Total Synthesis of Altenusin and Alterlactone

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Abstract: The resorcylic lactone alterlactone, a mycotoxin produced by *alternaria sp.*, was synthesized for the first time. The total synthesis was achieved in nine steps with 69% yield starting with acetal-protected phloroglucinic acid and 6-bromopiperonal, where the longest linear sequence consists of five steps. Key step is a Suzuki coupling used for the construction of the central biaryl bond. When the final deprotection with cleavage of benzyl ethers (yielding unprotected alterlactone) was performed in a less polar solvent the biaryl mycotoxin altenusin was obtained.

Key words: polyketides, fungal metabolites, resorcylic lactones, cross-coupling, Suzuki reaction

The resorcylic lactones¹ alternariol (1) and alternariol 9methyl ether (2) are main secondary metabolites of toxinproducing *alternaria* fungi (Figure 1).² Although the toxicity of these mycotoxins is low when compared with others (e.g., aflatoxins),³ infestations with *alternaria spp*. lead to significant crop losses by fouling of tomatoes, apples, and other fruits.⁴ Total syntheses have been provided for these compounds⁵ and for the structurally related toxins altenuene (3), isoaltenuene (4),⁶ and neoaltenuene (5).⁷

Minor toxins produced by *alternaria spp.*⁸ and other fungi⁹ are altenusin (6) and its oxidized derivative dehydroaltenusin (7). Altenusin is a selective pp60c-Src kinase inhibitor (IC₅₀ = 20 nM, 400 × more active than against other tested protein kinases),¹⁰ a moderate inhibitor of myosin light-chain kinase (IC₅₀ = 340 μ M),¹¹ and a specific inhibitor of neutral sphingomyelinase ($K_i = 20 \mu$ M),¹² inhibits trypanothione reductase from the parasite *Trypanosoma cruzi* (IC₅₀ = 4.3 μ M)¹³ and shows a broad antimicrobial activity against several multidrug-resistant bacterial and fungal strains,¹⁴ but is only moderately active against HIV-1 integrase.⁹⁶ It had originally been synthesized en route to dehydroaltenusin (7) in eight steps with 27% yield starting with commercially available phloroglucinic acid and 4-methylcatechol, where the longest linear sequence consisted of five steps.¹⁵

Alterlactone (8) was isolated from a fungal endophyte *alternaria sp.* It shows moderate cytotoxic in vitro activity against L5178Y cells (11.8% growth at 10 μ mol/mL).¹⁶ Further biological activities have not been reported for 8, nor has a total synthesis been published for this natural product. To supply sufficient amounts of material for further testings we aimed for a convergent total synthesis of

SYNLETT 2012, 23, 371–374 Advanced online publication: 19.01.2012 DOI: 10.1055/s-0031-1290135; Art ID: B60111ST © Georg Thieme Verlag Stuttgart · New York several resorcylic lactones. Herein we report on the total synthesis of alternusin ($\mathbf{6}$) and of alterlactone ($\mathbf{8}$).

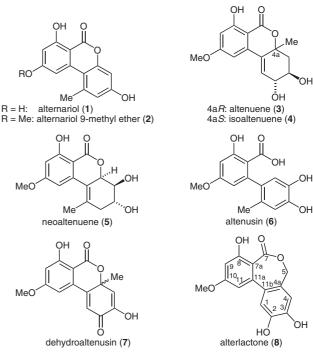
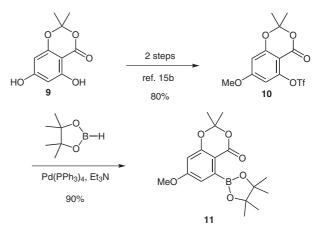


Figure 1 Selection of toxins produced by alternaria spp.



