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Thiophene containing triarylmethanes as antitubercular agents $\stackrel{\leftrightarrow}{\sim}$

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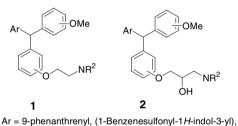
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Abstract—A new series of thiophene containing triarylmethane derivatives were synthesized from the Friedel–Crafts alkylation of diarylcarbinols followed by incorporation of amino alkyl chains. These were evaluated against *Mycobacterium tuberculosis* $H_{37}R_v$ and showed the activity in the range of 3.12–12.5 µg/mL in vitro. © 2007 Elsevier Ltd. All rights reserved.

Tuberculosis is the leading infectious disease caused by *Mycobacterium tuberculosis*.^{1,2} The situation is becoming alarming with the recent emergence of multi-drug-resistant (MDR) strains and its synergy with global human immunodeficiency virus (HIV).³ The search for more effective agents against *M. tuberculosis* (MT) is ongoing in an attempt to enhance survival and reduce morbidity, as proven by the high number of patents of new antitubercular agents in the past decade.⁴

Because of this, there is an urgent need for anti-TB drugs with improved properties such as enhanced activity against MDR strains, reduced toxicity, shortened duration of therapy, rapid mycobactericidal mechanism of action and the ability to penetrate host cells and exert anti-mycobacterial effects in the intracellular environment.

We have been involved in the design, synthesis and bioevaluation of new antitubercular agents and several triarylmethane (TRAM) derivatives 1 and 2 (Fig. 1) with antitubercular activity were reported.⁴ The structural pharmacophore is based on a triarylmethane nucleus containing phenanthrene, indole and anthracene rings and exhibited $1.56-25 \mu g/mL$ antitubercular activity in vitro. In order to optimize the anti-TB activity of



Ar = 9-phenanthrenyl, (1-Benzenesulfonyl-1*H*-indol-3-yl), pyridine-3-yl, 9-anthracenyl

Figure 1. Structures of antitubercular triarylmethanes with basic amino alkyl or amino hydroxy alkyl side chains.

TRAMs, we planned to systematically substitute the aryl group with rings having aromatic properties.

Thiophene containing compounds are well known to exhibit various biological activities such as BACE1 inhibitors,5 anti-inflammatory agents,6 anti-HIV PR inhibitors⁷ and anti-breast cancer.⁸ Thus, we decided to replace one of the aryl rings with thiophene in the triarylmethane nucleus. We also intended to incorporate chlorine and thiomethoxy substituted phenyl ring since it is generally observed that the presence of chlorine or thiomethoxy in a molecule profoundly affected its biological properties.^{9–11} Moreover, several amino alcohol derivatives such as ethambutol are well known to have antitubercular activity (Fig. 2).¹² Thus in our programme also, we intended to incorporate 2-hydroxy-amino functionality on thiophene containing pharmacophore for further diversification and designed to synthesize 3, 4 and 5 as our target molecules (Fig. 2).

Keywords: Thiophene; Antitubercular; Friedel–Crafts alkylation. * CDRI communication No. 7167.

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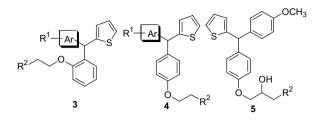


Figure 2. General structures of designed molecules.

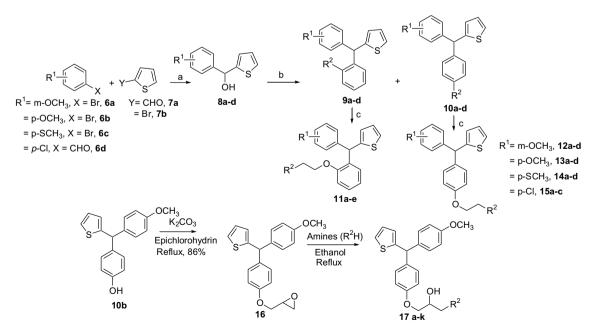
The title compounds were synthesized essentially following the steps as depicted below. Nucleophilic addition of Grignard reagents 6a-c and 7b onto thiophene-2-carbaldehyde 7a and p-chloro benzaldehyde 6d, respectively, in THF furnished the carbinol derivatives 8a-d in excellent vields. Out of various protic acids used in Friedel-Crafts alkylation, concd H₂SO₄ (cat.) in dry benzene gave better results. Thus, FC alkylation of 8a-dwith phenol led to the mixtures of 9a-d as minor and 10a-d as major products. The compounds 9a-d were characterized from its deshielded doublet aromatic protons and singlet methine proton resonances, respectively, than corresponding para isomers 10a-d due to the presence of -I inductive effect of *ortho*-hydroxy group. The structural identity of 9a-d and 10a-d was further confirmed by their ¹³C NMR and mass spectrum fragmentation analysis. In this competitive reaction, the generation of compounds 9a-d and 10a-d could be attributed to the fact that Friedel-Crafts alkylation occurs via attack of phenol at both ortho and para positions of the benzene nucleus with the carbinol 8a-d.

