



## Design of a novel pyrrolidine scaffold utilized in the discovery of potent and selective human $\beta_3$ adrenergic receptor agonists

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### ABSTRACT

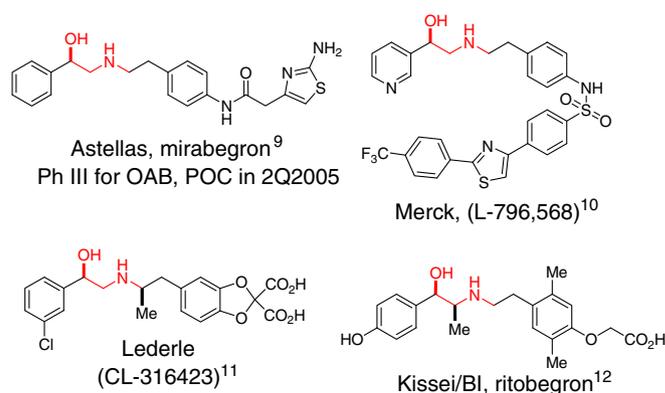
A novel class of human  $\beta_3$ -adrenergic receptor agonists was designed in effort to improve selectivity and metabolic stability versus previous disclosed  $\beta_3$ -AR agonists. As observed, many of the  $\beta_3$ -AR agonists seem to need the acyclic ethanolamine core for agonist activity. We have synthesized derivatives that constrained this moiety by introduction of a pyrrolidine. This unique modification maintains human  $\beta_3$  functional potency with improved selectivity versus ancillary targets and also eliminates the possibility of the same oxidative metabolites formed from cleavage of the N–C bond of the ethanolamine. Compound **39** exhibited excellent functional  $\beta_3$  agonist potency across species with good pharmacokinetic properties in rat, dog, and rhesus monkeys. Early de-risking of this novel pyrrolidine core (**44**) via full AMES study supports further research into various new  $\beta_3$ -AR agonists containing the pyrrolidine moiety.

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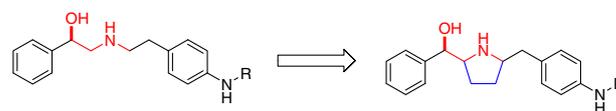
The  $\beta_3$ -adrenergic receptor ( $\beta_3$ -AR), also referred to as ADRB3, is present mainly in adipose tissue and is involved in the regulation of lipolysis and thermogenesis.<sup>1,2</sup> Consequently,  $\beta_3$ -AR agonists were extensively explored in the 1980's and 1990's for treatment of obesity and type 2 diabetes.<sup>3</sup> However, recent studies have shown that  $\beta_3$ -AR's are not only expressed in adipocytes but also in the bladder detrusor muscle and the lining of the gastrointestinal tract.<sup>4,5</sup> As a result, new therapeutic applications of  $\beta_3$ -AR agonists for the treatment of overactive bladder (OAB) and irritable bowel syndrome (IBS) have recently been explored.<sup>6,7</sup> Multiple companies have reported  $\beta_3$ -AR agonists over the past few decades, which has led to innovative ideas in the attempt to discover structurally unique  $\beta_3$ -AR agonists in this crowded competitive arena (Fig. 1).<sup>8</sup>

We present a recent discovery of a new pyrrolidine variation of the secondary amine moiety that appears as a conserved core of many reported  $\beta_3$ -AR agonists (Fig. 2). This modification affords analogs that maintain good potency and selectivity at the human  $\beta_3$ -AR while unlocking new prospects for research that could lead to a drug candidate.

In the late 1990's, Merck reported a series of potent and selective sulfonamide derived human  $\beta_3$ -AR agonists for the possible treatment for obesity.<sup>13</sup> Clinical candidate L-796,568 was discovered which possessed good functional human  $\beta_3$ -AR agonist



**Figure 1.** Examples of  $\beta_3$ -AR agonists (conserved acyclic ethanolamine pharmacophore in red). See above-mentioned references for further information.