Scheme 1 Synthesis of boronate 11

The projected strategy for a convergent synthesis of altenus in (6) and alterlactone (8) resembles a protocol, which had already been successfully applied to the total synthesis of other resorcylic lactones. Key reaction would be a Suzuki reaction of a suitably substituted aryl boronate **11** with a highly oxygenated aryl bromide. Starting with commercially available acetal-protected phloroglucinic acid **9** a pinacol-derived boronate **11** was obtained with 72% yield (Scheme 1).^{15b,17,18}

While bromination of *O*-methylvanilline (**12**, Figure 2) was successful,¹⁹ the required complete demethylation of the resulting carbaldehyde **13** to yield **15** was not possible with any of the methods investigated.²⁰

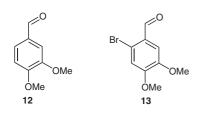
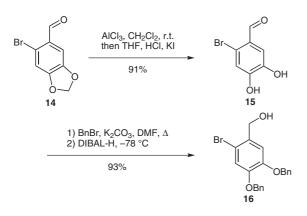


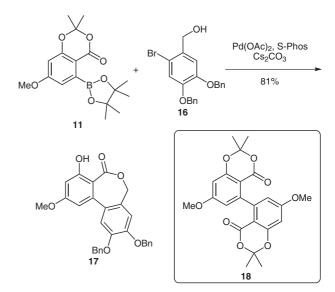
Figure 2

We therefore used commercially available bromopiperonal (14) as starting material, which was successfully deprotected with anhydrous aluminum chloride followed by refluxing with hydrochloric acid and potassium iodide in order to cleave the intermediately formed chloromethoxybenzaldehyde.^{20a} Double benzylation using a modified protocol of Reitz et al.^{20a} and reduction of the aldehyde function with diisobutylaluminum hydride yielding the unknown alcohol 16 (Scheme 2) turned out to afford better results than a previously published method.^{20b,21}



Scheme 2 Synthesis of brominated derivative 16

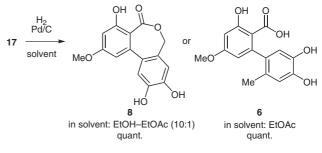
Suzuki coupling with concomitant translactonization was achieved with palladium(II) acetate as catalyst, cesium carbonate as base, and S-Phos as ligand in a 81% yield (Scheme 3).²² Minor amounts of the homocoupling product **18** (ca. 26% calculated on the boronate **11**, which was used in excess) were obtained with this protocol, as already reported for similar couplings.¹⁸ Though formation of this side product could not be prevented, it was easily separated by conventional chromatography. Since traces of the starting material **16** could not be removed even by medium-pressure liquid chromatography (MLPC) purifi-



Scheme 3 Suzuki coupling yielding protected alterlactone 17

cation with HPLC turned out to be necessary. This ultimately afforded the coupling product **17** in high yield.

The reaction outcome in the hydrogenolysis of benzylprotected alterlactone **17** was strongly dependent from the solvent used (Scheme 4). Hydrogenolysis in the presence of palladium on charcoal quantitatively led to alterlactone (**8**) when a mixture of ethanol and ethyl acetate (10:1) was used as solvent, while utilization of ethyl acetate as the sole solvent furthermore led to further reduction (cleavage of the benzylic C–O bond in the seven-membered ring) with formation of altenusin (**6**).



Scheme 4 Hydrogenolysis of benzyl-protected alterlactone 17

This first total synthesis of alterlactone (8) was achieved in 69% with nine steps starting with acetal-protected phloroglucinic acid 9 and 6-bromopiperonal (14), where the longest linear sequence consists of five steps. Similarly, a competitive further total synthesis of altenusin (6) was achieved with the same yield and in the same number of steps. Since dehydroaltenusin had previously been synthesized from altenusin (6),¹⁵ this work is furthermore a formal synthesis of dehydroaltenusin (7). NMR spectroscopic data of synthesized alterlactone (8) are in full agreement with published data of the natural product,¹⁶ giving unambiguous evidence that the proposed structure of alterlactone (8) is correct (Table 1).

