, ethyl ethoxythiocarbonylsulfanylacetate was successfully added to the allylic moiety of 12 under radical conditions to provide 13 (88%). 10,11 The xanthate was then removed using tributyltin hydride and AIBN, which afforded diester 5 (96%). Since the direct Dieckmann condensation of 5 for the construction of the spirocyclic skeleton failed, the tert-butyl ester of 5 was transformed to afford the corresponding methyl ester 14 by the removal of tert-butyl ester with TFA in methylene chloride, followed by methylation with an excess of diazomethane (98%). Dieckmann condensation of 14 was then successfully accomplished under LiHMDS base conditions in THF at 0 °C, resulting in the production of spirocyclic β -ketoester 15 (98%). The hydrolysis of 15 with 50% aqueous KOH in MeOH, followed by decarboxylation, afforded ketone 6 (93%).

Next, the C(7R) chirality of **2** was introduced by the diastereoselective reduction of **6**. The reduction of **6** using

DIBAL in THF at -78 °C afforded **16** as a single diastereomer of the desired 7R-form in 93% chemical yield. Notably, the diastereoselectivity was much higher than that in the reduction of the free amide analogue of **6** under the same reaction conditions, which was reported previously (87% de). The bulky benzhydryl group seems to play important roles in not only the high enantioselectivity of C(6R) in the PTC alkylation of **4** but also the high enantioselectivity of C(7R) in the diastereoselective reduction of **6**. The reduction of the lactam **16** using excess LiAlH₄ in THF readily afforded piperidine **17** (90%). Finally, (+)-isonitramine {**2**, $[\alpha]_D^{23} = +6.22$ (c 1.16, CHCl₃); $[\alpha]_D^{23} = +5.8$ (c 1.4, CHCl₃) ^{4f}} could be obtained by the catalytic hydrogenation of **17** with Pd(OH)₂ in methanol (87%).

In conclusion, a new efficient synthetic approach toward (+)-isonitramine (2) was developed. (+)-Isonitramine (2) was synthesized in 12 steps (43% overall yield, 98% ee) by enantioselective phase-transfer catalytic alkylation, Dieckmann condensation, and diastereoselective reduction from δ -valerolactam (7). Both the high enantioselectivity and chemical yield make this approach a practical route for the large scale synthesis of 2-azaspirocycles. Structure—activity relationship studies of (+)-isonitramine derivatives are now in progress.

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Supporting Information Available. Spectroscopic characterizations of compounds 2, 4–6, 8, and 10–17. This material is available free of charge via the Internet at http://pubs.acs.org.

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