**Figure 2.** Pyrrolidine derived scaffold (blue highlights modification to acyclic ethanolamine).

potency and ~1000-fold selectivity over  $\beta_1$  and  $\beta_2$  adrenergic receptors. In preclinical species, however, L-796,568 exhibited

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moderate oral bioavailabilities in rats and dogs and low oral bioavailability in rhesus.<sup>14</sup> To identify possible metabolic pathways which may be causing the high clearance and moderate oral bioavailability in rats, studies to determine oxidative metabolism in rat liver microsomes were performed.<sup>15</sup> These studies revealed that the majority of metabolites were derived from cleavage of the central C–N bond from the ethanolamine which ultimately led to identification of a carboxylic acid metabolite and a primary amine metabolite (Fig. 3). This oxidative metabolism may be a contributing factor to the high clearance and moderate oral bioavailability seen in rats. Moreover, the released metabolites may have untoward biological effects in animals. Blockade or modification of this metabolic pathway may therefore improve the pharmacokinetic profiles of acyclic  $\beta_3$ -AR agonists and prevent formation of metabolites such as those observed from L-796,568.

Previous reports have described the effects of simple substituents at either of the two carbons adjacent to the secondary amine,<sup>11,12</sup> and we sought to understand the effects of substituents on both sides of the amine. We therefore synthesized sulfonamide **4** (as a mixture of four diastereomers) which is a close analog of **3** (Fig. 4).<sup>16</sup> Unfortunately, the introduction of methyl substituents onto each of the carbons adjacent to the secondary amine resulted in a complete loss of  $\beta_3$ -AR activity which may be attributed to steric interactions between the two methyl substituents that induce a rotational conformation not favored by the  $\beta_3$ -AR.

If this hypothesis is correct, it is possible that restricting the rotation of the acyclic ethanolamine may result in a favorable conformation for  $\beta_3$ -AR agonist functional activity while

simultaneously modulating the metabolic pathways observed with L-796,568. Observation of molecular models suggested that a five-membered ring may best mimic the conformations of the acyclic ethanolamine. Consequently, chemistry was developed to synthesize a pyrrolidine core. Initially, the stereochemistry of the disubstituted pyrrolidine ring would need to be investigated to determine the preferred stereoisomer for  $\beta_3$ -AR agonist activity. A cross metathesis synthetic route using two substituted vinyl intermediates would afford the foundation from which all isomers could be obtained using a common synthetic sequence.

The synthesis of four distinct vinyl stereoisomer precursors began by treating commercially available 3-chlorobenzaldehyde (**5**) with vinyl magnesium bromide to afford vinyl alcohol **6** (Scheme 1). Protection of the hydroxyl group was accomplished with *tert*-butyl, dimethyl silyl chloride in the presence of triethylamine. Aldehyde **7** was obtained by oxidative cleavage of the vinyl group with ozone and subsequent reduction of the ozonide using triphenylphosphine. After formation of sulfinamide **8**<sup>17</sup> treatment with vinyl magnesium bromide afforded sulfonamide **9** as a mixture of all four diastereomers which were separated using silica gel column chromatography. Protecting group manipulation was then carried out on all four isomers to afford the *N*-Boc-protected compounds **10–13** which were then poised to undergo the desired cross metathesis reaction.

Preparation of the second vinyl intermediate is shown in Scheme 2. Intermediate **15** was obtained upon treatment of commercially available methyl (4-aminophenyl) acetate (**14**) with benzyl chloroformate followed by ester saponification with 5.0 N lithium hydroxide. The carboxylic acid was then converted to the Weinreb amide<sup>18</sup> (**16**) using standard EDC coupling with *N,O*-methylhydroxylamine. After purification via silica gel flash column chromatography, the Weinreb amide was then treated with vinyl magnesium bromide to afford the vinyl ketone **17**.