Position ^a	¹ H NMR data		¹³ C NMR data	
	Natural	Synthesized	Natural	Synthesized
1	7.03 (s)	7.04 (s)	115.5	115.5
2			146.6	146.6
3			145.9	145.9
4	6.90 (s)	6.91 (s)	115.5	115.5
4a			140.0	140.1
5	4.80 (d, $J = 11.0$ Hz), 4.85 (d, $J = 11.3$ Hz)	4.83-4.85	67.8	67.8
7			168.7	168.8
7a			109.5	109.5
8			159.9	160.0
9			100.8	100.8
10	6.45 (d, $J = 2.2$ Hz)	6.46 (J = 2.5 Hz)	162.2	162.2
11			105.0	105.0
11a	6.50 (d, J = 2.2 Hz)	6.51 ($J = 2.4 \text{ Hz}$)	126.6	126.6
11b			129.8	129.9
OMe	3.81 (s)	3.82	55.4	55.4
2-OH, 3 OH	9.37, 9.47 (br s)	9.38, 9.47 (br s)		
8-OH	10.21 (br s)	10.21 (br s)		

 Table 1
 Comparison of NMR Data for Alterlactone (8)¹⁶

^a The numbering is given in Figure 1. A different numbering had been given in the original literature.¹⁶

9,10-Bis(benzyloxy)-4-hydroxy-2-methoxydibenzo[c,e]oxepin-5(7H)-one (17)

Arylbromide 16 (210 mg, 0.526 mmol), boronate 11 (228 mg, 684 μmol), Cs₂CO₃ (0.514 g, 1.58 mmol), Pd(OAc)₂ (3.5 mg, 16 μmol), and S-Phos (purity 97%, 13.0 mg, 32.0 µmol) were dissolved under an Ar atmosphere in degassed dioxane-H₂O (7:1, 10 mL), and the mixture was stirred for 20 h at 80 °C (monitoring with TLC) and cooled. A sat. NH₄Cl solution (10 mL) was added, and the mixture was extracted with EtOAc (2×25 mL). The organic layers were dried (Na₂SO₄) and concentrated, and the residue was purified by chromatography (silica gel, cyclohexane-EtOAc = 10:1) to yield side product 18 (37 mg) and a mixture (240 mg) of coupling product 17 and bromide 16. Aliquots of this mixture (150 mg) were purified with preparative HPLC (MeCN-H₂O = 75:25 with 0.1% TFA to 98:2 with 0.1% TFA) to yield 17 as a colorless solid (119 mg, 0.254 mmol, 81%); mp 150–154 °C; $R_f = 0.54$ (hexanes–EtOAc = 2:1). IR (KBr): 2915, 1641, 1566, 1514, 1422, 1346, 1252, 1200, 1150, 1075, 1044, 987, 967, 916, 835, 753 cm⁻¹. ¹H NMR (500 MHz, CDCl₃): δ = 3.85 (s, 3 H, OMe), 4.79–4.81 (m, 1 H, CH₂), 4.98–5.02 (m, 1 H, CH₂), 5.23 (s, 4 H, Bn), 6.41 (d, ${}^{4}J$ = 2.6 Hz, 1 H, ArH), $6.55 \text{ (d, } {}^{4}J = 2.6 \text{ Hz}, 1 \text{ H}, \text{ ArH}), 6.96 \text{ (s, 1 H, ArH)}, 7.09 \text{ (s, 1 H, ArH)$ ArH), 7.47–7.32 (m, 10 H, Bn), 10.62 (br s, 1 H, OH) ppm. ¹³C NMR (101 MHz, CDCl₃): $\delta = 29.7$ (CH₃), 55.6 (CH₃), 71.3 (CH₂), 71.4 (CH₂), 71.5 (CH₂), 100.9 (CH), 106.3 (C),108.5 (CH), 113.7 (CH), 114.1 (CH), 116.0 (CH), 127.2 (C), 127.2 (CH), 127.4 (CH), 127.5 (C), 128.1 (CH), 128.1 (CH), 128.6 (CH), 128.7 (CH), 128.8 (CH), 132.6 (C), 136.6 (C), 136.7 (C), 141.0 (C), 149.4 (C), 149.9 (C), 163.8 (C), 163.9 (C), 172.8 (C) ppm. MS-FAB: m/z = 469 $[M + H]^+$. HRMS-FAB: *m/z* calcd for C₂₉H₂₅O₆ 469.1651 [MH⁺]; found: 469.1649.

