N-Glycosylation of 2,3-Dideoxyfuranose Derivatives Having Difluoromethylene-phosphonate and -phosphonothioate Functionality at the 3α -Position

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Abstract: TiCl₄-Mediated N-glycosylation of 2,3-dideoxyribofunanosides having a difluoromethylene-phosphonate or -phosphonothioate functional group at the 3α -position with silylated pyrimidines was examined. The phosphonate functional was a good directing group to induce α -N-glycosylation for α -N³-pyrimidinenucleotide analogue **13** in high diastereoselectivity. The phosphonothioate was an effective functional group to give β -N¹-pyrimidine-nucleotide analogues **18a**-c with good diastereoselectivity. The nucleotide analogue **18a** was transformed to the difluoromethylenephosphonate analogue **20** of thymidine-3'-phosphate by oxidation with MCPBA, followed by aqueous work-up.

Key words: nucleotide analogues, N-glycosylations, neighbouring-group effects, phosphorus, fluorine

Since the discovery of certain sugar-modified nucleosides and nucleotides having potential antiviral and antitumor effects, many useful strategies for modification of naturally occurring nucleosides and nucleotides have been devised in the search for novel nucleotide analogues having significant biological activity.¹ Replacement of 3'- or 5'phosphate moiety of naturally occurring nucleotides with metabolically stable methylenephosphonate functional group has been introduced to synthesize novel nucleotide analogues which act as a pseudoterminator of polymerase-catalyzed RNA-synthesis² or as important components of antisense oligonucleotides having increased resistance to nuclease.³

In the synthesis of a series of metabolically stable 2',3'dideoxynucleotide analogues having a methylenephosphonate functionality at the 3' α -position, previously, we have pursued Lewis acid-mediated N- and C-glycosylation of 2,3-dideoxyfuranose derivatives **1a,b** possessing a methylenephosphonate or methylenephosphonothioate functionality.⁸ In the N-glycosylation with the phosphonothioate **1a**, the phosphonothioate functional group effectively participates in forming the bicyclic cationic intermediate **A** to give the β -nucleotide analogue **2** stereoselectively (92% de) under the modified Vorbrüggen conditions (Scheme 1).^{8a,9} The nucleotide analogue **2** was transformed to the phosphonic acid analogue **3** of thymidin-3'-phosphate through exchanging the phosphonothioate functionality to phosphonic acid.

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BzC BzC BzC OEt T(TMS)₂ TiCl₄ ^ICH₂P(X)(OEt)₂ ĊH₂P(О)(ОН)₂ CH2P(S)(OEt)2 3 X=S 2 (92% de) 1a 1b X=O BzC OEt ÒEt

Scheme 1

With a view to obtaining metabolically stable nucleotide analogues which act as a mimic for the naturally occurring nucleotides, replacement of the phosphate moiety with a difluoromethylenephosphonic acid would be a good tactic.⁴ Many α , α -difluoromethylenephosphonate derivatives have been studied as a potential enzyme inhibitor and as a useful probe for elucidation of biochemical processes.⁵ Methods for synthesis of 5'-deoxy-5'-difluoromethylenephosphonate nucleotide analogues 4 have been developed.⁶ However, to the best of our knowledge, methods for stereoselective synthesis of 2',3'-dideoxy-3'-difluoromethylenephosphonate nucleotide analogues of general structure 5 have not been developed.⁷ In this paper, we report stereoselective synthesis of 2',3'-dideoxy-3'difluoromethylenephosphonate nucleotide analogues by the phosphnothioate-assisted N-glycosylation with pyrimidine bases as an extension of our previous work (Figure 1).



Figure 1

First, we examined N-glycosylation of the glycosyl donor **12** having the difluoromethylenephosphonate functionality (Scheme 3). The glycosyl donor **12** was readily prepared from 1,2-O,O-isopropylidene-(R)-glyceraldehyde **6** as shown in Scheme 2. Horner–Emmons–Wadsworth reaction of **6** with *tert*-butyl diethylphosphonoacetate gave the unsaturated ester 7 in 80% yield. Reaction between 7 and diethyl (lithiodifluoromethyl)phosphonate¹⁰ at -78 °C in THF gave the Michael adduct 8 in 81% yield as a 1:1 mixture of diastereoisomers. Treatment of the mixture with concentrated HCl in EtOH at 25 °C reproducibly gave the *trans*-lactone 9^{11} as the sole product in 50% yield. Under the conditions, the (3R,4S)-diastereoisomer of 8 was rapidly lactonized, but the (3S,4S)-diastereoisomer decomposed during the reaction.¹² The relative stereochemistry of 9 was deduced by a NOESYexperiment, which showed a strong correlation between H-3 and the methylene protons α to the hydroxyl. Reduction of 10 with BH₃·THF in THF at 0 °C gave the lactol 11 in 80% yield without affecting the phosphonate functional group. The lactol **11** was transformed to an anomeric mixture of 1-ethoxy derivative **12** (α : β = 1:1) in 95% yield on treatment with triethyl orthoformate in EtOH in the presence of a catalytic amount of ammonium nitrate.



