

A Bifunctional Ligand Enables Gold-Catalyzed Hydroarylation of Terminal Alkynes under Soft Reaction Conditions

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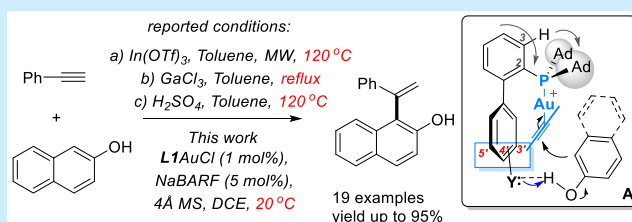


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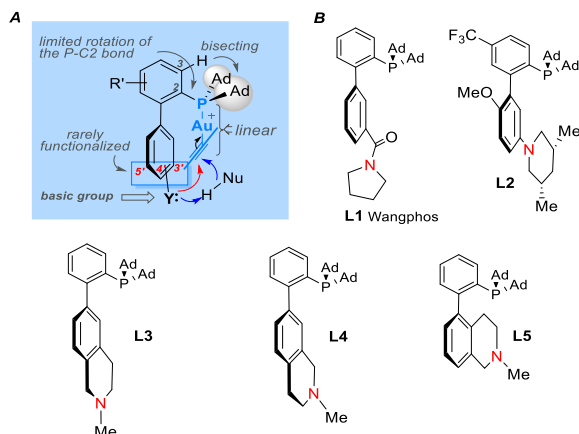
Supporting Information

ABSTRACT: An efficient gold-catalyzed hydroarylation of alkynes under soft reaction conditions is developed by utilizing a bifunctional ligand. This transformation features a broad substrate scope and thus exhibits moderate to excellent efficiency.



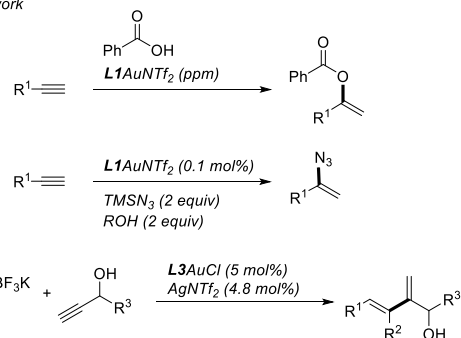
Over the past few years one of our groups has developed a range of new bifunctional phosphine ligands¹ in gold catalysis.² These ligands, as revealed in Scheme 1B, despite possessing the traditional biphenyl-2-ylphosphine skeleton similar to Johnphos,³ are unique and extraordinary by a remote basic group on the other benzene ring. With this kind of bifunctional ligands, cooperative interactions between the ligand and either the external nucleophiles or substrates by overcoming the solid linear character of the centroid in the gold(I) intermediate could be achieved. Meanwhile, rotation of the P–C2 bond in the ligand would also be restricted with the help of Ad groups (Scheme 1A). This metal–ligand cooperation⁴ has permitted the development of novel gold catalysis.¹ For example, the ligands L2–L5 (Scheme 1B) featuring a remote amino group enable soft propargylic deprotonations, which is the key transformation in several of our recent work.^{1a,b,f} Especially,

Scheme 1. Biphenyl-2-ylphosphine Ligands for Homogeneous Gold Catalysis

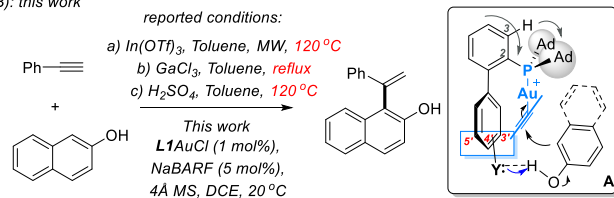


Scheme 2. Previous Work and Our Design

A): our previous work

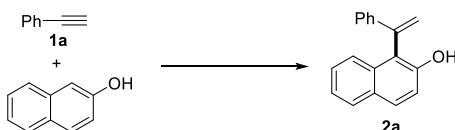


B): this work



the WangPhos (L1) ligand, which is installed with a remote amide group, could promote the gold-catalyzed transformations of terminal or interminal alkynes with the aid of the weak hydrogen bonding between the external nucleophiles and the ligand.^{1b,d} With such a ligand, facile addition of carboxylic acids