Cross metathesis<sup>19</sup> could now be accomplished between olefins **10–13** and **17** (Scheme 3). Diastereomers **10–13** were each cross-coupled with vinyl ketone **17** by mixing the two intermediates in the presence of a stoichiometric amount of Zhan I catalyst<sup>20</sup> under dilute conditions (0.05 M). Intermediates **18–21** were next converted to the pyrrolidine rings using a catalytic hydrogenation reaction which accomplished five separate reactions in a single pot. Careful tracking of the reaction using LC–MS allowed the sequence of reactions to be identified. Initially, the reduction of the double bond occurs with 90% conversion after 30 min under 50 PSI hydrogen. Next, removal of the chloride and Cbz protecting groups was observed, followed by formation of the cyclic imine from the ring closure of the ketone with the free amine. Finally, the rate-limiting step in this reaction appears to be the hydrogenation of the imine to form the pyrrolidine ring. The protection of the amines with *tert*-butyl-dicarbonate is the final reaction step to afford the four separate *cis/trans* (3:1 ratio) pyrrolidine diastereomers **26–29** (as shown in Scheme 3).

The pyrrolidine scaffolds were now available for synthesis of  $\beta_3$ -AR agonists. We first targeted a sulfonamide analogous to L-796,568. Sulfonyl chloride **30**<sup>21</sup> was reacted with each of the four diastereomers followed by simultaneous deprotection of the amine and hydroxyl groups using 4 N HCl in dioxane with 10% (v/v) water to furnish **31–34**.

The acyclic compound (**3**) is also shown to allow comparisons to the pyrrolidine analogs (Table 1). The four diastereomers vary greatly in h $\beta_3$ -AR agonist potency and h $\beta_1$ /h $\beta_2$ -AR selectivity.

Not surprisingly, analogs with the undesired (*S*)-hydroxyl stereochemistry were 100-fold less potent than the (*R*)-hydroxyl diastereomers. On the other hand, the two analogs with the (*R*)-hydroxyl stereochemistry both exhibited good potency in the h $\beta_3$ -AR agonist functional assay. Compound **31** was optimal and exhibited similar functional activity as acyclic analog **3**. Since **31** was a 3:1