Scheme 2 Reagents and conditions: (a) $(EtO)_2P(O)CH_2CO_2$ -*t*-Bu, NaH, THF, 80%; (b) LiCF₂P(O)(OEt)₂, THF, -78 °C, 81%; (c) concd HCl, EtOH, 50%; (d) TBDPSCl, imidazole, DMF, 85%; (e) BH₃·THF, THF, 0 °C, 80%; (f) CH(OEt)₃, cat. NH₄NO₂, EtOH, 80 °C, 95%.

N-Glycosylation of **12** with bis-trimethylsilylthymine T(TMS)₂ was carried out at 0 °C in CH₂Cl₂ in the presence of TiCl₄. The reaction gave the α -N³-nucleotide analogue **13** of a high diastereomeric excess (90% de) in 42% yield. In this reaction, T(TMS)₂ reacted with **12** preferably at the N³-nitorogen of the pyrimidine from α -face of the glycosyl donor.^{9c} The site of glycosylation on the pyrimidine could be assigned by characteristic downfield shift (δ = 6.83, t, *J* = 7.7 Hz) of the anomeric proton as well as a strong correlation between N¹-H and H-6 in the NOESY spectrum (Figure 2). α -Configuration of **13** was assigned on the basis of the observed NOESY-relay correlations among protons in the sugar moiety depicted in Figure 2.

With the results of N-glycosylation of **12** having a difluoromethylenephosphonate functionality in mind, we next



Scheme 3



Figure 2 NOESY-correlations of 13

examined N-glycosylation of the phosphonothioate analogues 16 and 17 with bis-trimethylsilylpyrimidines (Scheme 4 and Table 1). The glycosyl donors 16 and 17 were prepared from the lactone 10. Thionylation of 10 with Lawesson's reagent in refluxing toluene gave the phosphonothioate 14 in 80% yield, which was transformed to an anomeric mixture of 1-ethoxy derivative 16 $(\alpha:\beta = 1:1)$ via the lactol **15** in good overall yield (76%) in an analogous manner to that of synthesis of 12. The lactol 15 was also transformed to an anomeric mixture of 1-acetoxy derivative 17 (α : β = 1:1) by the standard protocol (Ac₂O, pyridine). The previous results of Lewis acid-mediated N-glycosylation of non-fluorinated glycosyl donor **1a** with $(T(TMS)_2)$ revealed that $TiCl_4$ was the best Lewis acid to induce the high stereoselectivity and yield.^{8a} Then, 1-ethoxy derivative 16 was treated with $T(TMS)_2$ in the presence of TiCl₄ in CH₂Cl₂ at 0 °C for 3 hours. In contrast to N-glycosylation of the phosphonate analogue 12, the desired β -N-glycosylation proceeded with a good diastereoselectivity (β : $\alpha = 89:11$) to give β -N¹-nucleotide analogue 18a in 93% yield (entry 1). The relative stereochemisty of 18a and the site of glycosylation on the pyrimidine were confirmed on the basis of the NOESYexperiments. The diagnostic NOESY-correlations are depicted in Figure 3. Under the same conditions, bis-trimethylsilyluracil [U(TMS)₂] reacted with 16 to give 18b in 63% yield with comparable diastereoselectivity (entry 2). However, significant decrease in diastereoselectivity was observed in the reaction with bis-trimethylsilyl-5fluorouracil [5-FU(TMS)₂] (entry 3). The 1-acetoxy derivative 17 seems to be less potent as a glycosyl donor than 1-ethoxy derivative 16; the reaction of 17 with $T(TMS)_2$ under the same conditions resulted in a significant loss of the diastereoselectivity and yield (entries 1 vs. 4). A similar trend was observed for the reaction with $U(TMS)_2$ (entries 2 vs. 5).