Friedel–Crafts alkylated products **9a–d** and **10a–d** were then aminoethylated with different dialkylaminoethyl chloride hydrochloride chains in the presence of K_2CO_3 and acetone and furnished **11a–e**, **12a–d**, **13a–d**, **14a–d** and **15a–c** in good yields (Scheme 1). The corresponding salts were obtained after treating the amines with ethanolic HCl. Initial bioevaluation of the compounds **11a–e** and **12a–d**, **13a–d**, **14a–d** and **15a–c** gave interesting antitubercular activity results in vitro. The compound **10b** was reacted with epichlorohydrin in presence of K_2CO_3 to furnish the oxirane **16** in good yield (86%). The oxirane was then reacted with selected primary and secondary amines to furnish a variety of 2-hydroxy amino alkyl derivatives (**17a–k**) (Scheme 1).

Biology determination of activity in vitro: The activity of the compounds against M. tuberculosis $H_{37}R_v$ was determined by agar microdilution technique.¹³ (details are described in experimental section). The activity of the compounds 11a-e, 12a-d, 13a-d, 14a-d, 15a-c and 17a-k in vitro is shown below (Table 1).

A closer look into the biological results of the above compounds reveals that in a given diarylmethylphenol series (10a–d), 10b containing *p*-OCH₃ gave better activity. On the other hand, 11a–e with *ortho* substituent did not exhibit good activity (>12.5 µg/mL). Among the pair of compounds 12a, 12b; 13a, 13b; 14a, 14b and 15a, 15b, increasing the alkyl chain on nitrogen gave better activity. Increase in polarity of the side chain reduces the activity (12c, 13c, 14c and 15c). Incorporation of 2-hydroxy-amino functionality on the pharmacophore for better activity was not encouraging.

In conclusion we have synthesized thiophene containing several triarylmethane (TRAM) derivatives as new series of antitubercular compounds with antimycobacterial activity. The compounds containing increased amount of alkyl group on nitrogen of the side chain were active



Scheme 1. Reagents and conditions: (a) Mg, THF, 0 °C, rt, 2 h, 8a (69%), 8b (71%), 8c (74%) and 8d (79%); (b) phenol, concd H₂SO₄ (cat.), dry benzene, reflux, 2 h, 9a (12%), 9b (9%), 9c (11%) and 9d (12%); (c) alkylaminoethyl hydrochloride (ClCH₂CH₂R²·HCl), anhyd K₂CO₃, dry acetone, reflux, 6–7 h, (yields given in Table 1).

Table	1.
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Entry	Compound	\mathbf{R}^1	\mathbb{R}^2	Yield ^a (%)	MIC (µg/mL) agar microdilution method ¹
1	10a	<i>m</i> -OCH ₃	ОН	61	>12.5
2	10b	p-OCH ₃	OH	59	3.12
3	10c	p-SCH ₃	OH	63	>12.5
4	10d	p-Cl	OH	58	>12.5
5	11a	p-OCH ₃	$N(CH_3)_2$	71	>12.5
6	11b	p-OCH ₃	$N(CH_2CH_3)_2$	76	>12.5
7	110 11c	<i>p</i> -OCH ₃ <i>p</i> -OCH ₃	N	78	>12.5
	11d	<i>p</i> -ОСН ₃		73	>12.5
8		-			
9	11e	<i>p</i> -OCH ₃	N	68	>12.5
10	12a	m-OCH ₃	$N(CH_3)_2$	71	12.5
11	12b	m-OCH ₃	$N(CH_2CH_3)_2$	78	3.12
12	12c	<i>m</i> -OCH ₃	NO	82	12.5
13	12d	<i>m</i> -OCH ₃	N	79	3.12
14	13a	<i>p</i> -OCH ₃	$N(CH_3)_2$	79	12.5
15	13b	<i>p</i> -OCH ₃	$N(CH_2CH_3)_2$	72	6.25
16	13c	<i>p</i> -OCH ₃	NO	76	>12.5
17	13d	<i>p</i> -OCH ₃	N	72	12.5
18	14a	p-SCH ₃	$N(CH_3)_2$	70	>12.5
19	14b	p-SCH ₃	$N(CH_2CH_3)_2$	71	3.12
20	14c	<i>p</i> -SCH ₃	NO	76	>12.5
21	14d	<i>p</i> -SCH ₃	Ŋ	78	3.12
			\bigcirc		
22	15a	p-Cl	$N(CH_3)_2$	71	6.25
23	15b	p-Cl	$N(CH_2CH_3)_2$	76	3.12
24	15c	p-Cl	NO	69	12.5
25	17a	<i>p</i> -OCH ₃	N_N-{_}	80	>12.5
26	17b	<i>p</i> -OCH ₃	HN	77	>12.5
			~		
27	17c	<i>p</i> -OCH ₃	N_N_	83	>12.5
28	17d	<i>p</i> -OCH ₃		70	>12.5
29	17e	<i>p</i> -OCH ₃	N S	86	>12.5
30	17f	<i>p</i> -OCH ₃	N_CH3	81	>12.5
31	17g	<i>p</i> -OCH ₃	HN ^N	77	>12.5
32	17h	<i>p</i> -OCH ₃	HN	76	12.5
33	17i	<i>p</i> -OCH ₃	N	79	12.5
34	17j	<i>p</i> -OCH ₃	N	85	>12.5
35	17k	<i>p</i> -OCH ₃	HN [∽] N _{≥N}	81	>12.5
• -					
36 37	Rifampin Isoniazid (INH)	_			0.1 0.05

^a Isolated yield after silica gel column chromatography.

with MIC $3.12 \mu g/mL$ and might be a lead for further optimization and development.

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Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.bmcl. 2007.10.083.

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