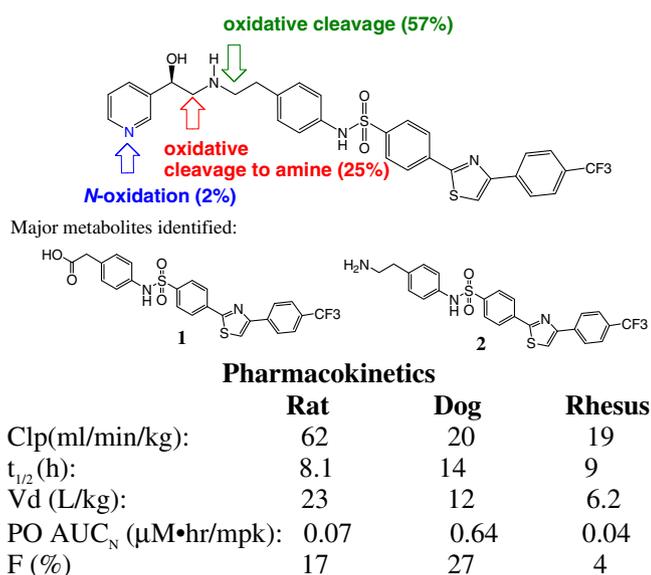


Figure 3. Metabolism of L-796,568 observed in rats.<sup>15</sup>

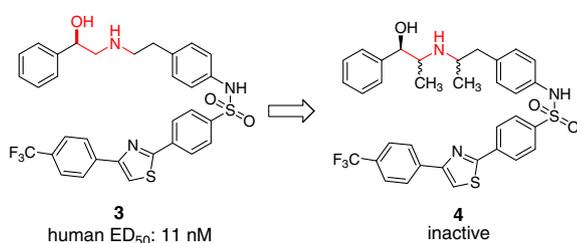
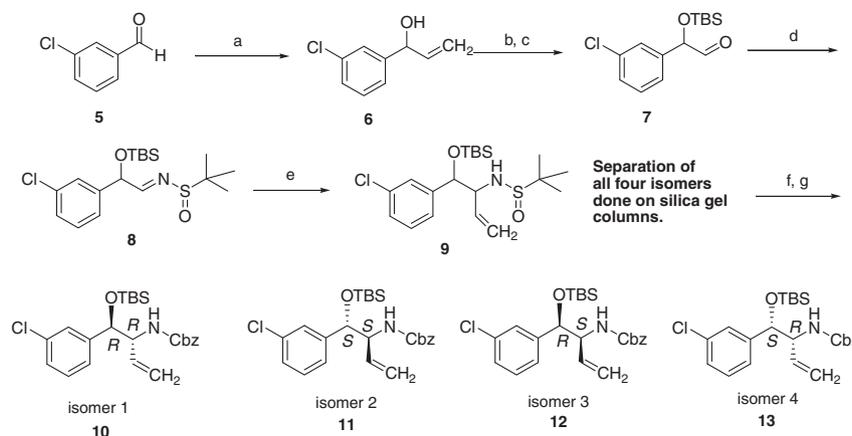
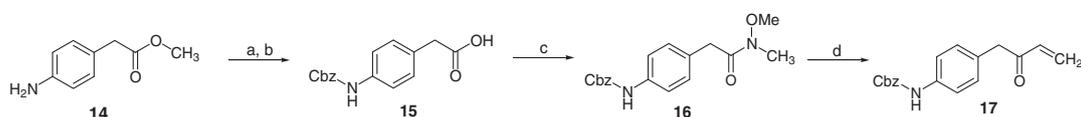


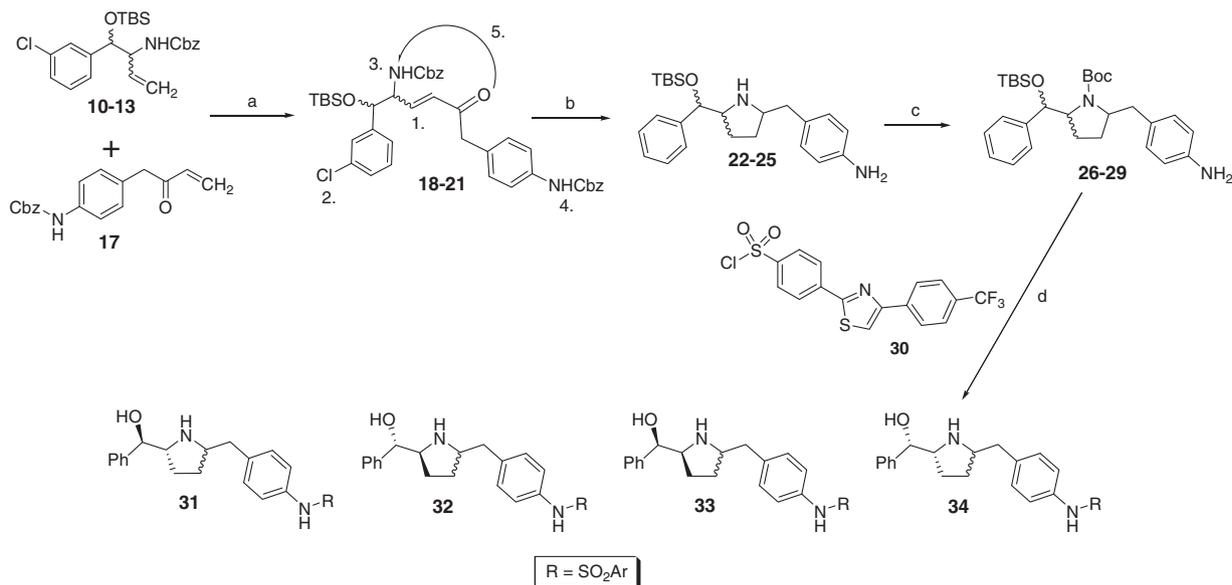
Figure 4. Di-methyl substituted analog **4**.



**Scheme 1.** Synthesis of ethanolamine intermediates.<sup>16</sup> <sup>a</sup>Reagents and conditions: (a) 1.6 M vinylmagnesium bromide in THF, THF (0–25 °C) 16 h (82%); (b) TBSCl, imidazole, DCM (68%); (c) ozone, DCM (–78 °C) followed by PPh<sub>3</sub> (–78 °C to 25 °C) overnight (71%); (d) *S*-(-)-2-methyl-2-sulfinamide, copper sulfate, DCM, overnight (72%); (e) vinylmagnesium bromide, THF (0 to 25 °C), 5 h (88% total yield); (f) 4 N HCl in dioxane (quantitative); (g) benzyl chloroformate, TEA, DCM (63–82%).