Scheme 4

Table 1N-Glycosylation of **16** and **17** with Silylated Pyrimidines in
the Presence of $TiCl_4$

Entry ^a	Glycosyl donor	Base ^b	18	R	Yield	Ratio (β:α) ^c
1	16	T(TMS) ₂	18a	Me	93	89:11
2	16	U(TMS) ₂	18b	Н	64	87:13
3	16	5-FU(TMS) ₂	18c	F	84	76:24
4	17	T(TMS) ₂	18a	Me	72	70:30
5	17	U(TMS) ₂	18b	Н	52	84:16

^a All reactions were carried out at 0 $^{\circ}$ C for 3 h in the presence of 5.0 equiv of TiCl₄ in CH₂Cl₂.

^b Prepared in situ from bis-trimethylsilylacetamide (2.0 equiv) and the base (2.0 equiv) in the solvent.

^c Determined by ¹H NMR (400 MHz, CDCl₃).



Figure 3 NOESY-correlations of 18a

The reactivity of the TiCl₄-mediated N-glycosylation of 1-acethoxy derivative **17** with T(TMS)₂ was significantly affected by the reaction temperature.¹³ On reaction of **17** with T(TMS)₂ (2 equiv) in CH₂Cl₂ in the presence of TiCl₄ (5 equiv) at -40 °C for 3 hours, T(TMS)₂ reacted at the N^3 -nitrogen to give a 66:34 mixture of α - N^3 -nucleotide analogue **19** and the corresponding β -anomer in 72% yield. However, when the reaction was carried out at -40 °C for 3 hours, and was quenched after being stirred at 0 °C for an additional 3 hours, β - N^1 -nucleotide analogue **18a** (β : α = 88:12) was produced in 78% yield (Scheme 5). The results may suggest that the N-glycosylation of **17** with

 $T(TMS)_2$ in the presence of $TiCl_4$ kinetically produces the α - N^3 -nucleotide analogue **19** with poor diastereoselectivity, which isomerizes to the thermodynamically more stable β - N^1 -nucleotide analogue **18a** with good diastereoselectivity under the conditions.



Scheme 5

The nucleotide analogue **18a** (β : α = 89:11) thus obtained was converted to the difluoromethylenephosphonate analogue **20** of thymidine-3'-phosphate in 64% yield by oxidation with MCPBA (5.0 equiv), followed by aqueous work-up. Diastereomerically pure **20** could be isolated by preparative HPLC [Inertsil (GL-science), hexane:EtOAc = 1:1, Scheme 6].



Scheme 6

In summary, we have developed an efficient method for the stereoselective synthesis of novel 2',3'-dideoxyribonucleotide analogues, in which the difluoromethylenephosphonate functionality is incorporated to the $3'\alpha$ position as a phosphate isostere.