Side Product 18: 7,7'-Dimethoxy-2,2,2',2'-tetramethyl-4,4'-dioxo-4H,4'H-[5,5']bi{benzo[d][1,3]dioxinyl}

¹H NMR (300 MHz, CDCl₃): δ = 1.73 (s, 6 H, CH₃), 1.76 (s, 6 H, CH₃), 3.85 (s, 6 H, OMe), 6.46 (d, 2 H, ⁴J = 2.5 Hz, ArH), 6.48 (d, 2 H, ⁴J = 2.5 Hz, ArH).

4,9,10-Trihydroxy-2-methoxydibenzo[*c,e*]**oxepin-5**(7*H*)-**one** [Alterlactone (8)]

Pd/C (10%, 16.0 mg, 154 µmol) was added to a solution of protected alterlactone 17 (18.0 mg, 38.0 µmol) in EtOAc (0.05 mL) and EtOH (0.5 mL). The atmosphere was replaced with H₂, and the mixture was stirred for 5 h at r.t. (monitoring with TLC). The mixture was filtered, and the filtrate was concentrated to yield alterlactone (8) as a colorless solid (10.9 mg, 38.0 μ mol, quant.). $R_f = 0.29$ (CH₂Cl₂-MeOH = 10:1). IR (KBr): 3080, 1647, 1604, 1566, 1525, 1458, 1421, 1370, 1268, 1201, 1152, 1112, 1075, 1022, 990, 824, 773, 621 cm⁻¹. ¹H NMR (400 MHz, DMSO- d_6): δ = 3.82 (s, 3 H, OMe), 4.83–4.85 (m, 2 H, CH₂), 6.46 (d, ${}^{4}J$ = 2.5 Hz, 1 H, ArH), 6.51 (d, ⁴J = 2.4 Hz, 1 H, ArH), 6.91 (s, 1 H, ArH), 7.04 (s, 1 H, ArH), 9.38 (br s, 1 H, OH), 9.47 (br s, 1 H, OH), 10.21 (br s, 1 H, OH) ppm. ¹³C NMR (101 MHz, CDCl₃): δ = 55.4 (CH₃, OCH₃), 67.8 (CH₂, C-5), 100.8 (C, C-9), 105.0 (CH, C-11), 109.5 (C, C-7a), 115.5 (2 CH, C-4, C-1), 126.6 (C, C-11a), 129.9 (C, C-11b), 140.1 (C, C-4a), 145.9 (C, C-3), 146.6 (C, C-2), 150.0 (C, C-8), 162.2 (C, C-10), 168.8 (C, C-7) ppm; MS–FAB: $m/z = 289 [M + H]^+$. HRMS–FAB: m/z calcd for C₁₅H₁₃O₆: 289.0712 [MH⁺]; found: 289.0709. The analytical data are in full accordance with published data.¹⁶

3,4',5'-Trihydroxy-5-methoxy-2'-methyl-biphenyl-2-carboxylic Acid [Altenusin (6)]

Pd/C (10%, 18.0 mg, 171 μmol) was added to a solution of protected alterlactone **17** (20.0 mg, 43.0 mol) in EtOAc (1 mL). The atmosphere was replaced with H₂, and the mixture was stirred for 8 h at r.t. (monitoring with TLC). The mixture was filtered, and the filtrate was concentrated to yield altenusin (**6**) as a colorless solid (11.9 g, 43.0 µmol, quant.). $R_f = 0.13$ (CH₂Cl₂–MeOH = 10:1). ¹H NMR (250 MHz, CD₃OD): $\delta = 1.92$ (s, 3 H, CH₃), 3.81 (s, 3 H, OMe), 6.17 (d, ⁴J = 2.6 Hz, 1 H, ArH), 6.43 (d, ⁴J = 2.6 Hz, 1 H, ArH), 6.49 (s, 1 H, ArH), 6.59 (s, 1 H, s, ArH) ppm. The analytical data are in full accordance with published data.^{96,15b}

Supporting Information for this article is available online at http://www.thieme-connect.com/ejournals/toc/synlett. Included are experimental and spectroscopic data for all compounds, and spectra of title compounds.

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