**Scheme 2.** Synthesis of vinyl ketone intermediate. <sup>a</sup>Reagents and conditions: (a) benzyl chloroformate, TEA, DCM (92%); (b) LiOH, 1:1:1 THF, water, MeOH, 60 °C 5 h (78%); (c) *N,O*-dimethyl-hydroxylamine hydrochloride, EDC, HOAt, DMF (69%); (d) 2 equiv vinylmagnesium bromide, THF (–78 °C to 0 °C, 72%).

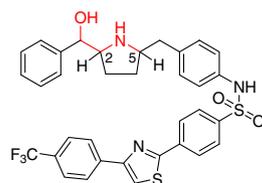


**Scheme 3.** Synthesis of the pyrrolidine core and sulfonamides **31–34**. <sup>a</sup>Reagents and conditions: (a) Zhan I catalyst, DCM, 40 °C 5 h (75%); (b) 10% palladium on carbon, H<sub>2</sub>, ethanol (92% total yield); (c) *tert*-butyl di-carbonate, TEA, THF, overnight (83%); (d) 4-[2-[4-(trifluoromethyl)phenyl]-1,3-thiazol-4-yl]benzenesulfonyl chloride, DIEA, DMF, 2 h, rt (62–84%) followed by 4 N HCl in dioxane (10% water v/v), 1 h, rt (88–96%).

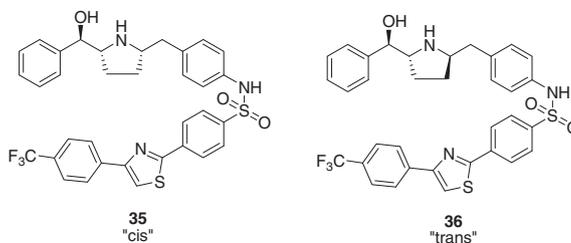
mixture of *cis/trans* pyrrolidine isomers, separation of the two diastereomers was important to determine the most desired pyrrolidine stereoisomer.

Separation of the *cis* and *trans* pyrrolidine isomers of compound **31** was accomplished by use of preparative SFC using a ChiralCEL OD chiral column eluting with 20% methanol in carbon dioxane as the elutant.<sup>22</sup> Characterization of the major isomer (compound **35**) was done with NOESY <sup>1</sup>H NMR which supports an assignment of the relative stereochemistry as a *cis* pyrrolidine.<sup>23</sup>

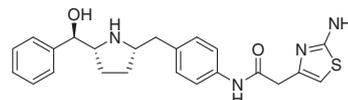
The *cis* pyrrolidine diastereoisomer **35** is 20-fold more potent than the *trans* isomer **36** (Table 2). Additionally, the *cis* diastereomer is more selective over both  $\beta_1$  and  $\beta_2$  ARs than the *trans* diastereomer. Further profiling of this compound was halted, however, due to the poor bioavailability observed in rats with compound **35**. After IV dosing, the compound exhibited a clearance of 13.2 mL/min/kg and a half-life of 11.4 h. After oral dosing, however, poor exposures and bioavailability was observed (dose normalized PO AUC = 0.07  $\mu$ M·h/mpk and oral bioavailability of 3.8%).