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- (11) Spectroscopic data for selected new compounds: Compound **9**: $[\alpha]_D^{20}$ +13.8 (*c* 1.0, CHCl₃). ¹H NMR (400 MHz, CDCl₃): δ = 4.90–4.83 (1 H, m), 4.38–4.23 (4 H, m), 4.00 (1 H, dd, J = 2.6, 12.5 Hz), 3.72 (1 H, dd, J = 2.9, 12.5 Hz), 3.40–3.22 (1 H, m), 2.85 (2 H, d, J = 8.4 Hz), 2.56 (1 H, br s), 1.40 (6 H, t, J = 7.1 Hz). ¹³C NMR (100 MHz, CDCl₃): $\delta = 175.1, 119.1 \text{ (dt, } J_{CP} = 215.4 \text{ Hz}, J_{CF} = 263.0 \text{ Hz}), 78.6$ (d, $J_{\rm CP}$ = 4.0 Hz), 65.2 (d, $J_{\rm CP}$ = 7.1 Hz), 65.0 (d, $J_{\rm CP}$ = 7.1 Hz), 63.0, 40.1 (dt, $J_{CP} = 15.8$ Hz, $J_{CF} = 21.0$ Hz), 20.3, 16.0 (d, $J_{CP} = 5.2$ Hz). ¹⁹F NMR (376 MHz, CDCl₃, benzotrifluoride): $\delta = -54.49$ (1 F, dd, $J_{\text{HF}} = 4.9$ Hz, $J_{\text{PF}} = 105.2$ Hz), -54.54 (1 F, $J_{\text{HF}} = 5.8$ Hz, $J_{\text{PF}} = 105.2$ Hz). ³¹P NMR (162 MHz, CDCl₃): δ = 5.76 (t, J_{PF} = 105.2 Hz). MS (EI): $m/z = 303 [M^+ + 1], 285 [M^+ - OH], 271 [M^+ - CH_2OH].$ IR(film): 3247, 2988, 1789, 1645 cm⁻¹. Anal. Calcd for C₁₀H₁₇O₆F₂P: C, 39.74; H, 5.67. Found: C, 39.43; H, 5.66. Compound **13**: $[\alpha]_D^{20}$ +60.4 (*c* 1.3, CHCl₃). ¹H NMR (400 MHz, CDCl₃): $\delta = 9.56 (1 \text{ H}, \text{d}, J = 5.6 \text{ Hz}), 7.74-7.62 (4 \text{ H}, J = 5.6 \text{ Hz}), 7.74-7.62 (4 \text{ H}, J = 5.6 \text{ Hz}), 7.62 ($ m), 7.48–7.31 (6 H, m), 6.91 (1 H, dd, J = 1.1, 5.6 Hz), 6.83 (1 H, t, J = 7.7 Hz), 4.90–4.83 (1 H, m), 4.34–4.19 (4 H, m), 4.02-3.94 (1 H, m), 3.82-3.74 (1 H, m), 3.64-3.45 (1 H, m), 3.02 (1 H, ddd, J = 8.4, 12.4, 12.4 Hz), 2.59–2.48 (1 H, m), 1.89 (3 H, d, J = 1.1 Hz), 1.37 (3 H, t, J = 7.1 Hz), 1.36 (3 H, t, J = 7.1 Hz), 1.07 (9 H, s). ¹³C NMR (100 MHz, CDCl₃): $\delta = 163.9, 152.6, 135.6, 135.5, 135.0, 133.4, 133.1, 129.6,$