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Table 1. Reaction Conditions Study^a


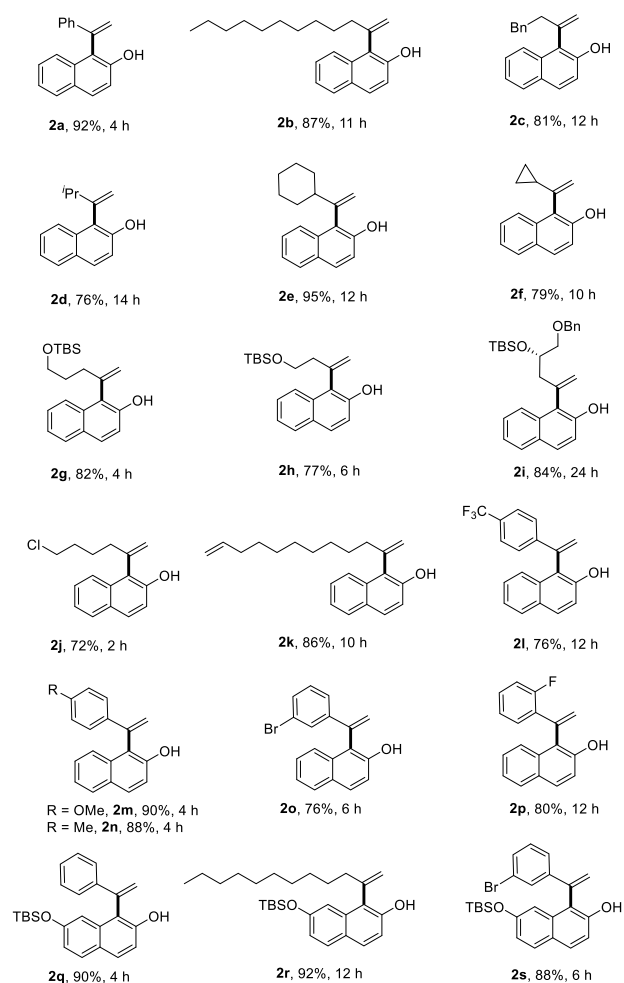
entry	conditions ^a	yield ^b
1	L1AuCl (1 mol %), NaBARF (5 mol %), DCE, 4 Å MS, 20 °C, 4 h	92
2	L2AuCl (1 mol %), NaBARF (5 mol %), DCE, 4 Å MS, 20 °C, 4 h	<5
3	L3AuCl (1 mol %), NaBARF (5 mol %), DCE, 4 Å MS, 20 °C, 4 h	<5
4	L4AuCl (1 mol %), NaBARF (5 mol %), DCE, 4 Å MS, 20 °C, 4 h	<5
5	L5AuCl (1 mol %), NaBARF (5 mol %), DCE, 4 Å MS, 20 °C, 4 h	<5
6	PPh ₃ , IPr or JohnPhos as Au ligand (1 mol %), NaBARF (5 mol %), DCE, 4 Å MS, 20 °C, 4 h	0
7	PPh ₃ , IPr or JohnPhos as Au ligand (1 mol %), NaBARF (5 mol %), Et ₃ N (1.0 equiv), DCE, 4 Å MS, 20 °C, 4 h	0
8	L1AuCl (1 mol %), DCE, 4 Å MS, 20 °C, 4 h	0
9	NaBARF (5 mol %), DCE, 4 Å MS, 20 °C, 4 h	0
10	L1AuCl (1 mol %), AgOTf (5 mol %), DCE, 4 Å MS, 20 °C, 4 h	55
11	L1AuCl (1 mol %), AgNTf ₂ (5 mol %), DCE, 4 Å MS, 20 °C, 4 h	72
12	L1AuCl (1 mol %), NaBARF (5 mol %), PhF, 4 Å MS, 20 °C, 2 h	77
13	L1AuCl (1 mol %), NaBARF (5 mol %), PhCF ₃ , 4 Å MS, 20 °C, 2 h	82