**Table 1**  
Sulfonamide analogs

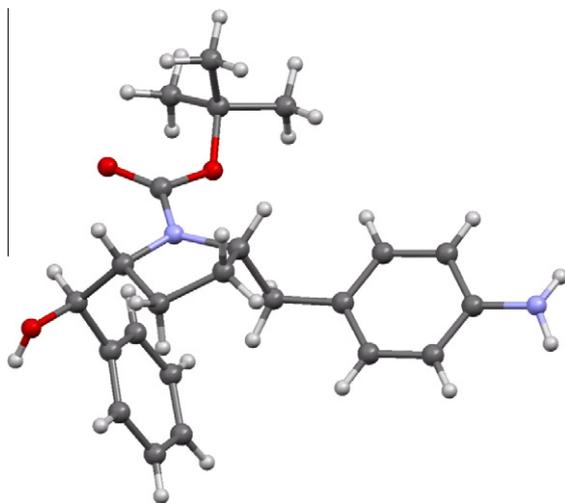
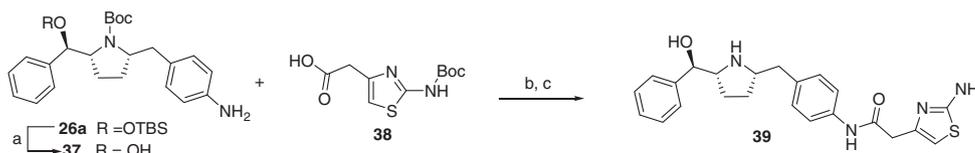
Compound	Chirality		hβ3		Off-target IC <sub>50</sub> 's (nM)	
	OH	Pyrrolidine	EC <sub>50</sub> (nM)	Act (%)	hβ1	hβ2
<b>3</b>	R	Acyclic	11	95	1040	780
<b>31</b>	R	2(R), 5(R,S)	15	101	2930	2180
<b>32</b>	S	2(S) 5(R,S)	2030	89	1780	3530
<b>33</b>	R	2(S), 5(R,S)	81	104	2150	3530
<b>34</b>	S	2(R) 5(R,S)	1307	104	1520	2790

**Table 2**  
Separated diastereomers of compound **31**

Compound	EC <sub>50</sub> (nM)	Act (%)	hβ1	hβ2	CYP3A4	CYP2D6
<b>35</b>	4.1	95	2189	2509	>50,000	>50,000
<b>36</b>	83	76	242	297	>50,000	>50,000

**Table 3**  
Aminothiazole amide analog (**39**)**Compound 39**  
EC<sub>50</sub>: 0.98 nM (99% Act)  
hβ1/hβ2 IC<sub>50</sub>: >20 μM

	Pharmacokinetics		
	Rat	Dog	Rhesus
Clp (ml/min/kg)	11	23	43.4
t <sub>1/2</sub> (h)	7.7	8.3	9.9
V <sub>d</sub> (L/kg)	5.8	14	26.4
PO AUCn (μM·h/mpk)	0.44	0.32	0.14
F (%)	12	18	14
Ancillary activity: CYP inhibition (nM)	Solubility (mg/ml)	Microsomal stability: % remaining (45 min)	
2D6 2200	@pH <5.2:>1.8	Dog 23	
3A4 >100,000	@pH 7.4:1.3	Human 40	
2C9 >100,000	hNa <sub>v</sub> 1.5 26%I @ 30 μM	Rat 13	
2C8 >100,000	hERG IC <sub>50</sub> 11.2 μM	Rhesus 77	

**Figure 5.** Crystal structure of *cis* pyrrolidine enantiomer (**37**).**Scheme 4.** Synthesis of the pyrrolidine aminothiazole amide. <sup>a</sup>Reagents and conditions: (a) 1.0 M TBAF in THF, rt 4 h (78%, intermediate synthesized for crystallization purposes only); (b) EDC, HOAt, DMF, rt 16 h (83%); (c) 4 N HCl in dioxane with 10% (v/v) water added, rt, 2 h (94%).



associated with this article can be found, in the online version, at doi:10.1016/j.bmcl.2010.12.087.

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- Zhan Catalyst I is available from Zannan Pharma LTD, catalog number RC-301. Patents CN1907992A, US 2007/0043180 A1, PCT WO 2007/003135 A1.
- This material was obtained from prior batches synthesized for the scale-up of L-796,568. The chemistry to prepare this intermediate has been documented previously.<sup>10</sup>
- The major isomer eluted first at retention time of 4.11 min, while the second isomer eluted at 4.56 min.
- The stereochemistry of the two pyrrolidine chiral centers was determined to be *cis* based on 2D NOE NMR data (see Supplementary data).
- Selectivity was assessed by SFC analytical analysis and the effort to improve selectivity was studied via trial and error of changes in the reaction conditions. Original less selective conditions: Dilution 0.02 M and stoichiometric amount (by weight) of the catalyst (resulted in 3:1 mixture of *cis/trans*).
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