127.6, 120.1 (dt, $J_{CP} = 216.2$ Hz, $J_{CF} = 263.7$ Hz), 110.3, 83.1, 79.8, 64.7 (d, $J_{CP} = 6.7$ Hz), 64.5 ($J_{CP} = 6.9$ Hz), 64.1, 42.3 (dt, $J_{CP} = 14.4$ Hz, $J_{CF} = 20.7$ Hz), 29.9, 26.8, 19.3, 16.4 (d, $J_{CP} = 5.2$ Hz), 12.9. ¹⁹F NMR (376 MHz, CDCl₃, benzotrifluoride): $\delta = -47.1$ (1 F, dd, $J_{FF} = 300.9$ Hz, $J_{\rm FP} = 105.7$ Hz), -58.5 (1 F, ddd, $J_{\rm FF} = 300.9$ Hz, $J_{\rm FP} = 111.5$ Hz, $J_{\text{FH}} = 26.1$ Hz). ³¹P NMR (162 MHz, CDCl₃): $\delta = 6.77$ $(dd, J_{PF} = 105.7, 111.5 \text{ Hz})$. MS (EI): $m/z = 593 [M^+ - t-Bu]$. HRMS (EI): m/z calcd for C₂₇H₃₂N₂O7F₂PSi [M⁺ - t-Bu]: 593.1697. Found: 593.1697. Anal. Calcd for C₃₁H₄₁N₂O₇F₂PSi: C, 57.22; H, 6.35; N, 4.30. Found: C, 56.61; H, 6.35; N, 4.22. Compound **18a**: $[\alpha]_{D}^{20}$ +30.2 (*c* 1.3, CHCl₃) for a sample of 78% de. ¹H NMR (400 MHz, CDCl₃): δ = 8.07 (1 H, br s), 7.71–7.62 (4 H, m), 7.49–7.34 (7 H, m), 6.19 (1 H, t, J = 6.8 Hz), 4.46–4.40 (1 H, m), 4.31–4.17 (4 H, m), 4.16–4.08 (1 H, m), 3.84 (1 H, dd, *J* = 2.4, 11.6 Hz), 3.68–3.50 (1 H, m), 2.82-2.72 (1 H, m), 2.21-2.11 (1 H, m), 1.56-1.53 (3 H, m), 1.36 (32 H, t, J = 7.0 Hz), 1.34 (3 H, t, J = 7.0 Hz), 1.11 (9 H, s). ¹³C NMR (100 MHz, CDCl₃): $\delta = 163.9, 150.3, 135.5,$ 135.2, 135.0, 133.1, 132.3, 130.0 (2 carbons), 127.9 (2 carbons), 120.9 (dt, $J_{CP} = 176.5$ Hz, $J_{CF} = 268.4$ Hz), 111.3, 84.5, 78.9, 64.9 (d, $J_{\rm CP}$ = 6.7 Hz), 64.5, 41.5 (dt, $J_{\rm CP}$ = 16.6 Hz, $J_{CF} = 20.4$ Hz), 33.1, 26.9, 19.4, 16.1 (d, $J_{CP} = 5.7$ Hz), 11.9. ¹⁹F NMR (376 MHz, CDCl₃, benzotrifluoride): $\delta =$ -50.3 (1 F, ddd, $J_{\text{HF}} = 13.1$ Hz, $J_{\text{FF}} = 289.5$ Hz, $J_{\text{FP}} = 110.6$ Hz), -54.7 (1 F, ddd, $J_{\rm HF} = 21.9$ Hz, $J_{\rm FF} = 289.5$ Hz, $J_{\rm FP} = 107.7$ Hz). ³¹P NMR (162 MHz, CDCl₃): $\delta = 74.9$ (dd, $J_{\rm PF} = 107.7$ Hz, 110.6 Hz). MS (EI): m/z = 609 [M⁺ – t-Bu], 541 [M⁺ – thymine]. Anal. Calcd for C₃₁H₄₁N₂O₆F₂PSSi: C, 55.84; H, 6.20; N, 4.20. Found: C, 55.65; H, 6.14; N, 4.01. Compound **20**: $[\alpha]_D^{20}$ +34.0 (*c* 1.0, CHCl₃). ¹H NMR (400 MHz, $CDCl_3$): $\delta = 8.24 (1 \text{ H, br s}), 7.71-7.61 (4 \text{ H, m}), 7.48-$ 7.33 (6 H, m), 6.19 (1 H, t, *J* = 6.9 Hz), 4.48–4.42 (1 H, m), 4.35-4.21 (4 H, m), 4.16-4.08 (1 H, m), 3.84 (1 H, dd, *J* = 2.5, 11.6 Hz), 3.46–3.25 (1 H, m), 2.85–2.73 (1 H, m), 2.24–2.12 (1 H, m), 1.54 (3 H, s), 1.38 (6 H, t, *J* = 7.1 Hz), 1.10 (9 H, s). ¹³C NMR (100 MHz, CDCl₃): $\delta = 163.9, 150.3,$ 135.5, 135.4, 135.2, 133.1, 132.3, 130.0 (2 carbons), 127.9 (2 carbons), 119.9 (dt, J_{CP} = 214.3 Hz, J_{CF} = 263.3 Hz), 111.3, 84.5, 78.5, 64.9 (d, $J_{CP} = 6.8$ Hz), 64.8 (d, $J_{CP} = 7.0$ Hz), 64.6, 42.3 (dt, $J_{CP} = 15.0$ Hz, $J_{CF} = 20.2$ Hz), 32.7, 26.9, 19.4, 16.3 (d, $J_{CP} = 5.2$ Hz), 11.9. ¹⁹F NMR (376 MHz, CDCl₃, benzotrifluoride): $\delta = -51.1 (1 \text{ F}, \text{ ddd}, J_{\text{HF}} = 14.5 \text{ Hz},$ $J_{\rm FF} = 302.9$ Hz, $J_{\rm PF} = 106.8$ Hz), -54.0 (1 F, ddd, $J_{\rm HF} = 21.5$ Hz, $J_{\rm FF} = 302.9$ Hz, $J_{\rm PF} = 106.8$ Hz). ³¹P NMR (162 MHz, CDCl₃): $\delta = 6.31$ (t, $J_{PF} = 106.8$ Hz). MS (EI): $m/z = 593 [M^+ - t-Bu]$. HRMS (EI): m/z calcd for C₂₇H₃₂N₂O7F₂PSi [M⁺ - t-Bu]: 593.1697. Found: 593.1672.

(12) Percy et al. reported that the diol i, in situ prepared stereoselectively from the corresponding olefin by OsO₄catalyzed dihydroxylation, was not readily cyclized to the *cis*-lactone ii under the conditions.^{7b} They were never able to isolate more than 11% of ii. The results are consistent with our findings (Scheme 7).



Scheme 7

(13) The 1-ethoxy derivative 16 did not react with T(TMS)₂ in the presence of TiCl₄ at -40 °C and the lactol 15 was recovered in 80% yield.

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