^aGeneral reaction conditions: reactions run in vials, alkyne (0.2 mmol), naphthol (0.3 mmol), 4 Å MS (50 mg), 0.1 M. ^bIsolated yields.

as well as in situ-generated hydrazoic acid to alkynes was achieved. This rate-accelerating phenomenon has also been harnessed to achieve highly enantioselective allenol cyclizations.^{1e} Very recently, an intermolecular propargylic alcohol hydroalkenylation was also achieved, which leads to the synthetically valuable conjugated dienyl alcohols (Scheme 2A).^{1g}

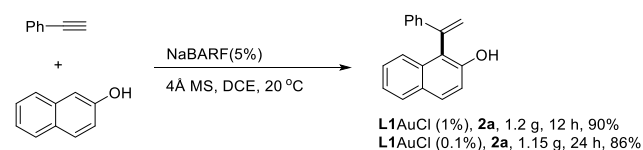
On the other hand, hydroarylation of an unactivated alkyne allows access to synthetically useful arylalkenes in an atom-efficient manner.⁵ With the pioneering work published by Yamaguchi about the phenol or naphthol alkenylation with terminal alkynes,⁶ a series of C-alkenylation transformations have been realized.^{7–9} These strategies often suffer from shortcomings including severe reaction conditions (strong inorganic acid or Lewis acid used as catalyst and reaction temperature mostly at 120 °C or even higher, Scheme 2B), low effectiveness, and bad selectivity. Therefore, it is still of need to develop new methods for the transformation of this kind under mild catalytic conditions. With our ongoing study in this bifunctional phosphine ligand chemistry, we envisioned that, by using phenol or naphthol as a nucleophile, hydroarylation of alkynes may be realized at ambient temperature with the help of the weak interactions between the aryl hydroxy group and the amide group, as outlined in A in Scheme 2B. Besides, such a ligand–substrate interaction would also help prevent the formation of O-alkenylated adducts. In this work, we describe such a bifunctional-ligand-enabled efficient gold-catalyzed hydroarylation of alkynes under soft reaction conditions.

At the outset, phenol was chosen as the nucleophile to implement the design of this chemistry. However, no reaction

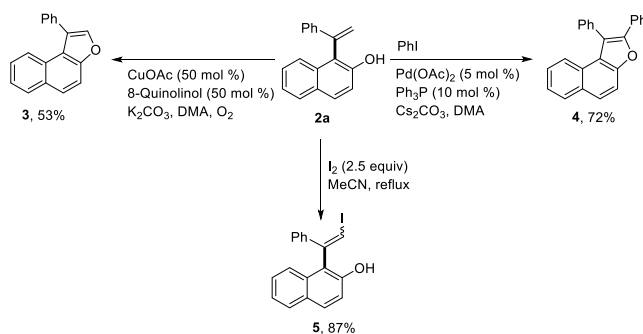
Scheme 3. Reaction Scope Study^a

^aGeneral reaction conditions: reactions run in vials, alkyne (0.2 mmol), naphthol (0.3 mmol), NaBARF (5 mol %), 4 Å MS (50 mg), L1AuCl (1 mol %), DCE (2 mL), 20 °C. Isolated yields are reported.

Scheme 4. Gram-Scale Reaction

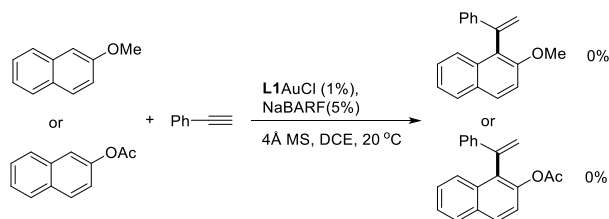


Scheme 5. Transformations of Product 2a



was observed after extensive exploration of ligands and reaction conditions while phenol was recovered intact. We then

Scheme 6. Mechanism Studies



employed naphthol as the nucleophile. Encouragingly, the expected compound **2a** was indeed formed in a good yield of 92% when **L1AuCl** was used as the catalyst and NaBARF was used as the chloride abstractor (Table 1, entry 1). Subsequent catalyst screenings revealed that our previously designed amine-functionalized ligands **L2–L5** were largely ineffective (entries 2–5). It should be mentioned that the use of common PPh_3AuCl , IPrAuCl , or JohnPhosAuCl in the presence of exogenous Et_3N or its absence resulted in no reaction at all (entries 6 and 7). Both the gold precatalyst and NaBARF were required for the reaction (entries 8 and 9).¹⁰ Replacing NaBARF by AgOTf (entry 10) or AgNTf_2 (entry 11) generated lower yields of the formation of **2a**. Besides, DCE was also found to be very significant for the optimal reaction conditions. With PhF or PhCF_3 as the reaction solvent, the yields of **2a** would reduce to only 77 and 82% (entries 12 and 13).

Then, the scope of this chemistry was examined. First, various aliphatic terminal alkynes were explored. As summarized in Scheme 3, when the terminal alkynes were installed by an alkyl group including 1-decanyl (**2b**), phenethyl (**2c**), isobutyl (**2d**), cyclohexyl (**2e**), or cyclopropyl (**2f**), the reactions all proceeded smoothly. Functionalized alkyl groups including OTBS (**2g**, **2h**), benzyloxy (**2i**), chloro (**2j**), and even C–C double bonds (**2k**) are also readily tolerated, highlighting the exceptionally mild nature of this chemistry. Several substituted phenylacetylenes were also examined. Substituents such as CF_3 , OMe, Me, Br, or F at the different positions on the benzene ring were tolerated, and the corresponding 1,1-diarylethylenes **2l–2p** were generated in excellent yields. Besides, 2-naphthol functionalized by a 7-OTBS group underwent reactions with different terminal alkynes smoothly to afford the desired products **2q**, **2r**, and **2s** in 90, 92, and 88% yields, respectively. With these results in hand, then the challenging internal alkynes were also tested. Early results revealed that replacing the terminal alkynes with internal alkynes including 6-dodecyne and diphenylacetylene resulted in no reaction at this point, even at a higher reaction temperature (90 °C).

A gram-scale formation of **2a** was also carried out under standard conditions, and 1.2 g of the desired product was furnished on a 10 mmol scale (Scheme 4). Moreover, when the reaction was performed with the catalyst loading lowered to 0.1 mol %, 86% yield of the desired product could be generated, albeit a longer reaction time was needed.

The synthetic utilities of this chemistry are also demonstrated by one-step conversions of **2a** to versatile naphtho[2,1-*b*]furan derivatives, as outlined in Scheme 5.¹¹ In addition, an additional function handle (i.e., iodo) was readily installed into **2a** in an excellent yield upon routine iodination.¹²

It should be noted that the hydroarylation transformation did not happen to provide the product at all when the naphthol hydroxyl group was transformed to acetoxy or methoxy (Scheme 6). This result is consistent with the mechanism we proposed.

In conclusion, we have presented an efficient strategy for bifunctional-ligand-enabled gold-catalyzed intermolecular hydroarylation of alkynes under soft reaction conditions. The WangPhos ligand, which was installed with a remote amide group, is shown to be enabling in this transformation. This chemistry exhibits gentle reaction conditions, furnishes excellent yields, and tolerates a broad substrate scope.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.0c02130>.

Experimental procedures, characterization data for all new products, and NMR spectra (PDF)

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Notes

The authors declare no competing financial interest.

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